



Structure elucidation and *in vitro* biological evaluation of sulfated exopolysaccharide from LAB *Weissella paramesenteroides* MN2C2

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ABSTRACT

The introduction of new bio-product for human diseases treatment is still a great challenge in drug delivery field. This study aimed to extract, purify and characterize exopolysaccharide (EPS) from *Weissella paramesenteroides* MN2C2 which identified by 16S rDNA under accession number of MK530206 in gene bank. The chemical and spectroscopic analysis using UV, FTIR, NMR and HPLC were applied. The results revealed that the product composed of 80% fructose, 9.3% glucose and 5.6% sucrose with acetyl groups. EDX analysis demonstrated that it formed of high ratios of carbon, nitrogen, oxygen, phosphorous and sulfur. The three-dimensional structure showed irregular compacted lumps with smooth surface. The anticancer activity revealed that the partially purified EPS was strong anticancer agent and highly selective against Colon Caco-2, Breast MCF-7 and Liver HepG-2 cancer cell lines compared to the crude ones. Both EPSs showed antiviral activity (99.99%) against Coxsackie virus (CVB3). Their DPPH antioxidant activity acted in a dose dependent manner. From the obtained results it was concluded that our EPS can serve as a promising candidate for drug delivery.

INTRODUCTION

Exopolysaccharides (EPS) are natural complex polymers that may exist as a homopolysaccharide (one type of sugar) or heteropolysaccharide (many types of monosaccharides) (Linares *et al.*, 2017). EPS can be produced from lactic acid bacteria (LAB) which are a group of bacteria found mostly in several fermented foods (Bajpai *et al.*, 2016). LAB fermentation byproducts (EPS) were found to have multiple bioactivities and health-promoting effects such as immune-modulatory, antioxidant, anti-allergenic, protection against infectious diseases, and anti-obesity effects (Linares *et al.*, 2017); thus, they have gained a great economic and therapeutic potential for application in industry, as anti-inflammatory, antitumor, antioxidant, and antiviral drugs (Bajpai *et al.*, 2016). Some LAB strains such as *L. plantarum* E9, *L.*

rhamnosus, *L. delbrueckii*, *L. bulgaricus* B3 (Bajpai *et al.*, 2016), and *Weissella* strains were considered as potential probiotic bacteria (Adesulu-Dahunsi *et al.*, 2017), which benefit their hosts through the production of functional EPS (Trabelsi *et al.*, 2009). Interestingly, the activity of EPS is closely related to its composition, as well as its physicochemical characteristics (Trabelsi *et al.*, 2009). For example, sulfated polysaccharides are characteristic with their high potent and wide spectrum effects against many viral infections (Kim *et al.*, 2010). There are different chemical and spectroscopic analyses that can detect EPS structure, for example, Fourier-transform infrared (FT-IR) and the more advanced techniques such as nuclear magnetic resonance (NMR), high-performance liquid chromatography (HPLC), and GC (Wu *et al.*, 2014). Therefore, a great effort must be spent to deliver new and safe drug from EPS that can be applied in disease treatment, especially cancer therapy to overcome the rapid increase in cancer cases over the world (Osuntoki and Korie, 2010). The mechanistic studies of antitumor activity of LAB-EPS indicated that it has indirect induction of immunity and destruction of tumor cells, activates phagocytosis process, enhances the secretion of interferons, and reduces the production of anti-inflammatory factors (Pan *et al.*, 2015). Furthermore, LAB can resist the free radicals of oxygen that

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cause the oxidative damage of cellular DNA, proteins, and lipids (Schieber and Chandel, 2014). Thus, LAB byproducts which have both antioxidant and anticancer activities can be applied as a safe and effective drug (Wang *et al.*, 2017).

On the other hand, LAB–EPS has a significant ability against viral infections caused by respiratory, gastroenteric, murine, influenza, herpes, and Newcastle disease viruses (Kim *et al.*, 2010). However, there were antiviral agents still not available for clinical use till now, for example, the myocarditis infectious agent called coxsackievirus (Wang *et al.*, 2012) and recently coronavirus (Harrison, 2020). Their mode of action depended on adsorption hindrance, internalization of the virus inside the cell, production of antiviral agents, and antiviral effect through immunomodulation. Therefore, the wide spectrum of antiviral compounds can effectively and rapidly reduce viral infection rates and combat possible pandemics (Harrison, 2020).

The aim of the current study was to isolate and identify a potential LAB strain which has the ability to produce EPS with valuable properties as a future drug. The produced EPS was extracted and partially purified and its structure was determined by the chemical and spectroscopic analysis, as well as its *in vitro* bioactivity estimation was tested as anticancer, antiviral, and antioxidant agents.

MATERIALS AND METHODS

Sampling and bacterial isolation

Samples of colostrums were collected from four buffalos after 24–48 hours of delivery. The samples were refrigerated at 4°C until processed within 24 hours. Nine ml of 0.9% sterile saline solution was added to 1 ml colostrum and mixed. Then, 0.1 ml was inoculated to DeMan–Rogosa–Sharpe (MRS) agar plates and incubated anaerobically at 37°C for 48 hours. Then, 20 colonies were selected randomly from each sample and sub-cultured onto MRS plates for further analysis (Sepp *et al.*, 1997).

Presumptive identification of LAB

Gram reaction and catalase production were checked for all isolates (Holt *et al.*, 1994). A presumptive identification was carried out by using the following tests: colony morphology and pigmentation; growth at 15°C for 10 days and 45°C for 48 hours in MRS broth and salt tolerance at 4%, 6.5%, and 8% NaCl in MRS. Furthermore, gas production from glucose was carried out using MRS broth containing 0.2% of bromocresol purple dispensed into tubes containing inverted Durham tubes. Then, 1% of 10⁶ cfu/ml of each organism was added and incubated at 37°C then observed after 24 hours (Davis, 1955).

Screening of LAB for EPS production

The positive bacterial isolates obtained from the previous step were inoculated separately in 100 ml of MRS broth and incubated aerobically in a static incubator at 37°C for 24 hours; then EPS production was tested as mentioned by Zisu and Shah (2003).

Extraction and purification of EPS

To isolate EPS from the positive strains, the supernatants were collected and neutralized to pH 6.8, then

heated at 100°C for 30 minutes, and recentrifuged for 20 minutes at 4°C. To precipitate EPS, an equal volume of cold absolute ethanol was added to each supernatant and stored overnight at 4°C. The pellets were harvested by centrifugation, washed with distilled water, and dialyzed against deionized water at 4°C for 24 hours and then freeze-dried and the highest EPS dry weight was selected for purification. Trichloroacetic acid 10% was added to EPS to remove proteins; then the supernatant was taken and dialyzed again as mentioned above and then freeze-dried. The purified EPS was dried in an oven at 50°C and weighted (Zisu and Shah, 2003). Then, the most producing isolate (2C2) was used for large-scale production by inoculating 6 l of MRS broth with 2% inoculum (10⁸cfu/ml) and the total dry weight was detected.

Molecular identification by 16S rDNA

The most potent strain was identified by 16S rDNA sequencing. PCR amplification was carried out using universal primers and conditions described by Sange *et al.* (1977). The products of PCR were purified using the QIAquick PCR purification kit according to the supplier's instructions (Qiagen, Hilden, Germany). DNA sequences were determined by the dideoxy chain termination method (Sange *et al.*, 1977). DNA similarity was determined by the BLAST search tool within the National Centre of Biotechnology Information (NCBI) GenBank.

Chemical characterization of EPS

The partially purified EPS (30 mg) was divided into six parts; each part (5 mg) was subjected to one analysis of the following.

Ultraviolet-visible spectroscopy (UV)

The UV spectrum analysis of EPS was determined by dissolving the sample in distilled water and then detected with a UV-visible spectrophotometer at wavelengths ranging from 190 to 800 nm using T80+UV/VIS Spectrometer, PG Instrument Ltd.

Fourier-transform infrared (FT-IR)

The IR spectrum analysis was carried out using FT-IR, Jasco 6100, Model Japan, Resolution: 4 cm⁻¹. The sample was ground with spectroscopic grade potassium bromide powder and then pressed into a 1 mm pellet for FT-IR measurement in the frequency range of 400–4,000 cm⁻¹.

Nuclear magnetic resonance (NMR)

The ¹H NMR spectra were recorded by using BRUKER 500 spectrometer. About 5 mg of our EPS sample was dissolved in 0.75 ml of DMSO and placed in a 5 mm NMR tube. The temperature of the sample was adjusted to 25°C. The chemical shift axis was calibrated with respect to the residual solvent proton at 4.78 ppm.

High-performance liquid chromatography (HPLC)

About 5 mg of EPS was hydrolyzed for about 4 hours; then the hydrolysate was dissolved in distilled water to be injected in HPLC (Boual *et al.*, 2012). The analysis was carried out using Agilent Technologies 1,100 series liquid chromatograph equipped with an autosampler and a refractive

index detector. The analytical column was a Shim-pack SCR-101N. The mobile phase was ultrapure water. The flow rate was kept at 0.7 ml/minutes for a total run time of 20 minutes with isocratic elution. Each carbohydrate concentration was determined after integration of respective areas and their comparison with standard curves obtained with sucrose, glucose, and fructose (Sigma, USA).

Scanning electron microscopy (SEM)

The microstructure and surface morphology of EPS were observed using a JEOL JEM-2100 Electron Microscope at an accelerating voltage of 20 keV. The EPS was fixed in 2.5% of glutaraldehyde (Sigma) and then dehydrated by ethanol. Next, samples were covered with gold using a sputter coater (Scancoat six, Oxford) and observed under scanning electron microscope operating at 20 keV. Micrographs were recorded at higher magnification to ensure clear images (Michalak *et al.*, 2011).

Energy dispersive X-ray (SEM-EDX) and mapping analysis

The elemental composition of the EPS was determined by EDX analyzer equipped with SEM. The X-ray spectrum of the elements was obtained at an accelerating voltage of 20 keV. The elemental analysis, as well as ions distribution (mapping) on investigated samples, was carried out using an EDX system according to Michalak *et al.* (2011).

EPS applications

In vitro anticancer activity

The *in vitro* cytotoxicity assay was assessed using five human cell lines (breast cancer cell line (MCF-7), lung cancer cell line (A549), hepatocellular carcinoma cell line (HepG-2), colorectal adenocarcinoma cell line (Caco-2), and human lung normal cell line (Wi-38)]. First, the cells were inoculated in Roswell Park Memorial Institute (1640) medium in 96-well microtiter plates at a concentration of 10^3 cell/well and then incubated for 24 hours at 37°C under 5% CO₂ incubator. After that, a fresh medium was added to the wells followed by the addition of different concentrations of our samples (0.5–25 mgml⁻¹) against the negative control (cells alone); then the plates were incubated for further 48 hours. Forty µl of 2.5 µg ml⁻¹ (3-(4, 5-dimethylthiazol-2-yl)-2,5-diphenyl tetrazolium bromide) was added to each well and incubated for 4 hours at 37°C. To stop the reaction, 200 µl of 10% sodium dodecyl sulfate was added to each well and incubated overnight. The positive control was prepared using the cytotoxic doxorubicin standard (50 µg ml⁻¹) to give 100% lethality under the same conditions (Thabrew *et al.*, 1997). Then, the absorbance was measured at 540 nm using a microplate multi-well reader.

Viability % = (Reading of extract/Reading of negative control – 1) × 100.

Selectivity index (SI)

The degree of selectivity of the compounds was expressed by its SI value. A high SI value (>2) indicates selective toxicity to cancer cells, while a low SI value (<2) is known to induce cytotoxicity in normal cells (Awang *et al.*, 2014). SI value = IC₅₀ normal cell/IC₅₀ cancer cell.

Antiviral activity by measuring of cytopathic effect

A safe dilution from EPS samples was prepared to be evaluated against coxsackievirus type B3 (CVB3) infection. Activated CVB3 was diluted in culture medium in a range of 10⁻⁴–10⁻⁷; then 50 µl from each dilution was incubated with an equal volume of EPS at 37°C for 1 hour in microtiter plate using CO₂ incubator against control (without EPS). The EPS samples were used at concentrations of 1.5 and 3 mg/ml. Then, a fresh medium was added to the treated and untreated virus and then the plates were incubated at 37°C in CO₂ incubator for 72 hours. The cytopathic effect was observed under an inverted microscope and virus titration was calculated as 50% tissue culture infection dose (TCID₅₀) as described by Shaheen *et al.* (2015). Differences between the values of treated and untreated virus gave the reduction in virus titer.

Antioxidant activity with 2,2-Diphenyl-1-picrylhydrazyl (DPPH) radical-scavenging assay

DPPH method was applied to detect the free radical-scavenging activity of our EPS samples (Yin *et al.*, 2010). Different concentrations (40, 80, 120, 160, 200, and 240 µg/ml) of crude and partially purified EPS samples were prepared in methanol, and then 100 µl was mixed with 900 µl of DPPH shaken vigorously and incubated in the dark for 30 minutes. The decrease in absorbance was measured at 517 nm against control.

% Free radical scavenging = [(A control – A sample)/A control] × 100,

where A sample is the absorbance of DPPH solution with sample and A control is the absorbance of DPPH solution in methanol without sample.

Statistical analysis

All of the determinations reported in this study were carried out in triplicate, and the results are presented as mean values.

RESULTS AND DISCUSSION

Screening, biochemical, and physiological identification of LAB

Eighty different isolates were isolated from four buffalo colostrum samples; only 53 were presumptively assigned to *Lactobacillus* genus (Collins *et al.*, 1993). All our lactobacilli formed white, cocci colonies with concave apex and regular margin. They were Gram-positive and catalase-negative. The isolates grew well at 15°C and 45°C; also they showed growth in MRS broth medium containing 2%–8% NaCl indicating that they are salt-tolerant. Most of them can produce gas from glucose indicating that they are heterofermentative type which can ferment sugars to produce different products including gases (Table 1). This is in accordance with Mulaw *et al.*'s (2019) study who obtained the same results with homo- and heterofermentative LAB.

Selection of high EPS-producing strain for large-scale production

The most potent isolate was strain number 2C2 (*Weissella paramesenteroides*) which gave the highest amount

Table 1. Identification of LAB and screening for EPS production.

Isolates no.	Gas from glucose	Growth at		Growth at NaCl (w/v)			EPS g/100 ml
		15°C	45°C	2%	6.5%	8%	
c1	+	+	+	+	+	+	0.00
c15	+	+	+	+	+	+	0.00
c16	+	+	+	+	+	+	0.00
c17	+	+	+	+	+	+	+ 0.12
c18	+	+	+	+	+	+	+ 0.12
c19	+	+	+	+	+	+	0.00
c20	+	+	+	+	+	+	0.00
2c1	+	+	+	+	+	+	0.00
2C2	+	+	+	+	+	+	+ 0.38
2c8	+	+	+	+	+	+	0.00
2c9	+	+	+	+	+	+	+ 0.15
2c10	+	+	+	+	+	+	0.00
2c11	-	+	-	+	+	-	0.00
2c12	+	+	+	+	+	+	+ 0.13
2c14	+	+	+	+	+	+	+ 0.14
2c15	+	+	+	+	+	+	+ 0.13
2c16	+	+	-	+	+	-	0.00
3c1	+	+	+	+	+	+	0.00
3c2	+	+	+	+	+	+	0.00
3c3	+	+	+	+	+	+	0.00
3c4	+	+	+	+	+	+	0.00
3c5	-	+	+	+	+	+	0.00
3c6	+	+	+	+	+	+	0.00
3c7	+	+	+	+	+	+	+ 0.14
3c8	+	+	+	+	+	+	+ 0.15
3c9	+	+	+	+	+	+	+ 0.13
3c10	+	+	+	+	+	+	0.00
3c11	+	+	+	+	+	+	0.00
3c12	+	+	+	+	+	+	0.00
3c13	+	+	+	+	+	+	0.00
3c14	+	+	+	+	+	+	0.00
3c15	+	+	+	+	+	+	0.00
3c16	+	+	+	+	+	+	0.00
3c20	+	+	+	+	+	+	0.00
4c1	+	+	+	+	+	+	0.00
4c2	+	+	+	+	+	+	0.00
4c3	+	+	+	+	+	+	0.00
4c4	+	+	+	+	+	+	+ 0.12
4c5	+	+	+	+	+	+	0.00
4c6	+	+	+	+	+	+	0.00
4c7	+	+	-	+	+	+	+ 0.17
4c9	+	+	+	+	+	+	0.00
4c10	+	+	+	+	+	+	+ 0.19
4c11	-	+	-	+	+	-	0.00
4c12	-	+	+	+	+	+	0.00
4c13	-	+	+	+	+	+	+ 0.18
4c14	-	+	+	+	+	+	+ 0.15
4c15	-	+	+	+	+	+	+ 0.17
4c16	-	+	-	+	+	-	0.00
4c17	-	+	-	+	+	+	0.00
4c18	-	+	+	+	+	+	0.00
4c19	-	+	+	+	+	+	0.00
4c20	-	+	+	-	+	-	0.00

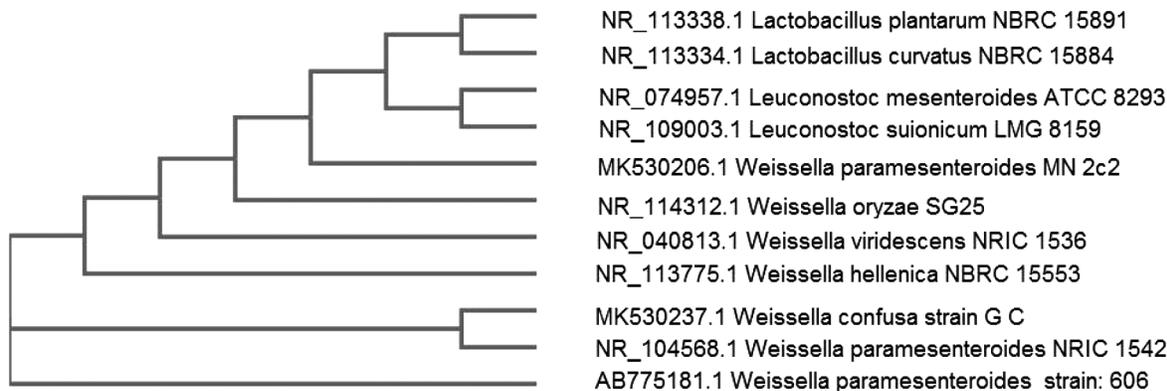


Figure 1. Phylogenetic tree of *Weissella paramesenteroides* MN2C2.

of EPS > 0.3 g% (Table 1). This strain was used for the large-scale production of crude EPS. Also, Jin *et al.* (2019) found that *Weissella confusa* VP30 strain produced the highest amount of EPS compared to other LAB strains which produced < 0.3 g% of EPS on MRS broth.

Molecular identification the most potent isolate by 16S rDNA

16S rDNA sequencing was applied to get accurate identification of the rare bacterial strain. A partial 420 bp linear DNA sequence for isolate number MN2C2 has been aligned and clustered with NCBI database sequences. The phylogenetic tree (Fig. 1) and the analysis of the 16S rDNA gene sequence revealed that this strain clearly belongs to the 16S rDNA strain *Weissella paramesenteroides* NRIC with 99% similarity. The locus sequences of DNA linear were deposited in the gene bank under the accession number of MK530206.

EPS characterization

Spectroscopic analysis

1. UV-visible absorption spectrum

This analysis can be applied to detect chromophore groups of atoms characterized by strongly absorbing electronic transitions of the purified EPS. Figure 2a shows the presence of two peaks in the UV area. The first peak at 192 nm had π - π^* transitions, which is related to many functional groups such as ester, carbonyl, and carboxyl or amine. A small peak (256 nm) found in the area from 250–280 nm π - π^* electronic transition is indicative of aromatic and polyaromatic compounds including protein and/or nucleic acid (Trabelsi *et al.*, 2009).

2. FT-IR analysis

The FT-IR analysis is a powerful tool for the identification of the functional groups of polymers, organic compounds, and many other compounds (Liu *et al.*, 2017). Figure 2b shows FT-IR spectrum of our polysaccharide in the range of 4,000–400 cm^{-1} . The broadband at 3,438 cm^{-1} is assigned to -OH group which is characteristic of polysaccharides, indicating the presence of a carbohydrate ring. The absorption bands at 2,929 cm^{-1} are attributed to C-H asymmetric stretching of methyl or methylene groups that almost found in hexoses (Kumar *et al.*, 2011). The stretching vibration of C=O bond is observed at the region 1,637

cm^{-1} , while the regions of 1,444 and 1,297 could be indicated for the presence of C-O that present in ether or alcohol groups (Liu *et al.*, 2017). The specific fingerprint region that identifies any EPS found between 950 and 1,200 cm^{-1} regions mostly represents pyranose ring vibrations overlapped with stretching vibrations mainly attributed to C-O-C glycosidic bond and C-O-H side groups (Dubey and Jeevaratnam, 2015). There was no absorption band in the region of 1,700–1,770 cm^{-1} indicating the absence of glucuronic acid and diacetyl ester (Wang *et al.*, 2015). This is in agreement with Kanamarlapudi and Muddada (2017) who obtained the same results with EPS from probiotic *Streptococcus thermophilus*.

3. NMR analysis

This method has been used to determine sugar units indicated by the chemical shifts (ppm) (Fig. 2c). Generally, the ^1H NMR polysaccharide spectrum consists mainly of three regions. The first one is formed of overlapping nonanomeric skeletal proton signals present between δH 3.3 and 4 ppm and it is characteristic for the ring proton region of several sugar residues of the polysaccharides (Ismail and Nampoothiri, 2010). The second region is called the anomeric proton region (δH 4.5–5.5 ppm) which indicates the presence of α - and/or β -anomers and the type of linkage is mostly 1 \rightarrow 3, 1 \rightarrow 4, or 1 \rightarrow 6, where our regions between 4.9–5.2 ppm and 4.4–4.8 ppm are due to the α -anomeric proton and β -anomeric protons, respectively (Dubey and Jeevaratnam, 2015). Finally, the high-field reporter region between 1.2 and 2.3 ppm showed the alkyl proton region of the third group which probably contains acetyl, acyl, or alkyl groups from glucosamine (Ismail and Nampoothiri, 2010). This result confirmed that the present product is a polysaccharide formed of α - and β -anomers of various hexoses and contained acetyl groups from glucosamine also; it had a pyranose ring overlapped with some side groups. This is in agreement with Jin *et al.* (2019) who isolated EPS from *W. confusa* VP30 and obtained the same peaks using one proton NMR.

High-performance liquid chromatography (HPLC)

HPLC technique was applied to separate and identify the detailed structure of our EPS. The refractive index chromatogram of EPS after the partial hydrolysis showed that there were three distinct peaks at the same retention times of the standard sugars

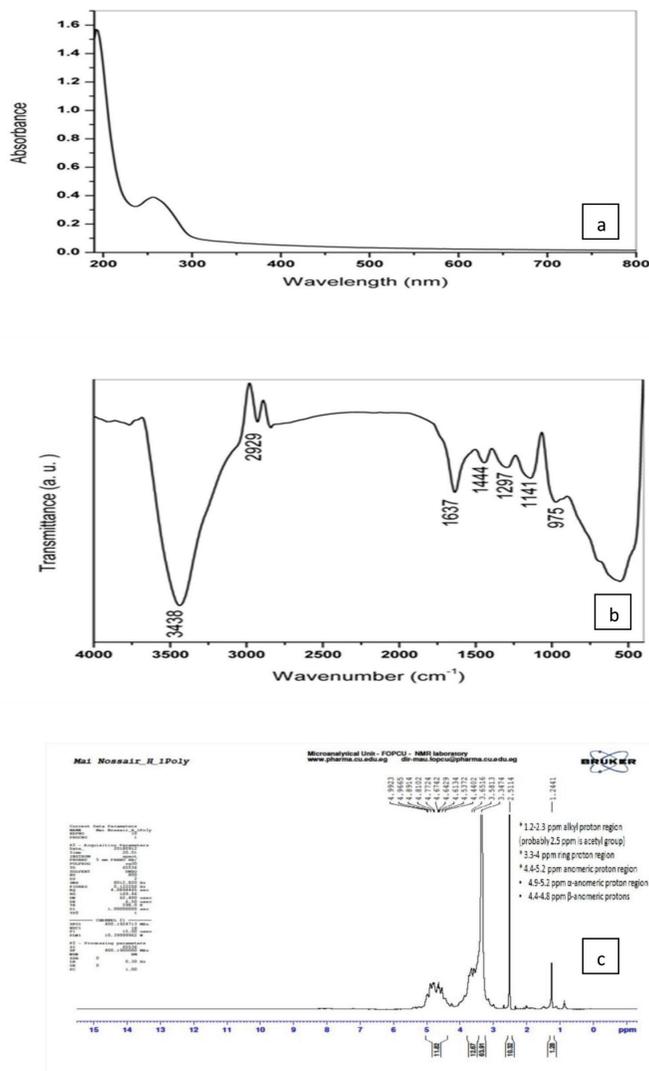


Figure 2. (a) UV-Vis spectrum; (b) FT-IR spectrum; (c) NMR spectrum of EPS.

(sucrose at 6.59 minutes, glucose at 7.96 minutes, and fructose at 8.85 minutes), in addition to the first elution peak at 5.01 minutes (Fig. 3). The highest concentration of our polysaccharide was fructose (259.1 mg ml⁻¹) recording about 80% of total polysaccharide, followed by 9.3% glucose (30.1 mg ml⁻¹) and 5.6% sucrose (18.1 mg ml⁻¹). Also, there were only 5.1% of unidentified peaks (12.7 mg ml⁻¹). In fact, the environmental and cultural conditions in addition to the strain type have a great influence on the type of the produced EPS and its monosaccharides ratios (Tukenmez et al., 2019). The same sugar content as ours was extracted from the fermented beverage of the red seaweed *Gracilaria fisheri* by Hayisama-ae et al. (2014). Furthermore, Ren et al. (2015) used HPLC successfully to separate mixtures of ten monosaccharides produced from *Pleurotus abalonus* and identified them as 87.9% d-glucose, 1.1% d-ribose, 4.4% d-galactose, 3.4% d-mannose, 1.9% l-rhamnose, and 1.4% d-glucuronic acid. Also, Jin et al. (2019) isolated and identified EPS from *W. confusa* VP30 which consists of dextran with glucose units. The EPS structure and monomer composition are very important because

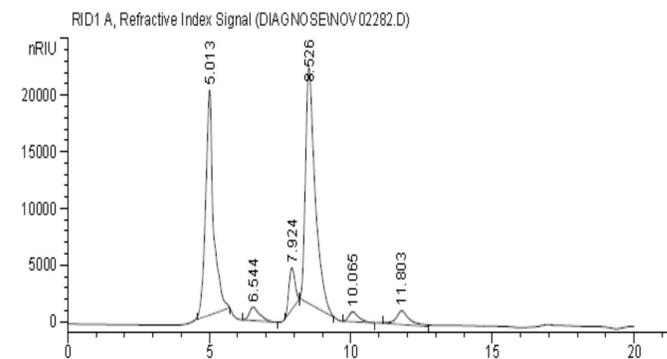


Figure 3. Chromatogram of the hydrolyzed EPS sample.

its application is dependent on its structure and it is specific for each strain (Tukenmez et al., 2019).

SEM of EPS

Generally, it was noted that different polysaccharides possessed different morphology and topography and this may be due to the difference in sample extraction, preparation, and purification. Also, it depends on the physicochemical characters of each EPS (Kanamarlapudi and Muddada, 2017). The scanning electron micrographs of our EPS are presented in Figure 4. Our EPS showed a three-dimensional structure of irregular highly compacted lumps with different sizes and a smooth surface (Liu et al., 2017). The highly compacted structures with flake-like units of EPS extracted from *Lactobacillus fermentum* CFR2195 were observed by Yadav et al. (2011). Also, the porous web-like and irregular structure of the EPS produced from *Lactobacillus plantarum* was noted by Wang et al. (2015).

EDX and mapping analyses

The EPS composed of some elements such as carbon, nitrogen, oxygen, phosphorous, and sulfur in high ratios of its weight which reached 42.31%, 10.11%, 42.68%, 4.13%, and 0.78%, respectively (Table 2 and Fig. 5a). Kanamarlapudi and Muddada (2017) obtained similar results with EPS from *Streptococcus thermophiles* which contains carbon, oxygen, nitrogen, and chlorine in ratios approximately near from ours. Also, our results were in harmony with Boukhelata et al. (2019) who detected the presence of carbon and oxygen as major components of EPS. This result was confirmed by mapping analysis of EPS as indicated in Figure 5b, where the mapping photo shows the presence of the same elements with their ratios which are only included in the selected field/area of the taken photo. Interestingly, the elemental composition and the percentage diversity were found to affect the structure and function of any EPS (Nwodo et al., 2012).

In vitro biological evaluation of EPS

Anticancer activity

Different concentrations (0.5–25 mg/ml) of both EPS from *W. paramesenteroides* MN2C2 were applied and exhibited different anticancer activities against five human cell lines, namely, breast cancer cell line (MCF-7), colorectal adenocarcinoma cell line (Caco-2), hepatocellular carcinoma cell line (HepG-2), lung

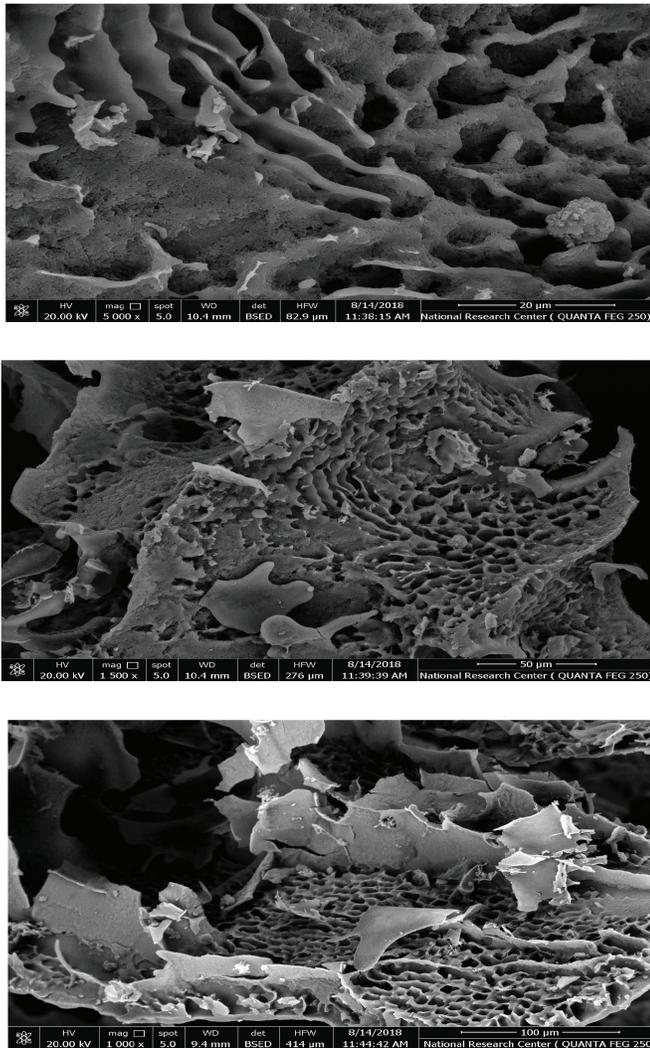


Figure 4. SEM analysis of EPS.

Table 2. EDX analysis of EPS.

Element	Weight %	Atomic %	Net int.	Error %
C K	42.31	49.83	112.89	8.83
N K	10.11	10.21	8.22	21.4
O K	42.68	37.73	108.08	11.43
P K	4.13	1.89	72.19	6.39
S K	0.78	0.34	13.83	18.96

cancer cell line (A549), and human lung normal cell line (Wi-38) (Table 3). The IC_{50} values of crude EPS were 4.2 for Caco-2, 7.2 for MCF-7, and 9.1 mg/ml for Wi-38, while IC_{50} values of the partially purified EPS were 3.5, 2.6, 4.8, 16.5, and 26.1 for Caco-2, HepG-2, MCF-7, A549, and Wi-38, respectively. HepG-2 and Caco-2 were the most sensitive cell lines, while A549 and Wi-38 were the most tolerant ones to our EPS. In this line, Haroun *et al.* (2013) obtained IC_{50} values which reached 9.07, 15.8, 17.6, 19.9, 24.2, and 34.7 for CACO (intestinal carcinoma), HELA (cervical carcinoma), HCT116 (colon carcinoma), HEPG2 (liver

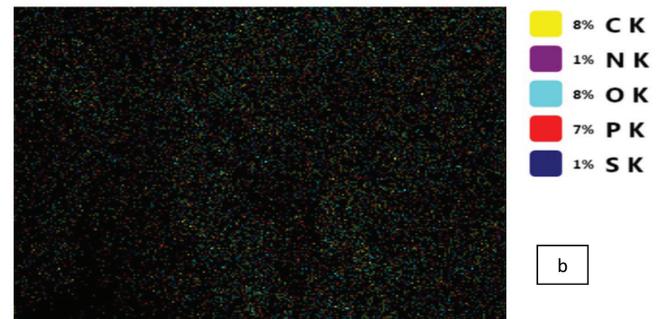
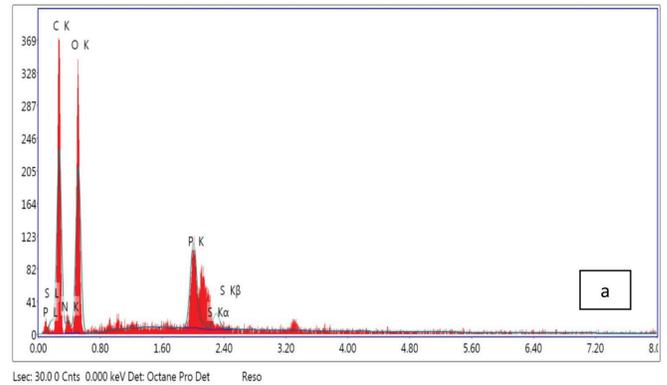


Figure 5. (a) EDX and (b) mapping analyses of EPS.

Table 3. Anticancer potential of crude and partially purified EPS.

Cell lines	Crude EPS		Pure EPS	
	IC_{50} mg/ml	(SI)	IC_{50} mg/ml	(SI)
Caco-2	4.20	2.15	3.50	7.46
HepG-2	0.00	0.00	2.60	9.99
MCF-7	7.20	1.27	4.80	5.39
A549	0.00	0.00	16.50	1.58
Wi-38	9.10	1.00	26.10	1.00

$TCID_{50}$ and selectivity index (SI).

carcinoma), MCF7 (breast carcinoma), HEP2 (larynx carcinoma), and HFB4 (normal melanocytes) cell lines, respectively, revealing that CACO was the most sensitive one to their EPS. Furthermore, our study demonstrated that the partially purified EPS was highly selective (SI > 2) against cancer cell lines suggesting that it had a great safety for the normal cells and this appeared especially for Caco-2, HepG-2, and MCF-7 (Haroun *et al.*, 2013).

The mode of action of EPS is not clear until now, and it may be attributed to the induction of apoptosis of many kinds of cancer cell lines and this could be associated with the main monomer structure of each EPS (Tukenmez *et al.*, 2019). Moreover, Zhou *et al.* (2019) stated that the antitumor activity is induced not only by an indirect effect of regulation of host immunity but also by a direct killing effect on cancer cells. Thus, the probiotic metabolites including polysaccharides, short-chain fatty acids, and protein inhibitory compounds can be used as alternative biotherapeutic cancer drugs (Bedada *et al.*, 2020).

Table 4. Antiviral activity of crude and partially purified EPS against CVB3.

Sample	Conc. (mg/ml)	Virus titers alone	Virus titers with EPS	Reduction value	Inhibition %
Crude EPS	3.0	10 ⁶	10 ²	10 ⁴	99.99
	1.5	10 ⁶	10 ²	10 ⁴	99.99
Pure EPS	3.0	10 ⁶	10 ^{2.5}	10 ^{3.5}	99.95
	1.5	10 ⁶	10 ³	10 ³	99.90

Reduction of virus titer was calculated as “virus titer without extract–virus titer with extract.”

Antiviral activity of EPS

To reduce or inhibit the viral reproduction without adverse effect on host cell viability, a low toxicity medicine with broad-spectrum activity was recommended (Harrison, 2020). The antiviral activity of crude and partially purified EPS was detected against CVB3 (Table 4). Both crude and partially purified EPS have potent antiviral activity which reached 3–4 logs reduction in virus titers with lethal percentage ranged from 99.90% to 99.99%, respectively, using safe concentrations for cell lines. Shaheen et al. (2015) isolated EPS from *L. bulgaricus* OLL1073R-1 with a powerful immune-stimulating effect against CVB3. Furthermore, Biliavska et al. (2019) applied EPS extracted from *Lactobacillus*, *Pediococcus*, and *Leuconostoc* against human adenovirus type 5 and the reduction in viral infection reached 85% and when they added EPS after viral adsorption, the viral formation and release were completely suppressed. The high antiviral potential of EPS against CVB3 can be explained by many ways including the blocking of viral replication at all its stages such as attachment, penetration, replication of genetic materials and capsid, assembly, and viral release from the infected cells (Biliavska et al., 2019), also enhancing host immune system or degrade viral particles (Zhou et al., 2019). Furthermore, our EPS is a sulfated polysaccharide which is characterized by high potent effect against many viruses such as herpes simplex virus, hepatitis B virus, influenza virus, and human cytomegalovirus by inhibition of virus–cell attachment meaning the first step of viral infection (Kim et al., 2010). Finally, the type of EPS monosaccharide can affect the viral replication, where it was found that EPS contain glucose and mannose which had a good effect against coronavirus, so it must be focused on the role of glycolysis metabolism in viral activity (Harrison, 2020).

Antioxidant activity by DPPH radical-scavenging method

The antioxidant capacity of EPS was determined by DPPH as shown in Table 5. Both crude and partially purified EPS showed high antioxidant activity (45.7%–62.1%) dependent on the sample concentration (40–240 µg/ml). The IC₅₀ reached 127.9 and 128.7 µg/ml for crude and pure EPS, respectively. Abdel-Fattah et al. (2012) obtained a strong free radical-scavenging activity with levan produced from *B. subtilis*. Generally, the antioxidant property may have an important role in prevention of several diseases including cancer, heart disease, and immune system decline and act as an anti-inflammatory agent (Wang et al., 2017). So, LAB-EPS has unique properties, particularly the antitumor, antioxidant, and immunostimulant properties which can be applied as strong and safe drugs (Biliavska et al., 2019).

CONCLUSION

LAB bioproducts such as EPS can help to lead a healthy life. The current study was conducted to isolate and identify a

Table 5. Antioxidant activity of crude and partially purified EPS.

Sample	Sample conc. (µg)	DPPH %	IC ₅₀ (µg/ml)
Crude EPS	40	45.7	127.9
	80	45.8	
	120	46.9	
	160	53.3	
	200	58.3	
	240	61.7	
Pure EPS	40	45.7	128.7
	80	45.8	
	120	46.6	
	160	53.6	
	200	58.8	
	240	62.1	

probiotic strain that can produce EPS that has suitable properties for prevention/treatment of some diseases. The EPS extracted from *W. paramesenteroides* MN2C2 was purified and identified and it was found to have qualified characters as an alternative biotherapeutic cancer drug with an antioxidant property which plays an important role in the treatment of tumor. Furthermore, our EPS could provide a good lead for the development of new antiviral drugs.

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AUTHOR'S CONTRIBUTION

Mai Amer contributed to conceptualization, formal analysis, funding acquisition, methodology, resources, and writing the original draft. Eman Elgammal contributed to conceptualization, formal analysis, supervision, writing the original draft, review, and editing. Nagwa Atwa, Ahmed Eldiwany, Insaf Dawoud, and Ferial Rashad contributed to conceptualization, formal analysis, and supervision.

CONFLICTS OF INTEREST

The authors report no financial or any other conflicts of interest in this work.

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