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Field and Gaussian-based 3D-QSAR models on 8-amino-imidazo [1, 5a] pyrazine derivatives as Bruton's tyrosine kinase inhibitors

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ABSTRACT

Bruton's tyrosine kinase (BTK) stands as a crucial enzyme implicated in various health disorders associated with B-cells, garnering significant attention over time. It presents a compelling molecular target for addressing autoimmune diseases and malignancies linked to B-cells. The inhibition of BTK has demonstrated efficacy in treating autoimmune diseases, B-cell leukemia, and lymphomas. In this study, we conducted a 3D-QSAR analysis on a series of derivatives of 8-Amino-imidazo [1, 5a] pyrazines and optimized CGI-1746 derivatives to pinpoint essential descriptors that augment their biological activity. Our study involved the creation of Gaussian and field-based 3D-QSAR models using target-based alignment derived from docked poses. These models exhibited statistical validity, with a predictive correlation coefficients (r^2) of 0.67 for the Gaussian model and 0.60 for the field-based model. Additionally, conventional correlation coefficients (r^2) were 0.93 and 0.92 for the Gaussian and field-based models, respectively. Contour map analyses unveiled the significance of steric and hydrophobic interactions as substantial contributing factors to the enhanced activity of these compounds. These discoveries hold significant promise for advancing research and development in the realm of BTK inhibitors.

INTRODUCTION

Bruton's tyrosine kinase (BTK) is identified as a member of the Tec family kinase, as established by Smith *et al.* [1] in 2001 and Gomez-Rodriguez *et al.* [2] in 2007. Its prominent expression in hematopoietic cells and classification as one of the largest cytoplasmic kinases have been documented by Smith *et al.* [3], Genevier *et al.* [4] in 1994, and Mohamed *et al.* [5] in 2009. BTK, according to the findings of Kurosaki [6] in 1997, Mahajan *et al.* [7] in 1999, and Mao *et al.* [8] in 2001, plays a pivotal role in multiple signal transduction pathways governing the activation, development, differentiation, growth, proliferation, and survival of B lymphoid cells. The activation of BTK, in conjunction with several essential transcription

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X-linked agammaglobulinemia (XLA) is a condition associated with mutations in the BTK gene. In males, XLA leads to a deficiency in mature B-cells and serum albumin, resulting in decreased levels of immunoglobulins in the serum and susceptibility to persistent infections, as reported by Tsukada and Witte [12] in 1994. Given that rheumatoid arthritis (RA) is characterized by polyclonal B-cell activation and B-cell growth, the selective suppression of BTK presents an attractive treatment approach for both RA and B-cell lymphoma. BTK's unique therapeutic potential stems from its selectivity for B-cells and its nonlethal nature, as outlined by Herman *et al.* [11] in 2011. In a previous study, we employed an extensive ligandbased virtual screening approach to identify prospective novel

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inhibitors, including ZINC85569481, BE015, ZINC08790868, and BE009, as highlighted in Sakthivel and Habeeb [13] in 2018. The use of QSAR as a tool has proven effective and promising in predicting molecular activity and elucidating chemo-biological interactions related to the studied biological activity, as discussed by Ghodsi and Hemmateenejad [14] in 2016.

However, a previous study conducted by Ali *et al.* [15] in 2021, which reported the potential of imidazo [1,5-a] pyrazine derivatives, relied solely on atom-based QSAR models, providing only rough approximations. Previous studies using different alignment strategies in QSAR modeling have demonstrated effective predictive power [3].Therefore, a more accurate approach for lead discovery and optimization can be achieved by both field-based and Gaussian-based 3D-QSAR models.

In another study, Ahmadi *et al.* [17] in 2021 utilized a hybrid descriptor-based approach involving Monte–Carlo methods to model BTK inhibitors. In our present study, we explored a 3D-QSAR analysis on a series of derivatives of 8-amino-imidazo [1, 5-a] pyrazines and optimized CGI-1746 derivatives, which were evaluated for their potential to inhibit BTK, as documented by Liu *et al.* [18] and Young *et al.* [19] in 2016.

MATERIALS AND METHOD

Dataset

A set of 66 potential inhibitors of BTK showing *in-vitro* activity was collected after a thorough literature review [18,19]. The experimental IC50 value of inhibitors was converted to pIC50 [20]. Chemsketch software was used to create a visual representation of the 2D structure of 66 BTK inhibitors [21]. The ligands were created using the Ligprep module of Maestro 10.2 version. This included creating possible tautomers and stereospecificity based on the chiral centers contained in the ligands, converting their 2D structures into 3D structures, and optimizing them using the OPLS 2005 force field. Finally, a low-energy ring conformation was produced for every ligand [22]. Table 1 provides the experimental and anticipated IC50 values for every drug that was developed using Schrodinger 10.2 to create 3D-QSAR models, which were based on Field and Gaussian techniques.

	Table 1	. Experimental	and QSAR	predicted	activity of	compounds.
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C N	Ligand	Experimental	Predicted a	ctivity		Ligand	Experimental	Predicted	l activity
S. No	name	activity	Field based	Gaussian	S. No	name	activity	Field based	Gaussian
1	b46*	9.826	9.268	9.540	34	b26	7.886	7.924	8.319
2	b41	9.77	9.373	9.54	35	b28*	7.824	8.697	7.966
3	b37	9.769	9.495	9. 74 4	36	32	7.699	8.049	7.439
4	b45*	9.721	9.967	8.812	37	9*	7.5223	7.889	7.338
5	b50	9.658	8.976	9.631	38	21	7.523	7.790	7.545
6	b1*	9.568	9,383	9.009	39	7	7.398	7.904	7.362
7	b42	9.523	9.471	9.625	40	26*	7.367	7.231	6.881
8	b3	9.509	9.404	9.662	41	14	7.301	7.715	7.412
9	b48	9.495	9.346	9.483	42	23	7.301	7.374	7.263
10	b2	9.494	9.316	9.265	43	31	7.301	7.549	7.397
11	b43*	9.443	9.613	9.621	44	12	7.222	7.934	7.616
12	b19	9.432	9.281	8.815	45	30*	7.222	7.933	7.917
13	b20	9.39	8.874	9.059	46	b25	7.187	7.668	8.225
14	b36*	9.387	9.299	9.167	47	16	7.155	7.456	6.784
15	b23	9.377	9.055	9.400	48	11*	7.046	8.094	6.515
16	b44	9.346	9.140	9.127	49	b27	6.983	7.609	7.153
17	b21*	9.284	9.030	8.820	50	b24*	6.975	8.525	8.500
18	b49	9.284	9.053	9.595	51	8	6.959	7.231	6.521
19	b22	9.276	9.392	9.182	52	b30	6.921	8.016	6.510
20	b47*	9.268	9.106	8.979	53	15*	6.886	7.552	6.380
21	b17	9.149	8.547	8.619	54	10*	6.824	7.291	6.224
22	b18*	9.131	9.024	8.210	55	20	6.824	7.184	7.202
23	b35	9.008	8.764	9.010	56	27	6.678	7.094	6.727
24	b33	8.959	8.810	8.634	57	5	6.648	7.496	7.170
25	b16	8.796	7.988	6.659	58	6*	6.638	7.608	6.674
26	b31	8.796	8.966	8.844	59	22	6.638	7.395	7.121

S. No	Ligand	Experimental	Predicted a	octivity	- C No	o Ligand name	Experimental	Predicted activity	
5. NO	name	activity	Field based	Gaussian	- 5. INU			Field based	Gaussian
27	b29	8.495	8.095	8.447	60	13	6.589	7.075	6.790
28	b40*	8.432	9.013	8.232	61	28	6.523	7.322	6.472
29	2*	8.222	7.095	6.285	62	19	6.468	7.139	6.409
30	b32	8.201	8.459	8.556	63	24*	6.31	7.061	6.989
31	17	8.097	8.172	8.055	64	25	6.187	7.668	7.306
32	Cgi	8.097	8.258	7.723	65	4	6.066	6.978	6.109
33	29	8	8.613	7.912	66	3	5.533	6.407	5.047

*Compounds in test set.

Alignment and protein preparation

The focus of this study is on developing fieldbased 3D-QSAR models, where the accurate alignment of compounds is crucial. For this purpose, the molecular docking process to establish a receptor-based alignment was enforced by downloading the experimental structure of BTK (PDB ID: 3OCS). To confirm that the structure was ready for analysis, Schrodinger 10.2v's protein preparation wizard was implemented to remove water molecules followed by the optimization of H-bond configurations. Careful and close attention was also paid to optimizing the target protein and OPLS 2005 force field-based minimization (heavy atom convergence = 0.5 A°) [23].

Receptor grid generation and docking

The receptor grid generating module in Schrodinger 10.2v was utilized to build the receptor grid used in the current study. The spatial organization of the receptors that are explored during the docking process is provided by this grid file. The residues around the active site, which includes LEU542, THR410, GLY411, THR474, LEU528, ALA428, TYR476, ALA479, ALA478, GLY480, SER453, ASN526, ASP459, GLU475, VAL416, and LEU408, were particularly included in the grid that was created [24]. The Glide XP technique was then used to dock the known inhibitors to BTK.

Field-based and Gaussian-based 3D-QSAR model calculations

The 3D-QSAR analysis in Schrödinger 10.2v employed both field-based and Gaussian-based methods, following the approach outlined by Myint and Xie [25] and Verma *et al.* [26]. The methodology involved establishing a 3D-QSAR model that correlates the 3D characteristics with the activities of aligned molecules. Construction of the QSAR model utilized partial least square (PLS) regression analysis, incorporating five factors. The independent variables for this analysis were the field and Gaussian intensities, computed based on the steric and electrostatic fields surrounding the ligand molecule within a 3D grid.

In the domain of multivariate statistical analysis, the field-based 3D-QSAR model operates as an alignmentdependent technique, establishing a relationship between molecular field interaction energy terms and biological activities. This model computes the electrostatic and steric fields within a grid surrounding the ligand molecule and subsequently correlates experimental pIC50 values with these interaction energy terms. Conversely, the Gaussian 3D-QSAR model incorporates hydrophobic, hydrogen bond donor (HBD), and hydrogen bond acceptor (HBA) fields, along with other potential fields, to calculate interaction energy. The determination of these potential fields relies on Gaussian equations. Both the field-based and Gaussian 3D-QSAR models share the common objective of generating predictive 3D-QSAR models. These models aim to forecast the biological activities of ligand molecules based on their interaction energies with the surrounding environment.

RESULTS AND DISCUSSION

Generation of field-based and Gaussian-based QSAR models

To gain a thorough understanding of the biological significance associated with each structural fragment and to establish a statistical relationship between activity and chemical attributes, we employed a receptor-based alignment method derived from Glide XP docking in our study. This method served as the basis for creating both field-based and Gaussian-based QSAR models. Figure 1A illustrates the positioning of inhibitors within the BTK binding site, while Figure 1B displays the alignment of ligands. The 3D-QSAR model was constructed using the most favorable docked configurations from Glide XP. We generated a field-based 3D-QSAR model with two key features, steric and electrostatic, and a Gaussian-based 3D-QSAR model with five features: Gaussian steric, Gaussian electrostatic, Gaussian hydrophobic, H-BA, and H-BD.

For model generation and validation, all compounds were randomly divided into training and test sets. PLS analysis was conducted with 46 compounds in the training set and 20 compounds in the test set, employing five PLS factors to produce a model with acceptable q^2 and r^2 values. The experimental and predicted IC50 values are provided in Table 1 for reference.

Field-based 3D-QSAR model

A force field-based 3D-QSAR model was created by establishing a correlation between the steric and electrostatic field intensities of the training set. The effectiveness of the model was assessed using various methods. A considerable degree of correlation was shown by the r^2 cv value of 0.6952, which was obtained using the Leave One Out (LOO) crossvalidation method. Furthermore, a study conducted without



Figure 1. (A) Alignment of known inhibitors in the binding site of BTK. (B) Ligand alignment after docking.

Parameters	Field based QSAR	Gaussian QSAR
SD^1	0.332	0.3216
R Squared $(R^2)^2$	0.9292	0.9335
$R^2 \operatorname{CV}^3$	0.6952	0.675
Stability ⁴	0.823	0.818
F^5	110.2	118
P^6	5.10E-23	1.34E-23
RMSE ⁷	0.7	0.77
Q Squared $(Q^2)^8$	0.6737	0.6079
Pearson-r 9	0.8473	0.8207

Table 2. Statistical summary of the 3D QSAR model.

1. SD: standard deviation of the regression

2. R squared: Coefficient of regression

3. R Squared CV: Cross validation R square

4. Stability: Stability of the model predictions to changes in the training set composition

5. F: variance ratio. Large values of F indicate a more statistically significant regression

6. P: significance level of variance ratio. Smaller values indicate a greater degree of confidence;

7. RMSE: root-mean-square error, RMS error in the test set prediction

8. Q squared: directly analogous to the R-squared, but based on the test set predictions.

9. Pearson-R: Pearson R value for the correlation between the predicted and observed activity for the test set.

cross-validation produced a r^2 value of 0.9292, indicating a robust connection. Listed in Table 2, the standard error of 0.33 and F ratio of 110 provide further insights into the model's accuracy and significance. The electrostatic and steric features of the compounds were analyzed using contour maps.

Steric contour maps

Steric contour maps visually depict favorable and unfavorable steric bulks. In Figure 2A and B, the green and yellow colors represent field-based steric interactions. To elaborate, the green regions within the molecule signify that the introduction of bulky substituents has the potential to augment its activity. Conversely, the yellow regions indicate that the inclusion of bulky components may diminish the molecule's activity. For instance, when examining the highly active molecule b46 (Fig. 2A), the presence of the green color in the benzene ring implies that incorporating bulky substitutions at this location could potentially enhance the molecule's activity. In the molecule with the lowest activity, molecule 3 (Fig. 2B), the presence of a green contour near C5 of the benzene molecule suggests that the addition of a bulky group at that position could potentially enhance the molecule's activity. The steric field's contribution to this observation was measured to be 0.73.

Electrostatic contour maps

In Figure 2C and 2D, field-based electrostatic interactions for the most reactive and least reactive molecules are depicted. The highly reactive molecule is represented by blue color, indicating electropositive regions, which have the potential to enhance its activity. On the other hand, the least reactive molecule is shown in red, signifying electronegative regions, suggesting that the presence of electronegative groups may increase or enhance its activity. It is worth noting that the contribution of the electrostatic force field was 0.27.

Steric and electrostatic contour maps for the highest, moderate, and lowest active compounds are shown in supplementary Figure S1. The combined electrostatic contour map unmistakably shows that incorporating both electropositive and electronegative groups will enhance the biological activity of the compounds.

Gaussian-based 3D-QSAR model

Table 2 presents the statistical details of the Gaussianbased QSAR model, encompassing the electrostatic, HBA, HBD, hydrophobic, and steric fields, along with their corresponding field contributions. The model's performance underwent evaluation through various methods. The LOO crossvalidation method yielded an r^2 cv value of 0.675, indicating



Figure 2. (A and B) Steric contour and (C and D) electrostatic contour map for the best field-based 3D - QSAR model. For the steric model, favorable regions were shown in green, and in the electrostatic contour map, favorable electropositive regions were shown in blue, and electronegative regions were shown in red color.

a moderate level of correlation. Furthermore, a non-cross-validation analysis resulted in an r^2 value of 0.9335, suggesting a robust correlation between the model's predictions and the actual data. The standard error of the estimate was measured at 0.32, and the *F* ratio was calculated to be 118, underscoring both the accuracy and significance of the model.

The contributions of the steric, electrostatic, hydrophobic, HBA, and HBD fields were 0.385, 0.099, 0.293, 0.161, and 0.063, respectively. In comparison, the steric (0.385) and hydrophobic

(0.293) field intensities exhibited higher contributions than the electrostatic, HBA, and HBD fields. This observation implies that in protein-ligand interactions, there is a greater emphasis on steric and hydrophobic fields, underscoring their significance in influencing the biological activity of compounds.

Gaussian steric contour maps

Figure 3A and 3a depicts the Gaussian steric contour maps for the highly active molecule (B46) and the least active



Figure 3. Gaussian steric contour maps for (A) most active (B46) and (a) least active (compound 3); Gaussian electrostatic contour maps for (B) most active (B46) and (b) least active (compound 3); Gaussian hydrophobic contour maps for (C) most active (B46) and (c) least active (compound 3); Gaussian HBA contour maps for (D) most active (B46) and (d) least active (compound 3); Gaussian HBD contour maps for (E) most active (B46) and (e) least active (compound 3).

molecule (compound 3). The green regions in the molecule indicate positions where the presence of bulky substituents is favorable and likely to enhance the activity. Conversely, the yellow regions suggest that incorporating bulky groups at these positions may diminish the activity. Notably, in the case of the least active molecule, it is evident that the introduction of bulky substitutions (indicated by the green color) near the first benzene ring of both B46 and compound 3 may enhance the compound's activity. The Gaussian steric contribution was determined to be 0.385, and this value was higher compared to the contributions of other factors, underscoring its significant influence on the compound's biological activity.

Gaussian electrostatic contour maps

The Gaussian electrostatic contour maps for the highly active molecule (B46) and the least active molecule (compound 3) are displayed in Figure 3B and 3b, respectively. In these figures, the blue and red contours symbolize electropositive and electronegative groups, respectively, which hold the potential to augment the biological activity of the compounds. The electrostatic contribution was determined to be 0.161, ranking as the second lowest when compared to other contributions. In case of the most active molecule, the presence of a large red

contour map around the hydrindane and a blue contour map at the benzene and methyl groups suggests that electronegative and electropositive atoms at these positions may positively influence the compound's activity, enhancing it. For the least active molecule, it is noteworthy that the addition of an electronegative group near the C4 methyl group of the first benzene ring could potentially increase the compound's activity.

Hydrophobic

In Figure 3C and 3c, the yellow color (hydrophobic groups) at that position is tolerated and the white contour represents unfavored hydrophobic groups. In the highest active molecule, the yellow contour near methyl benzene shows good activity. The white contour represents in second benzene ring represents the disfavored hydrophobic group. The overall contribution of hydrophobic in our model is 0.293.

HBA and HBD

In Figure 3D and 3d, an HBA contour map for the most and least active molecules is presented. The red contour in the HBA indicates improved activity, while the magenta contour suggests that the activity may be reduced due to the presence of HBA groups. Specifically, in compound 3,



Figure 4. Force field Scatter plot between observed and predicted activity of (A) training set and (B) test set compounds.



Figure 5. Gaussian Scatter plot between observed and predicted activity of (A) training set and (B) test set compounds.

a red contour near the oxygen atom of methoxybenzene is identified to favor the activity of the compound. This oxygen atom accepts electrons from LYS430 to form hydrogen bonds. Figure 3E and 3e represent the HBD contour. The purple contour represents the area favored by donors and unpreferred areas are shown in cyan color. The least activity of compound 3 may be due to the acceptor and donor (highlighted) forming a hydrogen bond with MET477. The contribution of HBA and HBD in the build model was 0.161 and 0.063, respectively.

Validation of field- and Gaussian-based 3D-QSAR models

The scatter plots, comparing predicted versus experimental activity for the training and test set inhibitors, are depicted in Figure 4 for the field-based 3D-QSAR model and in Figure 5 for the Gaussian-based 3D-QSAR model. These plots offer a visual representation of the model's performance in predicting the biological activity of the compounds.

Table 3. Field contribution of force field and Gaussian QSAR.

Field parameters	Force field QSAR	Gaussian QSAR
Force field steric	0.73	
Force field electrostatic	0.27	
Gaussian steric		0.385
Gaussian electrostatic		0.099
Gaussian hydrophobic		0.293
Gaussian HBA		0.161
Gaussian HBD		0.063

Gaussian Scatter plot between observed and predicted activity are provided in Supplementary Figure S2. The correlation coefficients (r^2) for the field-based model and Gaussian model are 0.9292 and 0.9335, respectively. The difference in r^2 values between the field-based and Gaussian models is not significant. The comparison between experimentally observed and predicted IC50 values reveals the Gaussian model's effectiveness in predicting the activities of both the training and test molecules. The statistical analysis underscores that both models exhibit robust predictive capabilities. Furthermore, they offer valuable insights into the chemical characteristics of the ligands, which can be instrumental in the development of inhibitors targeting BTK.

CONCLUSION

In conclusion, our study successfully developed robust 3D-QSAR models for BTK inhibitors, employing both fieldbased and Gaussian-based approaches. The field-based QSAR model, considering steric and electrostatic fields, demonstrated strong predictive power, reflected in an r^2 value of 0.92 and a q^2 value of 0.69. In parallel, the Gaussian-based model, incorporating five field intensities (Gaussian steric, Gaussian electrostatic, Gaussian H-BA, Gaussian H-BD, and Gaussian hydrophobic), exhibited outstanding performance, boasting an r^2 value of 0.93 and a q^2 value of 0.67. These models provide valuable insights for the rational design of novel compounds with potent BTK inhibitory activity.

AUTHOR CONTRIBUTIONS

All authors made substantial contributions to conception and design, acquisition of data, or analysis and interpretation of data; took part in drafting the article or revising it critically for important intellectual content; agreed to submit to the current journal; gave final approval of the version to be published; and agree to be accountable for all aspects of the work. All the authors are eligible to be an author as per the International Committee of Medical Journal Editors (ICMJE) requirements/guidelines.

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CONFLICTS OF INTEREST

The authors report no financial or any other conflicts of interest in this work.

ETHICAL APPROVALS

This study does not involve experiments on animals or human subjects.

DATA AVAILABILITY

All data generated and analyzed are included in this research article.

PUBLISHER'S NOTE

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USE OF ARTIFICIAL INTELLIGENCE (AI)-ASSISTED TECHNOLOGY

The authors declares that they have not used artificial intelligence (AI)-tools for writing and editing of the manuscript, and no images were manipulated using AI.

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The supplementary material can be accessed at the

link here: [https://japsonline.com/admin/php/uploadss/4601_pdf.pdf]

SUPPLEMENTARY MATERIAL

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