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Temperature and pH-responsive nanogels as intelligent drug delivery systems: A comprehensive review

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ABSTRACT

With the advancement of technology and the evolution of modern healthcare systems, the focus is shifting from conventional drug delivery vehicles to novel, nanosized carriers. Nanogels are an example of such nanosized drug delivery systems that have been researched at length. They can be designed to be sensitive to a multitude of physical and chemical stimuli and this endows them with the capability to deliver the drugs they carry in a site-specific manner. These intelligent drug delivery systems are biocompatible and capable of loading copious quantities of the drug. pH is a major chemical property and temperature is a major physical property of a biological system. So nanogels responsive to either pH or temperature or a combination of both, possess immense biomedical potential. This review encompasses synthetic techniques and evaluation tests to confirm the responsive nanogels. A closer look is also taken at their biomedical applications.

INTRODUCTION

Drug delivery systems designed using nanotechnology deal with a large variety of nanosized materials that have certain valuable characteristics that are dependent on their nano-size. Their ability to boost intracellular drug delivery and their subcellular targeting ability are some examples of these characteristics. They are also able to access certain inaccessible sites of the body due to their capability to overcome barriers to the anatomy and physiology of living beings (Banerjee, 2018). Targeted drug delivery to specific sites by nanocarriers reduces off-target effects or side effects. Fluctuations in the drug level in plasma are reduced, dosage frequency is lowered, the solubility of insoluble drugs is enhanced, and an overall increase in patient comfort is seen when nano-formulations are used for therapy (Wanigasekara *et al.*, 2016). Due to this, a substantial number of studies are undertaken

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to design different types of drug nanocarriers. Nanogels are an example of one such nanocarrier.

Nanogels are hydrogel nanoparticles consisting of crosslinked polymeric networks, with a tunable size of less than 100 nm. Their biocompatibility, sustainability, and versatility make them biomedically useful (Rahdar *et al.*, 2019). The integration of the properties of hydrogels with those of nanoparticles endows nanogels with the ability to deliver drugs in a targeted and timecontrolled manner. They act as ideal vehicles for drug delivery because of their superior drug loading efficiency, stability, and responsiveness to stimuli like pH, temperature, ionic strength, etc (Zhang *et al.*, 2016).

Stimuli-responsive nanogels have the potential to conduct controlled and site-specific drug delivery. They are also known as smart or intelligent nanogels. Chemical triggers like pH, enzymes, and ionic change, and physical triggers like temperature, pressure, and magnetic field, alter the nanogel's swelling behavior or lead to the breakdown of its polymeric network. This affects drug release from the nanogel (Ahmed *et al.*, 2020).

A variety of synthetic methods like polymerization and cross-linking are available for the fabrication of these nanogels

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(Liu *et al.*, 2014). The drug loading mechanism must be selected carefully to ensure that the large surface area provided by the nanogel network is utilized and the maximum amount of the drug is loaded (Kendre *et al.*, 2019). Physical entrapment, covalent conjugation, and self-assembly are some of the methods by which the drug is loaded into the nanogel (Zarekar *et al.*, 2017). The molecular weight of the polymer, the thickness of the cross-linked network of the gel, the degradation rate of the polymer, and the interaction of the drug and polymeric chains in the gel influence the drug release characteristics of the system (Suhail *et al.*, 2019). List of polymers used as a temperature and pH-responsive nanogels as intelligent drug delivery system are listed in Table 1.

Smart nanogels are used to deliver drugs in chemotherapy, diabetes, and pulmonary diseases, to name a few. They are also used as optical imaging agents and as thermo-chemotherapeutic agents (Vicario-de-la-Torre *et al.*, 2017). Some of the recent research based on the biomedical applications of single, dual, and multi-stimuli-responsive nanogels are listed in Table 2.

Temperature (a physical trigger) and pH (a chemical trigger) are two extremely significant environmental conditions in biological and biochemical systems (Hassanpour *et al.*, 2017). Therefore, nanogels responsive to these pH and temperature are of great interest, and research-based on these two formulations are on the rise as depicted in the citation report generated by Web of Science, depicted in Figure 1.

Nanogel pros

Because of their high-water content, nanogels are extremely biocompatible and biodegradable formulations that function in the system like real tissues. The drug loading capacity of Nanogel is usually high in nature. The nanogel's sustained release can be regulated by crosslinking a polymeric network, and these polymeric networks are responsible for the nanogel's tiny size, which improves permeation in biological membranes. The most significant advantage of nanogel is that it can incorporate both hydrophilic and hydrophobic drugs, with very little premature drug leakage from the solution. It is readily delivered by parenteral and mucosal routes, and it can reach the tiniest capillary capillaries due to their small volume, as well as infiltrate tissues via the paracellular or transcellular pathways.

Nanogel cons

While nanogels have many advantages, they do have certain drawbacks that may restrict their usage in particular instances. The following are some of the limitations of nanogels. (a) Complete removal of the solvent and surfactants necessitates the use of costly methods. (b) Surfactant or monomer traces may remain and cause toxicity. (c) Manufacturing variation, in which the usual qualities of nanogels are only feasible within a narrow range of sizes. (d) In some cases, a strong interaction between drug and polymer reduces the hydrophilicity of the nanogels and causes the structure to collapse, entrapping the drug molecules irreversibly and increasing the hydrophilicity of the nanogel matrix.

This review takes a closer look at nanogels which are singly responsive to pH and temperature as well as dual pH/temperature-responsive nanogels. The synthetic methods, evaluation techniques, and applications of both single pH and temperature-responsive nanogels as well as dual pH/temperatureresponsive nanogels are considered.

pH-Responsive Nanogels

pH-responsive materials can deliver drugs in a targeted manner by taking advantage of the differences in pH in the different regions of the body. Healthy tissues have a pH of 7.4, the stomach has a pH ranging from 1.0 to 3.0 while tumor tissues have a pH range of 6.5 to 7.0. Nanogels show a greater degree of stability in

Table 1. List of polymers used as a temperature and pH-responsive nanogels as intelligent drug delivery system.

S. No	Polymers used	Reference
1	Poly 6-O-vinyladipoyl-D-galactose-ss-N-vinyl caprolactam-ss-Methacrylic acid	Lou et al., 2015
2	Polyethylene glycol	Wang et al., 2015
3	Poly(amidoamine)	Wang et al., 2016
4	Chitosan and Pluronic F-127	Sahu et al., 2017b
5	Polyglycerol and polyethylenimine	Dimde et al., 2017a
6	Protamine, poly (acrylic acid), and poly (N-isopropyl acrylamide)	Don et al., 2017
7	Dextran	Curcio et al., 2017
8	Sodium alginate modified graphene oxide-grafted N-isopropyl acrylamide and acrylic acid	Sang et al., 2018
9	Chitosan-grafted poly(N-isopropylacrylamide)	Luckanagul et al., 2018
10	Sodium copper chlorophyllin loaded N-isopropyl acrylamide and N-(hydroxymethyl)acrylamide	Chang <i>et al.</i> , 2018
11	Hyaluronan	Zhang et al., 2018
12	Alginate and modified β-cyclodextrin	Hosseinifar et al., 2018
13	Poly(N-isopropylacrylamide)-co-poly (N, N- (dimethyl amino) ethyl methacrylate)	Salimi et al., 2018
14	Dendritic polyglycerol and poly(N-isopropyl acrylamide)	Rancan et al., 2019
15	Poly (ethylene glycol) diglycidyl ether and cystamine double cross-linked hyaluronic acid	Tian et al., 2019
16	3-acrylamidophenylboronic acid, 2-(acrylamido) glucopyranose, and boron dipyrromethene	Guo et al., 2019
17	Methacrylated monocarboxylic sugarcane bagasse cellulose and N-isopropyl acrylamide	Pan et al., 2020

Nanogel	Triggering stimuli	Drug carried	Target cells	Therapeutic application	Reference
Poly (6-O-vinyladipoyl-D- galactose-ss-N-vinyl caprolactam- ss-Methacrylic acid) nanogel	Glutathione (redox), pH, and temperature	Doxorubicin	HepG2 hepatoma cells	Chemotherapy	Lou <i>et al.</i> , 2015
Polyethylene glycol and graphitic carbon dots-integrated chitosan nanogel	pH and light	Doxorubicin	DU145 human prostate cells	Chemo-photothermal therapy	Wang et al., 2015
Poly(amidoamine) dendrimer- based nanogel	Elastase Enzyme	Doxorubicin	C6 glioma cells	Chemotherapy	Wang et al., 2016
Chitosan and Pluronic F-127 nanogel	рН	Bleomycin	HaCaT human keratinocyte cells	Chemotherapy	Sahu <i>et al.</i> , 2017b
Dendritic polyglycerol and polyethylenimine nanogel	pН	siRNA	HaCaT and HeLa cells	Gene Therapy	Dimde et al., 2017a
Protamine, poly (acrylic acid), and poly (N-isopropyl acrylamide)	Temperature, pH, and trypsin enzyme	Doxorubicin and rose Bengal	MCF-7/ADR (Multi- drug resistant breast cancer cells)	Chemotherapy and photodynamic therapy of multidrug-resistant breast cancer	Don et al., 2017
Dextran nanogel	pH and Glutathione (redox)	Methotrexate	HeLa cells	Chemotherapy	Curcio et al., 2017
Sodium alginate modified graphene oxide-grafted N-isopropyl acrylamide and acrylic acid nanogel	Temperature, pH, and magnetic field	Doxorubicin	MCF-7 cells	Chemotherapy	Sang <i>et al.</i> , 2018
Chitosan-grafted poly(N-isopropyl acrylamide) nanogel	Temperature	Curcumin	NIH-3T3 fibroblast cells and HeLa cells	Chemotherapy	Luckanagul <i>et al.</i> , 2018
Sodium copper chlorophyllin loaded N-isopropyl acrylamide and N-(hydroxymethyl)acrylamide nanogel	Temperature and light	5-fluorouracil	L929 fibroblast cells	Chemotherapy	Chang et al., 2018
Hyaluronan nanogel	Light	Chloroquine rhodamine B, cisplatin	MDA-MB-231 triple-negative breast cancer cells	Photochemotherapy based on chloroquine triggered lysosomal expansion and rhodamine B triggered photosensitive lysosome disruption	Zhang <i>et al.</i> , 2018
Alginate and modified β-cyclodextrin nanogel	Pressure	5-Fluorouracil	HT-29 colon cells	Chemotherapy	Hosseinifar <i>et al.</i> , 2018
Poly(N-isopropylacrylamide)- co-poly (N, N- (dimethyl amino) ethyl methacrylate) nanogel	pH and Temperature	Cisplatin	HepG2 cells	Chemotherapy	Salimi et al., 2018
Dendritic polyglycerol and poly(N-isopropyl acrylamide) nanogel	Temperature	Tacrolimus	The human abdomen and breast cells	Treatment of skin inflammation	Rancan <i>et al.</i> , 2019
Poly (ethylene glycol) diglycidyl ether and cystamine double crosslinked hyaluronic acid nanogel	Glutathione (Redox)	Doxorubicin	HaCaT and SCC (Squamous cell carcinoma cells)	Chemotherapy	Tian <i>et al.</i> , 2019
3-acrylamidophenylboronic acid, 2-(acrylamido) glucopyranose, and boron dipyrromethene-based nanogel	Glucose	Insulin	NIH3T3 cells	Detection and treatment of diabetes	Guo et al., 2019
Methacrylated monocarboxylic sugarcane bagasse cellulose and N-isopropyl acrylamide-based nanogel	Glutathione (redox), pH, and temperature	Doxorubicin	-	Chemotherapy	Pan <i>et al.</i> , 2020

 Table 2. Smart nanogels and their biomedical applications.



Figure 1. Web of Science citation reports depicting the sum of citations per year (2015–2020) for the search terms. (a) "pH-responsive nanogels" and (b) "thermoresponsive nanogels" OR "temperature-responsive nanogels" (Data as of February 2021).

physiological environments in comparison to other nano vehicles. Therefore, nanogels sensitive to such pH changes may act as ideal carriers of drugs to precise locations in the body (Hajebi *et al.*, 2019). The pH-responsive behavior of these nanogels enhances the release of the drug while minimizing its loss in off-target sites (Sim *et al.*, 2017).

pH-responsive nanogels may swell or collapse depending on their ionization state due to the addition or removal of protons from the positively charged or negatively charged moieties in the nanogel system. Anionic nanogels having groups like carboxylic acid show a high degree of swelling when their environmental pH is high due to the removal of protons from the anionic groups. However, cationic nanogels consisting of amine groups are protonated at low or acidic pH values leading to the enhancement of the swelling rate of the nanogel (Gonzalez-Urias *et al.*, 2019).

The pH-responsive behavior of nanogels may be utilized not only for drug release but also for drug loading (Qureshi *et al.*, 2019). Both synthetic and organic polymers may be used in the fabrication of pH-triggered nanogels (Karimi *et al.*, 2016).

Synthesis

A variety of techniques may be used to synthesize pHresponsive nanogels. Some of them are discussed in this section.

Inverse emulsion polymerization

The inverse emulsion technique is a commonly used synthetic technique for the preparation of pH-responsive nanogels. This process needs no complex apparatus and the size range of the nanogel formed may be controlled with ease. In the first step, the monomer or precursor polymer solution is emulsified in a continuous apolar phase. A suitable oil-soluble surfactant is used for this. In the next step, chemical crosslinking of the precursor polymer (which is enclosed in the droplets of the emulsion) is conducted. The monomers may also be directly polymerized. A suitable crosslinker or functional group of the polymer chain may be reacted to achieve crosslinking (Krisch *et al.*, 2017).

In a study performed by Zhong *et al.* (2018), inverse emulsion polymerization was used to produce a nanogel comprising of bisacrylamide-crosslinked acrylamide along with methacrylic acid. The co-monomers were first dissolved in an aqueous solution having the bisacrylamide crosslinker and then they were added to an organic phase containing hexane, Brij-30, and dioctyl sulfosuccinate. The nanogels were polymerized via free radical polymerization inside the aqueous droplets suspended in hexane. This allowed the control of both aggregation and particle size. The synthesis technique produced uniform and biocompatible, pH-responsive (due to methacrylic acid) nanogel.

Click chemistry crosslinking polymerization

Click chemistry is a highly reactive and selective process. The yield of nanogel from this process is also sufficiently high (Buwalda *et al.*, 2017).

For example, pH-responsive nanogels for cancer treatment were prepared in a study performed by Wang *et al.* (2017) using thiol-ene click chemistry. Ortho ester diacrylamide (OEAM), methoxy polyethylene glycol acrylate (mPEGA), and pentaerythritol tetra (3-mercapto propionate) (PT) were used as the reactants. PT and OEAM formed the 3D gel network of the formulation and the removal of free mercapto groups was done using mPEGA. This method was extremely efficient and free of by-products. Doxorubicin loaded into the nanogels was released at the pH of the tumor microenvironment.

Inverse nanoprecipitation

In the inverse nanoprecipitation method, hydrophilic nanogels are obtained by precipitating macromonomers in solvents such as acetone. The technique does not require surfactants or ultrasound so the processing conditions are considered to be very mild (Steinhilber *et al.*, 2013).

A study performed to fabricate pH-degradable nanogel used the inverse nanoprecipitation method in combination with inverse electron demand Diels-Alder crosslinking chemistry. Dendritic polyglycerol was utilized as a biocompatible, synthetic polymer to functionalize the macromonomers tetrazine, norbornene, and bicyclo [6.1.0] nonyne. Benzacetal and tetrahydropyran-based acetals were incorporated into the macromonomers to endow the nanogel with pH-degradability. The process was uncomplicated and required no toxic catalysts or expensive synthetic precursors. The macromonomer solution in water (a solvent) was injected into a suitable non-solvent (acetone was used in this study). Water was found to disperse within acetone while the macromonomers were found to precipitate out of the solution due to their insolubility. Small aggregates were formed initially which eventually formed larger conglomerates. Since the local macromonomer concentration was high inside the aggregates, the reaction between the dienophile and methyl tetrazine was found to proceed rapidly and this resulted in the formation of crosslinks between the aggregates. Thus, a hydrophilic, pH-degradable nanogel was formed. Over time, smaller gel networks were found to form crosslinks among themselves until all the macromonomers were depleted. This resulted in the formation of a stable nanogel (Oehrl et al., 2020).

Evaluation of responsiveness

pH-responsive nanogels are first evaluated to figure out their particle size and morphology using techniques such as scanning electron microscopy, transmission electron microscopy, and dynamic light scattering. Zeta potential and polydispersity index of the nanogel are other instrumentally evaluated parameters. The gelation time and temperature as well as the rheological profile of the nanogel, similarly, require evaluation. The efficiency of entrapment of the therapeutic agent or drug within the nanogel is also evaluated. Apart from this, other application-specific *ex vivo* and *in vivo* studies are performed (Sahu *et al.*, 2019).

However, to ensure that the nanogel formed is truly stimuli-responsive, specific tests must be performed to confirm its responsive behavior. Two of the in vitro tests for this purpose are swelling analysis and in vitro drug release studies. In both tests, the nanogel is exposed to buffer solutions of different pH values for specific periods (Sahu *et al.*, 2017b).

In a study, pH-responsive, naproxen sodium-loaded nanogels were developed for gastro-protective drug delivery. The swelling behavior of this nanogel was figured out by exposing the nanogel to two buffer solutions having a pH of 1.2 and 6.8, respectively, for a specific interval of time. The degree of swelling was computed as a ratio of the mass of swollen particles to the mass of dried particles after immersion. At pH 1.2, the nanogel had remained unionized and therefore a lower degree of swelling was seen. However, at pH 6.8, a remarkable degree of swelling was seen because of the ionization of the carboxylic acid groups of the nanogel. This proved that the nanogel was responsive to the environmental pH and was able to protect the drug from the gastric pH. United States of Pharmacopoeia (USP) type-2 dissolution apparatus was used to perform in vitro drug release studies. The nanogel was added to buffer solutions of pH 1.2 and 6.8 in the dissolution apparatus and at specific time intervals, samples were withdrawn, and the amount of drug released was determined using an ultraviolet (UV) spectrophotometer. A higher drug release was seen at pH 6.8 as compared to pH 1.2. Further proof of the pH-dependent behavior of the formulated nanogel was, therefore, obtained (Sarfraz *et al.*, 2019).

Applications

Smart nanogels facilitating drug release in response to environmental pH may conduct site-specific drug delivery. A few of the major applications of this intelligent drug delivery system are enlisted in this section.

Delivery of chemotherapeutic drugs

The pH of the tumor microenvironment is more acidic (ranges from pH 5 to 6.8) than normal tissues. This is due to the accumulation of compounds that are acidic within the tumor tissues (e.g., lactic acid). pH-sensitive polymers are used to fabricate pH-responsive nanogels that are capable of differentiating between the pH of normal cells and tumor cells, to deliver chemotherapeutic drugs specifically into the tumor tissue (Ahmadi *et al.*, 2020).

Poly (N-isopropyl acrylamide-co-acrylic acid) nanogels were formulated specially to carry the chemotherapeutic drug doxorubicin. The nanogel showed superior drug loading capabilities due to electrostatic interactions between its anionic groups and the cationic groups of the drug. It was seen that a minimal amount of drug was delivered in a plasma simulated medium (pH 7.4, 0.14 M NaCl) while a stimulus-responsive release occurred in a lysosomal simulated medium (pH 5, 0.14 M NaCl), as depicted in Figure 2. Apart from this, the nanogels displayed low cytotoxicity [confirmed by 3-(4, 5-dimethylthiazolyl-2)-2, 5-diphenyltetrazolium bromide (MTT) assay]. The gel was nanosized, so it accumulated in the tumor tissue via enhanced permeation and retention (EPR) effect. Thus, this smart nanogel was considered as an excellent candidate as a nanocarrier for doxorubicin, displaying pH-responsive drug release after endocytosis into tumor cells (Cuggino *et al.*, 2016).

However, targeting may be inefficient if the nanogels rely only on the EPR effect to accumulate in the tumor. Therefore, they could be coupled with targeting ligands to achieve active targeting of tumors. A pH-responsive, hyaluronic acid-functionalized smart nanogel was formulated wherein methacrylate hyaluronic acid was copolymerized with a cross-linker having acid-degradable orthoester groups. Doxorubicin was loaded into the nanogel. Cellular uptake studies showed that the nanogel specifically targeted tumor cells. Evaluation tests using 3D tumor spheroids showed that the formulated nanogel was able to completely penetrate the tumor spheroids. Thus, greater inhibition of tumor cells was seen. The nanogels showed enhanced stability and minimal premature release of doxorubicin at neutral pH and maximal drug delivery at the acidic pH of the tumor microenvironment. Thus, targeted cancer therapy was achieved with this formulation (Yang *et al.*, 2017).

pH-dependent, interpenetrating polymeric network nanogel carrying curcumin was also prepared using natural, nonimmunogenic gelatin and acrylamidoglycolic acid. The nanogel



Figure 2. Doxorubicin-loaded pH-responsive nanogels releasing the drug specifically at acidic pH (pH 5.0). Drug release does not take place at pH 7.4.

S. No	Product name	Applications
1	Skin perfect brightening nanogel	It brightens the skin and gives complete hydration. It repairs, tones, and protects the skin
2	NBF Gingival gel	It is a nano-bio fusion technology gel. This gel is absorbed at the gingival cellular level and maintained with a nano bioactive protecting film. It is a high-quality gum protector
3	Aqua Multi-Effect Nano Gel Cream	It is a moisturizing gel that gives complete hydration for a long time. It also works as an anti-wrinkle cream
4	HA nanogel	An excellent alternative to regular toothpaste. Reducing the risk of decay and also reduced bad breath
5	Zyflex nanogel	It relaxes the muscles and erases the body pain
6	Oxalgin nanogel	It gives deeper action and quicker penetration
7	Sane care nanogel	Reduces accumulated fats on the abdomen, arms, legs, thighs, and double chin
8	Revivagenix Pro-Collagen Nano Gel	It is an anti-wrinkle cream, gives complete hydration to the skin for a longer time
9	AugenNanogel Eye-care Gel	It is an eye care gel with deep penetration properties

 Table 3. Currently marketed nanogel formulations.

was able to effectively encapsulate curcumin. The nanogel showed slow swelling in solutions of pH 1.2–5.0. However, it was seen to swiftly swell in solutions of pH 5.0–8.5. The formulation was also biocompatible with the fibroblast cell culture and released curcumin in a controlled fashion. It had enhanced dispersibility in aqueous solutions and improved chemotherapeutic properties in colorectal tumor cells in comparison to free curcumin. Thus, the bioavailability and site-specificity of the drug were enhanced by using smart nanogels (Madhusudana Rao *et al.*, 2015).

A polymer nanogel comprising of a pH-responsive core made of poly (2-diisopropylaminoethyl methacrylate) and a pH-sheddable shell made of polyethylene glycol (PEG) was fabricated to enhance cellular association. At the low tumor microenvironment pH, the PEG shell was cleaved and the charge on the surface of the smart nanogel was reversed. This enhanced the cellular association of the nanogel. Thus, such nanogels can be effectively used as carriers of chemotherapeutic drugs (Sui *et al.*, 2019).

A biomimetic, temozolomide-loaded chitosan nanogel whose surface was modified with a penetration enhancer named transcutol was investigated as a possible nanocarrier system for the topical therapy of skin papilloma. The system offered pH-triggered release at the pH of the papilloma microenvironment. The formulation was stable, and a sustained, site-specific drug release took place in the tumor cells. Thus, the system could act as an alternative to needle-based chemotherapeutic systems for the treatment of skin cancer (Sahu *et al.*, 2020).

Gene therapy

Small-sized (less than 100 nm) cationic nanogels were seen to be useful for ribonucleic acid (RNA) interference-based gene silencing. Larger gels were found to accumulate in lysosomal compartments and thus in vitro transfection failed (Nuhn *et al.*, 2014). Cationic, pH-sensitive nanogels were investigated as carriers of genetic material. pH-cleavable dendritic polyglycerol and low molecular weight polyethylenimine-acrylamide were combined using the thiol-Michael nanoprecipitation technique. The method of synthesis was mild and did not require a catalyst, so sensitive small interfering ribonucleic acid (siRNA) could be encapsulated within the nanogel during the process of synthesis. This minimized the loss and degradation of siRNA. The nanogels formed had pH-responsive benzacetal bonds. The nanogel displayed gene silencing effects like modification-free polyethylenimine. However, the cytotoxicity of the formulated nanogel was significantly lower. It showed an enhanced transfection efficiency and controlled release of the genetic material. Thus, well-defined and user-friendly nano vehicles were formulated for gene therapy (Dimde *et al.*, 2017a).

Gemcitabine, a chemotherapeutic drug, has a structural similarity with the deoxycytidine nucleotides of deoxyribonucleic acid (DNA). So, it could replace the nucleotides when designed as a gemcitabine-loaded DNA nanogel. The formulated nano vehicles could respond to changes in pH due to the formation of an i-motif-like quadruplex. The nanosize of the formulation enabled quick uptake by cells. The nanogel was seen to rapidly disassemble in response to pH and intracellular nuclease-mediated drug delivery was achieved. Thus, superior chemotherapeutic activity was obtained through the use of these intelligent nanostructures (Pan *et al.*, 2019).

pH-sensitive hybrid nanogel particles were formulated via an oil-in-water emulsion process using fluorescently doped nanoparticles of silica encapsulated inside a positively charged poly (2-diethylamino ethyl methacrylate) hydrogel. siRNA loaded into the nanoformulation displayed acidic pH-triggered release. The formulation and the genetic materials were able to avoid sequestration by endosomes and degradation by enzymes. The nanogel was found to decrease the expression of the membrane receptor protein CXCR4 in MDA-MB-231 or human breast cancer cell lines. The formulation was preferentially accumulated in the tumor tissue and thus a potent chemotherapeutic effect was obtained (Khaled *et al.*, 2016).

Miscellaneous applications

A core–shell nanogel formulated using polyethylene glycol-polyaspartic acid proved pH-sensitive swelling behavior. The nanogel was found to be hydrophilic and was able to encapsulate recombinant human insulin. Despite the generous size of the protein drug, loading efficiency was high as enhanced swelling of the nanogel took place when water was present. A low or acidic pH lowered the release of the drug due to the deswelling of the smart nanogel. Thus, the nanogel could protect the protein from acid degradation. At a pH of 7.4, 100% drug release was observed. The nanogel was also biocompatible and biodegradable. Thus, pH-responsive nanogels may be used to deliver protein drugs such as insulin orally (Park *et al.*, 2013).

The oral route is non-invasive. Therefore, it is the route of choice for the administration of drugs. Thus, mannan-modified poly (2-hydroxyethyl methacrylate-co-methacrylic acid) nanogel was formulated as a pH-responsive oral vaccine nanocarrier. The nanogel was able to encapsulate and protect the vaccine at acidic pH and intestinal or basic pH was found to trigger the release of the vaccine. The presence of mannan further improved the internalization of the released protein by macrophages and the expression of required co-stimulatory moieties. Thus, the formulated intelligent nanogel system could be used for efficacious oral vaccine delivery (Durán-Lobato *et al.*, 2014).

In another study, Carbopol and pH-sensitive methacrylic acid were used to formulate nanogels. Ketoprofen, a gastro-irritant, was loaded into the nanogel and it was seen that the nanogel was able to shield ketoprofen from the low (acidic) pH of the stomach and released it at a higher or basic pH of the intestine. So, pHresponsive nanogels may carry out gastro-protective drug delivery of irritant drugs (Sarfraz *et al.*, 2020).

Dendritic polyglycerol was used to formulate a pHresponsive nanogel which was then used to measure pH values inside hair follicles. A porcine ear model was used for the *ex vivo* evaluation of the formulated nanogel. The nanogel was able to penetrate the skin at a great depth using the follicular pathway and the pH was found to increase from pH of 6.5 at the surface of the skin to pH 7.4 in deeper regions of the hair follicle. Additionally, the nanogel created was non-toxic. Thus, the system behaved as a pH-sensitive nanosensor and could be developed further as a smart delivery system for the intrafollicular delivery of drugs (Dimde *et al.*, 2017b).

Temperature-Responsive Nanogels

The temperature may function as a trigger for intelligent drug delivery to inflamed areas and tumor tissue when they have elevated temperatures between 40°C and 45°C. They display superior stability during circulation and faster drug release kinetics in comparison to hydrogels (Hajebi *et al.*, 2019). Temperatureresponsive nanogels show shrinking-swelling behavior triggered by environmental temperature. This facilitates controlled drug release from the smart nanogel (Yin *et al.*, 2020).

When heating alters the hydrophobic/hydrophilic ratio of polymers and reduces their solubility, they are characterized according to a value termed as lower critical solution temperature (LCST). Above LCST, these polymers become insoluble and the formulated nanogel collapses/shrinks, whereas below LCST the nanogel swells. Some polymers, on the other hand, become more soluble on being heated. They are characterized according to the upper critical solution temperature (UCST). Below UCST, these polymers become insoluble and the nanogel formed collapses, whereas above UCST, swelling of the nanogel occurs. A specific temperature known as volume phase transition temperature (VPTT) is correlated to the LCST and UCST values of the polymer used to create the nanogel. When the temperature is below the VPTT of the polymer, polar-polar interactions between hydrophilic portions of the polymer and water are predominant as compared to nonpolar-nonpolar interaction of the hydrophobic regions of the nanogels with each other. The degree of swelling of the gel is enhanced by improving the ratio of solvation of the chains of the polymer. Above VPTT, nonpolar-nonpolar interactions are predominant and water molecules are released into the solution phase. A deswelled polymer network is left behind (Ahmadi et al., 2020).

An example of a polymer used to create thermoresponsive, smart drug delivery vehicles is poly(N-isopropyl acrylamide) (PNIPAM) (Wang *et al.*, 2019). It exhibits temperature-responsive behavior below LCST, at a temperature near 32°C (Lang *et al.*, 2018). Below LCST, the polymer is hydrophilic, while above LCST the polymer aggregates and starts showing hydrophobic characteristics. The system is thus found to shrink above LCST. The LCST of the polymer may also be controlled by modifying the polymer chain (Yin *et al.*, 2020).

Synthesis

Of the substantial number of synthetic techniques used to fabricate thermoresponsive nanogels, a few are discussed in this section.

Reversible addition-fragmentation chain transfer (RAFT) polymerization

Reversible addition-fragmentation chain transfer or RAFT polymerization is a flexible and popular method for imparting living characteristics to radical polymerization. It applies to a large number of monomers and can withstand not only different reaction conditions but also different solvents. The reaction does not require any metal catalysts (Peng *et al.*, 2016b). So, it can be used to synthesize substances that have biomedical applications. Nanogels formulated using the RAFT polymerization technique can be controlled to a great extent concerning their structure and characteristics (Rajan *et al.*, 2017).

nanogel А thermoresponsive was formulated RAFT polymerization using by the monomer 3-dimethyl(methacryloyloxyethyl)ammonium propane sulfonate. Its polymer was found to display UCST behavior in the presence of water. The experiment was conducted in a mixture of ethanol and water, at a temperature of 70°C. Poly (poly (ethylene glycol) methyl ether methacrylate) was used as a macro-chain transfer agent (CTA) to polymerize the monomer. A difunctional monomer was used as a cross-linker. Hairy nanogels were formed at a high concentration of ethanol. The smart nanogel displayed reversible swelling (induced by heating) and shrinking (induced by cooling). This was found to be by UCST behavior that was hypothesized. The nanogel also possessed a tunable volume changing the ratio, size, and transition temperature based on the concentration of ethanol in the ethanol-water mixture, molar ratio of monomer: macro-CTA, and quantity of crosslinker (Fu et al., 2017).

Inverse miniemulsion polymerization

Miniemulsion polymerization methods involve stable mixtures consisting of hydrophilic polymer-surface active agent assemblies in continuous organic mediums (Chiriac *et al.*, 2019).

Inverse miniemulsion was used to fabricate thermoresponsive nanogels using N-vinyl caprolactam (NVCL) and 2-methoxyethyl acrylate (MEA) as monomers. Both monomers owned favorable biocompatibility. Poly (ethylene glycol) dimethacrylate (PEGDMA) functioned as a crosslinker, enhancing the thermoresponsive nature of the nanogel. An emulsifier solution of low polarity was made using the surfactants Span 80 and Tween 60. NVCL, MEA, PEGDMA, and sodium tetrafluoroborate were dissolved in water, resulting in a polar solution. Mixing and preemulsification were done, resulting in a crude emulsion. It was then sonicated and a monomer miniemulsion was formed. An initiator was added to it and nitrogen gas was used to purge the formulation. Polymerization was then conducted in an oil bath under magnetic stirring. The resultant smart nanogel displayed reversible temperature-responsive transition behavior (Gao *et al.*, 2019).

Precipitation polymerization

Emulsion polymerization may be a versatile synthetic method, but it suffers from some drawbacks such as a difficult surfactant removal process and poor control over the polymetric network of the nanogel due to self-crosslinking at high temperatures. To overcome this, a facile synthetic process free of heat and surfactants was investigated. In this process, the thermosensitive PNIPAM nanogel was prepared by precipitation polymerization. A co-initiator system consisting of potassium persulfate (KPS) and N, N, N', N"-tetramethyl ethylenediamine (TMEDA) was used and N, N'-methylene bisacrylamide (MBA) was used as a crosslinker. The mixture of N-isopropyl acrylamide (NIPAM), TMEDA, MBA, and deionized water was first taken in a three-neck round-bottomed flask. After the reagents were entirely dissolved, polymerization of NIPAM was initiated adding KPS to the solution dropwise. Purification of the nanogel obtained was done by dialysis. TMEDA functioned as a catalyst, speeding up KPS release, and the formation of free radicals. This led to the initiation of polymerization of NIPAM in the presence of an MBA. The nanogel displayed distinct thermoresponsive properties (Wei et al., 2018).

Evaluation of responsiveness

Similar to pH-responsive nanogels, before evaluation of the thermoresponsive behavior of the formulated nanogel, parameters such as zeta potential, size, and size distribution are determined using techniques such as dynamic light scattering (Gerecke *et al.*, 2017). FT-IR and protein nuclear magnetic resonance studies may be performed to study the formation of the nanogel. Differential scanning calorimetry may be used to determine the thermal properties of the nanogel (Morelli *et al.*, 2018).

Other evaluation tests to characterize thermoresponsive nanogels, such as the determination of the degree of polymerization of the nanogel by gel permeation chromatography studies, are also done. Transmission electron microscopy studies produce the average radius of the formulation. In a study, the thermoresponsiveness of the system was confirmed via its swelling and shrinking behavior on being exposed to successive cycles of heating and cooling. Five cycles of heating from 10°C to 60°C and cooling from 60°C to10°C were carried out and the hydrodynamic radius was measured to prove that the nanogel was sensitive to alterations in temperature (Ohshio *et al.*, 2019).

Thermoresponsiveness of glycerol-based nanogels could also be determined using UV-Vis spectroscopy. The phase transition behavior of the nanogel could be estimated by measuring its cloud point temperature (Giulbudagian *et al.*, 2014).

Apart from this, tests to determine the therapeutic efficacy and safety of the formulation may also be performed (Luckanagul *et al.*, 2018).

Applications

Like pH-responsive nanogels, thermoresponsive nanogels can also conduct site-specific, stimulus-responsive drug release. Some of the biomedical applications of the formulation are elaborated in this section.

Treatment of skin inflammation

Thermoresponsive nanogels may be designed using the natural properties of the skin. An increase in skin temperature from 32° C to 37° C is seen with increasing depth. So, drug-loaded nanogels possessing a transition temperature of 35° C could utilize this temperature gradient to release the drug (Cuggino *et al.*, 2019). Temperature triggers a reversible change in the volume of the nanogel when the VPTT is crossed. This allows controlled drug release by the nanogel (Tiwari *et al.*, 2019).

In inflammatory skin conditions like psoriasis, the skin may be deficient in a stratum corneum barrier. Thermoresponsive nanogels loaded with etanercept were designed to treat such skin conditions. Release of the drug from the formulation was seen to take place in response to temperature, after topical application. The drug was distributed throughout the stratum corneum and the viable epidermis and an enhanced anti-inflammatory activity was seen (Giulbudagian *et al.*, 2018b).

Polyglycerol-based thermoresponsive nanogels were also used for topical drug delivery across skin with both intact and disrupted barriers. The thermal trigger applied via infrared radiation enabled the formulation to penetrate the stratum corneum to a greater depth. The formulation also reached the epidermis and dermis of barrier-free skin in significant amounts. Thus, the nanogel could be used for treating inflammatory skin disorders (Rancan *et al.*, 2017).

Thermoresponsive nanogels for topical administration, comprising of dendritic polyglycerol and thermoresponsive glycerol, were used as nanocarriers for dexamethasone. β -cyclodextrin was used as a penetration enhancer in the formulation. The drug was found to be localized in the hydrophobic cavity of cyclodextrin. Thus, effective delivery of the drug to the epidermis and dermis of the skin was seen and this nanotechnology-based therapeutic system could be used to treat inflammatory skin diseases. It proved to be superior to marketed dexamethasone creams (Giulbudagian *et al.*, 2018a).

When gold nanoparticles were combined with thermoresponsive nanogels made of Pluronic F-127 and hydroxypropyl methylcellulose and investigated for its antibacterial and burn-induced wound healing capabilities, an enhancement in bioavailability, skin permeation, anti-bacterial, and antiinflammatory activity was seen. The nanogel displayed prolonged and sustained effects (Arafa *et al.*, 2018; Zafar *et al.*, 2014).

The immunotoxicology of various thermoresponsive nanogels was studied by Edlich *et al.* (2017) Despite superior cellular uptake, no cytotoxic or genotoxic effects were observed. No reactive oxygen species were induced. Thus, thermoresponsive nanogels were found to be safe for topical use.

Penetration into the hair follicle

Nanoparticles can penetrate the hair follicle but are unable to overcome the follicular barrier. So, they cannot reach the viable cells or release the drug they carry. Alternatively, smallsized drugs cannot penetrate the hair follicle to a great depth. Thus, studies showed that the most effective method to deliver drugs via the follicular route was to use nanocarriers capable of releasing the drug in response to a stimulus, in the vicinity of the structure being targeted. This could be achieved using thermoresponsive nanogels (Sahle *et al.*, 2017). Thermoresponsive nanogels labeled with indodicarbocyanine, owning a cloud point of 34°C, were evaluated for their ability to penetrate the hair follicle. The results of the studies performed proved that the smart nanogel may be useful for temperature-triggered penetration into the follicles of the hair under physiological conditions (Jung *et al.*, 2018).

Delivery of chemotherapeutic drugs

The onset of hyperthermia leads to increased vascular perfusion and tumor tissue permeability in cancer cells. Based on this property of cancer cells, thermoresponsive nanogels may be useful as site-specific, "on-demand" delivery vehicles for chemotherapeutic agents (Indulekha *et al.*, 2017).

Poly(L-lactide)-g-pullulan copolymer-based nanogels were studied as nanocarriers for the delivery of chemotherapeutic drugs. A doxorubicin-loaded nanogel was prepared and an increase in drug release was seen to take place with increasing temperatures for 50 hours. Thus, the drug-loaded nanogel displayed effective long-term drug release and chemotherapeutic activity at elevated temperatures. The thermoresponsive behavior of the smart nanogel was investigated via the MTT assay. *In vitro* cytotoxicity was studied at 37°C and 42°C. Results showed that the nanogel was biocompatible. The cytotoxic effect of the nanogel at the higher temperature was 1.6 times more than that at the lower temperature as the amount of drug released at 42°C was more than the drug released at 37°C. Higher temperature led to higher hydrophobicity inside the drug-loaded nanocarrier. This allowed the nanogel to be internalized into the tumor cells easily (Seo *et al.*, 2012).

5-fluorouracil and megestrol acetate were loaded into a fibrinogen-graft-poly (N-vinyl caprolactam) nanogel to obtain a biocompatible, temperature-sensitive formulation. The formulation was directed towards specific receptors present on breast cancer cells. The drug release was found to be significant above LCST rather than below it. The formulation displayed improved toxicity, apoptosis, and cellular uptake when studied using the MCF-7 cell line. The drug combination was released in a sustained manner *in vivo*. Thus, smart nanogels may effectively deliver chemotherapeutic drugs to breast cancer cells (Rejinold *et al.*, 2015).

Hybrid core-shell nanogels consisting of NIPAM and vinyl-modified silica were used for thermoresponsive drug delivery. Doxorubicin-loaded nanogels were prepared and the drug release at various temperatures was observed via UV-vis spectroscopy. The release of the drug was found to be higher when the temperature was lower than the temperature of the gel collapse of the polymer. The smart nanogel was found to be biocompatible and could release the drug in a controlled manner in tumor tissues, as confirmed by cytotoxicity studies (Hajebi *et al.*, 2020).

Nowadays, thermoresponsive nanogels are often combined with photosensitive materials (Strozyk *et al.*, 2017). In a study, a temperature-sensitive poly (N-isopropyl acrylamide-coacrylic acid) nanogel was used to encapsulate a heat shock protein 70 inhibitor, 2-phenylethynesulphonamide (PES). A photothermal coupling agent named poly (3,4-ethylene dioxythiophene) was also loaded in the nanogel. When irradiated with near-infrared (NIR) radiation, a temperature-triggered phase transition was found to take place and PES was released from the nanosystem. PES lowered the tumor cells' tolerance to heat. So, tumor tissues could be ablated at a greater depth with a small amount of NIR radiation (Liu *et al.*, 2016).

Gene delivery

Thermoresponsive nanogels have been studied as potential carriers of genes for the treatment of cancer and inflammatory diseases as a result of their capacity to transport the gene to specific regions in response to temperature changes (Wang *et al.*, 2015).

PNIPAM was integrated into the side chain of polyethyleneimine to obtain a thermosensitive nanogel. TRP53, a tumor suppressor gene that plays a crucial role in the regulation of the cell cycle and programmed cell death, was loaded in the nanogel. The nanogel showed significant transfection efficiency, *in vivo* accumulation, and tumor inhibition. When the temperature of PNIPAM was increased above its VPTT, the cellular uptake of the nanogel was enhanced. Thus, the tumor was passively targeted and inhibited (Cao *et al.*, 2015).

Genetic skin diseases which lead to the reduction of the synthesis of proteins, require the substitution of the specific proteins locally. The physicochemical properties of proteins exempt them from topical delivery. However, to prove the efficacy of intelligent nanogels for topical gene delivery, the protein transglutaminase 1-loaded nanogels consisting of temperature-responsive poly (N-isopropyl acrylamide)-polyglycerol were designed. It had a thermal trigger point of 35°C and this was suitable for cutaneous delivery based on the natural physiological temperature gradient of skin. When the temperature was greater than or equal to 35 C, particle size was seen to reduce, protein release was seen to increase. No alterations in the structure or the activity of the protein were seen. The formulation was also capable of penetrating the skin significantly (particularly in barrier-deficient skin). Transglutaminase 1 was efficiently delivered to skin models that lacked the protein and the skin model's barrier function was found to be restored. Thus, this Nano formulation could be used to topically deliver genes for the treatment of genetic skin conditions (Witting et al., 2015).

Miscellaneous applications

Local anesthetics have poor percutaneous absorption across the skin. This makes its topical delivery complex. To overcome this problem, a thermoresponsive mixed micellar nanogel loaded with lidocaine and prilocaine was fabricated. When a temperature change took place, the system changed its phase (sol to gel). This property of temperature-induced phase transition made the nanogel suitable for topical application. The formulation was found to be compatible with the skin when applied topically and an enhanced anesthetic effect was observed when *in vivo* studies were performed. Thus, the smart nanotherapeutic system could be utilized for topical anesthesia (Sharma *et al.*, 2017).

Thermoresponsive nanogels were also studied as potential ophthalmic drug delivery vehicles. A hydrophobic drug named muscone was loaded in the nanogel and the formulation exhibited phase transition at physiological temperature. The nanogel was also capable of resisting clearance because of blinking, leading to enhanced retention time on the surface of the cornea. It could permeate into the ocular surface more effectively. Additionally, it was found to have improved biocompatibility and bioavailability. Thus, such thermoresponsive systems could effectively deliver hydrophobic drugs to the eye (Wang *et al.*, 2016).

The process of transporting antigens across the stratum corneum is complex and thus transdermal immunization is difficult. A poly (N-vinyl caprolactam)-based nanogel was studied as a potential thermoresponsive, antigen delivery vehicle. The temperature of the transition of the nanogel was found to be close to the physiological skin temperature. The nanogel was loaded with ovalbumin. *In vitro* evaluation tests proved that the formulation was cytocompatible and capable of loading and delivering the protein in a temperature-triggered fashion. *Ex vivo* skin penetration studies were performed using a human skin model. Enhanced skin permeation was seen after the topical application of the nanogel as compared to free ovalbumin. Thus, intelligent, temperature-responsive nanogels have great promise as topical vaccination systems (Sonzogni *et al.*, 2018).

Smart nanogels comprising of poly[N-(2-hydroxypropyl) methacrylamide] and thermoresponsive poly[N(2,2-difluoroethyl) acrylamide] were developed as biocompatible, thermosensitive nanostructures. They possessed significant amounts of magnetically equivalent atoms of fluorine and could be used for 19F magnetic resonance imaging (Kolouchova *et al.*, 2018).

4. DUAL pH/Thermoresponsive Nanogels

Single stimulus-responsive nanogels are unable to effectively release the drug they carry in a controlled manner. Dual stimuli-responsive nanogels have been suggested as an alternative to overcome this problem (Yin *et al.*, 2020). Dual pH and temperature-responsive nanogels combine the advantages of both pH-responsive and temperature-responsive nanogels.

Pathological conditions such as cancer are associated with an increase in temperature locally $(2^{\circ}C-5^{\circ}C)$ increase) and a local decrease in pH (by 1–2.5 units of pH). Smart nanogels dually sensitive to pH as well as temperature may be used for site-specific therapy of such conditions, as depicted in Figure 3 (Salehi *et al.*, 2015). Dual pH/redox-responsive (Yang *et al.*, 2016), pH/glucose-responsive (Si *et al.*, 2019), redox/thermal-responsive (Peng *et al.*, 2018) are some of the other dual stimuli-responsive formulations that have been investigated.

An example of a dual pH and temperature-responsive polymer is poly (N, N-diethylamino ethyl methacrylate) (PDEAEM). The tertiary amino moieties of the polymer could be easily changed to quaternary amino groups and this gave the polymer a positive charge. The pKa of the polymer was found to be close to 7 and the polymer displayed a physiologically significant volume phase transition pH ranging from 5.0 to 7.4. The smart polymer was also observed to have a pH-dependent temperature of transition (Manzanares-Guevara *et al.*, 2018). Alternatively, acidic monomers such as acrylic acid may be copolymerized with thermoresponsive substances like PNIPAM to obtain dual responsive nanogels (Bardajee *et al.*, 2017).

Synthesis

Some of the synthetic methods reported in the literature for the formulation of dual pH/thermoresponsive nanogels are detailed in this section.



Figure 3. Drug-loaded dual pH/thermoresponsive nanogel releases the drug at the specific pH and temperature of the tumor microenvironment. Side effects are decreased as no drug is released at the physiological pH and temperature of normal cells.

Free radical polymerization

Free radical polymerization could be used for the formulation of nanogels comprising PNIPAM and chitosan. MBA was used as the crosslinking agent and N, N, N'-ammonium persulfate (APS) functioned as the initiator. The monomers were converted into free persulfate radicals by the initiator. This resulted in the induction of polymerization. The initiator and catalyst (N'tetramethyl ethylenediamine) were first dissolved in distilled water and the solution was heated to a temperature of 70°C. Degassing was done by using nitrogen. The mixture was stirred continuously. Sodium dodecyl sulfate was used to stabilize precursor particles and this led to the reduction of the particles to the nano-range. Reactant solutions consisting of NIPAAM, MBA, and sodium dodecyl sulfate were dissolved in distilled water. Glacial acetic acid was used to dissolve chitosan. The reactant solutions were then added dropwise to the mixture of initiator and catalyst and polymerization took place on constant stirring (Štular et al., 2018).

Free radical polymerization was also used to synthesize monodispersed nanogels consisting of oligo (ethylene glycol) methacrylate. Ultrasonication was used to assist the process as it lowered the reaction time and the desired nanoformulation could be obtained in the absence of an inert atmosphere (Macchione *et al.*, 2019).

Emulsion polymerization

Acrylic acid and butyl acrylate were used as comonomers in the fabrication of dual responsive nanogels by emulsion polymerization in a study performed by Ashrafizadeh *et al.* (2019) Acrylic acid functioned as a highly hydrophilic, pH-sensitive, costeffective, and biocompatible substance. Hydrophobic butyl acrylate, owning a low glass transition temperature, improved the chain flexibility and stretching ability of acrylic acid. The hydrophobicity of butyl acrylate also enhanced the strength of the crosslinking of the monomers. The presence of both hydrophilic and hydrophobic components in the nanogel allowed the loading and delivery of a larger variety of drugs. Chemical crosslinking was done using a bifunctional comonomer, ethylene glycol dimethacrylate (EGDMA) which also prevented the nanogel from disintegrating. Sodium persulfate functioned as the initiator. The reactants were fed into the reaction vessel and heating and continuous stirring led to the initiation of nucleation. Cooling resulted in the formation of a dispersion. It was filtered and purified by ultrafiltration to obtain the nanogel. Direct emulsion polymerization provided an easy synthetic route for the fabrication of the nanogel, free of multiple synthetic steps and costly, toxic solvents (Li *et al.*, 2018).

To avoid using toxic surface-active agents, a soap-less or surfactant-free emulsion polymerization was adopted to fabricate nanogels comprising of a PDEAEM core and PEG shell (An et al., 2015). This is a "green" method of synthesis because the process is conducted in the water and no solvent-based purification stages are involved. A simple dialysis method is sufficient for purification. The method is scalable and uniformly sized nanogels are obtained. Diethylaminoethyl methacrylate (the main monomer), polyethylene glycol methyl ether methacrylate (stabilizer), and EGDMA (crosslinker) were mixed and dissolved in deionized water (Lipowska-Kur et al., 2020). The process of polymerization took place at a temperature of 85°C under continuous stirring. Potassium persulfate (initiator) functioned as the initiator of polymerization. The process was halted through cooling and the resultant dispersion was purified by dialysis to obtain the desired dual responsive nanogel (Manzanares-Guevara et al., 2018).

Evaluation of responsiveness

The nanogels must be evaluated to ensure that they are responsive to both pH and temperature. To that end, dual responsiveness of the smart nanogels comprising of crosslinked poly [itaconic anhydride-co-3,9-divinyl-2,4,8,10-tetraoxaspiro (5.5) undecane] and 1,12-dodecandiol was evaluated using dynamic light scattering technique. The effect of temperature on zeta potential and hydrodynamic radius of the nanogel was investigated using a Peltier device. The nanogel was exposed to a pH range of 4–11 and the effect of pH on the zeta potential and radius of the nanogel was studied to prove the pH responsiveness of the nanogel (Nita *et al.*, 2016).

NIPAM nanogels fabricated via differential microemulsion polymerization were evaluated for their ability to respond to pH by using acid-base solutions and phosphate buffer solutions. The particle size was studied after exposure of the nanogels to different acidic and alkaline pH values. Similarly, acid-base and buffer solutions of certain fixed pH values were taken and the effect of increasing solution temperatures on the size of the nanogel particles was investigated to prove thermoresponsive behavior (Pruettiphap *et al.*, 2017).

In another study, dual temperature and pH-responsive, 5-fluorouracil-loaded nanogels formulated using poly (N-vinyl caprolactam) were evaluated for their responsiveness to both stimuli by performing in vitro drug release studies using a tablet dissolution tester. The formulation was exposed to 0.1 M HCl solution of pH 1.2 and phosphate buffer solution of pH 7.4 at two different temperatures. 25°C and 37°C. Samples withdrawn at regular intervals were evaluated for the drug content using a UV spectrophotometer. The nanogels were fully swollen at the lower temperature (below its LCST) and they were present in a collapsed state at the higher temperature (above LCST). So, drug release was more at 25°C than 37°C. This confirmed the thermoresponsive behavior of the nanogel. The release of the drug was also higher at intestinal pH or pH 7.4 rather than at gastric pH (pH 1.2) due to greater swelling of the polymer network at higher pH. Thus, the nanogels were proved to be pH-responsive as well (Madhusudana Rao et al., 2013).

Applications

The advantages of pH and temperature-responsive nanogels are combined to obtain dual pH/temperature-responsive nanogels. These novel nanoformulations have a variety of biomedical applications, some of which are discussed in this section.

Delivery of chemotherapeutic drugs

Certain malignant cancers give rise to alterations in both pHs and the temperature around the tumor simultaneously. So, a small rise in local temperature and a slight lowering of extracellular pH (pH 6–7) are observed. The pH is found to decrease further within the cells, primarily within endosomes (pH 5–6) as well as lysosomes (pH 4–5.0). So, based on this property of the tumor, dual pH/thermoresponsive nanogels can be formulated. These nanogels can intelligently differentiate between tumor cells and normal, healthy cells. Thus, the targetability and therapeutic efficacy of the formulation is enhanced (Salehi *et al.*, 2015).

Additionally, both pH and temperature can be regulated with ease with the help of external triggers (Soni *et al.*, 2016).

Nanogels consisting of thermoresponsive NIPAM and pH-responsive N, N-(dimethylamino) ethyl methacrylate crosslinked using MBA were loaded with cisplatin and magnetic nanoparticles made of Fe_3O_4 . As per the results of the MTT assay,

the drug-loaded magnetic nanogel displayed increased cytotoxicity on HepG2 cell lines as compared to the free drug. Cytotoxicity assay results also proved the biocompatibility of the nanogel. A greater drug release was seen at the pH (pH 5.7) and temperature (40°C) of the hepatocellular carcinoma microenvironment rather than at physiological pH and temperature. Thus, the novel, dual stimuliresponsive nanogel designed in this study could be used for the efficient treatment of hepatocellular carcinoma (Salimi *et al.*, 2018).

Bionanogels responsive to dual stimuli were formulated using pH-responsive carboxymethyl cellulose grafted with thermoresponsive oligo (-ethylene oxide)-based copolymers. At low pH (pH 6.5 extracellularly and pH 5.0–6.5 inside endosomes and lysosomes of cancer cells) and high temperature (higher than LCST), doxorubicin carried by the nanogel was released. Thus, the drug-loaded, intelligent bionanogel possessed enhanced chemotherapeutic potential (Wen *et al.*, 2015).

A dual pH/thermoresponsive nanogel comprising of (N-isopropyl acrylamide-co-2-acrylamido-2-methyl-1poly propanesulfonate-co-1-propene-2-3-dicarboxylate) crosslinked by ethylene glycol dimethacrylate was designed to deliver doxorubicin specifically to tumor tissue by bypassing several physiological barriers. The nanogel was conjugated to N. O-carboxymethyl chitosan to improve its pH-responsive behavior. The nanocarrier exhibited high drug loading capabilities and the drug exhibited slow and sustained drug release at the temperature (37°C) and acidic pH (pH 5.5) of the tumor microenvironment. The nanogel was found to be taken up by MCF-7 cells rapidly and exhibited preferential cytotoxic behavior towards breast cancer cell lines (MCF-7 and MB231) as compared to normal human breast epithelial cell lines (MCF10A). Thus, targeted and controlled delivery of the entrapped chemotherapeutic drugs took place leading to an enhancement in the efficacy of treatment (Verma et al., 2016).

Treatment of diabetes

Controlled and prolonged insulin delivery systems are needed to reduce the frequency of injections taken by diabetic patients. This, in turn, will enhance patient compliance with therapy (Yao *et al.*, 2014).

Hydroxypropyl methylcellulose crosslinked with methacrylic acid using poly (ethylene glycol) diacrylate as a crosslinking agent, was used to fabricate hybrid nanogels. At a pH of 6, the LCST of the smart nanogel was found to be near the normal temperature of the human body. Thus, sustained release of insulin may take place. The nanogel also had a high insulin loading capacity and entrapment efficiency. By adjusting the pH and temperature, control over drug release from the nanogel can be achieved. Thus, a controlled release of insulin from the nanogel can be achieved in patients suffering from diabetes (Zhao *et al.*, 2016).

Miscellaneous applications

Poly-ε-caprolactone nanoparticles carrying triclosan (antimicrobial drug) and chitosan-based hydrogels carrying flurbiprofen (an anti-inflammatory drug) were entwined to obtain smart nanogels. The formulation displayed both pH-responsive and thermoresponsive behavior. The nanogel was also sufficiently bioadhesive and *in vivo* studies proved that the formulation exhibited enhanced antibacterial and anti-inflammatory activity. The smart nanocarrier was thus a suitable therapeutic system for the treatment of periodontitis using triclosan and flurbiprofen (Aminu *et al.*, 2019).

Nanogels possessing PDEAEMA cores and poly (N-vinyl caprolactam) (PVCL) shells were fabricated for the delivery of genes. The nanogel formed complexes with siRNA via electrostatic interactions. Extensive evaluation tests proved the dual responsive behavior of the nanogel. The novel, intelligent, biodegradable, cationic nanogel provided greater sophistication to the therapeutic system and was capable of carrying out controlled delivery of genes (Aguirre *et al.*, 2016).

There are several nanogel formulations on the market. The majority of the marketed formulations are esthetic treatments, many tooth paste formulations minimize tooth decay issues, and other formulations are personal care items that are extensively used for skin care concerns. Currently marketed nanogel formulations are listed in Table 3.

CONCLUSION AND FUTURE PERSPECTIVE

The changing face of healthcare is giving rise to a demand for more efficient therapeutic systems. Stimuli-responsive nanogels were studied extensively and results proved that they were suitable alternatives to conventional drug delivery vehicles. They could intelligently deliver drugs to specific targets in the body in response to different physical and chemical stimuli. Therefore, they reduced unwanted off-target effects. pH and temperature, being the main chemical and physical properties of a biochemical system, respectively, are two of the major stimuli used to trigger drug release from nanogels. Therefore, nanogels singly responsive to pH and temperature were investigated at length. The advantages of these two types of nanogels were combined and dual pH/temperatureresponsive nanogels were also fabricated. These nanogels could be synthesized in several diverse ways. Each synthetic process owned its unique advantages. By conducting evaluation tests to confirm their responsiveness to specific stimuli, it was seen that the nanogels could swell or collapse in response to specific pH or temperature or a combination of both. Thus, they can be used to intelligently deliver drugs in several different conditions which are associated with a change in pH, temperature, or both. Further research on ways to enhance the biomedical value of nanoformulation lead to the design of multi responsive nanogels. These nanogels may respond not only to dual pH and temperature stimuli but also to other stimuli like glucose concentration, redox conditions, light, pressure, etc. Greater control over drug delivery can be obtained with multi responsive nanogels leading to an enhancement in the therapeutic efficacy of the formulation. Fine-tuning of these intelligent nanocarriers to reduce problems related to toxicity and premature drug delivery and to enhance its site-specificity, biocompatibility, and therapeutic efficacy is needed. Methods to upscale the production of these nanogels must also be investigated. Once these formulations are optimized to possess the greatest therapeutic potential with the least chances of toxicity, steps may be taken to introduce them into clinical settings. This will revolutionize the treatment of many different diseases like cancer, inflammatory disorders, diabetes, etc.

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The authors (Adrija Jha, Annamalai Rama, Bhautik Ladani, Natasha Verma, Sivakumar Kannan, and Anup Naha) declare no conflict of interest, financial, or otherwise.

ETHICAL APPROVALS

Not Applicable

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