



# Development of levofloxacin glycosylated mesoporous silica nanoparticles for urinary tract infections

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## ARTICLE HISTORY

Received on: 16/06/2024  
Accepted on: 14/10/2024  
Available Online: 05/11/2024

### Key words:

Antimicrobial resistance,  
Levofloxacin, D-Glucuronic  
acid, Glycosylation, Lectin,  
Nanoparticle.

## ABSTRACT

Nanoparticle drug delivery for infectious disease has paved the way for effective treatment; mesoporous silica nanoparticles (MSNs) with versatile design options, tunable pore size, high surface area, and adequate pore volume have garnered widespread interest. The current study deals with the synthesis of MSNs loaded with levofloxacin (LVF), followed by glycosylation to enhance the uptake by bacterial cells. The MSNs were prepared by modifying Stober's process. The amino-modified MSNs were glycosylated using D-Glucuronic acid by EDAC-NHS coupling chemistry. The LVF-loaded glycosylated MSNs (GLY-MSN) were characterized for various parameters and *in vitro* antimicrobial efficacy study. The surface functionalized MSNs had a particle size of  $673.6 \pm 60.81$  nm and were found to be spherical from the SEM images. The drug loading capacity for plain MSNs and GLY-MSN was found to be  $10.41\% \pm 0.81\%$  and  $11.43\% \pm 0.93\%$ , respectively. The LVF release from GLY-MSN was found to be  $21.02\% \pm 3.38\%$  whereas that from plain MSN was  $7.50\% \pm 1.31\%$  at the end of 48 hours. The minimum inhibitory concentration of LVF-GLY-MSN on *Escherichia coli* was found to be lesser than that for LVF and LVF-MSN. Hence, the synthesized GLY-MSN may be an effective drug delivery system for the treatment of drug-resistant bacterial infections.

## INTRODUCTION

Healthcare-associated infections, caused by superbugs such as *Staphylococcus aureus*, *Pseudomonas aeruginosa*, and *Klebsiella pneumoniae*, have become primary mortality causes in significant populations. Classified as ESKAPE bugs under the World Health Organization's Priority I, these microorganisms pose a significant challenge due to their heightened antimicrobial resistance [1–3]. The rise of multidrug resistance has emphasized the inadequacy of current antibiotics, with high doses leading to potential toxic effects. This situation has intensified the research for alternate treatment strategies.

Nanoparticulate drug delivery technology emerges as a promising alternative, particularly against multidrug-resistant infections. Their distinct chemical and physical properties have demonstrated beneficial outcomes. Various nanoparticles, including liposomes, niosomes, nanotubes, silver nanoparticles, solid lipid nanoparticles, and polymeric nanoparticles, have been explored for targeted drug delivery and controlled drug release at infection sites [4,5]. Such nanoparticles offer the advantage of delivering multiple therapeutics simultaneously, potentially mitigating resistance development. Furthermore, they can enhance the pharmacokinetic profiles of drugs, reduce required doses, and thus decrease potential toxic effects and dosing frequency.

Of the nanoparticles studied, mesoporous silica nanoparticles (MSNs) stand out due to their design versatility, tunable pore size, extensive surface area, optimal pore volume, and ease of functionalization. This makes them particularly valuable in combating bacterial infections [6–8]. The most

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widely explored type of mesoporous material is Mobil Composition of Matter No. 41 (MCM-41), characterized by its 2.5–6 nm pore diameter and hexagonal arrangement [6]. In addition, MSNs are recognized for their biocompatibility and substantial drug-loading capacity.

Understanding that bacteria interact with host cell glycoconjugates using various cell wall proteins is crucial. Carbohydrate-binding proteins, such as lectins, primarily facilitate bacterium-host adhesion [9,10]. This interaction has paved the way for glycomimetics, molecules that mimic carbohydrate activity [11]. Various multivalent neo-glycoconjugates, like glycoclusters using multivalent scaffolds such as di-, tri-, tetravalent sugars, cyclodextrin core, porphyrins, calixarenes, and cyclic peptides have been explored. Mannoside clusters used as ligands for targeting type-1 fimbrial lectin found in *Escherichia coli* (FimH), extended tetramannosylated clusters for increasing multivalency, FimH binding, glyconanotubes, glycodendrimers involving PPI dendrimers, carbosilane, glycopeptide, and hyperbranched boltorn dendrimer have been developed for effectively addressing the problem of antimicrobial resistance [12,13]. Incorporating sugar moieties in carrier molecules augments interaction points between bacterial sugar-binding proteins (like lectins) and carbohydrates, enhancing binding and efficient antimicrobial agent delivery [14,15].

Based on this proof-of-concept, in the present study, we have designed glycosylated MSNs (GLY MSNs) loaded with levofloxacin (LVF) for its plausible use in combating bacterial resistance observed in urinary tract infections (UTIs). LVF, a third-generation fluoroquinolone, has a 6–8 hours half-life and acts by inhibiting the bacterial DNA gyrase enzyme, crucial for bacterial DNA replication (topoisomerase II inhibition) [16]. Despite its efficacy in treating complicated UTIs [17], resistance development remains a concern [18] due to various factors like mutations at the topoisomerase gene and upregulated drug efflux [19,20]. While LVF delivery via MSNs and chitosan nanoparticles has been explored [21], our study is a pioneering effort to harness glycosylation for targeted LVF delivery using MSNs in UTI treatments. This approach seeks to reduce both dosages and the potential for bacterial resistance.

## MATERIALS AND METHODS

### Materials

LVF was procured from Yarrow Chemicals, Mumbai, India. From Loba Chemie, India, we procured N-cetyl-N, N, N, trimethylammoniumbromide (98%, CTAB). Tetraethylorthosilicate (TEOS), 3-aminopropyltriethoxysilane (APTES), and D-glucuronic acid were all obtained from TCI Chemicals, India. N(3-dimethylaminopropyl)N'-ethylcarbodiimide hydrochloride (EDC.HCl) and N-hydroxysuccinimide were obtained from Spectrochem Pvt. Ltd., Mumbai. All other reagents utilized in the study were of analytical grade.

### Methodology

#### Preparation of MSNs

MSNs were synthesized using a method previously described by Narayan *et al.* 2021 [22,23]. Briefly, CTAB (1.37

mmol) was dissolved in distilled water containing a 2.5 M sodium hydroxide solution and stirred at 80°C for 30 minutes. TEOS (11.28 mmol) was added to the above solution under vigorous stirring at 950 rpm. After allowing the mixture to react overnight, ethanol was added. The resultant product was filtered and dried. The dried MSNs were subjected to calcination at 500°C for 6 hours.

#### Preparation of amine functionalized MSNs (MSN-NH<sub>2</sub>)

To prepare amine-functionalized MSNs, a post-grafting method was utilized. MSNs were dispersed in dry toluene, and APTES (2.6 mmol) was added dropwise with continuous stirring. This mixture was subjected to reflux at 120°C for 10 hours. The product was washed with toluene and residual toluene was removed with the help of rotavapor (Buchi, Labortechnik AG, Flawil, Switzerland). The resultant product was dried under vacuum. Figure 1 illustrates the synthesis of MSNs followed by their amino functionalization.

#### Glycosylation of amine-functionalized MSNs

The amine-functionalized MSNs were glycosylated by coupling them with a sugar acid moiety, D-glucuronic acid, in aqueous media using the EDAC-NHS coupling chemistry [24]. Initially, an equimolar amount of EDAC.HCl, NHS, and D-glucuronic acid (0.16 mmol each) were dissolved in water. The pH was adjusted to 7.5–8, and the solution was heated to 50°C while stirring for 2 hours. Subsequently, 100 mg of MSN-NH<sub>2</sub> was added, and the reaction was allowed to proceed for 24 hours. The resulting product, designated as GLY-MSN, was filtered, washed with ethanol and water, and dried under a vacuum.

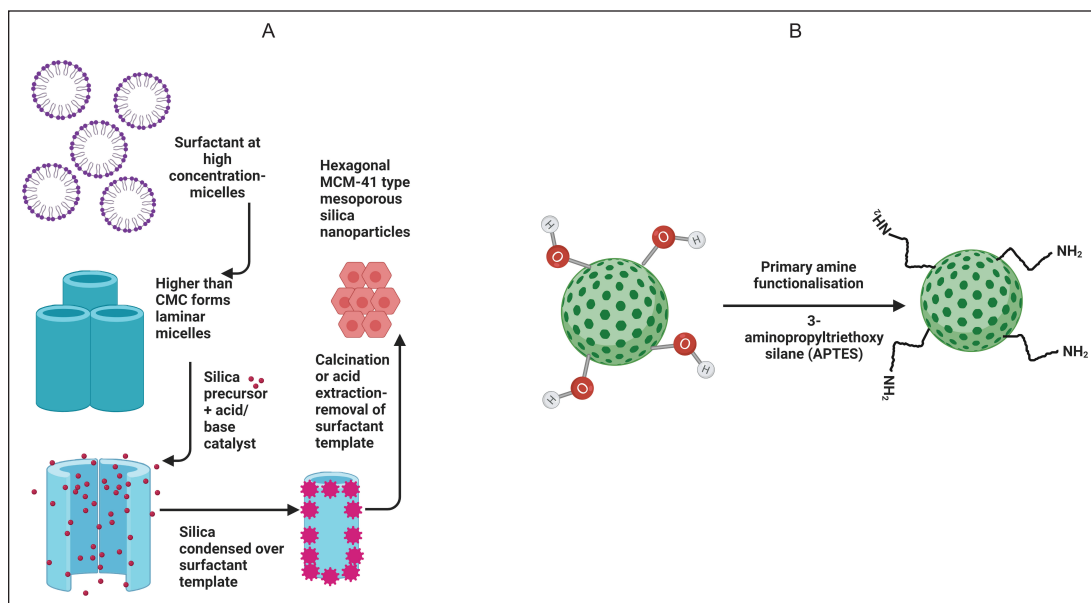
#### Drug loading into MSNs

MSN-NH<sub>2</sub> and GLY-MSN particles were loaded with LVF using the wet-impregnation method, as described in prior literature [24]. In brief, MSNs were dispersed in an aqueous solution of LVF and stirred at room temperature for 18 hours. The drug-loaded MSNs were then separated by centrifugation at 18,000 rpm for 15 minutes. The supernatant was analyzed using UV-visible spectrophotometry (Shimadzu, Kyoto, Japan) at 288 nm, allowing the analysis of untrapped LVF. Different batches were tested with varying drug-to-MSN ratios: 1:2, 1:5, and 1:8. The batch demonstrating optimal LVF loading was chosen for GLY-MSN loading. The loading capacity was determined using the following formula:

$$\text{Loading capacity} = \frac{\text{initial amount of drug added-recovered amount in supernatant}}{\text{amount of MSN added}} \times 100$$

#### Characterization of the prepared formulation

The nanoparticles were characterized using various techniques. Field emission scanning electron microscopy (FESEM) was carried out using a GEMINI SEM300. Transmission mode FTIR spectra were recorded using an FTIR–8,300 spectrometer (Shimadzu Corporation, Japan). Differential Scanning Calorimetry (DSC) analyzed the LVF's



**Figure 1.** Illustration of A) Preparation of plain mesoporous silica nanoparticle by liquid template crystal mechanism; B) Amine functionalization of MCM-41 by post grafting.

physical state within the MSNs. Using nitrogen adsorption, the Barrett–Joyner–Halenda (BJH) method assessed the MSNs' pore volume and size. The Brunauer–Emmett–Teller (BET) method determined the specific surface area. Dynamic light scattering measured particle size distribution and zeta potential with a Malvern Zetasizer (Malvern Nano Series NanoZS, UK).

### *In vitro* drug release profile

*In vitro* LVF release was gauged using the dialysis bag method [12,13]. Drug-loaded MSNs were dispersed in 2 ml of phosphate buffer (pH 7.4) and placed into a dialysis bag (MWCO 12 kDa). After sealing both ends, the bag was submerged in pH 7.4 phosphate buffer and agitated in a shaking incubator at 100 rpm and 37°C. At predefined intervals, samples were collected and replenished with fresh media. The LVF content was determined spectrophotometrically at 288 nm.

### *In vitro* antimicrobial efficacy study

#### *Inoculum preparation for the broth dilution test*

The antimicrobial efficacy was evaluated using the reference strain *E. coli* (ATCC 25922). For pure culture isolation, the bacteria were cultured on 5% sheep blood agar with incubation at 37°C for 24 hours. Subsequently, 2–3 isolated colonies were introduced into Mueller–Hinton broth and incubated for 4 hours to achieve the logarithmic phase. The broth's turbidity was adjusted to match the 0.5 McFarland standard.

#### *Minimum inhibitory concentration (MIC)*

The MIC was determined using the broth microdilution method and sterile microtiter plates. Fresh stock solutions of antimicrobial agents, including LVF, MSN LVF, and LVF-GLY-MSN, were prepared with a final concentration of 5,120 µg/ml. These solutions underwent a two-fold dilution. 50 µl of the bacterial inoculum (adjusted to 0.5 McFarland standard) was added to each

well of the microtiter plates. The plates were incubated at 37°C for 16–20 hours. A growth control well was included for each antimicrobial agent, containing the broth and bacterial inoculum without the antibiotic. The MIC endpoint was identified as the lowest concentration, preventing visible bacterial growth.

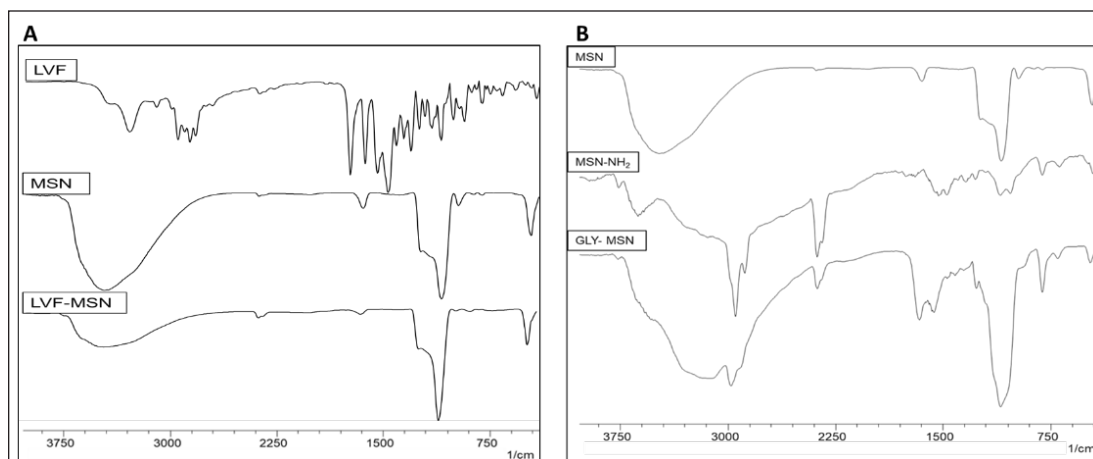
#### *Minimum bactericidal concentration (MBC)*

Post-MIC determination, broth from the microtiter plate wells with no visible microbial growth was subcultured onto sterile Mueller–Hinton agar plates. These plates were incubated at 37°C for 24 hours. The MBC was defined as the minimum concentration required to kill 99.9% of the bacteria after the 24-hour incubation [25–27].

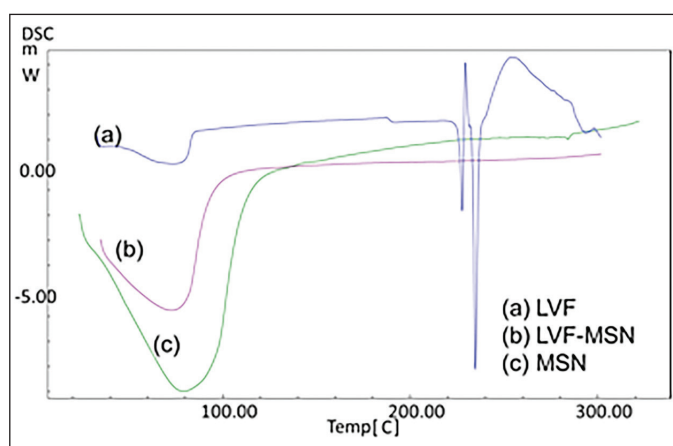
## RESULTS AND DISCUSSION

### Characterization of MSNs

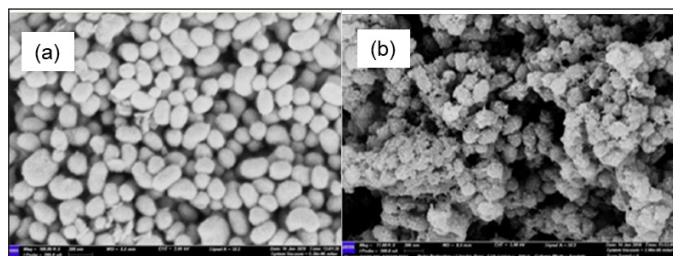
FTIR analysis of MSNs displayed distinct peaks of Si-O-Si stretching at 1,085.92 cm<sup>-1</sup>, Si-OH stretching and bending at 3,450.65 cm<sup>-1</sup> and 1,629.85 cm<sup>-1</sup>, respectively. Evidence of amine functionalization on MSNs was evident from the N-H stretching at 3,113.11 cm<sup>-1</sup>. The aminopropyl chain is attached to the MSNs by aliphatic C-H stretching between 2,922.16 and 2,858.51 cm<sup>-1</sup>, N-H bending at 1,678.07 cm<sup>-1</sup>, and C-N stretching at 1,089.78 cm<sup>-1</sup>. GLY-MSN showed the presence of two distinct peaks at 1,651.07 cm<sup>-1</sup> and 1,548.84 cm<sup>-1</sup> corresponding respectively to C=O str (amide-I) and N-H bending vibration (amide-II) of amide. The presence of O-H and secondary amine was ascertained from the bands between 3,506.59 and 3,109.25 cm<sup>-1</sup>, and C-H of alkane at 2,953.02 cm<sup>-1</sup>. The FTIR spectra are presented in Fig. 2. The FTIR spectrum of LVF showed characteristic peaks at 3,269.34 cm<sup>-1</sup> (OH), 2,931.80 cm<sup>-1</sup> (C-H), 1,722.43 cm<sup>-1</sup> (C=O), 1,300.02 cm<sup>-1</sup> (C-N), and 1,083.99 cm<sup>-1</sup> (C-F) [28], which are presented in Fig. 2. Notably, MSNs loaded with LVF did not showcase characteristic



**Figure 2.** FTIR spectra for A) Levofloxacin (LVF), synthesized Mesoporous silica nanoparticles (MSNs) and LVF loaded MSNs; B) Mesoporous silica nanoparticles, Amine-functionalized MSNs (MSN-NH<sub>2</sub>), and Glycosylated MSNs.



**Figure 3.** DSC thermograms for (a) Pure levofloxacin, (b) levofloxacin-loaded mesoporous silica nanoparticles, and (c) mesoporous silica nanoparticles.



**Figure 4.** FESEM images of (a) plain mesoporous silica nanoparticles and (b) glycosylated mesoporous silica nanoparticles.

peaks of LVF, suggesting the complete adsorption of LVF within the MSN pores. These observations align with previous studies [29]. The DSC thermograms for LVF and LVF-MSN are shown in Fig. 3. DSC thermograms of LVF displayed a pronounced endotherm at 234.70°C, signifying its crystalline character. Upon its incorporation into MSNs, a transformation to the amorphous form was observed, as indicated by the lack of peaks near the melting point of LVF. FESEM imagery, illustrated in Fig. 4,

**Table 1.** Particle size and zeta potential of MSNs.

Batch	Average size (nm) (Mean ± SD)	Zeta potential (mV) (Mean ± SD)	PDI (Mean ± SD)
MSNs	383.8 ± 23.45	-30.1 ± 5.64	0.307 ± 0.10
MSN-NH <sub>2</sub>	577.5 ± 32.34	34.6 ± 5.12	0.475 ± 0.52
GLY-MSN	673.6 ± 60.81	-15.3 ± 4.05	0.342 ± 0.12

portrayed plain MSNs as spherical, while GLY-MSN depicted a rougher surface, likely due to glucuronic acid attachment. The particle size and zeta potential of the MSNs, MSN-NH<sub>2</sub>, and GLY-MSN were ascertained via a Malvern zeta sizer (Malvern Instruments, Malvern, UK). GLY-MSN demonstrated an enlarged size relative to plain MSNs (Table 1). The positive zeta potential in amine-functionalized MSN is likely attributed to protonated amine groups. Metrics from nitrogen adsorption highlighted an average pore size of 3.81 nm, a pore volume of 0.678 cm<sup>3</sup>/g, and a BET surface area of 993.8811 m<sup>2</sup>/g [30].

#### Drug loading and entrapment efficiency

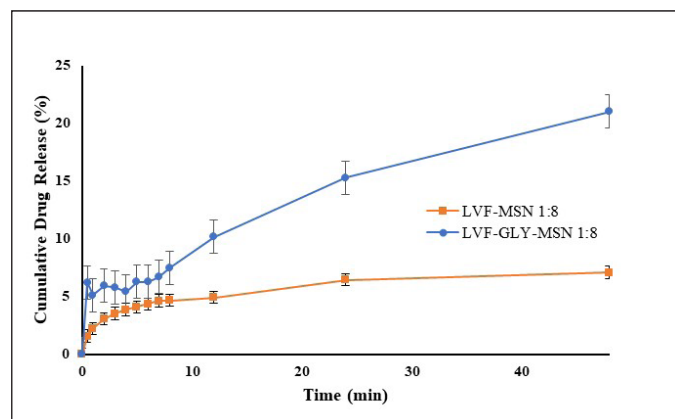
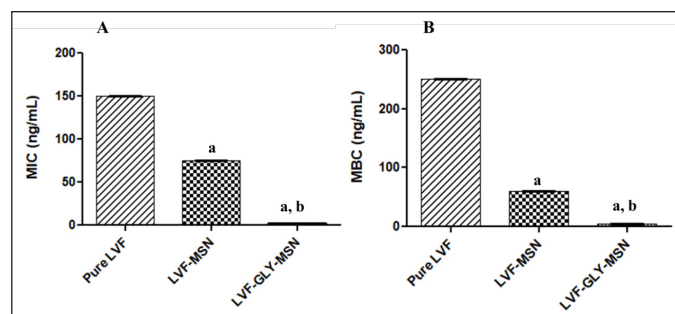
The amount of LVF entrapped in the MSNs was quantified using an indirect method. Exploring various LVF: MSN ratios revealed an increased carrier proportion relative to LVF, amplifying entrapment efficiency. Interestingly, GLY-MSN and plain MSNs presented comparable loadings. This may result from the sugar acid entities near MSN pores hindering drug molecule penetration. Associated data are tabulated in Table 2.

#### *In vitro* drug release study

The release of LVF from MSNs in phosphate buffer saline (pH 7.4) yielded a sustained release from GLY-MSN of 21.02% ± 3.38 % over 48 hours. In contrast, the plain LVF-MSN showed a release of just 7.50% ± 1.31% in the same duration. This enhanced release from GLY-MSN may be due to the presence of LVF on the surface and at the pore openings. Release dynamics are graphically represented in Fig. 5.

**Table 2.** Data for drug loading efficiency.

Carrier	LVF: MSN weight ratio	Loading (%)
MSNs	1:2	18.07 ± 2.64
MSNs	1:5	12.02 ± 2.57
MSNs	1:8	10.41 ± 0.81
GLY-MSN	1:8	11.43 ± 0.93

**Figure 5.** *In vitro* release profile of levofloxacin loaded MSN (1:8 of LVF and MSN), and levofloxacin loaded glycosylated MSN (1:8 of LVF and glycosylated MSN) in phosphate buffer saline pH 7.4.**Figure 6.** (A) Minimum inhibitory concentration (MIC) and (B) Minimum bactericidal concentration (MBC) of pure LVF, LVF-loaded MSNs, and LVF-loaded glycosylated MSNs. All values are expressed as mean ± SD, <sup>a</sup>*p* < 0.05 in comparison to pure LVF, <sup>b</sup>*p* < 0.05 in comparison to LVF-MSN.

### *In vitro* antibacterial efficacy study

The LVF-loaded MSNs showed a noticeable decline in MIC and MBC levels compared to pure LVF, indicating enhanced antimicrobial potency when incorporated into MSNs. Remarkably, the glycosylated MSNs (GLY-MSN) containing LVF exhibited even lower MIC and MBC levels, significantly outperforming both the pure LVF and the LVF loaded into MSNs (*p* < 0.05). This enhanced efficacy is likely due to the enhanced uptake of the GLY MSNs. The presence of the glucuronic acid moiety enhances interactions with lectin receptors, promoting improved adhesion between the bacterial cell and MSNs. The attached sugar units increase their multivalent nature, thus facilitating more effective

interactions with bacterial sugar-binding proteins [31,32]. The precise MIC and MBC data can be referenced in Fig. 6.

### CONCLUSION

MSNs have emerged as a promising platform for delivering antimicrobial agents, offering a potential solution to bacterial resistance. Their appeal lies in their porous structure, flexibility for surface modifications, and inherent biocompatibility, making them especially conducive to antimicrobial drug delivery.

In our study, the GLY MSNs loaded with LVF showcased markedly lower MIC and MBC values than the pure LVF and the LVF encapsulated in MSNs. This highlights the superior efficacy of the glycosylated formulation against *E. coli*. This enhanced performance can be attributed to amplified sugar-lectin interactions, which likely boost the bacterial cellular uptake of the drug. Our findings suggest that glycosylated LVF-loaded MSNs could present a viable strategy for addressing drug resistance in UTIs. However, to further validate these promising results, it is essential to conduct studies on multidrug-resistant strains. In addition, adhesion studies are necessary to understand the underlying mechanisms by which glycosylation counteracts antimicrobial resistance.

### ACKNOWLEDGMENTS

The authors would like to express their sincere gratitude to the Manipal Academy of Higher Education, Manipal, for providing facilities for the literature search. The authors are also thankful to the Indian Council of Medical Research (ICMR), New Delhi, Ref. No. 35/13/2020-Nano/BMS and Ref. No. AMR/Fellowship/25/2022-ECD-11 for the fund support.

### AUTHOR CONTRIBUTIONS

Songhita Mukhopadhyay—Data Acquisition, Data Analysis/Interpretation, Drafting Manuscript, Critical Revision of Manuscript

Reema Narayan—Data Analysis/Interpretation, Critical Revision of Manuscript, Statistical Analysis

Shivaprasad Gadag—Data Analysis/Interpretation, Critical Revision of Manuscript

Padmaja A Shenoy—Data Analysis/Interpretation, Critical Revision of Manuscript

Sanjay Garg- Data Analysis/Interpretation, Critical Revision of Manuscript

Ashwini T—Data Analysis/Interpretation, Funding

Usha Y Nayak—Concept and Design, Data Analysis/Interpretation, Critical Revision of Manuscript, Funding, Statistical Analysis, Admin, Technical or Material Support, Supervision, Final Approval

### CONFLICTS OF INTEREST

All authors declare that they have no conflict of interest.

### ETHICAL APPROVALS

This study does not involve experiments on animals or human subjects.

## DATA AVAILABILITY

All data generated and analyzed are included in this research article.

## PUBLISHER'S NOTE

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## USE OF ARTIFICIAL INTELLIGENCE (AI)-ASSISTED TECHNOLOGY

The authors declares that they have not used artificial intelligence (AI)-tools for writing and editing of the manuscript, and no images were manipulated using AI.

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### How to cite this article:

Mukhopadhyay S, Narayan R, Gadag S, Shenoy PA, Garg S, Ashwini T, Nayak UY. Development of levofloxacin glycosylated mesoporous silica nanoparticles for urinary tract infections. *J Appl Pharm Sci*. 2024;14(12):174–179.