



α -Glucosidase inhibitor compounds of *Uncaria sclerophylla* leaves' most active chromatography fraction: *In vitro*, *in silico*, and ADMET analysis

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ABSTRACT

Diabetes mellitus is a noncontagious disease and a leading cause of death globally. It is characterized by a disorder in glucose metabolism, resulting in uncontrolled hyperglycemia. Various classes of type 2 DM therapy have demonstrated therapeutic benefits, but the global incidence continues to rise, necessitating the discovery of new therapeutic agents for type 2 diabetes. This study aims to determine the most potent *Uncaria sclerophylla* Roxb leaf fraction as an α -Glucosidase inhibitor, accompanied by the Absorption, Distribution, Metabolism, Excretion, and Toxicity (ADMET) profile and interaction mechanism of compounds in *U. sclerophylla* leaves. Extraction involved using four-graded maceration to obtain compounds with different polarities, followed by column chromatography of the most potent extract. The bioassay utilized spectrophotometry techniques and a microplate reader. Liquid chromatography quadrupole time-of-flight mass spectrometry (LC-QToF-MS) was employed to identify compound profiles in the most promising fractions, followed by *in silico* analysis of ADMET profiles and interaction mechanisms. The study found that ethyl acetate and methanol extract potency in α -Glucosidase inhibition with an IC_{50} of 75.75 and 42.70 μ g/ml, respectively. The most promising fraction, USMeth5, was obtained through column chromatography with better activity than acarbose with an IC_{50} of 22.85 μ g/ml. LC-QToF-MS analysis revealed the presence of various phenolic compounds, flavonoids, and alkaloids in USMeth5, along with ADMET profiles and interaction mechanisms *in silico*. The results of the study have successfully unveiled the potential of *U. sclerophylla* leaves and the most active fractions to be developed as a promising antidiabetic.

INTRODUCTION

Diabetes mellitus (DM) is a noncontagious disease and a leading cause of death globally. It is characterized by a disorder in glucose metabolism, resulting in uncontrolled hyperglycemia [1]. Chronic uncontrolled, abnormal high glucose level is a metabolic disorder caused by either a compromised insulin

action, an insufficiency of insulin secretion, or both [2]. The most common cases of DM are type 2 DM (T2DM), which reaches 90% of all DM cases [3]; continued T2DM raises the risk of mortality, diminishes the quality of life, and leads to higher treatment expenses [4]. Complications linked to T2DM include microvascular issues such as diabetes, kidney disease, retinopathy, and peripheral neuropathy, as well as macrovascular problems such as coronary heart disease, stroke, and peripheral arterial disease [5]. Cohort studies have shown that DM is associated with various cancers, functional cognitive disabilities, liver disease, affective disorders, and

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sleep disorders. These studies have also provided new insights into infection-related complications of DM [6]. According to the International Diabetes Federation, there were 10.5% of the global population, or 537 million people with diabetes, in 2021, which cost \$966 billion for diabetes treatment. This cost is estimated to continue to increase in 2024 to \$1054 billion. The global incidence of diabetes is estimated to increase alarmingly, reaching 643 million, or 11.3% of the worldwide population, in 2030 and 783 million (12.2% of the global population) in 2045 [1,7]. Various classes of T2DM therapy have demonstrated therapeutic benefits, but the global incidence continues to rise, necessitating the discovery of new therapeutic agents for type 2 diabetes [7,8].

α -Glucosidase inhibitors are commonly used as one of the initial therapies for T2DM. They are widely used to prevent T2DM in people at high risk, in pre-diabetic conditions, and in stage 1 and stage 2 diabetes; they are also combined with other T2DM therapies [9–11]. α -Glucosidase inhibitors can be given to individuals with kidney disease, and this makes them a suitable therapy for people living with diabetes complicated by nephropathy. Therefore, this class of therapy is more beneficial for people living with T2DM with kidney disorders compared to other therapy classes, such as sulfonylureas, metformin, SGLT2 inhibitors, and glinide, which are not recommended for individuals with kidney disorders [10,12]. α -Glucosidase inhibitors are vital enzymes that are important in converting polysaccharides into glucose, where these therapeutic agents can suppress the glucose absorption rate by delaying the digestion of polysaccharides. This delay in converting polysaccharides to glucose has proven very important in treating hyperglycemia, making it beneficial in treating T2DM [13,14].

Various studies have demonstrated that traditional medicinal plants used for treating diabetes show potential in both *in vitro* and *in vivo* studies and have also successfully reduced blood sugar levels in pre-clinical and clinical trials [15–18]. One genus of medicinal plants with potential antidiabetic properties is *Uncaria*, which is reported to contain phenols, flavonoids, terpenoids, and alkaloids [19,20]. Various studies on the antidiabetic activity of the *Uncaria* genus have shown the potential of this genus in treating diabetes, including *Uncaria laevigata*, *Uncaria tomentosa*, *Uncaria cordata*, *Uncaria gambier*, *Uncaria longiflora*, *Uncaria acida*, and *Uncaria callophylla*, both *in vitro* and *in vivo* [21–26].

Our previous study reported that *Uncaria sclerophylla* Roxb, a popular medicinal plant in Kalimantan, Indonesia, is used as a traditional anti-diabetic medicine, and studies have focused on the plant's potential as an antidiabetic using its stems and twigs [27]. However, the anti-diabetic properties of the leaves as an α -Glucosidase inhibitor have never been reported. This study explores the potential of *U. sclerophylla* leaves as an antidiabetic agent with inhibitory effects against α -Glucosidase. The study used a four-grade maceration technique and column chromatography fractionation to obtain the most promising fraction for further development as an antidiabetic therapy. The compounds in the selected fraction will be identified using UPLC-QToF/MS-MS, and their inhibitory effects against α -Glucosidase will be assessed using *in silico* methods. Additionally, the identified compounds' Absorption,

Distribution, Metabolism, Excretion, and Toxicity (ADMET) profiles will be evaluated *in silico*.

MATERIALS AND METHODS

Plant material

Uncaria sclerophylla leaves were collected from Meratus Forest in Kalimantan, Indonesia, during the dry season. The cleaned fresh leaves (8 kg) were dried in a blower oven at 35°C pulverized with a grinder, and sieved through a 40-mesh sieve, yielding 1.92 kg powdered leaves (24% yield), then refrigerated at 8°C while pending analysis. Plant authenticity was determined and deposited in the Pharmacognosy-Phytochemistry Laboratory, Faculty of Pharmacy, Universitas Indonesia (voucher specimen number 237/LB/XI/2021).

Chemicals and instrumentation

Chemicals: methanol, ethyl acetate, dichloromethane, n-hexane, silica Gel 70-230 mesh, TLC plate 254GF (Merck, Germany). Acarbose, α -Glucosidase enzyme, para-nitrophenyl- α -D-Glucopyranoside, bovine serum albumin, potassium dihydrogen phosphate, sodium carbonate, dimethylsulfoxide (Sigma-Aldrich, USA). Instrumentations: Microplate reader (Glomax Promega, UK), UPLC-QToF-MS/MS from Acquity UPLC H-Class System; Xevo G2-S QToF (Waters, USA).

Extraction

Uncaria sclerophylla leaf simplicia (1 kg) was extracted by four-level maceration using n-hexane, dichloromethane, ethyl acetate, and methanol as extraction solvents (1:20 ratio). The extraction process begins with n-hexane solvent, and the unextracted part is then extracted using dichloromethane. The same extraction process was continued with increasing polarity solvents such as ethyl acetate and methanol. Extract finishing is processed using a rotary evaporator and a blower oven at a temperature of 35°C. The resulting extracts are n-hexane extract, dichloromethane extract, ethyl acetate extract, and methanol extract.

Fractionation

Column chromatography was carried out in the selected extract fractionation with a combination of n-hexane, ethyl acetate, and methanol as solvent using a column with a length of 50 cm and a diameter of 3.5 cm. Silica gel (70–230 mesh)

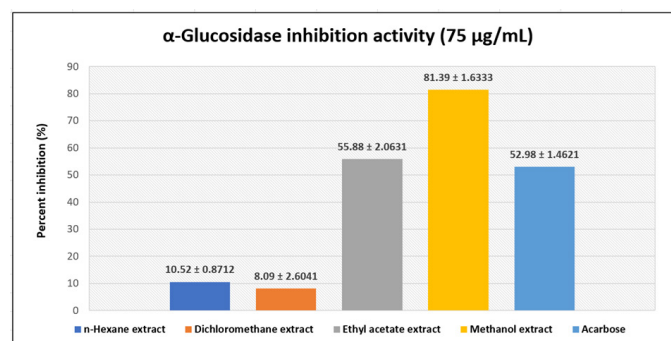


Figure 1. α -Glucosidase inhibition activity of extracts and acarbose.

Table 1. IC₅₀ of α -Glucosidase inhibition for selected extracts and acarbose.

Sample	Concentration ($\mu\text{g/ml}$)	Percent inhibition (%) \pm SD	Equation and R^2	IC ₅₀ ($\mu\text{g/ml}$)
Acarbose	60	47.35 \pm 2.2002	$Y = 0.3834X + 24.352$	66.90
	75	52.98 \pm 1.4621	$R^2 = 0.9998$	
	90	59.11 \pm 3.0027		
	120	70.18 \pm 1.1120		
	135	76.19 \pm 3.2245		
Methanol extract	12	6.79 \pm 2.1214	$Y = 1.3599X - 8.0637$	42.70
	24	26.33 \pm 1.6820	$R^2 = 0.9979$	
	36	41.19 \pm 2.1405		
	48	57.20 \pm 2.2645		
	60	72.95 \pm 1.8030		
Ethyl acetate extract	15	13.71 \pm 3.5866	$Y = 0.6218X + 2.8995$	75.75
	30	21.01 \pm 2.5157	$R^2 = 0.9951$	
	60	38.62 \pm 5.1110		
	90	57.48 \pm 6.0547		
	105	70.23 \pm 2.5449		

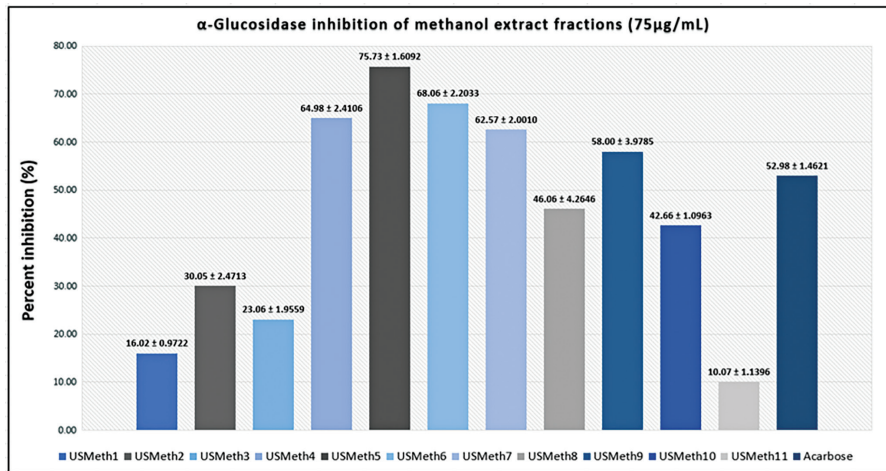
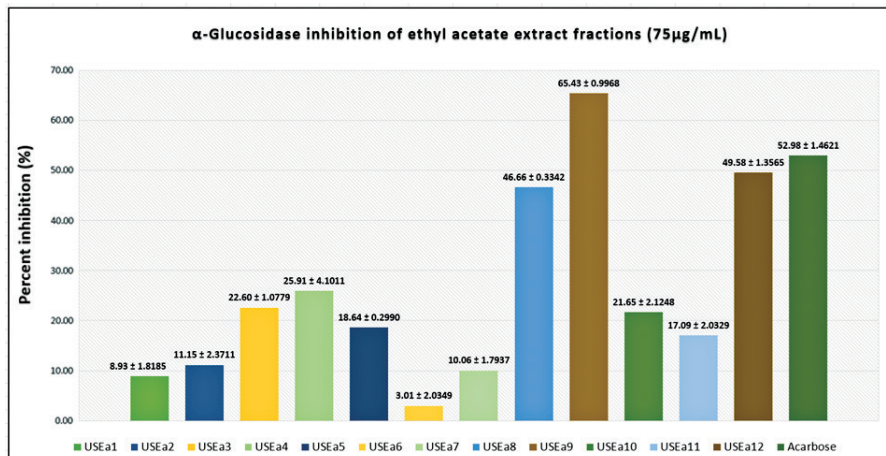
**Figure 2.** α -Glucosidase inhibition activity of methanol extract fractions and acarbose.**Figure 3.** α -Glucosidase inhibition activity of ethyl acetate extract fractions and acarbose.

Table 2. IC₅₀ of α -Glucosidase inhibition for selected fractions and acarbose.

Sample	Concentration (µg/ml)	Percent inhibition (%)± SD	Linear regression curve	IC ₅₀ (µg/ml)
Acarbose	60	47.35 ± 2.2002	$Y = 0.3834X + 24.352$	66.90
	75	52.98 ± 1.4621	$R^2 = 0.9998$	
	90	59.11 ± 3.0027		
	120	70.18 ± 1.1120		
	135	76.19 ± 3.2245		
USMeth4	22.5	32.50 ± 2.1136	$Y = 0.709X + 17.761$	45.47
	30	39.25 ± 4.4774		
	37.5	44.61 ± 0.2618	$R^2 = 0.9819$	
	45	52.03 ± 3.1480		
	52.5	54.72 ± 1.4506		
	60	58.95 ± 0.3631		
USMeth5	7.5	34.22 ± 4.1638	$Y = 0.9727X + 27.776$	22.85
	15	41.63 ± 4.8294		
	22.5	50.12 ± 2.1682	$R^2 = 0.986$	
	30	59.91 ± 1.4955		
	37.5	64.20 ± 2.1981		
	45	69.79 ± 2.1712		
USMeth6	7.5	20.36 ± 3.0884	$Y = 0.6106X + 16.826$	54.33
	15	27.45 ± 3.2513		
	22.5	29.22 ± 4.8009	$R^2 = 0.9863$	
	37.5	40.27 ± 0.9444		
	45	46.55 ± 4.6140		
	52.5	48.05 ± 1.1235		
USMeth7	7.5	19.73 ± 2.2258	$Y = 0.6381X + 15.878$	53.47
	15	25.19 ± 0.2519		
	30	36.27 ± 2.2282	$R^2 = 0.9902$	
	45	46.14 ± 4.4105		
	60	52.56 ± 0.2622		
USEa9	37.5	34.25 ± 0.7343	$Y = 1.0206X - 4.0001$	52.91
	45	40.02 ± 0.8344		
	52.5	51.09 ± 2.5814	$R^2 = 0.9904$	
	60	58.81 ± 0.6182		
	67.5	69.00 ± 0.9405		
	75	71.30 ± 0.6631		

in a ratio of 1:15 was chosen as the stationary phase during the fractionation process. The extracts (40 g of methanol extract and 15 g of ethyl acetate extract) were fractionated using a gradient mobile phase system starting with a combination of n-hexane and ethyl acetate at a ratio of 9:1, 8:2, 7:3, up to 0:10, further with a combination of ethyl acetate and methanol as a solvent in the ratio 9:1, 8:2, 7:3, up to 0:10. Thin-layer chromatography is used to analyze the chromatogram pattern for every 100 ml of elution results. The same or similar chromatogram patterns are combined into the same fraction.

α -Glucosidase inhibition activity

The α -Glucosidase inhibitory potential of extracts and fractions was measured by spectrophotometric principles using a microplate reader, according to the adopted method [27]. The procedure involved mixing 30 µl of samples (positive control, extract, and fraction) with 36 µl of phosphate buffer pH 6.8 and 17 µl of pNP-G (5 mM) in 96 wells, followed by an incubation at 37°C for 5 minutes. Subsequently, 17 µl of α -Glucosidase enzyme was added to the mixture, and the incubation continued at 37°C for 15 minutes. The enzyme reaction was terminated by adding 100 µl of Na₂CO₃ (267 mM). The p-nitrophenol compound, generated during the enzyme reaction, was measured at 405 nm with a microplate reader. Each test was performed in triplicate, and the standard deviation for each test was calculated.

Compound profiling using UPLC-ESI-QToF-MS/MS

The compound profile in the most active fraction was identified using a combination of UPLC with mass spectrometry and ESI as an ionizer. The liquid chromatography separation utilized a reverse phase technique with a C18 column (Acquity UPLC®, Waters, USA) at 40°C and autosampler temperature of 15°C. The mobile phase was a gradient system consisting of 0.1% formic acid in water and 0.1% formic acid in acetonitrile, with a 0.6 ml/minute flow rate. After separation via UPLC, the compounds were converted into ions using electrospray ionization in positive mode. The compound ions were then analyzed using Quadrupole Time-of-Flight mass spectrometry with a mass analysis range of 50–1,200 *m/z*. Mass spectrometry condition: the capillary voltage was 3 KV, cone voltage was 100 V, low collision energy of 6 eV, and high collision energy of 15–40 eV, source temperature was 120°C, desolvation temperature was 500°C, cone gas flow was 30 l/hour, desolvation gas flow was 1,000 l/hour, acquisition time was 20 minutes. Data processing and analysis were performed using Masslynx software (Waters) and the Unifi database.

Molecular docking

A total of 22 phytochemical compounds identified through compound profiling were analyzed. The 3D structures of 20 compounds were downloaded from the PubChem database, while the 3D structures of the remaining two compounds were generated using OpenBabel. The structure of the α -Glucosidase protein was downloaded from the research collaboratory for structural bioinformatics protein data bank (PDB) with the PDB ID: 3A4A [28]. The protein was processed by separating the native ligand from the protein and removing water molecules. Molecular docking was performed using AutoDock Vina through PyRx [29,30]. The center was set at *x* 21.595, *y* -7.436, and *z* 24.042, and the grid box for docking was configured to dimensions of 30 × 30 × 30 Å with a spacing of 0.375 Å. The exhaustiveness parameter was set to 200. Docking validation was performed by observing alpha-D-glucopyranose (native ligand) through redocking, and the results showed a root mean square deviation of less than 2 Å. The docking results were visualized using Discovery Studio 2021.

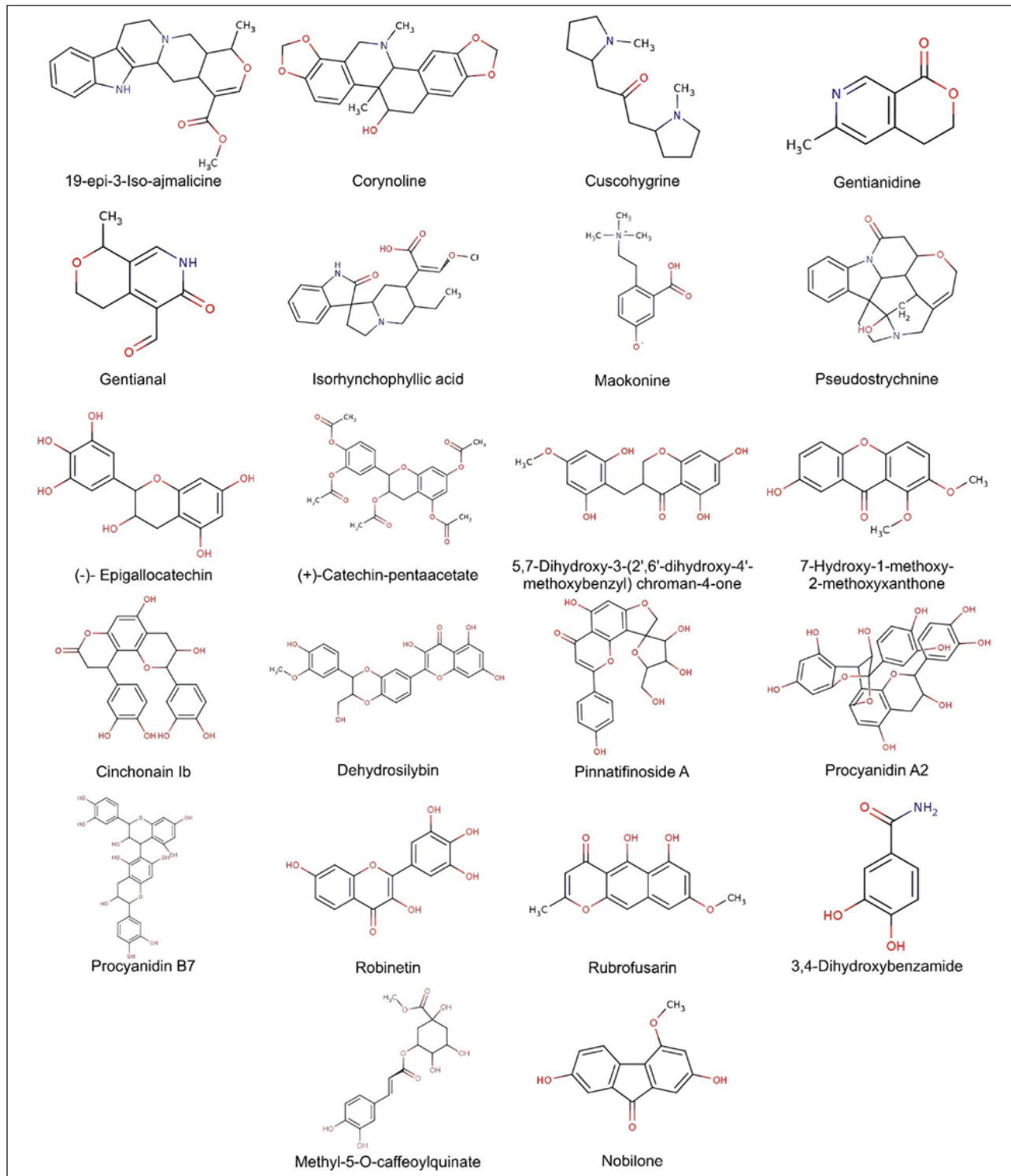


Figure 4. Structure of *U. sclerophylla* compounds used for docking.

ADMET analysis

Twenty-two phytochemical compounds from *U. sclerophylla* were analyzed for their ADMET profiles. This analysis utilized the Deep-Pk prediction tool (<https://biosig.lab.uq.edu.au/deppk/prediction>) to evaluate pharmacokinetic predictions and drug-likeness based on Lipinski's rule of five [31,32].

RESULTS

α -Glucosidase inhibition activity of extracts

The antidiabetic potential of *U. sclerophylla* leaves was investigated *in vitro*, focusing on their ability to inhibit α -Glucosidase. The study found that all extracts exhibited varying inhibitory activities. However, the ethyl acetate and

Table 3. Compounds detected in USMeth5 using LC-MS/MS.

Compound	Formula	Retention time (minute)	Ion mass (<i>m/z</i>) [M+H] ⁺	Total fragments found	Response
3,4- Dihydroxybenzamide	C ₇ H ₇ NO ₃	2.18	154.0494	2	1,527
Gentianal	C ₁₀ H ₁₁ NO ₃	3.69	194.0807	1	1,087
(-)- Epigallocatechin	C ₁₅ H ₁₄ O ₇	4.13	307.0805	1	699
Procyanidin B7	C ₃₀ H ₂₆ O ₁₂	5.85	579.1488	39	251,996
(+)-Catechin-pentaacetate	C ₂₅ H ₂₄ O ₁₁	7.00	501.1389	15	5,481
Pseudostrychnine	C ₂₁ H ₂₂ N ₂ O ₃	8.01	351.1690	23	798
Pinnatifinoside A	C ₂₁ H ₁₈ O ₉	8.38	415.1019	14	569
Methyl-5-O-caffeoylquininate	C ₁₇ H ₂₀ O ₉	8.60	369.1163	7	905
Robinetin	C ₁₅ H ₁₀ O ₇	8.99	303.0500	6	1,184
5,7-Dihydroxy-3- (2',6'-dihydroxy-4'-methoxybenzyl) chroman-4-one	C ₁₇ H ₁₆ O ₇	9.55	333.0969	15	8,366
Nobilone	C ₁₄ H ₁₀ O ₄	9.66	243.0650	3	2,329
Gentianidine	C ₉ H ₉ NO ₂	9.88	164.0701	2	1,858
Isorhynchophyllic acid	C ₂₁ H ₂₆ N ₂ O ₄	10.03	371.1958	44	89,173
Corynoline	C ₂₁ H ₂₁ NO ₅	10.02	368.1488	13	44,761
Dehydrosilybin	C ₂₅ H ₂₀ O ₁₀	10.52	481.1120	23	32,842
7-Hydroxy-1- methoxy-2- methoxyxanthone	C ₁₅ H ₁₀ O ₆	10.76	287.0538	15	147,426
Rubrofusarin	C ₁₅ H ₁₂ O ₅	10.78	273.0743	14	929
19-epi-3-Isoajmalicine	C ₂₁ H ₂₄ N ₂ O ₃	10.90	353.1853	131	169,787
Procyanidin A2	C ₃₀ H ₂₄ O ₁₂	11.29	577.1352	19	78,362
Cinchonain Ib	C ₂₄ H ₂₀ O ₉	11.72	453.1197	16	10,308
Maokonine	C ₁₂ H ₁₇ NO ₃	12.18	224.1290	56	4,207
Cuscohygrine	C ₁₃ H ₂₄ N ₂ O	16.65	225.1953	34	1,692

methanol extract of *U. sclerophylla* demonstrated superior activity compared to the other extracts; details refer to Figure 1. The ethyl acetate and methanol extracts of *U. sclerophylla* leaves were then analyzed for IC₅₀ values and compared with acarbose when the methanol extract of the leaves exhibited superior activity compared to acarbose (Table 1).

Fractionation and α -glucosidase inhibition activity of fractions

Ethyl acetate and methanol extracts of *U. sclerophylla* leaves have shown potential in α -Glucosidase inhibitory activity and were further explored for fractionation. These extracts were fractionated using column chromatography, producing 11 methanol and 12 ethyl acetate fractions. All fractions exhibited inhibitory activity against α -Glucosidase (Figs. 2 and 3), and several fractions with inhibition percentages above 60% were further analyzed for their IC₅₀ values. The fraction with the highest activity (USMeth5) was found to have an IC₅₀ value of 22.85 μ g/ml, outperforming acarbose as a positive comparator (Table 2).

Compound profiling of the most active fraction in α -glucosidase inhibition

The most promising fraction as α -Glucosidase inhibitor, USMeth5, underwent further analysis to determine its

compound profile using the UPLC-QToF-MS/MS technique. The compound profiling of USMeth5 revealed that this fraction contains alkaloid, flavonoid, and phenol compounds. The alkaloid compounds in USMeth5 are characterized by nitrogen atoms in their structure [33,34], the phenol structure is characterized by the presence of an aromatic ring with one or more hydroxyl group substituents [35], and flavonoid compounds are more specifically characterized by a C6-C3-C6 carbon backbone structure, a benzo- γ -pyrone, and a phenyl ring [36]. The chemical structure of the compounds found in USMeth5 is depicted in Figure 4. Table 3 presents information on compound profiling in USMeth5, including compound formulas, retention times, ion masses, total fragments found, and response.

Molecular docking

The compounds detected from *U. sclerophylla*'s most active fraction (Fig. 4) were studied to understand their interaction with the α -Glucosidase enzyme. Using molecular docking techniques, the researchers analyzed the active compounds and their binding with the enzyme. The findings revealed the specific compounds from *U. sclerophylla* that exhibited the most robust binding to α -Glucosidase with acarbose as drug control. Table 4 and Figure 5 summarize the details of these interactions and their binding potential.

Table 4. Molecular docking of compounds contained in USMeth5.

Compound	Compound group	Pubchem CID	Binding affinity (Kcal/mol)	Hydrogen bond	Hydrophobic bond	Electrostatic bond
Acarbose (Drug control)	-	41774	-8.83	ARG213, ARG442, GLU277, ASP352, ASP69, HIS280, TYR158, GLU411, LYS156	TYR158	-
19-epi-3-Iso-ajmalicine	Alkaloid	179461	-9.01	SER157	TYR158	-
Corynoline	Alkaloid	177014	-9.47	LYS156, TYR158,	TYR158, LYS156	-
Cuscohygrine	Alkaloid	1201543	-6.50	ARG442, TYR158, GLU441, ASP215, ASP69, HIS351, ASP352	TYR72, VAL216, PHE178	-
Gentianidine	Alkaloid	362908	-6.14	GLU277, ASP352, ARG315	VAL216, PHE178	-
Gentianal	Alkaloid	171304	-5.98	ASP307	PHE303	-
Isorhynchophyllilic acid	Alkaloid	102004526	-8.86	ARG315, GLU411, TYR158	TYR158, PHE178, ARG315	GLU277, ASP352
Maokonine	Alkaloid	54704413	-6.22	ARG315, ARG442, ASP352, GLN353, ASP307		
Pseudostrychnine	Alkaloid	21723446	-9.45	SER240	TYR158, LYS156	-
(-)-Epigallocatechin	Flavonoid	72277	-8.54	ASP215, GLU277, GLU411, ARG442, PHE178	ARG315, VAL216	ARG442, ASP352
(+)-Catechin-pentaacetate	Flavonoid	5315742	-9.26	ARG315, ARG442, SER240		GLU411
5,7-Dihydroxy-3-(2',6'-dihydroxy-4'-methoxybenzyl) chroman-4-one	Flavonoid	-	-8.19	ARG442, HIS280, SER311, GLU277	TYR158, PHE178, VAL216, ARG315	-
7-Hydroxy-1-methoxy-2-methoxyxanthone	Phenol	-	-7.63	GLN279, ARG442, TYR158	TYR158, ARG315	ASP352
Cinchonain Ib	Phenol	10456516	-10.40	ASN415, ARG442, ASP307, SER240,	TYR158, ARG315	-
Dehydroisilybin	Flavonoid	5467200	-9.50	GLU277, GLU411, ASP352, GLN353	PRO312, PHE303, HIS280, ARG315	ASP307
Pinnatifinoside A	Flavonoid	44257853	-8.99	ASN415, SER240, PRO312, HIS280	ARG315	-
Procyanidin A2	Flavonoid	124025	-8.36	PRO312, THR306, ASP69, GLU277, ASP352	TYR158, PHE178, HIS280, ARG315, VAL216	-
Procyanidin B7	Flavonoid	13990892	-9.79	HIS280, GLU411, ASP69, GLN353, GLN279, ARG442, LEU313	TYR158	ARG442, ASP352, GLU411
Robinetin	Flavonoid	5281692	-8.94	ASP242, ASP307	TYR158, LYS156, ARG315	-
Rubrofusarin	Phenol	72537	-8.24	GLN279, GLU277, ASP215	TYR158, PHE178, ARG315	ASP352
3,4-Dihydroxybenzamide	Phenol	148675	-6.15	ARG442, GLU277, ASP352, GLN353	-	ASP352
Methyl-5-O-caffeoylquinamate	Phenol	6476139	-8.46	ARG442, SER240, ASP307, ARG315	TYR158	-
Nobilone	Phenol	16104871	-7.65	-	TYR158, ARG315, PHE314, LYS156	-

Table 5. ADMET analysis of compounds contained in USMeth5.

Name compound	Lipinski's rule Violations	Absorption		Distribution		Metabolism				Excretion		Toxicity AMES
		HIA		VDss		Substrate		Inhibitor		Half-life of drug		
						CYP2D6	CYP3A4	CYP2D6	CYP3A4			
Acarbose (Drug control)	3	No Abs	0.2	No	No	No	No	No	No	>3	No	No
19-epi-3-Iso-ajmalicine	0	Abs	7.1	Yes	Yes	Yes	Yes	Yes	Yes	<3	No	No
Corynoline	0	Abs	4.2	Yes	Yes	Yes	Yes	Yes	Yes	<3	No	No
Cuscohygrine	0	Abs	1.8	Yes	No	No	No	No	No	<3	No	No
Gentianal	0	Abs	1.2	No	No	No	No	No	No	<3	No	No
Gentianidine	0	Abs	1.2	Yes	No	No	No	No	No	<3	No	No
Isorhynchophylllic acid	0	Abs	0.6	No	No	Yes	No	No	No	<3	No	No
Maokonine	0	Abs	0.0	No	No	No	No	No	No	<3	No	No
Pseudostrychnine	0	Abs	2.6	Yes	Yes	Yes	No	No	No	<3	No	No
(-)-Epigallocatechin	1	Abs	0.4	No	No	No	No	No	No	<3	No	No
(+)-Catechin-pentaacetate	2	Abs	2.5	No	No	No	No	No	Yes	<3	No	No
5,7-Dihydroxy-3-(2',6'-dihydroxy-4'-methoxybenzyl) chroman-4-one	0	Abs	0.9	No	No	No	No	No	Yes	<3	No	No
7-Hydroxy-1-methoxy-2-methoxyxanthone	0	Abs	1.0	Yes	No	No	No	No	Yes	<3	Toxic	
Cinchonain Ib	1	No Abs	0.5	No	No	No	No	No	No	<3	No	No
Dehydroilybin	0	Abs	0.6	No	No	No	No	No	No	<3	No	No
Pinnatifinoside A	0	Abs	0.8	No	No	No	No	No	No	<3	No	No
Procyanidin A2	3	No Abs	0.5	No	No	No	No	No	No	<3	No	No
Procyanidin B7	3	No Abs	0.4	No	No	No	No	No	No	<3	No	No
Robinetin	0	Abs	0.2	No	No	No	No	No	No	<3	No	No
Rubrofusarin	0	Abs	0.8	No	No	No	Yes	Yes	Yes	<3	Toxic	
3,4-Dihydroxybenzamide	0	Abs	0.8	No	No	No	No	No	No	<3	No	No
Methyl-5-O-caffeoylquininate	0	Abs	0.9	No	No	No	No	No	No	>3	No	No
Nobilone	0	Abs	0.8	No	No	Yes	Yes	Yes	Yes	<3	No	No

HIA: Human Intestinal Absorption.
VDss: Volume of Distribution at Steady State.

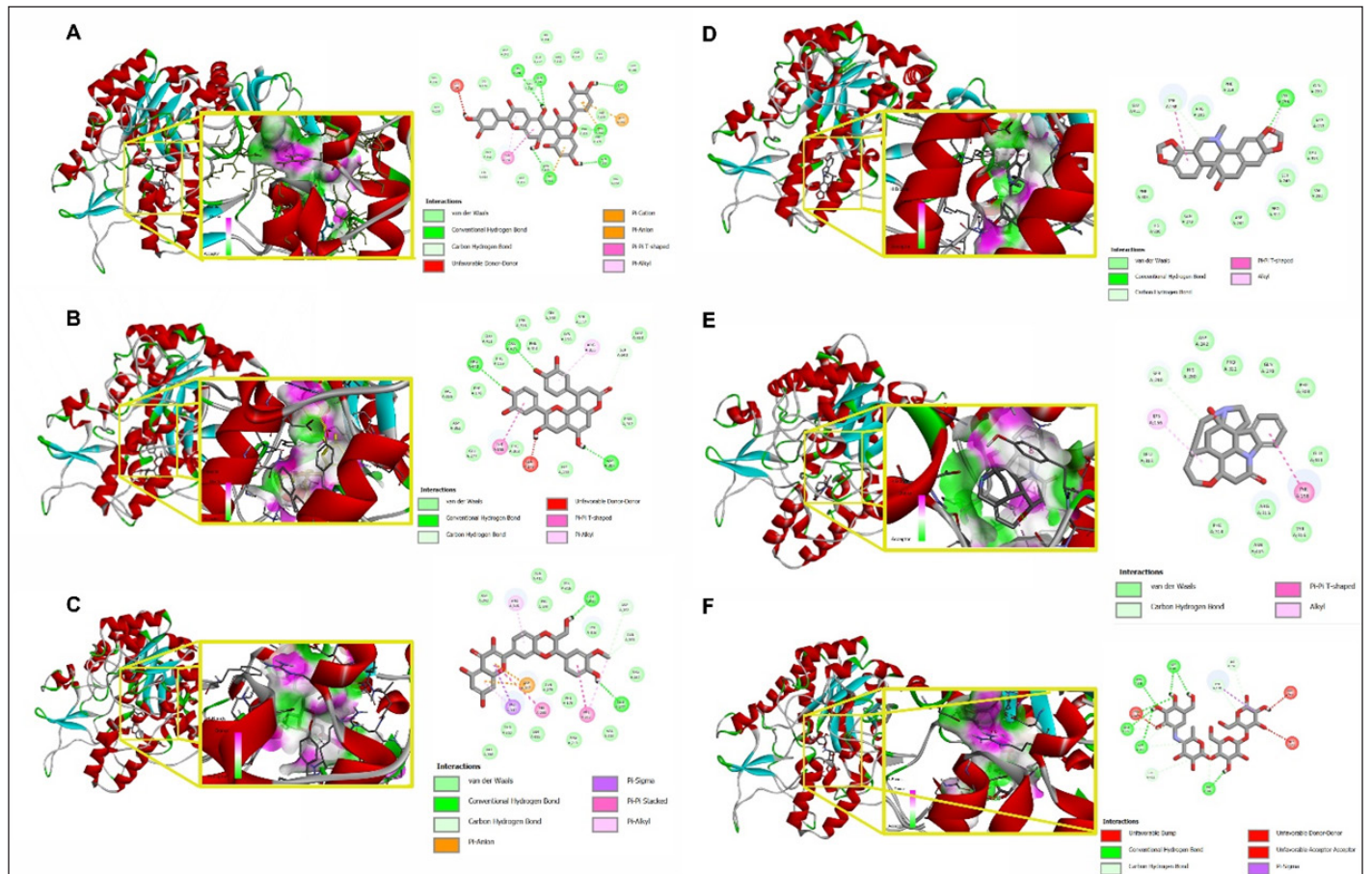


Figure 5. Molecular docking interaction of the five compounds with the best binding affinity (A) Cinchonain Ib, (B) Procyanidin B7, (C) Dehydrosilybin, (D) Corynoline, (E) Pseudostrychnine, along with Acarbose (F) as standard.

ADMET analysis

The evaluation of ADME and toxicity properties for the studied compounds adhered to standard pharmacokinetic criteria, with Lipinski's rule of five as a foundational guideline. This rule stipulates that compounds with molecular weights (MWs) ≤ 500 , hydrogen bond donors (HBDs) ≤ 5 , hydrogen bond acceptors (HBAs) ≤ 10 , and $\log P \leq 5$ are more likely to exhibit favorable oral bioavailability [32]. The compounds evaluated in this study largely conformed to these parameters, highlighting their potential as viable drug candidates.

Human intestinal absorption (HIA) values for nearly all compounds exceeded 30%, indicating good absorption capabilities. The volume of distribution at steady state (VD_{ss}) was greater than 0.45 log l/kg, further supporting their potential for effective systemic distribution. Moreover, the absence of significant metabolism by cytochrome P450 enzymes, specifically CYP2D6 and CYP3A4, suggests minimal interaction risks with these metabolic pathways, which can reduce adverse drug reactions. Importantly, none of the compounds tested positive for AMES toxicity, underscoring their safety profile in terms of mutagenicity [31]. These results provide a promising pharmacokinetic and safety outlook for the compounds, laying a strong foundation for further drug development investigations. The detailed ADMET parameters,

as summarized in Table 5, underscore their suitability for progressing into advanced stages of drug development.

DISCUSSION

The *U. sclerophylla* plant has long been renowned as a medicinal plant traditionally used by Kalimantan, Indonesia, for many years. Our previous study reported that the stem and twig parts of the *U. sclerophylla* plant have inhibitory activity against α -Glucosidase [27]. This current research explores the potential of *U. sclerophylla* leaves as an α -Glucosidase inhibitor, both *in vitro* and *in silico*. The leaves were extracted using four organic solvents simultaneously, employing a four-graded maceration technique to obtain compounds with various polarities contained in the leaves. The results showed that the methanol and ethyl acetate extracts exhibited the best activity in inhibiting α -Glucosidase, with an inhibition percentage of $81.39\% \pm 1.6333\%$ and $55.88\% \pm 2.0631\%$ at a sample concentration of 75 $\mu\text{g/ml}$.

Activity exploration continued to obtain the most effective fraction in α -Glucosidase inhibition using column chromatography fractionation on ethyl acetate and methanol extracts. Several fractions exhibited inhibitory activity above 60%, and the analysis continued to determine the IC₅₀ value to confirm the extent of its activity. The fractions analyzed for IC₅₀

(USMeth4-7 and USEa9) showed better activity than acarbose, a positive comparator, with an IC₅₀ range of 54.33–22.85 µg/ml. This demonstrates the superior potential of *U. sclerophylla* leaves as an α-Glucosidase inhibitor in treating diabetes.

USMeth5 is the most active fraction with IC₅₀ 22.85 µg/ml, indicating the potential of this fraction to be further investigated for the compounds contained therein, including prediction of the potential of these compounds *in silico*, including ADMET analysis. USMeth5 consists of alkaloid, phenol, and flavonoid compounds known for their potential as antidiabetics [35–40] and as inhibitors of α-Glucosidase [41–43]. In addition to being detected from USMeth5 as the most active fraction of *U. sclerophylla* leaves, the compound 19-epi-3-iso-ajmalicine was also detected from *U. hirsuta* [44] and *U. attenuata* [45]. The compound cuscohygrine found in certain plants shows potential in treating diabetes [46–48], as does the compound pseudostrychnine [49]. Phenol and flavonoid compounds contained in USMeth5 have been reported to have antidiabetic activity, namely the compound cinchonain 1b, which is reported to be able to induce insulin secretion *in vivo* and *in vitro* [50] and is also able to inhibit proteins that are important in diabetes therapy, namely α-Glucosidase and dipeptidyl peptidase-4 and has activity in glucose uptake [51], as well as rubrofusarin with its activity in inhibiting the PTP1B protein [52]. (-)-Epigallocatechin also has the potential for antidiabetic activity; this is supported by a study report where compounds with a catechin backbone have the potential for antidiabetic activity [53–55], including in the inhibition of α-Glucosidase [56,57]. Dehydrosilybin has been reported for its *in vivo* activity in overcoming diabetes complications and cardiomyopathy [58] and also has the potential as a glucokinase and PPARγ dual-target agonist [59]. Robinetin with antidiabetic potential [60] was also detected in USMeth5; this compound can alleviate liver metabolic failure and has hypoglycemic effects [61,62]. The compound 7-hydroxy-1-methoxy-2-methoxyxanthone was reported to be contained in fractions that can lower blood sugar levels [63] and in plants with the potential as α-Glucosidase inhibitors [64]. Procyanidins were reported to be contained in natural materials that can prevent postprandial hyperglycemia and have the potential to inhibit dipeptidyl peptidase-4 [65–67]. Some of the compounds detected in USMeth5 have never been reported for their antidiabetic activity, including inhibition of α-Glucosidase; this opens opportunities for exploring these compounds' activity in treating diabetes.

The *Uncaria* genus has the potential to be developed as an antidiabetic agent. Various studies have been conducted to ensure its activity, including *in vitro* studies on the leaves and stems of *U. gambir*, *U. acida*, *U. cordata*, *U. longiflora*, *U. lucida*, and *U. callophylla* [26,68]. *In vivo* studies on *U. tomentosa* showed hypoglycemic activity and the ability to delay diabetes progression [69]. Various α-Glucosidase inhibitor compounds have also been successfully isolated from the genus *Uncaria*; this strengthens the potential of this genus as an antidiabetic agent [70].

The molecular docking profiling of compounds from *U. sclerophylla* was conducted by comparing them to acarbose as a drug control. Ten compounds exhibited binding affinity values ranging from -10.4 to -8.86, namely cinchonain 1b, procyanidin

B7, dehydrosilybin, corynoline, pseudostrychnine, (+)-catechin-pentaacetate, 19-epi-3-iso-ajmalicine, pinnatifinoside A, robinetin, and isorhynchophylllic acid, which showed better binding processes compared to acarbose with a value of -8.83 (Table 4). Meanwhile, the other 12 compounds had binding affinity values ranging from -8.54 to -5.98. Lower binding affinity values indicate stronger binding and more effective inhibition potential against α-Glucosidase, suggesting that some compounds from *U. sclerophylla* may be more effective than the drug control in inhibiting the activity of this enzyme.

The docking results show that acarbose interacts through hydrogen bonds with the amino acids ARG213, ARG442, GLU277, ASP352, ASP69, HIS280, TYR158, GLU411, and LYS156, as well as through hydrophobic interactions with TYR158. Compounds from *U. sclerophylla* exhibit similar interaction patterns. For example, cinchonain 1b interacts with the residues ASN415, ARG442, ASP307, and SER240 through hydrogen bonds and with TYR158 and ARG315 through hydrophobic interactions. Procyanidin B7 forms more hydrogen bonds with residues such as HIS280, GLU411, ASP69, and GLN353, which are also involved in interactions with acarbose, and adds strong electrostatic interactions with ARG442, ASP352, and GLU411. Other compounds, such as dehydrosilybin, also show interactions with some of the same residues as acarbose, such as GLU411 and ASP352, which may contribute to their inhibitory potential. Although corynoline and pseudostrychnine exhibit simpler interaction patterns than acarbose, they still bind to the residue TYR158, which is key in acarbose's inhibition mechanism. This is shown in Table 4 and Figure 5.

Cinchonain 1b exhibits the lowest interaction energy, with hydrogen bonding at the ARG442 residue and hydrophobic interaction at the TYR158 residue, which is also key in the inhibitory mechanism of acarbose against the α-Glucosidase enzyme. This suggests that cinchonain 1b has the potential to be a more effective inhibitor than acarbose, with stable interactions at the enzyme's active site. In addition to cinchonain 1b, several other compounds from *U. sclerophylla* also show similar interaction patterns, indicating the presence of competitive inhibitor potential within these compounds.

The physicochemical properties of compounds from *U. sclerophylla* and acarbose (drug control) were analyzed based on Lipinski's rule, which includes criteria such as MW ≤500, HBD ≤5, HBA ≤10, and log P ≤5 [32]. The analysis revealed that almost all compounds comply with these rules, except for (-)-epigallocatechin, (+)-catechin-pentaacetate, cinchonain 1b, procyanidin A2, procyanidin B7, and acarbose which exhibit some deviations. Acarbose has a local effect in the small intestine, with less than 2% absorption into the systemic circulation [71]. This differs from other α-Glucosidase inhibitors, such as miglitol, with a high absorption profile in the systemic circulation [72]. A compound's ADMET properties are crucial parameters in drug discovery [73]. A compound is considered to have a favorable ADMET profile if it meets established criteria. One critical ADMET parameter is HIA, which measures how much a compound can be absorbed from an orally administered solution. The results of this study indicate that nearly all compounds have an HIA value above

30%, demonstrating good absorption capability through the human small intestine [31]. Another critical parameter is the VD_{ss}, which represents the theoretical volume required for the total drug dose to be uniformly distributed and produce the same concentration as in blood plasma [74]. A higher VD_{ss} value suggests that more of the drug is distributed into tissues rather than plasma. The predicted VD_{ss} value for a given compound is expressed in log l/kg. In this study, a compound is considered to have a favorable volume of distribution if its VD_{ss} value exceeds 0.45. The results show that nearly all compounds exhibit high VD_{ss} values, with none showing values below -0.15. The detailed results can be found in Table 5.

In the metabolism phase, Cytochrome P450 is a crucial class of detoxification enzymes predominantly found in the liver [75]. CYP450 enzymes play a key role in the catabolism of xenobiotics, facilitating their excretion through urine. Subgroups of these enzymes, such as CYP2D6 and CYP3A4, are particularly significant in drug metabolism [76,77]. This study identified that 19-epi-3-iso-ajmalicine, corynoline, and pseudostrychnine act as substrates and inhibitors for CYP2D6 and CYP3A4 enzymes. Additionally, to assess the mutagenic potential of these compounds, AMES toxicity tests were conducted using bacterial models [78]. The results indicated that 7-hydroxy-1-methoxy-2-methoxyxanthone and rubrofusarin are predicted to have toxicity based on the AMES test. Detailed results can be found in Table 5.

CONCLUSION

The leaves of *U. sclerophylla* have demonstrated potential as an antidiabetic agent by inhibiting α -Glucosidase. Methanol and ethyl acetate extracts from the leaves effectively inhibited α -Glucosidase, with IC₅₀ values of 42.70 and 75.75 μ g/ml, respectively. Further analysis of these extracts using column chromatography led to the discovery of USMeth5 as a more potent α -Glucosidase inhibitor with an IC₅₀ of 22.85 μ g/ml, outperforming acarbose. USMeth5 was found to contain various alkaloids, phenols, and flavonoids, and molecular docking revealed five compounds with better binding affinity than acarbose, including cinchonain Ib, procyanidin B7, dehydrosilybin, corynoline, and pseudostrychnine. These compounds were determined to be non-toxic and exhibited diverse pharmacokinetic profiles based on ADMET analysis. Further research is needed to explore additional antidiabetic mechanisms and to solidify the plant's potential in developing antidiabetic treatments.

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AUTHOR CONTRIBUTIONS

The authors have made significant contributions to the research, including the conception, design, data collection, analysis, and interpretation. The authors have been involved in drafting the article or critically revising it for intellectual content. The authors have agreed to the publication in the current journal, have given final approval for the version to be published, and have decided to be responsible for all aspects of the work. According to the guidelines of the International Committee of Medical Journal Editors (ICMJE), the authors are eligible to be considered as authors.

CONFLICT OF INTEREST

The authors declare that no conflict of interest could have influenced the result reported in this paper.

ETHICAL APPROVALS

This research does not involve human subjects or animals in the experiments.

DATA AVAILABILITY

The research data and analysis have all been presented in this article.

PUBLISHER'S NOTE

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USE OF ARTIFICIAL INTELLIGENCE (AI)-ASSISTED TECHNOLOGY

The authors declares that they have not used artificial intelligence (AI)-tools for writing and editing of the manuscript, and no images were manipulated using AI.

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