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A critical review of analytical methods for quantification of rivaroxaban in biological samples and pharmaceutical formulations

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ABSTRACT

Rivaroxaban (RXN) is a direct oral anticoagulant (direct inhibitor of factor Xa) used in the treatment of atrial fibrillation to reduce the risk of stroke caused by clot, periphery, and coronary heart disease. The foremost goal of the pharmaceutical drug is to exert its therapeutic action and cure the disease. Considering the therapeutic importance of RXN, this review focuses on the various spectrophotometric (UV) and chromatographic methods (thin-layer chromatography, high-performance thin-layer chromatography, high-performance liquid chromatography, ultraperformance liquid chromatography, liquid chromatography-tandem mass spectrometry, ultra-performance liquid chromatography-tandem mass spectrometry) available for the quality control and pharmacokinetic studies of RXN. The analytical methods published provide an overview of the various instrumental techniques that establishes the identity, quality, and purity of RXN. Stability-indicating methods anticipate the degradation products (DPs)/pathways of RXN which would help researchers to elucidate the structure of DP. The proposed review explores the greenness profiles of the reported analytical methods (2010-2024) based on the National Environmental Method Index in view of sustainable development goals 2030. The published bioanalytical methods provide a roadmap for potential researchers in developing more robust, efficient, and eco-friendly methods incorporating sorbent-based microextraction techniques to enhance biomedical research. Also, researchers incorporate analytical quality-based design (AQbD) which augments the quality and shrinks the out-of-trend and out-of-specification to expedite the regulatory flexibility in the analytical method. Overall this review illuminates the future canvassers to develop greener methods using eco-friendly solvents incorporating AQbD for the determination of RXN in biological fluids and drug products.

INTRODUCTION

Direct oral anticoagulants (DOACs) are widely prescribed in general practice to prevent clot formation. Rivaroxaban (RXN) an oxazolidine derivative is a novel DOAC usually referred to as a blood thinner agent which is extensively used to prevent deep vein thrombosis or pulmonary embolism. RXN is chemically 5-Chloro-N-({(5S)-2-oxo-3[4-(3-oxomorpholinyl)phenyl]-1,3oxazolidin-5-yl} methyl)2-thiophene carboxamide with an empirical formula of $C_{19}H_{18}C1N_3O_5S$ and 435.881g/mol as the molecular weight (Fig. 1). It was developed by Bayer and marketed by Janseen Pharmaceuticals, initially approved by FDA on July 1, 2011 [1–3]. Film-coated tablets of RXN are sold under the trade name Xarelto (2.5 mg, 10 mg, 15 mg, and 20 mg). It competitively inhibits free clot bound factor Xa and decreases the clotting ability of blood and prevents the formation of blood clots. In 2019, USFDA has approved RXN as an addon drug with aspirin for secondary prevention of peripheral arterial disease and acute coronary syndrome [1,2,4]. The development of new analytical methods is flourishing up day by day due to the advancement of hyphenated analytical instruments. The advancement of analytical instruments has led to cost-effective methods with improved precision and

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accuracy. An extensive review of various databases such as Science Direct, Springer, Pubmed, Scopus, Taylor & Francis, and Web of Science was accomplished. The published review lacks the detailed information required for the researchers with respect to the various published methods (Analytical, stability indicating, and bioanalytical methods), their merits and demerits, extraction techniques involved in the bioanalytical method, green assessment of the solvents used in various techniques [5]. Hence the present review surpasses the



Figure 1. Chemical structure of RXN.



Figure 2. Total number of analytical methods reported during the year 2012–2023. Databases: Science Direct, Springer, Pubmed, Scopus, Taylor & Francis and Web of Science.



Figure 3. Analytical methods for the estimation of RXN.

disadvantages of the existing review and embraces the future investigators with the overview of numerous simple and sophisticated analytical/bioanalytical methods for estimating RXN in different formulations (single and combined dosage forms) and biological matrixes. Also, since most of the methods [high-performance liquid chromatography (HPLC), HPLC/ultra-performance liquid chromatography-tandem mass spectrometry (UPLC-MS/MS)] developed for the estimation of RXN is stability indicating, a special emphasis on the forced degradation conditions and their results are incorporated separately which in turn help to improve the formulation, manufacturing process and determine the storage conditions of RXN. Bioanalytical methods determine the concentration of RXN and its metabolites in biological fluids, i.e., blood, serum, plasma, and urine by HPLC, HPLC/UPLC-MS/MS. Challenges associated during method development include employing suitable extraction techniques such as solid phase extraction (SPE), liquid-liquid extraction (LLE), and protein precipitation to extract RXN from the complex composition of a biological sample (presence of many interfering substances). This review discusses the pros and cons of extraction techniques essential to removing biological matrices and the applicability of the method. In addition, prominence is sited on the sustainability of the reported methods and assessment of greenness profiles as per the National Environmental Method Index (NEMI). Last, a comprehensive view of all the published analytical methods, consisting of chromatographic conditions and the results obtained, merits/demerits, application of the bioanalytical methods along with the greenness profile, is outlined in tabular form for a quick summary. Figure 2 reveals the different analytical methods for the estimation of RXN in single and combined dosage forms from 2012 to 2023. The graphical representation of a number of articles published from 2012 to 2023 is provided in Figure 3.

ANALYTICAL METHODS FOR DETERMINATION OF RXN

Spectrophotometry

Ultra- violet (UV)-Visible spectrophotometry is a simple, non-destructive, cost-effective versatile technique used to analyse pharmaceutical substances. Because of its simplicity, precise, reliability, and minimum solvent usage, it is widely used for the estimation of RXN. Sekaran et al. [6] developed a novel method using dimethyl sulphoxide as the diluent and reported the apparent molar absorptivity ($4.825 \times 104 \text{ l/mol/}$ cm) and Sandell's sensitivity (2.262 \times 10⁻³ µgcm⁻²) which represents the sensitiveness of the method (Fig. 4). Mustafa et al. [7] quantified RXN and also estimated the rate of drug release by in vitro study (Dissolution-USP Apparatus 2) medium. The drug release was 78% at 15 minutes using pH 4.5 acetate buffer containing 0.4% of sodium dodecyl sulfate as the dissolution medium. Muralikrishna et al. [8] reported a simple, sensitive, reproducible, and accurate method with methanol as the diluent. By using methanol as the diluent Kasad Pinaz *et al.* [9] reported a novel Area Under Curve method which would be better than zero-order UV spectrophotometric methods. Seshamamba et al. [10] reported derivative spectroscopy

with two different reagents, 4-Chloro-7-nitrobenzo-2-oxa-1, 3-diazole (Method I) and p-Dimethylaminocinnamaldehyde (Method II) for the precise and accurate determination of RXN in bulk and in its tablet dosage forms. El bagary et al. [11] reported a thermodynamic simultaneous method for the estimation of cilostazol and RXN based on the oxidation of iron (III) in the presence of 1, 10-phenanthroline to form tris (1, 10-phenanthroline) iron (II) complex (ferroin). The authors evaluated the thermodynamic parameters (free energy, enthalpy, and entropy) of complexation, also estimated the molar absorptivity $(53.02 \times 10^3 \text{ and } 1.28 \times 10^3 \text{ l/mol/ cm})$ and sandell sensitivity (0.0069 and 0.28 µg/cm⁻²) for cilostazol and RXN, respectively. Simultaneous estimation of RXN and Clopidogrel (CPL) using three UV Spectrophotometric methods (First derivative spectrophotometric method, ratio derivative spectrophotometric method, and absorbance ratio method) with methanol as the diluent was reported by Sharaf et al. [12] The reported methods were novel and distinct from each other, whereas methanol is the solvent of choice in most of the methods. The different methods of analysis were Area under the Curve, Derivative spectrophotometry, and thermodynamic method which indicated the applications and importance of the UV spectrophotometric method for the Quality Control of RXN



Figure 4. UV spectrum for RXN (adapted from reference 7).

S.No	Method/Matrix	Spectro	ophotometry Method		Re	sults		Ref no
		λ max (nm)	Diluent	Accuracy (%)	Linearity range (µg /ml) / R ²	Precision % RSD	LOD/ LOQ (µg /ml)	
1	UV-visible bulk & tablet	270	DMSO	99.82 – 100.50%	2-20/0.999	0.074%- 1.076%	0.212/0.642	Sekaran <i>et al.</i> [6]
2	UV-visible tablet dissolution	248	acetonitrile:Water(50:50 v/v) Acetate buffer with 0.4% SDS (pH 4.5)	102.38	0.5–5/0.9997	0.5– 50/0.9997	0.3/0.5	Celebier et al. [7]
3	UV-visible bulk & tablet	248.6	Methanol	100.85 %	2–12 μg/ ml/0.9998	0.08-0.915	0.09842/ 0.2982	Murali krishna <i>et al</i> . [8]
4	UV visible	241-260	Methanol	99.31%	2-12/0.999	0.537 % -0.297 %	0.059/0.179	Kasad Pinaz et al. [9]
5	UV-visible bulk & tablet	405	4-chloronitrobenzo- oxa-1,3-diazole in alkaline medium	99.74 - 100.51	2-20 µg/ml	0.621%- 0.900%	0.110 µg/ml	Seshamba et al. [10]
		545	p-Dimethyl aminocinnamaldehyde in acidic medium		25–125 µg/ml		0.483 µg/ml	
6	UV-visible tablet	510 ± 1 nm	1, 10-phenanthroline and formation of tris (1, 10-phenanthroline) iron (II) complex (ferroin)	NA	Cilastazol – 0.5–10 RXN – 30–400	NA	Cilastazol–0.046 RXN–4.719	El bagarey <i>et al.</i> [11]
7	Simultaneous estimation RXN and CPL first derivative spectrophotometric method	RXN :289 CPL:249.5	Methanol	NA	RXN :2.0–20.0 CPL:5.0-60	NA	RXN :0.211 CPL:0.361/ RXN:0.641 CPL:1.095	Sharaf <i>et al</i> . [12]
	Ratio derivative spectrophotometric method	RXN :256 CPL: 214.5	Methanol	NA	RXN : 2.0–20.0/ CPL: 5.0-60	NA	RXN: 0.137 CPL:0.485/ RXN:0.417 CPL :1.471	
	Absorbance Ratio method	RXN :232 CPL :249	Methanol	NA	RXN : 2.0–20.0 CPL:5.0-60	NA	RXN:0.272 CPL:0.485/ RXN:0.826 CPL:1.471	

Table 1. Spectrophotometry methods for determination of Rivaroxaban.



Figure 5. HPTLC densitogram of standard RXN (adapted from reference 15).

in pharmaceutical dosage forms [6-12]. The method and the spectrophotometric results were tabulated in Table 1.

Chromatography methods

High-performance thin-layer chromatography (HPTLC)

HPTLC an effective tool in qualitative analysis, is rapid and economically friendly compared to the hyphenated techniques. Prawez et al. [13] developed an eco-friendly method using RP-18 Silica gel 60 F254S as the stationary phase and Ethanol: Water (7:3 v/v) as the mobile phase using a densitometric detection at 253 nm which resulted in an Rf of 0.71 minutes. The method was highly expedient with a shorter run time, minimal sample preparation, and excellent recovery of 99.20% for the nano formulation (Fig. 5). Darshna et al. [14] developed a simple, specific, and cost-effective HPTLC method using Silica Gel 60 F254 under pure nitrogen stream as the stationary phase and Methanol: Toluene: Triethanolamine (7: 2.5:0.5 v/v) as the mobile phase. RXN was detected at 249 nm with an Rf of 0.60 minutes, linearity of 500-3,000 ng/spot, the limit of detection (LOD), and limit of quantification (LOQ) of 127.56 ng/spot and 386.57 ng/spot, respectively. The additives in the pharmaceutical formulations of the assayed analyte did not interfere with the determination of RXN. Both methods used an exclusive novel stationary phase, though the method showed a good recovery without compromising the usage of green solvents [13,14].

HPLC

HPLC, the most prevalent chromatographic method due to its simplicity, versatility, and broad scope. RP-HPLC uses a monolithic stationary phase and mixtures of organic solvents with buffers containing acidic/basic additives as mobile phases to separate moderately hydrophilic/hydrophobic compounds. Amelia *et al.* [15] Sahoo and Mekap *et al.* [16] and Eswarudu *et al.* [17] developed and validated a sensitive and accurate RP-HPLC method for the estimation of RXN. The chromatographic conditions of all three methods were similar, whereas optimization was done by varying the ratio of the mobile phase (Acetonitrile: Water). A sensitive and less time-consuming method with good recovery was achieved by Amelia *et al.* [15] Sunny *et al.* [18] developed an Ion Pair-RP-HPLC method using triethylamine as the ion pair reagent which improved the separation efficiency and peak shape of RXN. The method was cost-effective (Triethyl Amine Buffer (pH: 11.06): Acetonitrile (85:15 v/v)) and less time-consuming (R. 1.56 minutes) than the reported RP- HPLC methods. A sensitive method using a novel peak additive (Glacial acetic acid) in the mobile phase to improve the peak shape and efficiency of the method was developed by Meenakshi and Nageswara [19] Shivsankar et al. [20] developed an accurate, sensitive, precise, linear method but with a longer retention time (Rt) of 7.4 minutes. However, optimization of the mobile phase ratio would have resulted in a rapid method. Chandra shekar et al. [21] using the C18 column coupled with the guard column developed an RP-HPLC method that eluted RXN at 3.32 minutes. Awatade et al. [22] reported a method with a shorter run time and performed the qualitative analysis by IR Spectroscopy. A costeffective and sensitive method using Buffer : Acetonitrile (70 : 30) as the mobile phase and C18 column was developed by Khan et al. [23] Chromatographic results obtained from Simultaneous estimation by Rupali et al. [24] (RXN with CPL) and Sarkis et al. [25] (RXN with aspirin) resulted in a rapid, simple, reliable, precise and robust methods. The majority of the methods used C18 as the column with isocratic mode by use of different buffers (Acetate and phosphate buffer), peak modifiers (Triethyl amine, glacial acetic acid), varying the aqueous/organic composition (Water, Buffer/Acetonitrile, methanol), pH (Acidic/Basic) of the mobile phase and detected RXN by UV/PDA detector (220-280 nm). Researchers to develop a method employing eco-friendly solvents that would meet the SDGs. Analytical quality by design (AQbD), an extension of QbD offers a robust approach in the development of analytical methods leading to higher quality products and improved patient safety. An exclusive AQbD method by Santos et al. [26] for the enantiometric estimation of S - RXN and its chiral impurity R- RXN by RP-HPLC using a chiral stationary phase. The method was selective, linear, precise, accurate, and sensitive with a LOQ of 0.68 µg/ml for S-RXN and $1.0 \,\mu\text{g/ml}$ for the chiral impurity [15–26]. The chromatographic condition of the RP-HPLC methods and their results are presented in Table 2

STABILITY INDICATING METHOD

An ideal stability-indicating method discriminates the analyte and its degradation products (DPs) efficiently. Forced degradation studies aids in synthesis, formulation development, and packaging where knowledge of chemical behavior may be applied to enhance a therapeutic product [27].

UV Spectrophotometry

UV spectrophotometry is a cost-effective method that can be used for stress degradation studies of RXN and its degraded products. Girishchandra *et al.* [28] developed an accurate and reliable method in which the assay limit of RXN was as per the WHO limit (NLT 97 % and NMT 101.05 %). RXN degrade much especially on exposure to UV light and acidic medium but do not degrade in basic medium.

Chromatography methods

HPLC

HPLC methods can separate, detect, and quantify the various drug-related degradants of RXN formed on

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Table

S.No	Method		Chromatog	graphic Cor	ıditions			Res	ults		Greeness	Merits/	Ref. no
		Stationary Phase	Mobile Phase	Flow rate mL/ minute	Temperature (°C)	Detector/ detection wavelength (nm)	Rt (minute)	Linearity range (μg/ml)/ R2/ %RSD	% Recovery	LOD/	profile	Demertis	
	RP-HPLC	C18 (250 × 4.6 mm,5 μm)	Acetonitrile: water (80:20 v/v)	-	40	UV/247	2.986	10-70/0.996/<2%	99%100%	0.009381/0.07862		Sensitve method with shorter run time	Amelia <i>et al.</i> [15]
	HPLC	Phenomene× C18 (250 × 4.6 mm, 5 µm)	Acetonitrile : Water (55:45 v/v)	1.2	35	UV/251	3.8	5.0-40	0.99	V N	\bigcirc	Simple, linear, accurate method	Suraj and Mekap [16]
ц	HPLC Method	C18 (250 × 4.6 mm, 5 µm)	Acetonitrile: Water (50:50)	-	Ч	PDA/251	4.893	5 -50/0.9992/	98.84	0.054/0.164/0.2	\bigcirc	Simple, linear method with PDA detection	Eswarudu <i>et al.</i> [17]
	НРСС	Agilent Eclipse C8 LC (150 × 4.6 mm, 5 µm)	Triethyl Amine Buffer (pH: 11.06) : Acetonitrile (85:15 v/v)	1.2	25	UV/251	1.56	500-1/0.999 /<2	98–102.4	0.675/0.0786		Use of Peak modifiers, very less time consuming method, One and only method with C8 as the stationary phase	Sunny <i>et al.</i> [18]
	RP-HPLC	C18 (250 × 4.6 mm, 5 µm)	0.1% Glacial acetic acid : Acetonitrile (30:70v/v)	-	NA	PDA/250	3.44	2-10/0.9993/~2%	98%102%	0.008/0.248	\bigcirc	Acetic acid improves the peak shape and efficiency of the method	Meenakshi <i>et al.</i> [19]
	RP-HPLC	C18 (250 × 4.6 mm, 5 μm)	Potassium di hydrogen phosphate Buffer (PH3): ACN (60:40%)	-	NA	UV/248	7.45	1-5/ 0.9978/<1	100.8	0.093/0.2768	\bigcirc	Sensitive method with a longer run time	Shivasankar et al. [20]
	RP-HPLC Method	C18 (250 × 4.6 mm,5 µm) coupled with Silica guard column	Acetonitrile: Methanol : Ortho phosphoric acid (90:8:2) pH : 4.06	1.5	ΥN	UV/234	3.32	50 –200/0.997/0.47 – 0.609	99.708	0.75/2.47	\bigcirc	Only method coupled with guard column with a greater flow rate	Shekar <i>et al.</i> [21]
<u>н</u>	HPLC Method	Sun Q C18	Acetonitrile : Buffer 80 : 20 (Sodium Acetate)	-	ambient	UV/249	3.69	5-30/0.994	100.941	0.439/1.33/0.288	\bigcirc	Efficient column at low pH	Awatade <i>et al.</i> [22]

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Continued



storage/manufacturing [29,30]. Hyphenated techniques (HPLC/UPLC-MS/MS) have become a golden standard in the characterization and quantitative analysis of degradants with improved selectivity, sensitivity, and efficiency. The chromatographic conditions of the stability-indicating methods are represented in Table 2. Kasad Pinaz et al. [31] developed and validated a simple, cost effective, less time-consuming, precise and accurate method in which the acidic (3 DP), basic (4 DP), photolytic, and thermal degradation peaks was clearly separated from RXN peak. However, characterization of the degradants would help researchers to understand the degradation pattern of RXN. Seshamamba et al. [32] Souri et al. [33] Zareen et al. [34] Çelebier et al. [35] performed a sensitive method with UV detection and isocratic elution. The methods were efficient in detecting the degradants. Seshamamba et al. [32] used a cyano column which has led to an efficient and sensitive than the other methods. Souri et al. [33] proposed a stability-indicating method that revealed the unstability of RXN in stress degradation conditions, also reported the dissolution profile which indicated >90% release of RXN within 45 minutes (Dissolution medium: 0.6% sodium lauryl sulfate in acetate buffer, Apparatus: paddle at 75 rpm). Walter et al. [36] developed a specific and stability-indicating method with an accuracy of 99.77%, hence the method can be applied for the analysis in human plasma and tablet dosage forms. Girase *et al.* [37] Badroon and Sreeramulu *et al.* [38] Sachin et al. [39] reported a stability-indicating analytical method using gradient elution and diode array detector (DAD) detection. The peak obtained during the forced degradation condition was homogenous and unaffected by degradation impurities [38]. Monitoring of the commercial route of synthesis and impurity profiling which is very important for the quality of RXN by RP-HPLC was performed by Girase et al. [37] and Sachin et al. [39] RXN was found to degrade significantly in Acid condition, but impurities (1-10) were well separated with an less RRT and the mass balance of all the peaks was found to be satisfactory. The method developed by Sachin et al. [39] eluted all the potential impurities of RXN (Impurity A-H) within 14 minutes [31–41].

UPLC

Ultra-performance liquid chromatography (UPLC) using shorter columns improves analyte resolution and sensitivity, lowers solvent consumption, and shortens run times. Rajan and Anver Basha et al. [42] and Rao et al. [43] reported a UPLC-DAD method for the estimation of RXN using gradient elution. Both the authors characterized the impurities. Rao et al. [43] developed a novel, rapid, stable, sensitive ion pair -UPLC - DAD method using ion pair reagents (octane sulphonic acid sodium) as the mobile phase for estimating RXN in tablet dosage forms. The anionic pair reagent improves the peak shape and retention of RXN and the impurities were characterized. Jabaliya et al. [44] transferred the HPLC method to UPLC owing to increased resolution, sensitivity, and reduction in run time [42–44]. The results of UPLC indicated an accurate, precise, linear method than the developed HPLC method. The method would be a great advantage if the impurities were characterized by UPLC-DAD.

Liquid chromatography-tandem mass spectrometry (LC-MS/MS)

LC-MS/MS is an ideal technique for detecting nanomolar/picomolar quantities of various drugs, degradants, and food metabolites. Cholleti et al. [45] developed and validated an HPLC method and identified the degradants by LC-MS/MS method which separated enantiomers and all the process-related impurities of RXN. Arous et al. [46] investigated and found RXN susceptible to acid and base hydrolytic stress conditions. The authors identified, isolated, and characterized the degradation pathways using LC-MS, thin layer chromatography (TLC)densitometry, NMR spectroscopy, and FT-IR. Three major DPs (DP-1, DP-2, DP-3) were detected, separated, and determined and two (DP-2 and DP-3) of them were further characterized by NMR spectroscopy and FT-IR. Ramisetti and Kuntamukkala et al. [47] presented a simple, rapid, and isocratic stability indicating HPLC method for the determination of RXN which detected 3 DPs formed under stress conditions. The structures of Process related impurities and the DPs (DP-1 and DP-3) were quantified by high-resolution ESI-MS/MS analysis and characterized by NMR spectroscopy (Process-related impurities and DP-3 [45-47].

The stability-indicating methods reported were highly sensitive with shorter Rt indicating a cost-effective green approach (Table 3). Most of the methods were stability indicating and detected the degradants employing a gradient program; however, characterization of the degradants formed in the HPLC method could be a better way ahead in these stress degradation studies. Characterization of impurities of RXN and degradation pathways by LC-MS/MS methods establishes biological safety which reveals the need and scope of impurity profiling of RXN in pharmaceutical research.

COMBINED ANALYTICAL TECHNIQUES

RXN is estimated by combined analytical techniques to indicate the accuracy of a method with other methods. These articles provide us the collective information in a single paper. Lories et al. [48] reported a HPLC, TLC, and UV method for the estimation of RXN in single dosage form. Method A : Stability indicating analytical method-HPLC method with phenomenex-C18, $(150 \times 4.6 \text{ mm}, 5 \mu\text{m})$ column and 1.2% w/v Potassium Dihydrogen Phosphate pH 3.5 ± 0.2 and Acetonitrile (70:30, v/v) as the mobile phase at a flow rate of 1.5 ml/minute with UV Detection at 280 nm to detect the impurities. RXN, Timolol Maleate degradant and Bimatoprost degradant eluted at 8.5, 5.3, and 4.2 minute, respectively. The Correlation Coefficient (R²), % recovery, LOD, and LOQ of Timolol Maleate degradant and Bimatoprost degradant were 0.999, 100.84, 0.62 µg/ml, 0.65 µg/ml, and 0.999, 99.67, 1.3 μg/ml, 3.15 μg/ml, respectively. Method B: TLC method used precoated Silica Gel GF254, 0.25 mm as the stationary phase and Chloroform: Isobutyl Alcohol (50:50 v/v) as the mobile phase with Densitometric detection at 280 nm. The R_c of RXN was 0.60 ± 0.02 with a % RSD of $0.50-4.80 \mu g/spot$. Method C: First derivative method at a wavelength of 237.4 nm was linear over the range of 1.6–22.4 µg/ml. Method D: First derivative ratio method using acetonitrile as the solvent at a wavelength of 236 nm showed a R², %RSD, %Recovery, LOD, LOQ of 0.9999, 1.072 %, 99.99 ± 0.242 , 0.52μ g/ml and 1.85μ g/ml [48]. HPLC method emerged to be the sensitive and effective method in detecting the degradants at a short run time. Combinatorial methods help researchers explore the pros and cons of various methods from multiple angles and get more confined results.

BIO ANALYTICAL METHODS

Bio-analytical method development is useful in the identity, purity, potency, and bioavailability of drugs. The data that are obtained from bioequivalence and pharmacokinetic studies are required for the successful evaluation of drugs in animal models and human trials. The development of effective bio-analytical methods is therefore crucial for the successful implementation of drug development. Hyphenated techniques have turned out to be quintessential in the assessment of pharmaceuticals in various biological samples. Bio-analytical methods employing HPLC/UPLC/hyphenated techniques are ephemeral and assist in controlling the quality of pharmaceuticals as well as in obtaining pharmacokinetic and toxicokinetic data of RXN in single and combined dosage forms. RXN was extracted from biological samples (Rat Plasma, human plasma/serum, patient samples) employing SPE, LLE, and protein precipitation. Numerous variations depending on the nature and amount of solvent/matrix in the extraction method will be optimized to achieve a better separation of RXN from the biological matrix. Structurally related internal standards used in the research were chloramphenicol, Prednisolone, carbamazepine, Apixaban (AXN), Diazepam, and Risperidone. Few researchers have used stable isotopically labeled (SIL) internal standards and deuterated internal standards which yield robust and reproducible results compared to structurally related internal standards.

UV Spectrophotometry

UV Spectrophotometry has emerged as an effective method for estimating the analyte from biological samples. Bhavyasri *et al.* [49] developed a novel UV Spectrophotometric bioanalytical method using human plasma as the matrix and detected RXN at 252 nm with Acetonitrile: Water (60: 40) as the diluent. De proteination of plasma with protein precipitation extraction using Acetonitrile depicted a good recovery (99.04 %–100.74 %), linear (1–20 ppm), LOD/LOQ (0.0121 μ g/ml/ 0.0368 μ g/ml), and precision (0.02%–0.03 %) which proved to be a simple, cost-effective and sensitive method in the estimation of RXN in spiked human plasma [49].

Chromatography Methods

HPTLC

HPTLC is the most flexible, reliable, and cost-efficient separation technique so become one of the most popular methods for bio analysis. Shukla *et al.* [50] reported a bioanalytical method for the estimation of RXN in human plasma. Protein precipitation with acetonitrile as the precipitating reagent gave a compact band of $R_rat 0.44$ minutes with a recovery of 66.95%–69.03 %. This method was applied to a pharmacokinetic study of RXN and estimated the C_{max} (63.83 ng/ml) and t_{max} (47.3 ng/ml), area under the curve_{0-t} (290 and 219.0 ng/ml/hour), and

hod		Chromatogra	phic Cond	itions			Re	esults		Greeness	Merits/Demerits	Ref. no
Statio	nary Phase	Mobile Phase	Flow rate ml/ min	Temperature (°C)	Detector/ detection wavelength (nm)	Rt (min)	Linearity range (µg/ml)/ R2/ %RSD	% Recovery	001 ГОД	profile		
н _{СI}	enomenex $8(250 \times 4.6 \text{ m}, 5 \mu \text{m}),$	Methanol: Acetonitrile (50:50, v/v)	-	NA	UV/250	1.8	20–100/ 0.9999/ 0.143%	% 68.66	0.1277/0.387 µg/ml		Simple HPLC-UV method for acidic degradation/ Characterization would have improved the effectiveness of the method	Pinaz and Muralikrishna [31]
Cl Cl	omenex Luna 8 (250 × 4.6 mm,5 μm)	Methanol: acetonitrile (50:50 v/v)	1.2	NA	PDA/250	2.725	20–100 / 09995/0.143	100.5%	0.1277/0.3872	\bigcirc	No characterization of the degradants	Kasad [40]
Ph (2)	enomenex C18 $50 \times 4.6 \text{ mm}$, μ m), 100°A particle size	Methanol : Acetonitrile (50:50v/v)	Т	NA	PDA/250	1.7	20–100 /0.99995/ 0.143%	100.85%	0.1277/0.3872	\bigcirc	No characterization of the degradants	Kasad Pinaz and Muralikrishna [41]
0	N (250 × 4.6 mm, 5 μm)	0.1M sodium acetate: Methanol (60.40, v/v)	1	30	UV/247	4.46	1–120/0.9992/ 0.421	99.74%	0.194/0.648	\bigcirc	A very sensitive stability indicating cost effective HPLC-DAD method	Seshamamba et al. [32]
4	Column:C18 (150 × 4.6 mm, 5 μm)	Acetonitrile : Water (70:30 v/v)	0.7	40	PDA/249	2.9 minutes	0.04-200/ 0.9992	77.66	NA	\bigcirc	No characterization of the degradants	Walter <i>et al.</i> [36]
•	C18 (250 × 4.6 mm, 3.5 μm)	0.02M po tassium hydrogen phosphate (Acetonitrile: Methanol) Gradient elution	-	45	PDA/240	RRT -1.0	50–250 /0.99/0.72%– 2.44%	80%-120 %	0.018/ 0.03		Efficient method to characterize the degradants with good resolution	Girase et al. [37]
0 8	Solumn: Inertsil 8 (250 × 4.6 mm, 5 μ)	potassium Phosphate Buffer (pH3) : Acetonitrile Gradient elution	-	40	PDA/250	23.05 minutes	300– 700/0.9987/ 0.89 %	99.5%-97.5%	NA	\bigcirc	No characterization of the degradants	Badroon and Sreeramulu [38]
	łova-Pak C8 (4 n, 150 mm × 3.9 mm)	Acetonitrile: Potassium Dihydrogen phosphate 50 mM (pH 3.0) (40:60, v/v)	-	ΥN	UV/270	2.5	1–50/ 0.999/0.61	ΥN	0.58/0.19	\bigcirc	Degradation with respect to 80°C and room temperature	Souri <i>et al.</i> [33]
C	$218 (250 \times 4.6 \text{ mm}, 5 \mu)$	0.01 M potassium dihydrogen phosphate buffer : Acetonitrile (20 : 80) Gradient Elution	-	35	PDA/250	3.2	50–500/0.999/ 0.3%–0.9 %	98.9 %-110.1 %	ΥN	\bigcirc	Efficient HPLC-PDA method to characterize the impurities.	Garad and Mane [39]

Continued

Sayeda and Hangad [34]	Çelebier <i>et al.</i> [35]	Jebaliya <i>et al.</i> [44]	Rajan and Anver Basha [42]	Rao <i>et al.</i> [43]	Cholleti <i>et al.</i> [45]	Arous <i>et al.</i> [46]	Ramishetti and Kuntamukkala [47]
Eco-friendly HPLC-UV method	No characterization of the degradants	One and only stability indicating method with successful technology transfer to UPLC	Sensitive UPLC- PDA method with characterization of impurities	UPLC-PDA method with efficient characterization of impurities	Development of HPLC method and Characterization of impurities by LC-MS/MS	Elucidation of structure of the degradants	Elucidation of structure of the degradants
	\bigcirc		\bigcirc	\bigcirc	\bigcirc	\bigcirc	\bigcirc
NA	0.005/0.01	0.08/0.075 0.175/0.150	0.0005/ 0.0015	0.05-0.30	¥ Z	0.676/2.230	0.05/0.15
98.31–101.60 %	NA	99 to 101/99 to 101%	98.4% and 103.5%	93.9-107.9	ΥZ	NA	99.29 %
0.005-40	0.005-40.0/ 0.9997/1.17	20-80/10- 40/0.999	666.0	50–150 /0.9995/ <6.5%	90-100 %	300–1,000 /0.9997	50–150/0.9997/ 0.9%
3.37	Caffeine : 2.21 RXN : 3.37	NA	15.0	4.089	۲Z	25.67 Impurity A : 20.76	4.15
UV/250	UV/249	PDA/280	PDA/254	PDA/ 248	254	UV/254	RXN: m/z 436 DP-1: m/z 454 DP-3 : m/z 410 and 408
30	40	30	30	25	27	55	30
0.8	1.2	1/0.25	-	0.45	0.5	1.5	1.0
Methanol : Water (50:50)	Acetonitrile :Water (55:45 v/v) mixture	0.1% Orthophosphoric acid : Acetonitrile (60: 40 V/V)/0.1% Orthophosphoric acid : acetonitrile (50:45 v/v)	Mobile phase A - 0.05M diammonium hydrogen phosphate (pH 3.0) : acetonitrile (80.20, v/v) and Mobile phase Mobile phase Mobile phase : water (90:10, v/v) gradient programme	ImL Ortho phosphoric acid and 10 mM Octanel- sulphonic acid sodium Buffer and acetonitrile A (90:10) B(20:80) Acetonitrile (60:40%) Gradient elution	A:BA: Acetonitrile : Ethanol: n-butylamine B: Water : Methanol : n-butylamine (50 : 50 : 0.5)	Acetonitrile and Water	20 mM ammonium acetate and acetonitrile (65:35 v/v)
C18 (250 × 4.6 mm, 5 μ)	Phenomenex Luna C18 LC (250 × 4.6 mm,5 μm)	C8 100A (250 × 4.6 mm id, 5 µm)/ UPLC Column : Acquity UPLC (R) BEH C8 (100 × 2.1 mm id, 1.7 µm particle size)	Acquity BEH C8 column (100 mm × 2.1 mm, 1.7 µm)	Acquity UPLC BEH HSS T3 (100 mm × 2.1-mm, 1.8-µm)	Chiralpak IC (250 \times 4.6 mm, 5 μ)	C18 (250 × 4.6 mm 5 μm)	Kinetex C18 (150 ×4.6 mm, 5 µm) column
Stability Indicating Method RP- HPLC	RP-HPLC Internal standard: Caffeine	Stress study HPLC method: Technology transfer to UPLC	Stability – indicating UPLC	Stability – indicating UPLC	RXN Impurities -Forced degradation by LC-MS/MS	Liquid chromatography LC-MS/MS	Stress Degradation Products of Rivaroxaban LC-MS/MS
10	11	12	13	14	2	16	17

 $t_{1/2}$ (6.87 and 6.64 hours) for test and marketed formulation of RXN. Also, the study could be extended for the estimation of RXN in urine [50].

HPLC

Bioanalytical method development and validation using HPLC assist in performing quality control of pharmaceuticals as well as in obtaining pharmacokinetic and toxicokinetic data of drugs in less time. Protein precipitation which provides a quick sample clean up and better recovery was utilised by Yadav et al. [51] to develop a cost-effective bioanalytical method for the determination of RXN. Human plasma was spiked with RXN and the precipitating agent (acetonitrile) and centrifuged for 10 minutes at 1,000 rpm. The supernatant was injected into the HPLC and the results were monitored. Another author used SPE which effectively extracted RXN from human plasma with high recovery efficiency. Selective extraction of RXN based on their affinity to Phenomenex Strata-X 33-µm polymeric reversed phase cartridge facilitating the removal of interfering substances and enrichment of RXN by washing (methanol: water (50: 50, v/v)). RXN was eluted by methanol: glacial acetic acid (99: 1, v/v), followed by evaporation with a stream of nitrogen and finally reconstituted with the mobile phase before analysis [52]. This method was sensitive and increased the RXN concentration in samples by 7.5 times. Gouveia et al. [53] developed a fast and simple HPLC-DAD and applied to quantify the four currently marketed DOAC (AXN, EXN, dabigatran, and RXN). SPE using Oasis PRIME HLB cartridges (1cc/30 mg) for conditioning, the solid residue reconstituted with 0.1 % formic acid, and methanol (50:50, v/v) showed efficient recovery of all the anticoagulants. Also, the developed cost-effective HPLC-DAD bio analytical method can be more widely used in clinical laboratories than expensive LC-MS/MS [51-54].

LC-MS/MS

LC-MS/MS, a selective and sensitive technique used for pharmacokinetic studies, metabolites identification in biosamples. Reddy et al. [54] reported a precise, reproducible LC/MS/MS method that can be used for the TDM of RXN. RXN was found to be for about 5 freeze-thaw cycles and reconstituted samples were stable up to 72 hours post to extraction which indicates the effectiveness of the method. Varga et al. [55] Derogis et al. [56] Rohde [57] Rodina et al. [58] developed and validated a rapid, sensitive, highthroughput LC-MS/MS method for the estimation of RXN in human plasma and applied the method successfully in clinical studies. The authors adopted the protein precipitation method using methanol for the extraction of RXN from the biological matrix. The protein precipitation method was effective in extracting the analyte from clinical samples. Also incorporating gradient elution techniques leads to enhanced peak resolution, faster analysis times, and better detectability [56–58]. Simultaneous determination of RXN and dabigatran by LC-MS/MS in human plasma employing protein precipitation with zinc sulphate as the precipitating agent results in effective precipitation through hydrophobic interactions of protein molecules. Also, this method extracted the analyte by precipitating the proteins efficiently even when applied to 65 blood samples obtained from humans [59]. Simultaneous estimation of DOAC (AXN, dabigatran, RXN) in human plasma by LC-MS/MS employing protein precipitation technique was reported by Lagoutte-Renosi et al. [60] To compare the pharmacokinetics of RXN and TAK-438 in rats when being administered alone or being co-administered, Wang *et al.* [61] developed a simple, rapid and reliable method for simultaneous determination of the drugs in rat plasma by LC-MS/MS. Lee et al. [62] reported the first analytical method employing LLE for determining RXN in rat plasma which can be applied in special clinical situations. LLE is the classical method used for isolation, especially from water and biological fluid samples. Ethyl acetate, dichloromethane, and their mixtures are among the preferred extraction solvents. LLE was performed by mixing aliquot of a rat plasma and RXN with the IS and vortexed for a minute with the addition of Ethyl acetate. After centrifugation, the supernatant was collected, dried, and reconstituted with acetonitrile. The resulting solution was centrifuged and the supernatant solution was injected into the LC-MS/MS system and the results were monitored [55–63].

UPLC-MSMS

In recent years, a UPLC coupled with mass spectroscopy (MS) has been developed to estimate the biological samples, with better accuracy, sensitivity, precision, and high throughput. Iqbal et al. [64] developed UPLC/MS/MS method for the estimation of RXN in pharmaceutical dosage forms in the biological matrix (human Plasma) employing protein precipitation. All the other methods were simultaneous estimation of RXN in combined dosage forms by UPLC/MS/ MS. Foerster et al. [65], Zhang et al. [66], Zhao et al. [67], Schmitz et al. [68], Sheng et al. [69], and Kuhn et al. [70] developed UPLC/MS/MS method for the estimation of RXN in combined pharmaceutical dosage forms in the biological matrix (human plasma/rat plasma) employing protein precipitation. Foerster et al. [65] developed a dried blood spot (DBS) assay that facilitates the sufficient extraction of all DOAC (AXN, dabigatran, EXN, and RXN). Consistent extraction recoveries and matrix effects of all samples were achieved in a dual extraction process of DBS (ultrasonic bath and shaking), which were treated with extraction solvent (methanol/water, 95/5 % v/v). The method was highly advantageous such that the patients can sample DBS at home within the dosing interval. Kuhn et al. [70] reported a rapid method with a run time of 2.5 minutes which was found to be one of the fastest methods in the estimation of DOAC. The method reported by Schmitz et al. [68] presented the highest precision for the DOAC compared to all the methods [64-70]. Table 4 overviews the LC parameters, validation, and application of the bioanalytical methods. All the UPLC/MS/MS methods were applied for the pharmacokinetics studies in patients, and the results obtained revealed the sensitiveness of the method. Most of the methods used protein precipitation as the major extraction technique followed by the SPE technique, the % recovery obtained with both methods was within the limits. Whereas, micromethods with a limited amount of extraction phase than the sample have

ytical	Hydrol	ysis	Oxidation	Photolysis	Thermal	Impurities/	Reference
ue	Acidic	Alkali				Degradation products	
	68.17 %	No degradation		95.27 % Possible by UV light	79.04%		Mandake et al. [28]
	2.21%-32.29% (0.1 M HCl was used and refluxed for 1,2,4,6 hours and 1 Day and 3 Day at RT, 40 0 C and 60 0 C)	NA	NA	Ч	NA	NA	Pinaz and Muralikrishna [31]
	NA	NA	NA	17.47 % - 32.25 % (UV	' light and sunlight)	51.29 %59.01 % (oven at 60°C and 80°C for 1 hour)	Kasad [40]
	ΥN	5.04 % to 47.35 % (0.1M NaOH at 1,2,4,6 hours and 1 Day and 3 Day at 40 0 C and 60 0 C)	NA	ΥN	ΥN	NA	Kasad and Muralikrishna [41]
	7.705 % (0.1 N HCl for 2 hours at 80°C)	8.851 %(0.1 N NaOH for 2 hours at 80°C)	6.55 % (3 % H2O2 for 2 hours at 80°C)	9.186 % (sunlight for 24 hours)	7.304 % (105 °C for 2 hours)	NA	Seshamamba <i>et al</i> . [32]
	NA	NA	NA	NA	NA	NA	Walter et al. [36]
	(1 M HCL for 4 hours at 80°C)	1 M NaOH for 4 hours at 80°C	5 %v/v H2O2 for 4 hours at 40°C	1.2 million lux h	80°C for 12 hours	Impurity 1–10	Girase et al. [37]
	93.5% (0.1N HCL 700C/ 2 hours)	85.6 % (0.1N NaOH RT for 1 hour)	99.5 % (3 % H2O2 at 700C/ 48 hours)	99.1 % (1.2 & 106 LUX hours)	99.4% (600C/ 7 days)	NA	Badroon and Sreeramulu [38]
	7% (at room temp.) 45% (at 800C)	89 % (at room temp.) 73.8% (at 800C)	39 % (at room temp.) 15% (at 800C)	Solid form- 30% Water form- 84%	Solid form- 21% Water form- 30%	NA	Souri et al. [33]
	5.9% (1 N HCI/ 5 ml/60°C/2 hours)	9.9% (1 N NaOH /5 ml/R.T/20 minute)	1.6% (30% H202/ 5 ml/ 60°C/ 12 hours)	0.4% (1.2 million lux hours)	0.3 % (60°C for 7 days)	Impurities A - H	Garad <i>et al.</i> [39]
	0.72 % (0.1 N HCL for 4 hours at 70°C)	18.4 % (0.1 N NaOH for 4 hours at 70°C)	NA	4.2 % (UV 254 nm for 24 hours)	0.67 % (Oven at 150°C for 4 hours)	NA	Sayeda <i>et al.</i> [34]
	83.2 % (0.1 N hydrochloric for 2 hours at 40°C)	83.0 % (0.1 N sodium hydroxide for 2 hours at 40 °C)	NA	35.1 % (UV 254 nm for 24 hours)	77.7 % (2 hours at 80 °C)	NA	Çelebier <i>et al.</i> [35]

Table 4. Degradation Results of Rivaroxaban by UV/HPLC/LC-MS/

Continued

Jebaliya *et al*. [44]

NA

3 % (80°C for 36 hours in hot air oven)

1.5 % (72 hours)

50 % (0.1 N NaOH 6 % (30% v/v H2O2 for 60 minutes at at 60°C for 2 hour) 80°C)

13% (0.5 N HCl at 80°C for 45 minutes)

Rao <i>et al.</i> [43]	Cholleti et al. [45]	Arous <i>et al.</i> [46]	Ramisetti <i>et al.</i> [47]
Impurity 1-*10	NA	Acid-Appearance of DP2 & DP3. Alkali- Appearance of DP1, DP2 & DP3.	DP 1 s (S)-2-(2-((4- (5-((5-chlorothiophene-2- carboxamido)methyl)-2- oxooxazolidin-3- yl)phenyl)amino) ethoxy) acetic acid DP 2 No Structural confirmation DP 3 (R)-5-chloro-N-(2-hydroxy-3- ((4-(3-oxomorpholino) phenyl)amino)
N.P (105°C for 10 days)	NA	0.32 % (105°C for 7 days)	N.P (60 °C for 7 days)
N.P (UV light 254 & 365 nm for 10 days)	NA	0.09* (7 days at 200 W h/ m2 of UV light and 1.2 million lux hours of visible light)	N.P (UV light for 7 days)
N.P (3.0 % H2O2 at 70 °C) for 48 hours, 1 hour, 2 hours, and 48 hours	NA	0.92 % (3% H2O2 solution over a period of 72 hours)	18.4 % (10% H2O2 for 24 hours)
11.2% (0.1-N NaOH at RT for 48 hours, 1 hour, 2 hours, and 48 hours)	NA	34.83 % (0.1 N NaOH for 72 hours)	6.1 % (1 N NaOH for 1 hour)
6.4 % (0.1-N HCl at 70 °C for 48 hours, 1 hour, 2 hours, and 48 hours)	NA	27.61 % (0.1 N HCI for 72 hours)	(1 N HCl for 1 hour)
UPLC		Detection of impurities by LC- MS/MS	

emerged as an advantageous method than the usual methods. Syringe-based SPE (Microextraction by packed sorbent), disposable-pipette extraction, liquid-phase microextraction, thin-film microextraction, single-drop microextraction, sorbent-based microextraction techniques, dispersive liquidliquid microextraction, electromembrane extraction and stirbar sorptive extraction are the numerous micromethods used in clinical investigations. In a nutshell, the incorporation of microsampling techniques and microextraction would be a tremendous spike in the area of bioanalytical research.

GREENER ASSESSMENT OF THE REPORTED LIQUID CHROMATOGRAPHIC METHODS

Sustainable Development Goals 2030 were implemented to fulfill human development goals without exhausting resources for future generations [71,72]. Among the 17 SDGs, SDG Goal 12 (ensure sustainable consumption and production patterns) espouses Green analytical chemistry which deals with the development of efficient and sustainable chemical processes, thus reducing the environmental impacts [73]. NEMI, eco-scale assessments, green analytical procedure index, analytical method volume intensity, and analytical greenness metric are the several tools to evaluate the greenness of analytical methods. NEMI owing to its easiest and rapid greenness assessment, represented by a pictogram provides an adequate inference about the environmental effects of the investigated method. The circular pictogram divided into four quadrants (Fig. 6), where A = persistent, bio-accumulative, and toxic (PBT), B = hazardous, C = corrosive, and D = waste (run time \times flow rate) employs the summary of the greenness of the method. The greenness profile of all the reported analytical methods for RXN was assessed and summarised in a tabular form, the quadrants were filled green for the greener methods while left blank for the non-greener ones. All the methods used acetonitrile (PBT and hazardous) and methanol (Not PBT but hazardous) as the mobile phase. Hence, the method with methanol as the mobile phase would show three of the quadrants green (PBT, corrosive, and waste), whereas acetonitrile as the mobile phase would show two quadrants green [74–77]. All the reported methods were found to be noncorrosive (pH 2–12) and generated a low amount of waste (less than 50 g). The greenness assessment using NEMI suggested that among the HPLC methods (12), one method was 75 % and the remaining 11 methods were 50 %. The stability indicating methods (18) was 75 % for 2 methods and 50 % for the



Figure 6. Greeness assessment quadrants by NEMI.

propyl)thiophene-2-carboxamide

Ref. no		Yadav and Dubey [51]	Çelebier <i>et</i> al. [52]	Gouveia <i>et</i> al. [53]	Reddy et al. [54]	Varga <i>et al.</i> [55]	Derogis <i>et</i> al. [56]	Rohde G [57]
Application	:	An greener approach for the quantification of RXN in human plasma	Can be adapted on liver perfusion and intestinal permeability studies in animal models	Applied successfully to analyse patient samples in TDM	Applied for therapeutic drug monitoring of RXN, pharmacokinetic or bioequivalence studies	Therapeutic drug monitoring of RXN in 29 patients	Applied to 49 patients treated with RXN	Applied for the accurate determination of RXN and pharmacokinetics in human plasma.
Greeness	profile	\bigcirc	\bigcirc	\bigcirc	\bigcirc	\bigcirc	\bigcirc	\bigcirc
	DOD/ TOD/	1.5/4.5	0.005/0.01	LLOQ: 12.98%- 5.79 %	LLOQ :2.01 ng/ mL	LLOQ : 24.0 ng /ml	LLDQ :4 pg/ml ULOQ: 500 ng/ml	2μg/l
s	% Recovery	99.38 - 100.44%		AXN -77.75 % dabigatran - 97.41 %, EXN -77.61 %, RX N - 70.70	Overall recover: >96%	96.5 %108.5%	Accuracy : 95.2 - 101.0 %Acceptable Stability : 99.2 -110.3 %	96.3% and 102.9%
Result	Linearity range (µg/ ml)/ R2/ %RSD	5-40/0.975 /0.57%-1.15 %	0.005-40.0/ 0.997/ 1.17 %	AXN: 0.017- .28/ 0.017 - 5.28/ 11.89 %; EXN - 0.035-5.28/ 13.58 %; RXN: 0.017- 5.28/ 13.39 % R2: 0.993	2.00- 500.93 ng/ ml/0.99/3.8% -0.9 %	24.00-960.00 ng/ml/0.9927	2–500 ng/ml R2: 0.9996	0.50 and 500 g/l/
	Rt (minute)	6.2	Prednisolone: 2.21 RXN : 3.37	Dabigatran – 1.6 minutes, EXN- 3.6 minutes, AXN – 4.5 min, RXN – 4.7 min	RIV – 0.82 minutes \rightarrow m/z144.80 RIV D4 – 0.82 min	1.4	1.8	RXN: 3.30 IS: 3.37
	Internal Standard	Chloramphenicol	prednisolone	Chloramphenicol	RXN D4	₹ Z	RXN D4	([2H5, 15N]) RXN
ditions	Detector/detection wavelength (nm)	PDA/252	UV/249	PDA/ IS-278 nm, dabigatran and AXN – 300 nm , EXN-289 nm, RXN - 249 nm	m/z - 436.20>144.80 1S m/z - 440.20>144.70	positive electrospray ionization and MS/ MS mode (sum of m/z 231.1; 289.2 and 318.2 from m/z 436.3)	RXN: 435.9–144.9 IS: 440.1–144.9	RXN: m/z 436.1 to 144.9 IS: 464.2→144.9)
tographic Cond	Sample Sample extraction Technique	Protein Precipitation with acetonitrile	Solid Phase Extraction	Solid Phase Extraction	Solid Phase Extraction	Protein Precipitation with methanol	Protein Precipitation method	Protein Precipitation
Chrom	Flow rate ml/ min	-	1:2	1.2	0.8 mJ/ minute	-	0.5	-
	Mobile phase	methanol: water: dimethyl sulfoxide (50:45:5, v/v/v)	Acetonitrile : Water (55:45 v/v) mixture	Formic acid (0.1 %, v/v): acetonitrile (55:45 v/v) Gradient elution	Acetonitrile: 5 mm Ammonium Formate buffer (pH 3.5 \pm 0.3) (85:15 v/v)	0.2 % Formic acid in water : Acetonitrile (65:35v/v)	Ultra-pure water containing 0.01% Formic acid : 60% Methanol	acetonirile: 0.01 mol/l, ammonium acetate (Gradient elution)
	Stationary phase	Phenomenex luna C8 (25 cm × 5 μm 4.6 mm)	Phenomenex Luna C18 LC $(250 \times 4.6$ mm,5 μ m)	Column: C18 (55 mm × 4 mm, 3 µm) Temperature : 30o C	Phenomenex Gemini C8 $(50 \times 4.6 \text{ mm} \times 5 \mu)$ $\times 5 \mu$	Zorbax SB-C18 (3.0 × 100 mm, 3.5 µm),	Kinetex C18 (100 × 3 mm, 2.6 µm) ESI	C18 (125* 4 mm, 5 µm)
Method/Matrix		RP-HPLC/ Human plasma	RP-HPLC/ Human plasma	Simultaneous method HPLC/ Human Plasma	HPLC-MS/MS/ Human Plasma	LC-MS/MS/ Human Plasma	Pharmacokinetic Study Patients Plasma Sample HPLC-MS/MS	LC-MS/MS matrix: Human plasma
S.no		-	0	Ś	4	Ś	9	-

Table 5. Bioanalytical methods for determination of rivaroxaban.

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Continued

et al. [59]	Lagoutte- Renosi et al. [60]	Rodina <i>et</i> al. [58]	Wang, <i>et al.</i> [61]	Shaikh et al. [63]	Lee <i>et al.</i> [62]
Applied to 65 blood samples obtained in humans.	The method can be used during clinical studies and in daily routine practice for the management of specific clinical situations at reasonable cost	Applied at Davydovsky City Hospital, for TDM	Applied to pharmacokinetic interactions study between RXN and TAK-438 for the first time.	Applicable for supporting the pharmacokinetic, bioavailability, and bioequivalence studies of RXN	first analytical method employing LLE for estimation RXN in Rat plasma and can be applied in special clinical situations
\bigcirc	\bigcirc	\bigcirc	\bigcirc	\bigcirc	\bigcirc
2.5 ng/ml	RXN: 5μg/l 25 μg/l for apixaban and dabigatran	1.0 ng/ml	0.5 ng/ml	5.96 ng/ml	0.5 ng/ml
78.9 - 105.6	¥ Z	NA	RXN: 92.26 -109.42 TAK- 438:90.68 - 99.55	88.9% and 89.2%	89.2 - 105.7%
2.5-500 ng/ ml/0.995/1.4 - 9.6 %	RXN: 5 to 1000 μg/l, AXN and dabigatran: 25 to 1,000 μg/l	1 – 1000 ng/ ml/1.72–7.06/	RXN: 0.5-275.0 ng/ ml/0.994 TAK-438: 0.5-100.0 ng/ ml/0.997 %RSD: 2.66 -11.87	5.96–801 ng/ml/ 0.99/ 1.08%–-3.75%	0.5 - 500 ng/ m1/0.99/
dabigatran and 13C6- dabigatran – 1.45 RXN and 13C6-RXN – 2.15 min	Dabigatran - 2.75 AXN - 3.20 RXN - 3.25	RXN: 2.42 Promethazine: 2.41	RXN: 2.452, TAK-438; 1.678 and IS: 2.673 min	RXN: 1.8 RXN D4: 1.18	RXN: 2.9 minutes IS: 2.6 minutes
13C6-dabigatran and 13C6-RXN	13C6- dabigatran 13C, 2H7 AXN 13C6-RXN	Promethazine	carbamazepine	RXN D4	Apixaban
RXN: m/z 436.1→144.9,13 C6 RXN: m/z 442.1→144.9, dabigatran-m/z 472.2→289.0 and m/z 478.2→294.9, 13C6-dabigatran m/z 436.1 to 144.9	m/z 436.1–144.9	RXN (436.10 m/z) and promethazine (285.10 m/z)	RXN: m/z 436.1 → 144.9, TAK-438: 346.1 → 173.0, IS: 237.1 → 194.2	RXN: 436/145 (m/z) RXN D4: 440/145 (m/z)	436.1m/z and 460.1 m/z
protein precipitation by ZnSO4 solution	Protein precipitation with methanol	Protein Precipitation with Methanol	Protein Precipitation with acetonitrile	Solid phase extraction	Liquid liquid Extraction using ethyl acetate
0.3	0.35	0.1	0.5	1.5	0.25
0.07 g ammonium acetate with 0.1% formic acid: acetonitrile with 0.1% formic acid (90:20, V/V)	0.1% Formic acid: Acetonitrile with 0.1% Formic acid (Gradient elution)	0.1% formic acid: Acetonitrile with 0.1% Formic acid (Gradient elution)	 0.1% formic acid in acid in accontirile (A) and 0.1% aqueous formic acid (B) (A: B = 70: 30% v / v) 	Acetonitrile: 5mM ammonium acetate (80:20 v/v)	5% (v/v) aqueous acetonitrile solution including 0.1% (v/v) formic acid
CI8 (100* 2.1 mm, 1.7 µm)	С18 (100* 2 лпп, 3 µm)	C18 (50* 4.6 mm, 3.5 µm)	Agilent Eclipse plus C18 column (100 mm × 2.1 mm, 3.5 µm)	Gemini C18 (150 mm × 4.6 mm, 5 mm)	Luna C18 column (2.0 ×150 mm, 5 mm)
Simultaneous determination of RXN and dabigatran lev by LC-MS/MS in Human plasma	LC-MS/MS simultaneous determination of direct oral anticoagulants AXN, dabigatran, RXN	LC-MS/MS in Human Blood Serum	RXN and TAK- 438 in rat plasma LC-MS/MS	LC-MS/MS - RXN in human plasma and its application to a pharmacokinetic study	LC-MS/MS/ Rat plasma
~	0	10.	=	12	5

Schmitz et al. [68]	Sheng <i>et al.</i> [69]	Kuhn et al. [70]	Iqbal et al. [64]	Foerster et al. [65]	Zhang <i>et al.</i> [66]	Zhao et al. [67]
UPLC-MS/MS method was compared with coagulation methods by the use of patient samples for dabigatran and RXN	applied to study the pharmacokinetic interaction investigation in rats	Applied to determine the concentrations of the two drugs independent of inteference flactors such as hemolysis or lipaemic plasma	UPLC–MS/ MS assay was successfully applied to a single dose, oral pharmacokinetic study of RXN (1 mg/kg) suspension in healthy male rats.	Correlation between Dried blood spot and plasma concentrations was assessed in 33 patients	Applied successfully to the pharmacokinetic investigations of RXN, AXN and edoxaban in rats	Applied to the analysis of clinical samples in a clinical lactation study involving AXN and RXN
\bigcirc	\bigcirc	\bigcirc	\bigcirc	\bigcirc		\bigcirc
LOD :≤ 0.025 ng/ mL	RXN: 1.0 ng/ml Enalapril: 0.5 ng/ml	Dabigatran - 0.46 μg/l RXN - 0.54 μg/l	0.57 ng/ml	2.5 ng/ml	5 ng/ml	5 ng/ml
NA	81.7% 89.4%	Dabigatran - 104.5% RXN - 87.0%	95.3-102.6	65-81 %	92.8–108.0 %	106.13%- 109.05% in plasma and from 93.40% to 107.91% in breast milk
23–750 ng /ml R2 :≥ 0.99	RXN :1.0- 200 ng/ml/ 0.9944/2.2 -8.4 Enalapril: 0.5-100 ng/ ml/0.997/2.6- 7.9	0.8 and 800 µg/l/0.99	0.57–625 ng/ ml/0.997/	2.5 (AXN, RXN), 4.4 (dabigatran), and 9.3 ng/ml (edoxaban) to 750 ng/ml	RXN: 1.0–200 ng/ml/ 0.9948 AXN: 1.0–100 ng/ml/0.9971 Edoxaban : 1.0–500 ng/ ml/0.9956	5–500 ng/ml in plasma and 5–250 ng/ml in breast milk
Ч. Ч.	RXN: 2.13, Enalapril: 1.90 Diazepam: 2.56 min	[13C6]- Dabigatran - 1.47 minutes [13C6]-RXN - 1.97 min	RXN:0.81 minutes IS: 0.85 minutes	NA	RXN: 1.71 minutes, AXN: 1.69 minutes, edoxaban: 1.58 IS: 1.89 IS: 1.89	RXN: 1.84 AXN: 1.81 Edoxaban: 147
[13C6]Dabigatran [13C6] RXN, [13C, 2H7] Apixaban	Diazepam	[13C6]-Aabigatran [13C6]-dabigatran	Risperidone	isotope-labeled internal standard	diazepam	AXN-13C, 2 H8, EXN-2 H6 and RXN-13C6
MS/MS	RXN: 436.1→145.1 m/z Enalapril: 377.3→234.2 m/z Diazepam: 285.2→193.1 m/z	Dabigatran - 472.2 > 289.2 RXN - 436.1 > 145.0	RXN: m/z 436.00 - 144.87 IS: m/z 411.18 - 191.07	m/z 436.2–144.9	RXN - m/z 436.1 - 145.1 AXN: m/z 460.0 - 443. EXN: m/z 548.2 - 366.1 IS: m/z 285.2 - 193.1	m/z 436.1-144.9
Precipitation	Protein Precipitation with acetonitrile	Protein Precipitation	Protein Precipitation by Acetonitrile	Solid Phase Extraction	Protein precipitation	Protein Precipitation by methanol
800	0.3	0.35	0.0	0.5	0.4	0.5
2.5 mM ammonium Formate (pH 3.0): A cetonitrile	Formic Acid (0.1 %, v/v) and Acetonitrile (55:45v/v)	5%/95% water/ methanol containing also 0.1% formic acid and 2 mmol/I ammonium acetate.	Acetonitrile: 10mM ammonium acetate (80:20 v/v)	(95%5mM ammonium bicarbonate in water with 5% Acetonitrile gradient)	Acetonitrile and 0.1% formic acid in water	0.1% formic acid in deionized water (A) and acetonitrile (B) Gradient Elution
Acquity UPLC BEH C8 column (100 × 2.1 mm, 1.7 m; Waters)	Acquity UPLC BEH C18 (50 × 2.1 mm, 1.7 µm)	Acquity UPLC BEH Phenyl, (50 × 2.1 mm, 1.7 µm)	С18(100*2.1 mm, 1.7 µm)	Phenyl(2.1 *50 mm, 1.7 μm)	Acquity UPLC BEH C18 chromato- graphy column (2.1 mm 9 50 mm, 1.7 lm)	C18(50*2.1 mm, 1.8 µm)
UPLC-MS/MS/ plasma	UPLC-MS/MS/ plasma	UPLC- MRM Mass Spectrometry Method/Plasma Simultaneous method Dabigatran and RXN	UHPLC-MS/ MS/ plasma	UPLC-MS/ MS Dried blood spot technique to monitor direct oral anticoagulants	UPLC–MS/ MS/ Rat Plasma Determination of RXN, AXN and edoxaban in rat plasma	UHPLC-MS/MS/ Human Plasma - Simultaneous estimation of apixaban, EXN and RXN in human plasma and breast milk
14	15	16	17	18	19	20

remaining. The greenness profile of the bioanalytical methods was 75 % for 3 methods and 50 % for 17 methods. HPTLC methods reported a 100 % greenness profile among all the other analytical techniques. Subsequently, the replacement of toxic solvents like acetonitrile with ethanol or water could be more ecologically to make the existing analytical methods more sustainable meeting the requirement of SDG12. The reported methods pave a roadmap for future researchers that there is a need to use eco-friendly solvents for the chromatographic separation of RXN without compromising the sensitivity of the analytical methods.

CONCLUSION

The review elaborates on the different analytical methods of RXN which will be helpful to the researchers. Validation of all the methods was performed as per ICH guidelines and found to be linear, accurate, precise, specific, and robust. Stability-indicating methods revealed the degradation profile of the drug. Protein precipitation was found to be the most preferred extraction technique adopted for sample pretreatment. Miniaturized novel sample extraction methods can be adopted in the future for the extraction of complex matrices. A substantial emphasis on examining the greenness profile by NEMI, which indicated 50% green for 90% of the methods. Also, concepts of "blueness" and "whiteness" assessment using the recently introduced blue applicability grade index and Red-Green-Blue 12 (RGB 12) algorithms can be incorporated. To put it briefly there is a pressing need for developing eco-friendly methods and applying them in the analytical techniques for the separation of RXN from biological matrices and finished drug products. Future trends incorporating miniaturized techniques would pave a roadmap for the estimation of RXN.

LIST OF ABBREVIATIONS

AQbD: Analytical Quality by Design; AXN: Apixaban; CPL: Clopidogrel; DAD: Diode Array Detector; DMSO: Dimethyl Sulphoxide; DOAC: Direct oral anticoagulants; *DP:Degradation Product;* EXN: Edoxaban; HPLC: High Performance Liquid Chromatography; HPTLC: High Performance Thin Layer Chromatography; LOD: Limit of Detection; LOQ: Limit of Quantification;

LC-MS/MS: Liquid Chromatography Tandem Mass Spectrometry; PBT: *Persistent, Bioaccumulative, Toxic;* R²: Correlation Coefficient; Rt: Retention Time; RXN: Rivaroxaban; TLC: Thin Layer Chromatography; UPLC: Ultra Performance Liquid Chromatography; UPLC-MS/MS: Ultra Performance Liquid Chromatography Tandem Mass Spectrometry; UV: Ultra-Violet.

AUTHOR CONTRIBUTIONS

All authors made substantial contributions to conception and design, acquisition of data, or analysis and interpretation of data; took part in drafting the article or revising it critically for important intellectual content; agreed to submit to the current journal; gave final approval of the version to be published; and agree to be accountable for all aspects of the work. All the authors are eligible to be an author as per the International Committee of Medical Journal Editors (ICMJE) requirements/guidelines.

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This study does not involve experiments on animals or human subjects.

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All data generated and analyzed are included in this research article.

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USE OF ARTIFICIAL INTELLIGENCE (AI)-ASSISTED TECHNOLOGY

The authors declares that they have not used artificial intelligence (AI)-tools for writing and editing of the manuscript, and no images were manipulated using AI.

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