



Simultaneous isolation of gallotannins and a related phenolic from *Mangifera indica* kernels and assessment of their anti-*Trichomonas vaginalis* activities

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ABSTRACT

Therapeutic alternatives are being searched for trichomoniasis as a result of the increased prevalence of metronidazole-resistant infections. *Mangifera indica* L. (Anacardiaceae) is an important tree with a long history in medicine. Traditionally, it has been used as an anti-diarrheal and anti-diabetic, and recently, its gallotannin-rich leaves and stem bark extracts have shown antiparasitic activities against various parasites. Aiming at exploring the anti-*Trichomonas vaginalis* activity of mango's gallotannins, an aqueous ethanol extract of fresh kernels of *M. indica* was phytochemically investigated. Based on a simple gel chromatographic procedure, ethyl gallate (**2**), a group of five isomeric tetragalloyl-glucoses (**3–7**), and a pentagalloyl-glucose (**8**) were simultaneously isolated from a single fraction by a preparative Reversed-phase-high performance liquid chromatography. The isolates were identified based on spectroscopic analyses and comparison with reported data. They showed structural-dependent inhibitory effects on the growth of *T. vaginalis* trophozoites in an *in vitro* investigation. Ethyl gallate and 1,2,4,6-tetra-*O*-galloyl- β -D-glucose (**7**) exhibited elevated anti-*T. vaginalis* activity ($IC_{50} = 1.3, 2.4 \mu\text{g/ml}$, respectively). This is the first report exploring the potential of gallotannins as trichomonocidal agents.

INTRODUCTION

Trichomonas vaginalis (*T. vaginalis*) is the most common sexually transmitted parasite. It affects the external genitalia of men and women leading to a genitourinary disease trichomoniasis. The infections are sometimes associated with preterm birth of pregnant women (Silver *et al.*, 2014) and prostatic tumor of men (Sutcliffe *et al.*, 2012). The development of resistance, allergy, and other side effects due to repeated treatment with imidazole derivatives urged exploring safe alternative anti-trichomoniasis (Cudmore *et al.*, 2004). The gallotannins-rich leaves and stem bark of the tree

Mangifera indica L. (Anacardiaceae) (Núñez Sellés *et al.*, 2002) have shown antiparasitic activities against *Entamoeba histolytica* (Tona *et al.*, 1998); *Histomonas meleagridis*, *Tetratrichomonas gallinarum*, and *Blastocystis* sp (Grabensteiner *et al.*, 2008); *Plasmodium falciparum* (Awe *et al.*, 1998; Ruiz *et al.*, 2011; Zirihi *et al.*, 2005); and *Giardia lamblia* (Amaral *et al.*, 2006). The mango kernels largely produce gallotannins, gallic acid, and benzoic acid derivatives (Masibo and He, 2008). The kernels are a very rich source of gallotannins (15.5 mg/g dry kernel; Berardini *et al.*, 2004). It was traditionally employed to expel tapeworms and as an anti-diarrheal agent (Sairam *et al.*, 2003). Extracts of the kernels have also demonstrated antimicrobial activities against a wide range of the Gram-positive organisms which were ascribed to its gallotannins content (Ahmed *et al.*, 2007; Ka buki *et al.*, 2000). In our previous report, gallotannins, benzophenones, and xanthone C-glucosides from the mango stem bark were shown

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together with their antiviral and cytotoxic activities (Abdel-Mageed *et al.*, 2014). To explore the anti-*T. vaginalis* activity of the tannin constituents of the mango seed kernels, we conducted a phytochemical investigation focusing on the isolation of gallotannins. As a result, gallic acid (1), ethyl gallate (2), 2,3,4,6-tetra-*O*-galloyl- β -D-glucose (3), 1,2,3,4-tetra-*O*-galloyl- β -D-glucose (4), and a combination of almost equal proportions of 1,2,3,6-tetra-*O*-galloyl- β -D-glucose (5) and 1,3,4,6-tetra-*O*-galloyl- β -D-glucose (6), 1,2,4,6-tetra-*O*-galloyl- β -D-glucose (7), and 1,2,3,4,6-penta-*O*-galloyl- β -D-glucose (8) were purified. This study reports an easy chromatographic procedure for the simultaneous isolation of ethyl gallate (2), and the six gallotannins (3-8, Fig. 1) from seed kernels of *M. indica* variety *Tymor*. The effect of these seven compounds on the viability of *T. vaginalis* trophozoites is also reported here for the first time. A growth inhibitory mechanism of the trophozoite is suggested in view of the known protein binding and iron-binding affinity of the tannins.

MATERIALS AND METHODS

General experimental procedures

The one-dimensional (1D) and two-dimensional (2D) nuclear magnetic resonance (NMR) spectra were recorded on a Varian INOVA AS 600 instrument (Agilent, Santa Clara, CA, USA; 0.151 GHz for ^{13}C and 0.6 GHz for ^1H). The chemical shift values were shown in δ (ppm) relative to the solvent signals [$(\text{CD}_3)_2\text{CO}-d_6$ (δ_{H} 2.04; δ_{C} 29.8) on the tetramethylsilane (TMS) scale. Reversed-phase-high performance liquid chromatography (RP-HPLC) was done on a YMC-Pack ODS-A A-303 (product of YMC, Japan) column (4.6 i.d. \times 250 mm) using acetonitrile – water (2:8, v/v) with 0.1% acetic acid. The flow rate was set at 1 ml/min and the oven temperature at 40°C. The eluates were monitored by a Ultraviolet (UV) detector at $\lambda_{\text{max}} = 280$ nm. Preparative RP-HPLC was carried

out on a YMC-Pack ODS-A, A-324 column (10 i.d. \times 300 mm) using the mobile phase acetonitrile – water (2:8, v/v) with 0.1% acetic acid. The flow rate of 2 ml/min at column oven temperature 40°C, and UV detection ($\lambda_{\text{max}} = 280$ nm) were applied. The gels, Dia-ion HP-20 and MCI-gel CHP-20P (products of Mitsubishi Chemical, Japan), were used for the chromatographic experiments.

Plant material

Fresh seed kernels of *M. indica* variety *Tymor* were collected from mature ripe fruits which were purchased from a private farm in Assiut. The plant species was identified by Prof. Dr. Ayman Kotb, Horticulture Department, Faculty of Agriculture, Assiut University. An authentic sample (No. 2012MT) was kept in the Department of Pharmacognosy, Faculty of Pharmacy, Assiut University, Assiut, Egypt.

Extraction and isolation procedures

Mangifera indica fresh seed kernels (~2.5 kg) were extracted by maceration in EtOH/water [7:3, v/v, (3 L \times 4)] at ambient temperature. The obtained extract was concentrated under vacuum at ~40°C. The obtained concentrate (~400 ml) was applied to a Diaion HP-20 column (5.5 i.d. \times 63 cm), which was eluted with distilled water (6 L), methanol/water (2.5:7.5, v/v, 5 L), methanol/water (5:5, v/v, 5 L), methanol/water (7.5:2.5, v/v, 5 L), and methanol (5 L) successively. The respective eluates were dried under vacuum and yielded the respective water (54 g), methanol/water (2.5:7.5, v/v) (98.5 g), methanol/water (5:5, v/v) (55 g), methanol/water (7.5:2.5, v/v) (4.7 g), and methanol (4.4 g) fractions. A portion (3 g) of the methanol/water (2.5:7.5, v/v) fraction was subjected to chromatographic purification on MCI-gel CHP-20P (1.1 i.d. \times 37 cm) with water, water/methanol (9:1 \rightarrow 8.5:1.5 \rightarrow 8:2 \rightarrow 7.5:2.5 \rightarrow 7:3 \rightarrow 6.5:3.5 \rightarrow 6:4 \rightarrow 3:7 and 0:10, v/v). All the 700 drops of

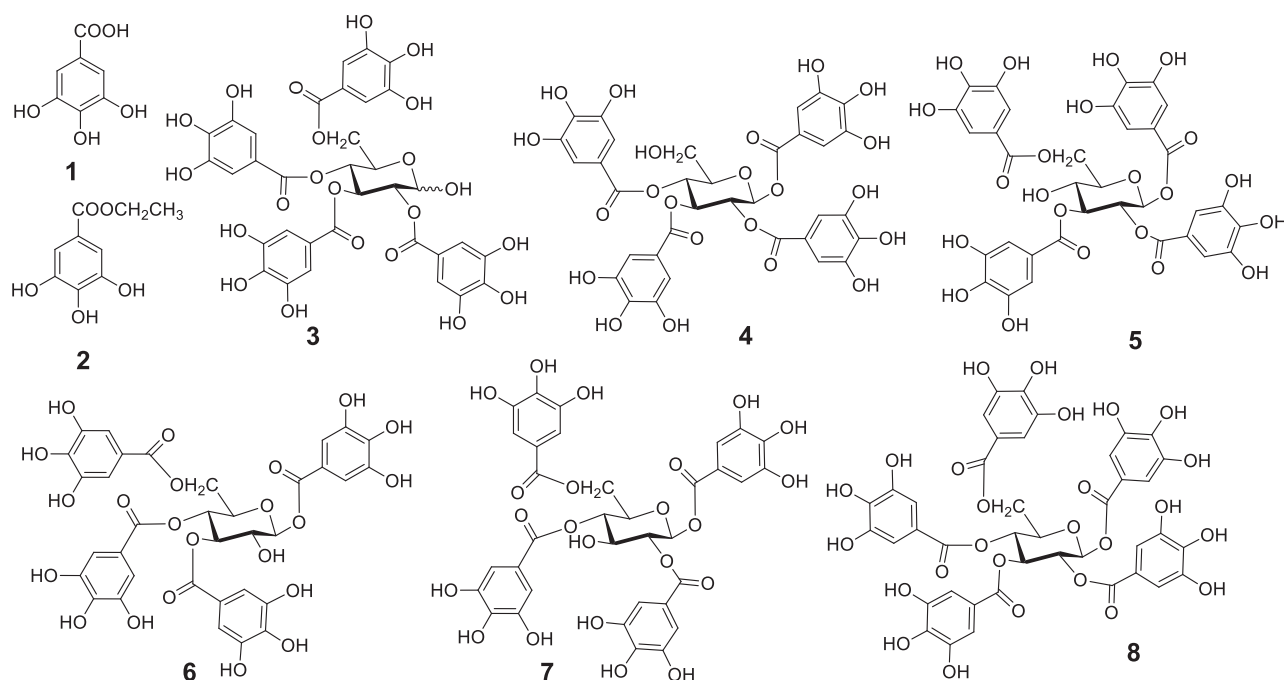


Figure 1. Structures of gallic acid (1) ethyl gallate (2), and the gallotannins 3-8.

the column eluates were gathered separately in a test tube using a fraction collector. The fractions were visualized by monitoring the RP-HPLC. The water eluate furnished crude gallic acid [1 (519 mg)]. The water/methanol (8.5:1.5, v/v) eluate provided pure ethyl gallate [2, (383 mg)]. The water/methanol (8:2, 133 mg), (7.5:2.5, 222 mg), and (7:3, v/v, 221.4 mg) eluates demonstrated identical RP-HPLC profiles (Fig. 2), indicating the presence of the same compounds in these elutes. A preparative HPLC purification of part (300 mg) of the water/methanol (7.5:2.5, v/v) eluted fraction led to purification of 2,3,4,6-tetra-*O*-galloyl- β -D-glucose [3, (4.2 mg)], 1,2,3,4-tetra-*O*-galloyl- β -D-glucose [4, (3 mg)], mixture (13.4 mg) of almost equal proportions of 1,2,3,6-tetra-*O*-galloyl- β -D-glucose (5) and 1,3,4,6-tetra-*O*-galloyl- β -D-glucose (6), 1,2,4,6-tetra-*O*-

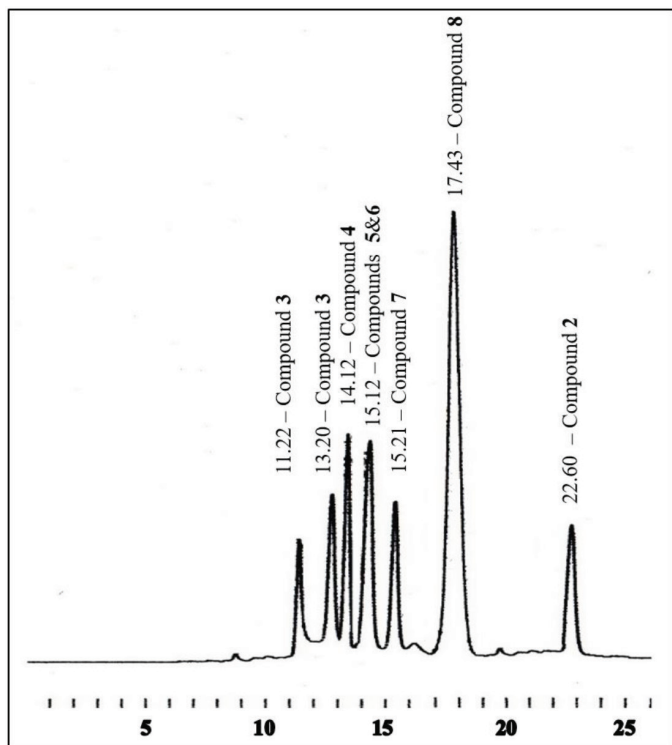


Figure 2. RP-HPLC chromatogram (at $\lambda_{\max} = 280$ nm) of the main compounds in the water/methanol (8:2—6.5:3.5, v/v) eluates from an MCI-gel CHP-20P column, from the water/methanol (8:2, v/v) eluate of the dia-ion HP-20 chromatographic fractionation for the aq. EtOH extract of mango seed kernels.

galloyl- β -D-glucose [7, (5 mg)], 1,2,3,4,6-penta-*O*-galloyl- β -D-glucose [8, (29 mg)], and another pure sample of the ethyl gallate [2, (34.8 mg)]. In addition, the water/methanol (6:4, v/v) eluted fraction also afforded 1,2,3,4,6-penta-*O*-galloyl- β -D-glucose [8, (188.5 mg)].

Compounds physicochemical and spectroscopic data

Gallic acid (1): White, non-crystalline powder; RP-HPLC analysis and co-chromatography with authentic sample ($t_R = 3.80$ min); $^1\text{H NMR}$ [0.6 GHz; $(\text{CD}_3)_2\text{CO}-d_6/\text{D}_2\text{O}$, 9:1] δ_{H} : 7.08 (2H, s, H-2,6); Electrospray ionization mass spectrometry (ESIMS) m/z 171 $[\text{M} + \text{H}]^+$ ($\text{C}_7\text{H}_7\text{O}_5$).

Ethyl gallate (2): Off-white, non-crystalline powder; $^1\text{H NMR}$ [0.6 GHz; $(\text{CD}_3)_2\text{CO}-d_6/\text{D}_2\text{O}$, 9:1] δ_{H} : 7.07 (2H, s, galloyl H-2,6), 4.22 (2H, q, $J = 7.2$ Hz, CH_2), 1.27 (3H, t, $J = 7.2$ Hz, CH_3); $^{13}\text{C NMR}$ [0.151 GHz; $(\text{CD}_3)_2\text{CO}-d_6/\text{D}_2\text{O}$, 9:1] δ_{C} 167 (C-7), 145.9 (2C, C-3,5), 138.7 (C-4), 121.6 (C-1), 109.6 (2C, C-2,6), 61 (C-8) and 14.5 (C-9); ESIMS m/z 199 $[\text{M} + \text{H}]^+$ ($\text{C}_9\text{H}_{11}\text{O}_5$).

2,3,4,6-Tetra-*O*-galloyl- β -D-glucose (3): Off-white, non-crystalline powder; $^1\text{H NMR}$ [0.6 GHz; $(\text{CD}_3)_2\text{CO}-d_6/\text{D}_2\text{O}$, 9:1] δ_{H} : 7.13, 7.02, 7.00, 6.92 [each s, 2H, galloyl H-2,6 \times 4], galloyls of the β -anomer of the D-glucose, 7.15, 7.03, 7.028, 6.98 [each s, 2H, galloyl H-2,6 \times 4], galloyls of the α -anomer of the D-glucose, glucose protons (Table 1); ESIMS m/z 789 $[\text{M} + \text{H}]^+$ ($\text{C}_{34}\text{H}_{29}\text{O}_{22}$).

1,2,3,4-Tetra-*O*-galloyl- β -D-glucose (4): Off-white, non-crystalline powder; $^1\text{H NMR}$ [0.6 GHz; $(\text{CD}_3)_2\text{CO}-d_6/\text{D}_2\text{O}$, 9:1] δ_{H} : 7.07, 7.025, 6.98, 6.95 (each s, 2H, galloyl H-2,6 \times 4), glucose protons (Table 1); ESIMS m/z 789 $[\text{M} + \text{H}]^+$ ($\text{C}_{34}\text{H}_{29}\text{O}_{22}$).

1,2,3,6-Tetra-*O*-galloyl- β -D-glucose (5): Off-white, non-crystalline powder; $^1\text{H NMR}$ [0.6 GHz; $(\text{CD}_3)_2\text{CO}-d_6/\text{D}_2\text{O}$, 9:1] δ_{H} : 7.13, 7.06, 7.04, 6.98 (each s, 2H, galloyl H-2,6 \times 4), glucose protons (Table 1); ESIMS m/z 789 $[\text{M} + \text{H}]^+$ ($\text{C}_{34}\text{H}_{29}\text{O}_{22}$).

1,3,4,6-Tetra-*O*-galloyl- β -D-glucose (6): Off-white, non-crystalline powder; $^1\text{H NMR}$ [0.6 GHz; $(\text{CD}_3)_2\text{CO}-d_6/\text{D}_2\text{O}$, 9:1] δ_{H} : 7.16, 7.10, 7.05, 7.02 (each s, 2H, galloyl H-2,6 \times 4), glucose protons (Table 1); ESIMS m/z 789 $[\text{M} + \text{H}]^+$ ($\text{C}_{34}\text{H}_{29}\text{O}_{22}$).

1,2,4,6-Tetra-*O*-galloyl- β -D-glucose (7): Off-white, non-crystalline powder; $^1\text{H NMR}$ [0.6 GHz; $(\text{CD}_3)_2\text{CO}-d_6/\text{D}_2\text{O}$, 9:1] δ_{H} : 7.14, 7.11, 7.08, 7.05 (each s, 2H, galloyl H-2,6 \times 4), glucose protons (Table 1); ESIMS m/z 789 $[\text{M} + \text{H}]^+$ ($\text{C}_{34}\text{H}_{29}\text{O}_{22}$).

1,2,3,4,6-Penta-*O*-galloyl- β -D-glucose (8): Off-white, non-crystalline powder; $^1\text{H NMR}$ [0.6 GHz; $(\text{CD}_3)_2\text{CO}-d_6/\text{D}_2\text{O}$,

Table 1. $^1\text{H NMR}$ data (δ in ppm, J in Hz) of glucose moiety of the galloylglucoses 3–8 (0.6 GHz, $(\text{CD}_3)_2\text{CO}-d_6/\text{D}_2\text{O}$, 9:1, 27°C).

glc. H	3		4	5	6	7	8
	3 α	3 β					
1	5.55 (d, 3.6)	5.16 (d, 7.8)	6.19 (d, 8.4)	6.10 (d, 8.4)	5.96 (d, 8.4)	6.03 (d, 8.4)	6.27 (d, 8.4)
2	5.11 (dd, 3.6, 10.2)	5.26 (dd, 7.8, 10.2)	5.55 (dd, 8.4, 9.6)	5.45 (dd, 8.4, 9.6)	4.023 (dd, 8.4, 9.6)	5.37 (dd, 8.4, 9.6)	5.62 (d, 8.4, 9.6)
3	6.04 (t, 10.2)	5.75, t	5.92, t	5.64, t	5.65, t	4.35, t	6.00, t
4	5.55 (t, 10.2)	5.47 (t, 10.2)	5.46 (t, 9.6)	4.04 (t, 9.6)	4.04 (t, 9.6)	5.38 (t, 9.6)	5.65 (t, 9.6)
5	4.58 (ddd, 1.8, 6.6, 10.2)	4.24 (ddd, 1.8, 6.6, 10.2)	4.16 (ddd, 1.8, 4.8, 10.2)	4.14 (ddd, 2.4, 5.4, 9.6)	4.35 (ddd, 2.4, 5.4, 9.6)	4.29 (ddd, 1.8, 4.8, 9.6)	4.54 (ddd, 1.8, 4.8, 9.6)
6	4.24 (dd, 4.8, 12)	4.24 (dd, 4.8, 12)	3.73 (dd, 1.8, 12)	4.63 (dd, 1.8, 12)	4.51 (dd, 2.4, 12)	4.52 (dd, 1.8, 12.6)	4.57 (dd, 1.8, 12.6)
	4.50 (dd, 1.8, 13.2)	4.50 (dd, 1.8, 13.2)	3.65 (dd, 4.8, 13.2)	4.46 (dd, 4.8, 12)	4.21 (dd, 4.8, 12)	4.19 (dd, 4.8, 12.6)	4.3 (dd, 4.8, 12.6)

9:1] δ_{H} : 7.13, 7.07, 7.04, 6.99, 6.97 (each s, 2H, galloyl H-2,6 \times 5), glucose protons (Table 1); ESIMS m/z 941 [M + H]⁺ (C₄₁H₃₃O₂₆).

Anti-*T. vaginalis* activity of the isolated compounds

Culture of *T. vaginalis*

Fresh *T. vaginalis* was isolated from a female patient visiting the Patient Treatment Center, Gynecology and Obstetrics Hospital, Faculty of Medicine, Assiut University. After explaining the aim of the study, informative consent was signed by the patient. This study was endorsed by the Research Ethics Committee, Faculty of Medicine, Assiut University, Egypt. The live trophozoites were grown in trypticase–yeast extract–maltose (TYM) medium (Innocente *et al.*, 2014) in 15-ml screw-stoppered glass tubes at 37°C. The medium was complemented with 1-ml heat-inactivated fetal calf serum, crystalline penicillin (1,000,000 IU/ml), and streptomycin sulfate (100,000 µg/ml), and the pH was adjusted at 6.0. The isolate was regularly transferred to fresh TYM medium every 48 hours, and maintained in the Parasitology Department, Faculty of Medicine, Assiut University. The growth of *T. vaginalis* was monitored until it reached the logarithmic growth phase. The trophozoites exhibiting normal viability and morphology were harvested, isolated from the medium by centrifugation, and regrown in fresh TYM medium for the investigations; 10⁵/ml cells were used for the evaluation of the anti-trichomonal effect of the isolated compounds. The organism count number was monitored using hemocytometer slides.

The trichomonacidal assay

The anti-*T. vaginalis* activity of the pure compounds (2–8) was determined *in vitro* based on the reported method (Ibrahim, 2013). The test compounds were diluted by DMSO as solubilizing vehicle at a concentration not more than 0.01% in the experiment medium. A series of two-fold dilutions for all compounds tested were obtained and used at final concentrations of 50, 25, 12.5, 6.25, 3.12, 1.56 µg/ml. A 50 µl of 1 \times 10⁵ trophozoite/ml was inoculated in 150 µl of the test compound in a sterile 96-well culture plate at 37°C (Innocente *et al.*, 2014). Three controls were considered: negative control with parasites only; vehicle control (DMSO); and the positive control [metronidazole (MTZ) (Sanofi-Aventis, Egypt). MTZ was dissolved in distilled H₂O, diluted by the incubation medium giving four concentrations, 3.12, 1.56, 0.75, and 0.0037 µg/ml, and used as the reference standard. The 96-well plate was incubated for 72 hours at 37°C, 5% CO₂. The number of living organisms was determined by counting the parasites by a hemocytometer using the trypan blue dye (0.2%) exclusion method. Control cultures were treated by the same procedure as that of the test cultures. All assays were carried out in triplicate and repeated three times. For evaluation of the activity of the compound at different contact times, samples from the cultures were taken after 24-, 48-, and 72-hours incubations. The mortality percentage of the parasites was determined from the equation {100 \times [100–a/b]} where “a” is the living organisms count in test wells and “b” is the living organisms count in the control wells (Ali, 2007).

Statistical analysis

The IC₅₀ values were computed from a non-linear regression curve using the GraphPad Prism 5.0 software.

RESULTS AND DISCUSSION

Simultaneous isolation of gallotannins

The development of analytical technique Liquid chromatography mass spectrometry/mass spectrometry (LCMS/MS) made profiling and identification of multi-components of plant extracts more rapid. Identification of a series of gallotannins in extracts of morphological parts of mango (peel, pulp, and kernel) by such technique has recently been reported (Berardini *et al.*, 2004). However, estimation of the activity of the individual gallotannins, as well as its pharmacokinetic and pharmacodynamic parameters, are largely dependent on the tannin structure (Berardini *et al.*, 2004, Gan *et al.*, 2018), which necessitate obtaining the tannin in pure form. We herein succeeded to simultaneously purify a group of five isomeric tetragalloyl glucoses (3–7), a pentagalloyl glucose (8), and a tannin related compound, ethyl gallate (2), based on a preparative RP-HPLC purification of a crude tannin mixture obtained from gel chromatography. Briefly, a concentrated aqueous EtOH extract of fresh mango kernels was fractionated on a Diaion HP-20 column with methanol/water gradients (0:10 \rightarrow 2.5:7.5 \rightarrow 5:5 \rightarrow 10:0, v/v). The methanol/water (2.5:7.5, v/v) eluted fraction was further fractionated on an MCI-gel CHP-20P with water, water/methanol gradients (9:1 \rightarrow 8.5:1.5 \rightarrow 8:2 \rightarrow 7.5:2.5 \rightarrow 7:3 \rightarrow 6.5:3.5 \rightarrow 6:4 \rightarrow 3:7), and methanol. Then, simultaneous isolation of the compounds 2–8 (Fig. 1) was attained by RP-HPLC purification of the water/methanol (7.5:2.5, v/v) eluate on A-324 (10 i.d. \times 300 mm) ODS-A column (YMC-Pack) using CH₃CN–water (2:8, v/v) acidulated with 1 ml/L acetic acid. A 2 ml/min flow rate at column oven 40°C was used, and the eluates were monitored by a UV detector set at λ_{max} 280 nm.

The easy and fast isolation procedures described here for gallotannins could be useful for the rapid purification of such compounds from other gallotannins-rich plants including other morphological parts of *M. indica*.

Identification of the isolated compounds

Analysis of the data obtained from ESIMS and NMR spectroscopic experiments, ¹H, ¹³C ¹H–¹H correlation spectroscopy (COSY), heteronuclear single quantum correlation (HSQC), and heteronuclear multiple bond correlation (HMBC), and comparison of the found data with literature ones, allowed identification of compounds 1 and 2 as gallic acid (Lee *et al.*, 2005) and ethyl gallate (Cui *et al.*, 2002), and galloylglucoses (3–8) as 2,3,4,6-tetra-*O*-galloyl- β -D-glucose (3) (Haddock *et al.*, 1982), 1,2,3,4-tetra-*O*-galloyl- β -D-glucose (4) (Berardini *et al.*, 2004), mixture of almost equal proportions of 1,2,3,6-tetra-*O*-galloyl- β -D-glucose (5) (Hagenah and Gross, 1993) and 1,3,4,6-tetra-*O*-galloyl- β -D-glucose (6) (Haddock *et al.*, 1982), 1,2,4,6-tetra-*O*-galloyl- β -D-glucose (7) (Haddock *et al.* 1982), and 1,2,3,4,6-penta-*O*-galloyl- β -D-glucose (8) (Haddock *et al.*, 1982). Worthy, gallic acid (1), ethyl gallate (2), and the pentagalloylglucose (8) are the most abundant isolates.

Anti-*T. vaginalis* activities of the isolated compounds

Secondary metabolites from natural sources have played a unique role in the discovery of anti-infectious compounds. Studies on the inhibitory effects of plant extracts on the growth of *T. vaginalis* trophozoites evidenced that extracts from the volatile oils producing plants *Mentha piperita*, *Salvia officinalis*,

and *Tanacetum parthenium* demonstrate anti-*T. vaginalis* activity identical to MTZ (Ezz Eldin and Badawy, 2015; Sharafi *et al.*, 2013). A commercial garlic-based product (Tomex®) has been also shown to significantly reduce the multiplication and motility of the *T. vaginalis* trophozoites (Ali, 2007). In another study, the hydrolyzable tannins containing ethyl acetate fraction of a *Eucalyptus* extract exhibited MTZ-like activity on the growth of the *Trichomonas* trophozoites (Hassani *et al.*, 2013). Meanwhile, limited research on the effects of pure compounds from natural sources, such as berberine alkaloid (de Brum Vieira *et al.*, 2015; Sharafi *et al.*, 2013) and piperazinyl derivatives of betulinic acid (Innocente *et al.*, 2014), have been reported.

Herein, the effect of a group of gallotannins and ethyl gallate from kernels of *M. indica* on the viability of *T. vaginalis*

clinical isolates was estimated (see experimental section). Although the common chemical nature of the investigated compounds, galloyl esters of a glucose core (Fig. 1), their antiprotozoal activities were varied based on the molecular structure (Fig. 3). All the examined compounds (2–8) exhibited a remarkable dose-dependent and time-dependent decrease in the percentage of living trophozoites (Fig. 3). After 24 hours of incubation, ethyl gallate, a tannin-related compound, showed the highest inhibitory effect on the *T. vaginalis* trophozoites viability (IC_{50} 1.3 $\mu\text{g/ml}$, Fig. 4). The gallotannin with an unacylated O-3 position of the glucose core, 1,2,4,6-tetra-O-galloyl- β -D-glucose (7), exerted a potent effect (IC_{50} 2.4 $\mu\text{g/ml}$), while the inhibitory effect of ~1:1 mixture of 5 and 6 on the *T. vaginalis* trophozoites was relatively low (IC_{50} 36.1 $\mu\text{g/ml}$). The other gallotannins (3, 4 and 8) exhibited

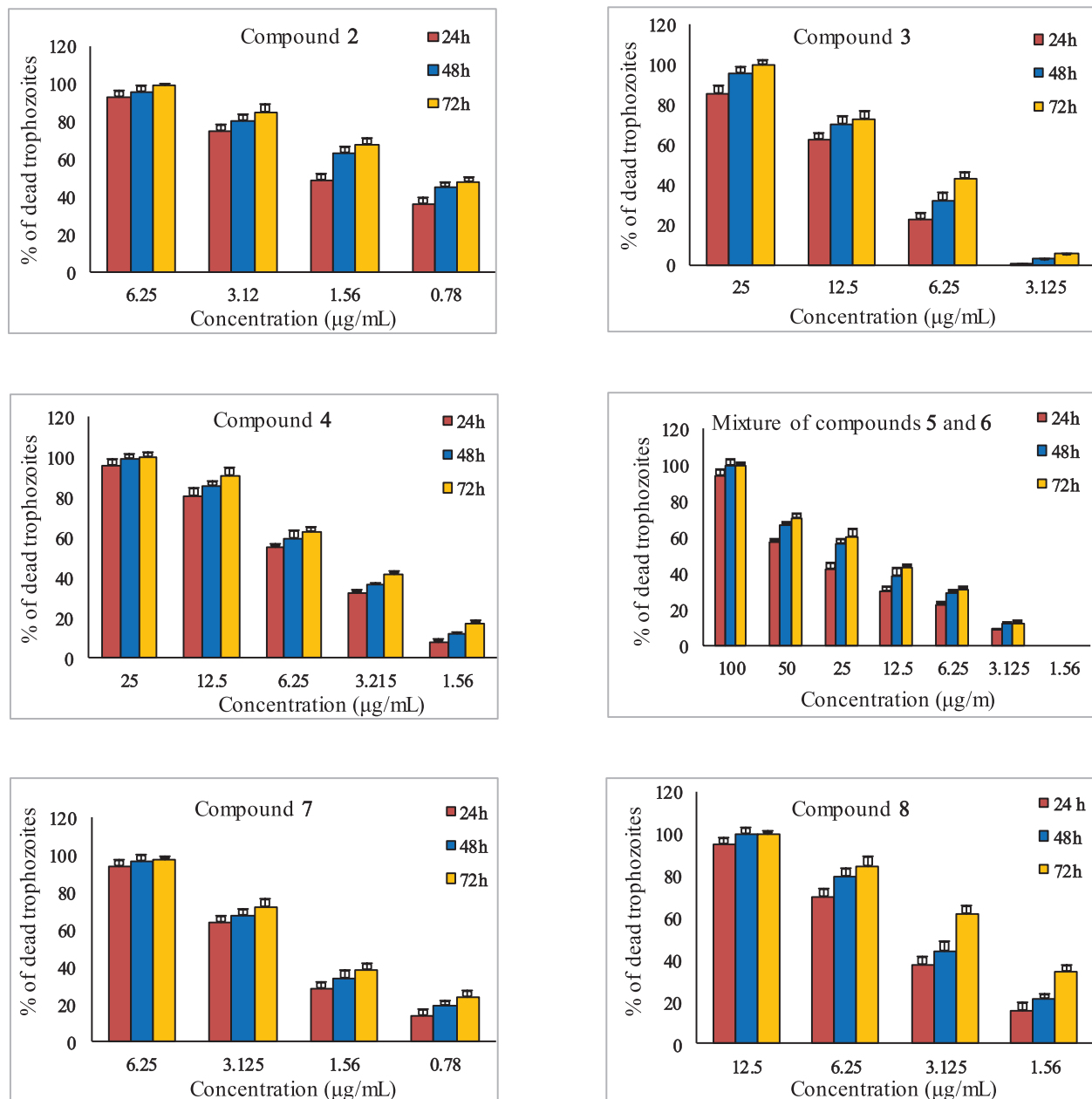


Figure 3. *In vitro* inhibitory effects of the compounds 2–8 on viability of *T. vaginalis* trophozoites. The bars represent standard deviation (SD) of the means.

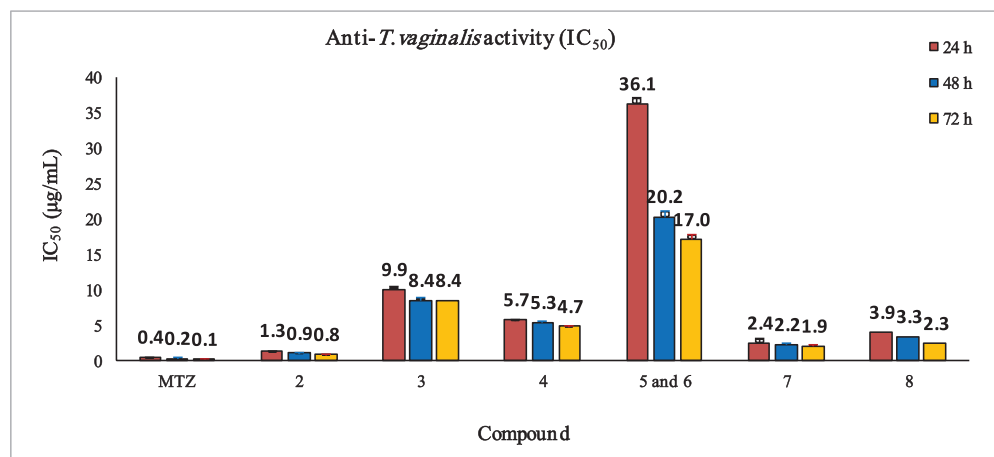


Figure 4. IC₅₀ of the compounds 2–8 on *T. vaginalis* clinical isolate compared to a standard control (MTZ). The bars represent SD of the means.

noticeable anti-*T. vaginalis* activities with comparable potencies (IC₅₀ 3.9 – 9.9 µg/ml). Due to the presence of some differences in the activity of the examined gallotannins, a structural activity relationship can't be generated because of the limited number of the investigated compounds.

The broad-spectrum antiprotozoal activities of the mango leaf and stem bark extracts (Núñez Sellés *et al.*, 2002) against *Entamoeba histolytica* (Tona *et al.*, 1998), *Histomonas meleagridis*, *Tetratrichomonas gallinarum* and *Blastocystis* sp (Grabensteiner *et al.*, 2008), *Plasmodium falciparum* (Awe *et al.*, 1998; Ruiz *et al.*, 2011; Zirihi *et al.*, 2005), and *Giardia lamblia* (Amaral *et al.*, 2006) agree with our herein obtained results. In addition, the reported leishmanicidal activity of several tannins against amastigotes of *Leishmania donovani* (Kolodziej *et al.*, 2001) also supports our results. Likewise, the antibacterial activity of gallotannins and/or extracts of mango kernels on different bacterial species have been also shown in several reports (El-Gied *et al.*, 2012; Engels *et al.*, 2010; Ka buki *et al.*, 2000; Rajan *et al.*, 2011; Rakholiya *et al.*, 2015; Shabani and Sayadi, 2014; Subbiya *et al.*, 2013). The antimicrobial properties of the kernel gallotannins were ascribed to its ability to intermingle with proteins, hinder the enzyme activity (Rajan *et al.*, 2011), and/or its ability to make a complex with metal ions such as iron (Engels *et al.*, 2009). Altogether, the anti-*T. vaginalis* activity of the isolated gallotannins may be attributed to either or all of the aforementioned mechanisms. *T. vaginalis* uses the iron-containing proteins lactoferrin and hemoglobin (Sehgal *et al.*, 2012) delivered by the menstruation blood (Figuroa-Angulo *et al.*, 2012). A study on the effect of iron deficiency in the host on *T. vaginalis* demonstrated changes in the parasite propagation, cytotoxicity, and immune evasion (Alvarez-Sánchez *et al.*, 2007). Therefore, iron deficiency by complexation with gallotannins may be the cause of the trophozoite cellular damage and the parasite survival inhabitation at the experimental conditions.

CONCLUSION

We are reporting here on the accumulation of gallotannins (galloylglucoses), gallic acid, and gallate derivatives in mango kernels. The procedure as described here is an easy and fast

isolation for gallotannins that could be useful for the preparation of these compounds either as a crude fraction or single pure sample from the kernels and from gallotannins-rich plant extracts including other morphological parts of *M. indica*. The present study demonstrated, to the first time, that the mango kernels along with its isolated gallotannins and ethyl gallate could be used for further studies on the development of novel preventive or therapeutic agents for the treatment of trichomoniasis. However, it is still required to achieve animal lab-work and more mechanistic studies to approve our *in vitro* finding.

CONFLICTS OF INTEREST

The authors declare that they have no conflict of interest.

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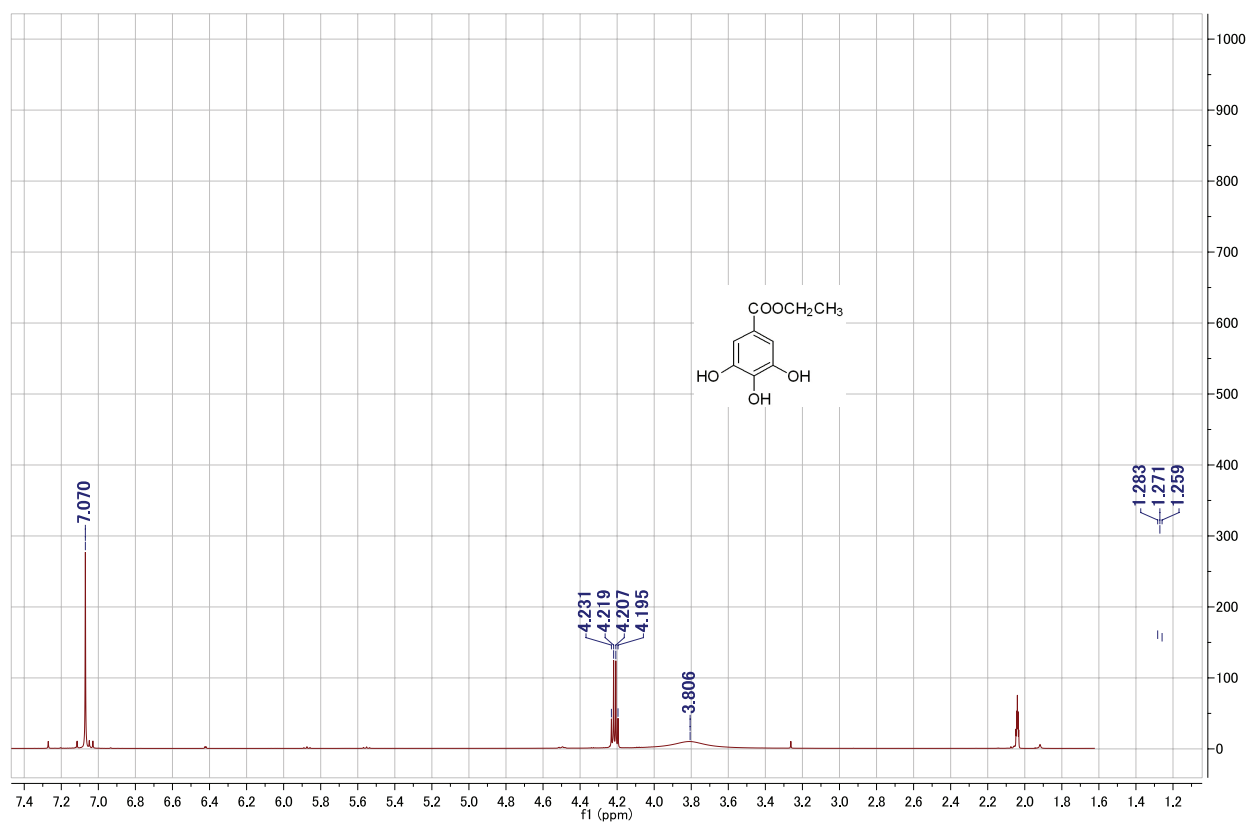
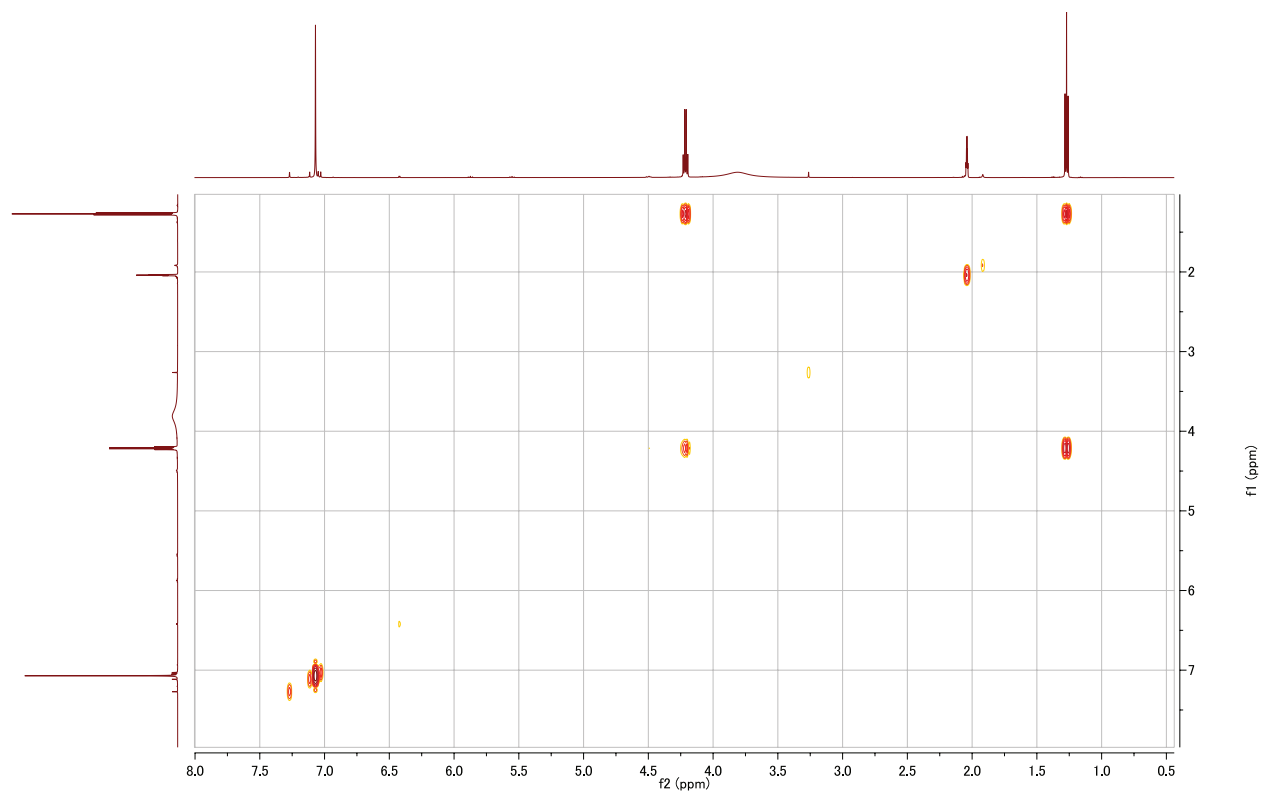
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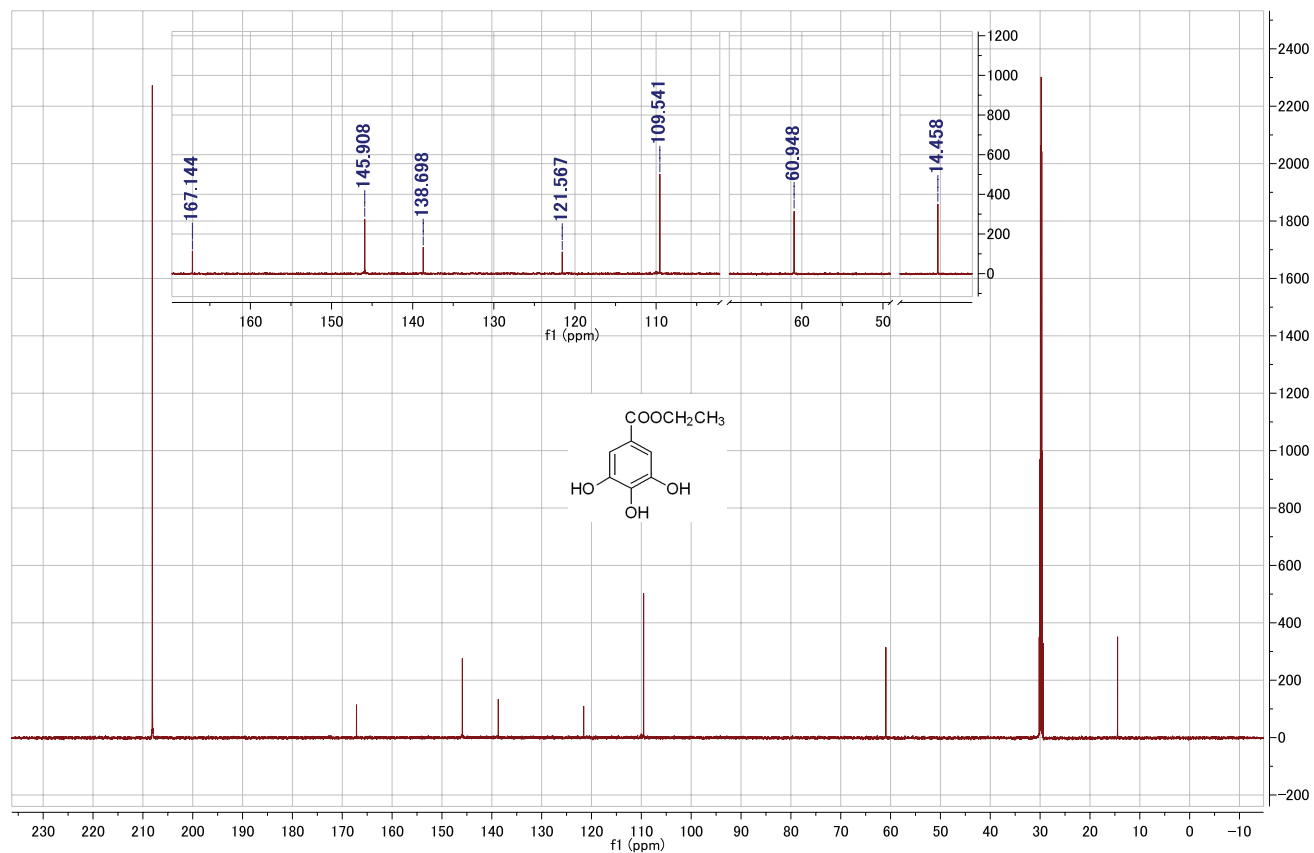
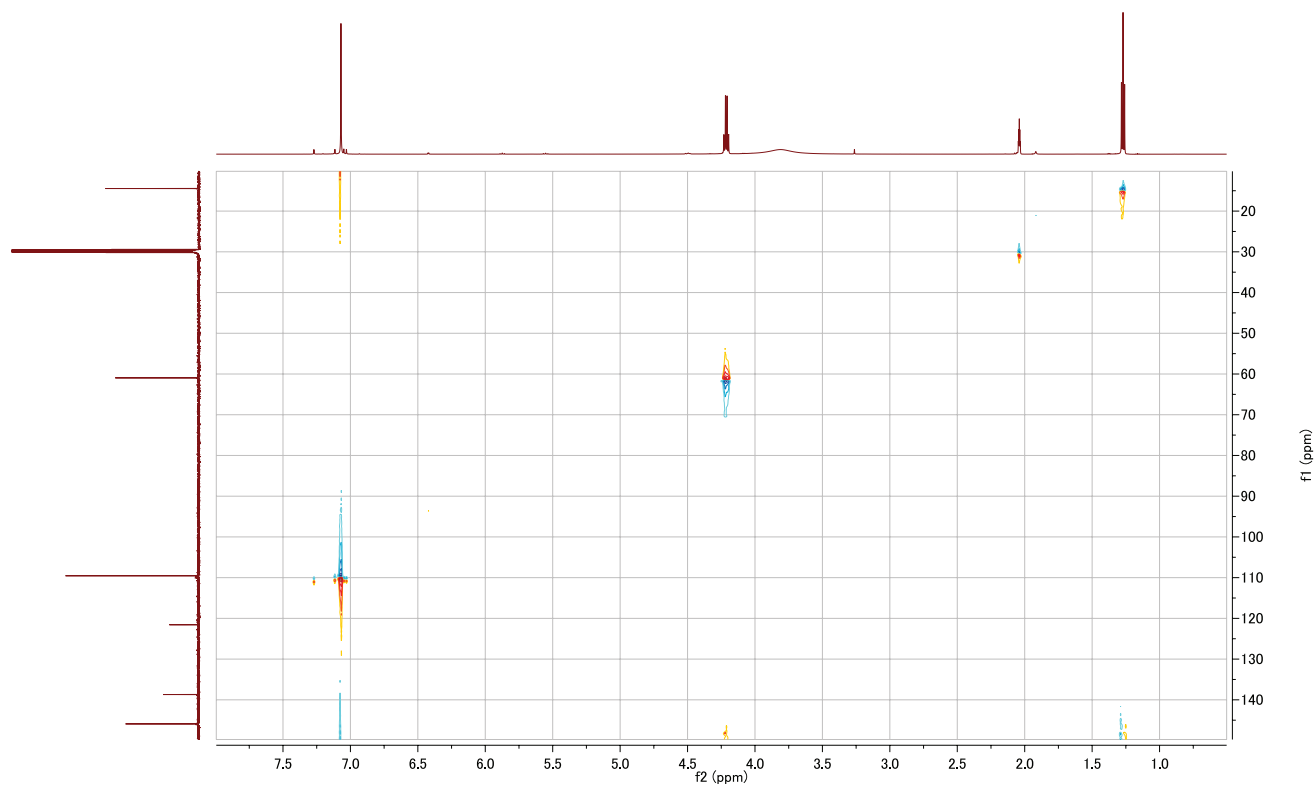
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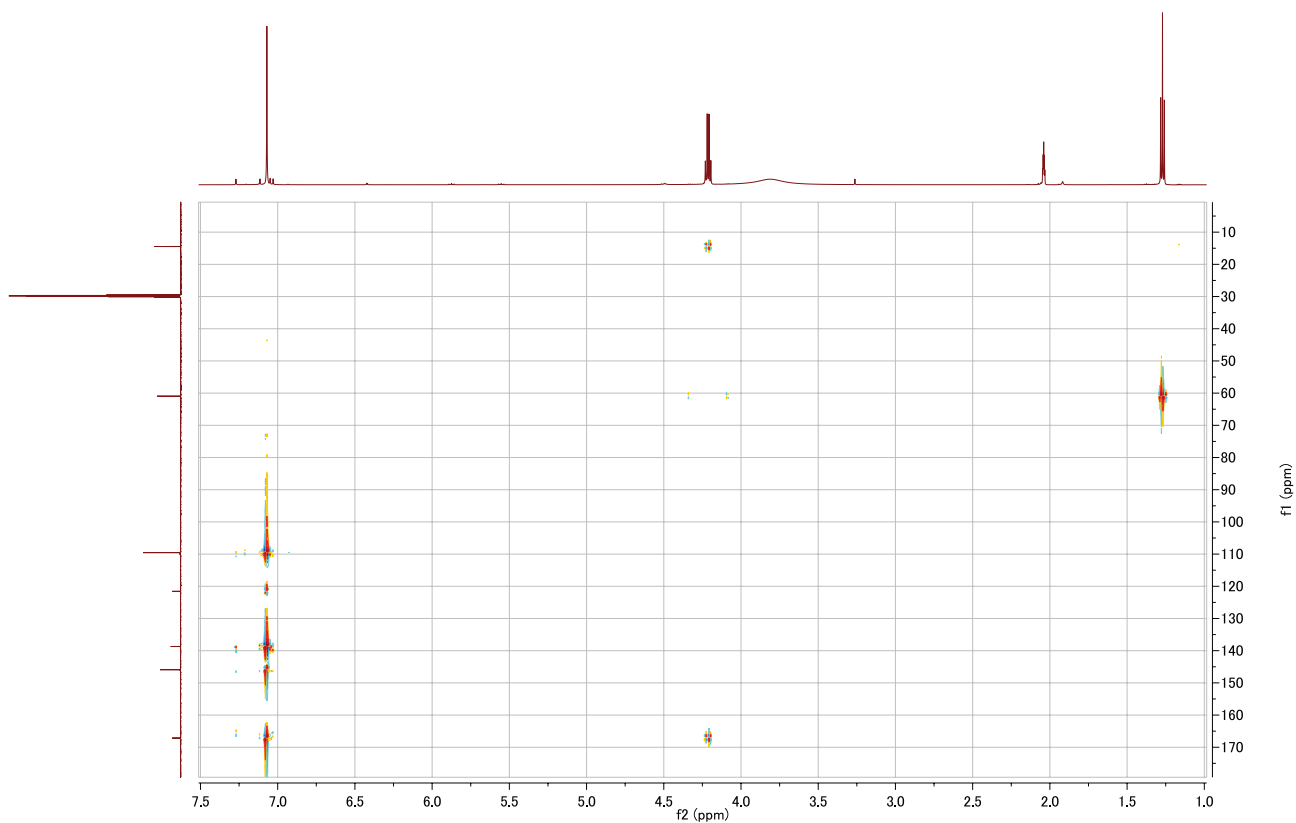
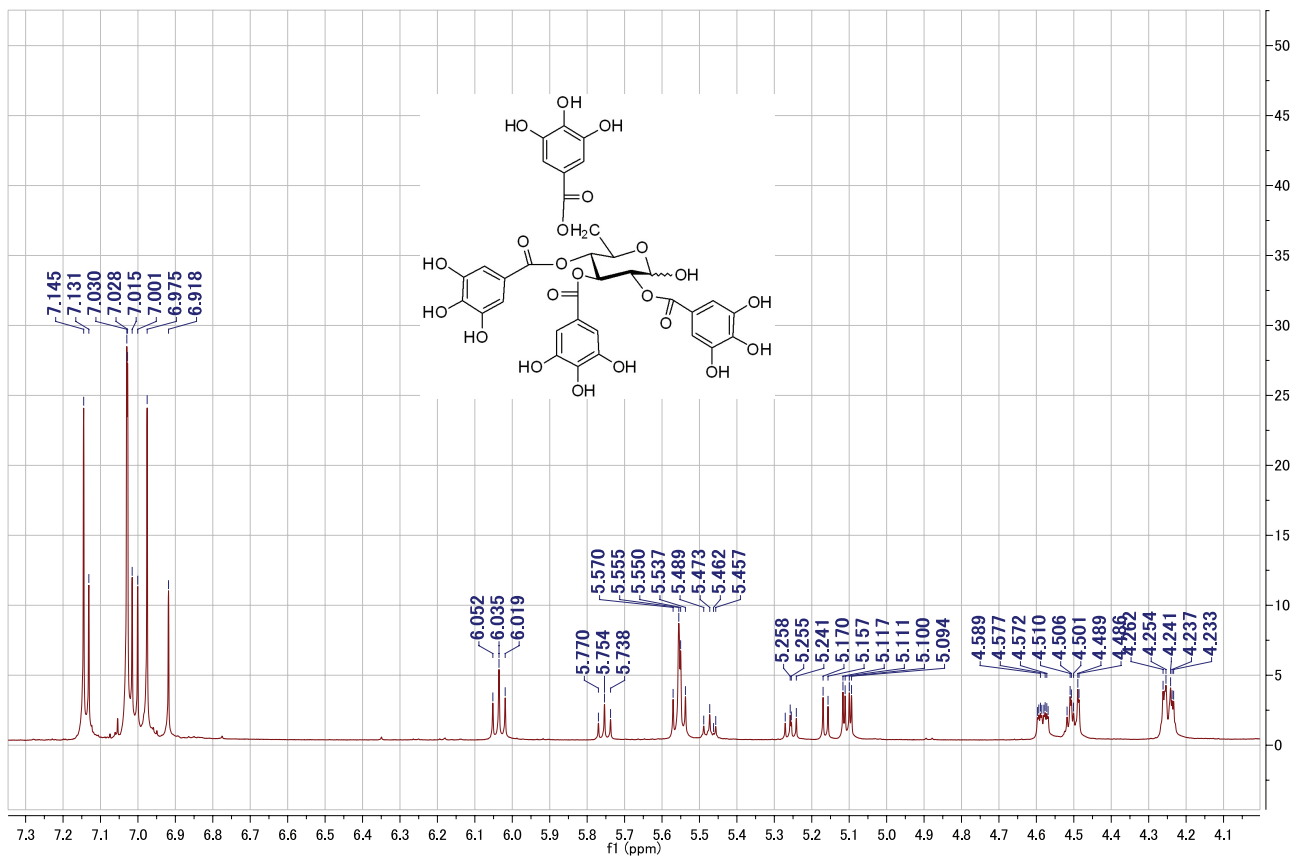
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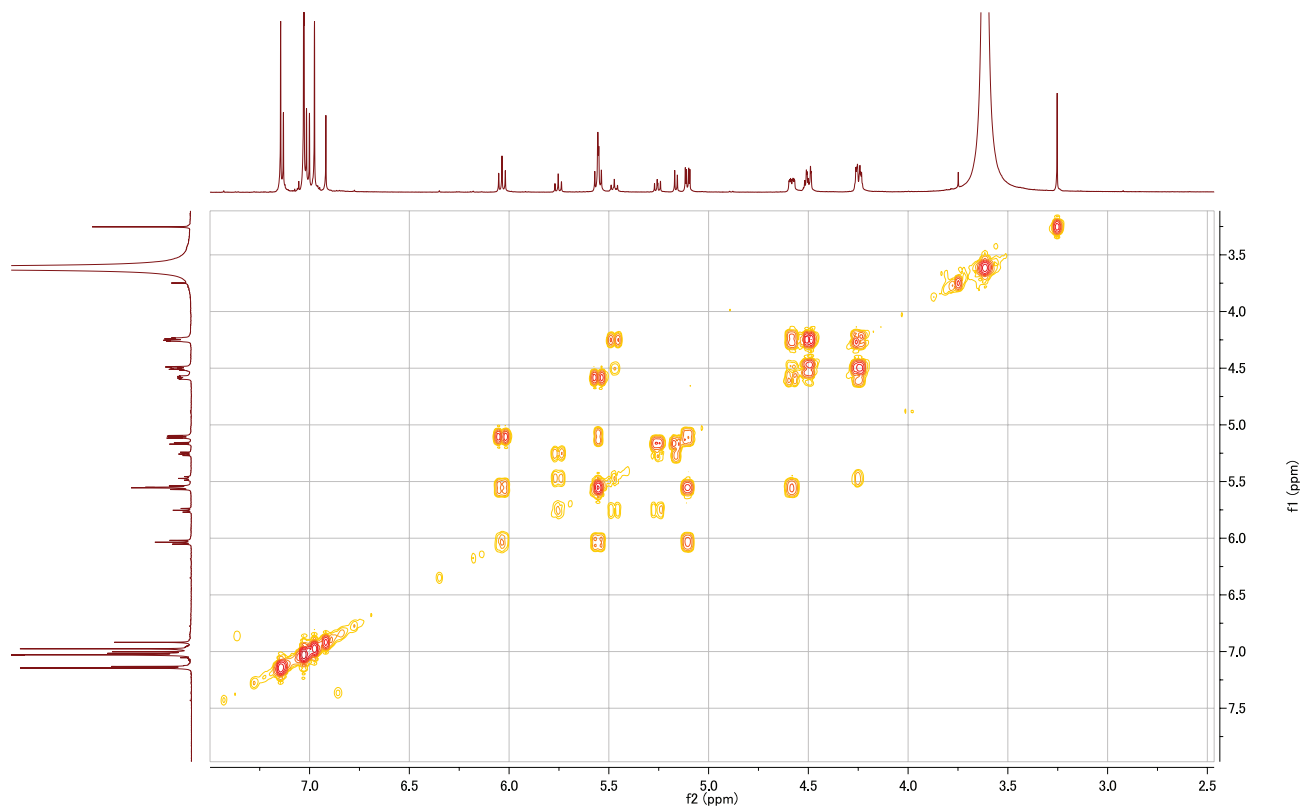
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SUPPLEMENTARY MATERIAL

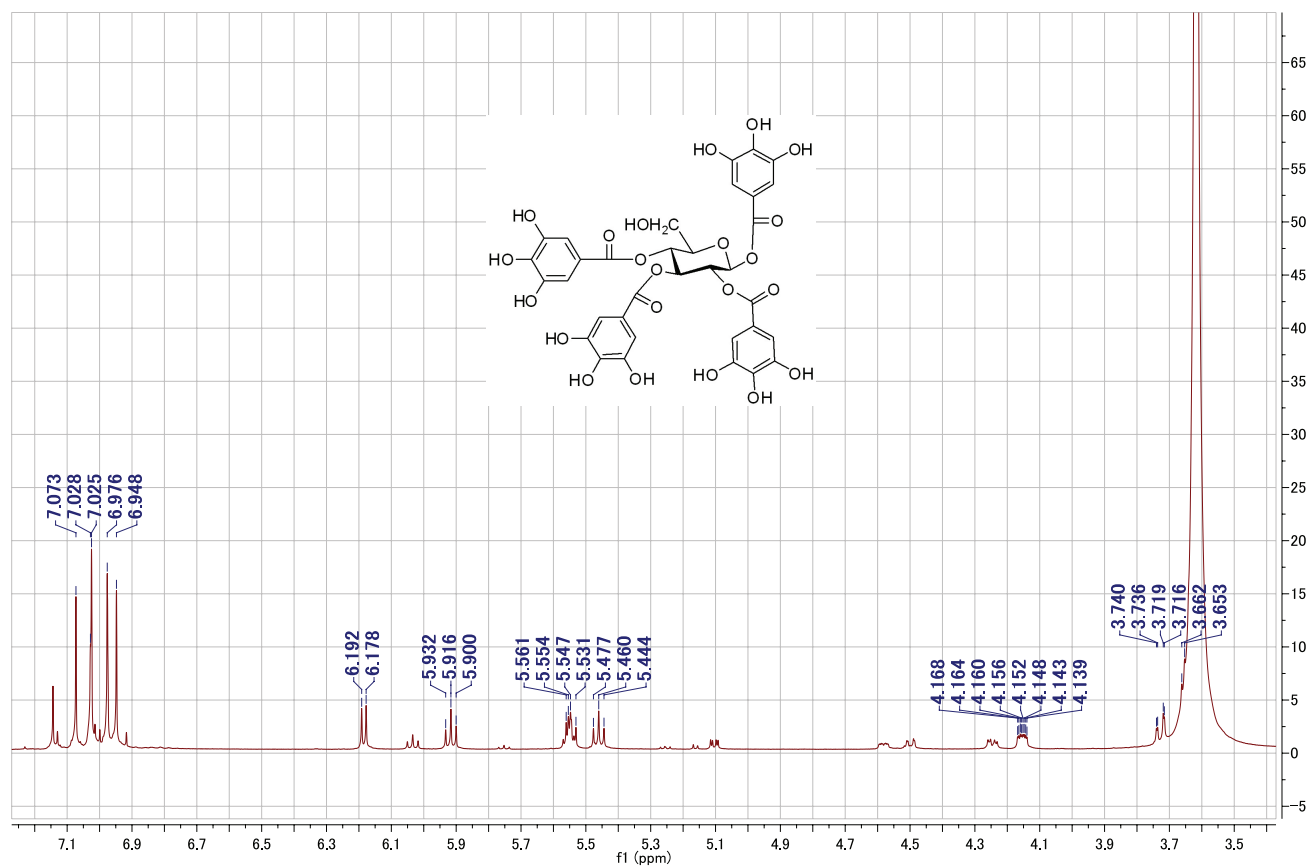
S1. ¹H NMR spectrum of 2 [600 MHz, (acetone-*d*₆ – D₂O, 9:1), 27 °C].S2. ¹H-¹H COSY spectrum of 2 [600 MHz, (acetone-*d*₆ – D₂O, 9:1), 27 °C].

S3. ^{13}C NMR spectrum of 2 [151 MHz, (acetone- d_6 – D_2O , 9:1), 27 °C].S4. HSQC spectrum of 2 [600 MHz, (acetone- d_6 – D_2O , 9:1), 27 °C].

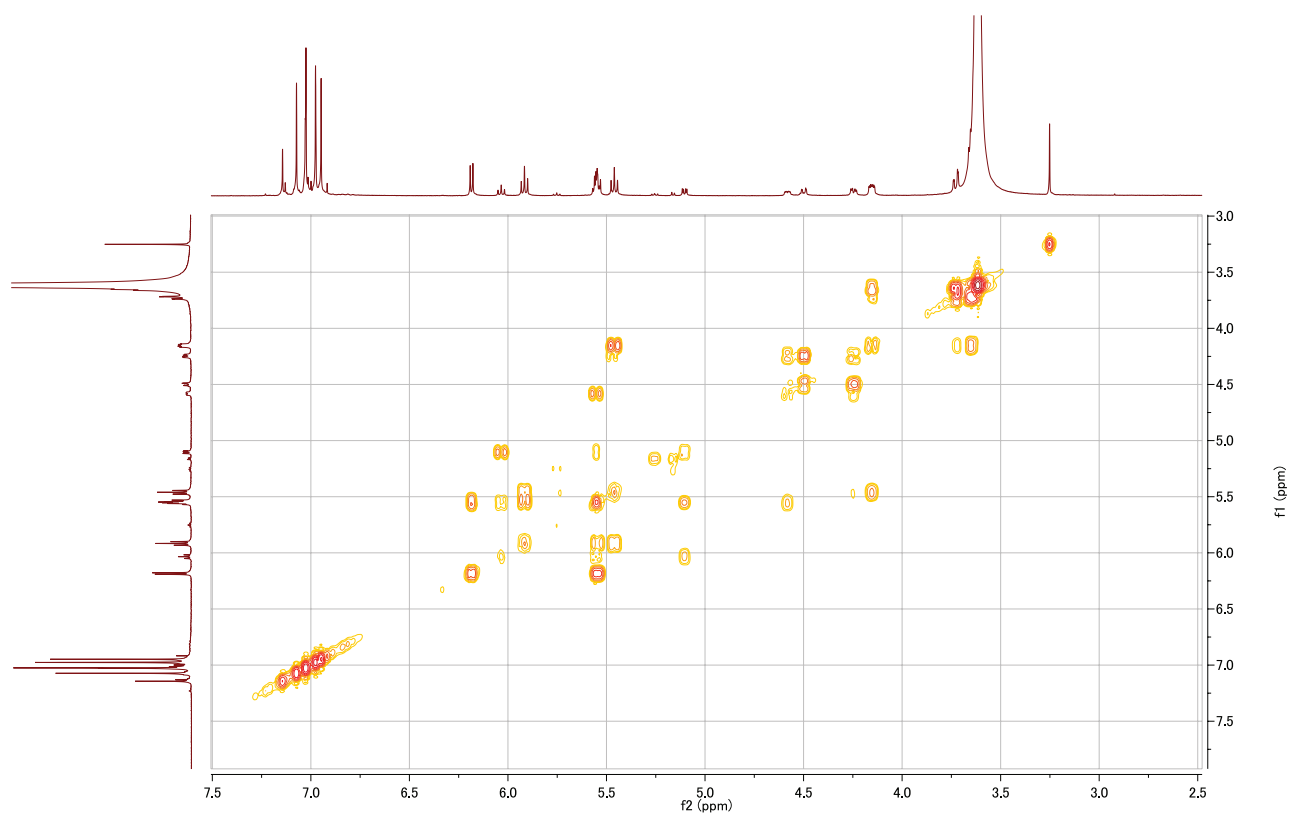
S5. HMBC spectrum of **2** [600 MHz, (acetone- d_6 - D_2O , 9:1), 27 °C].S6. 1H NMR spectrum of **3** [600 MHz, (acetone- d_6 - D_2O , 9:1), 27 °C].



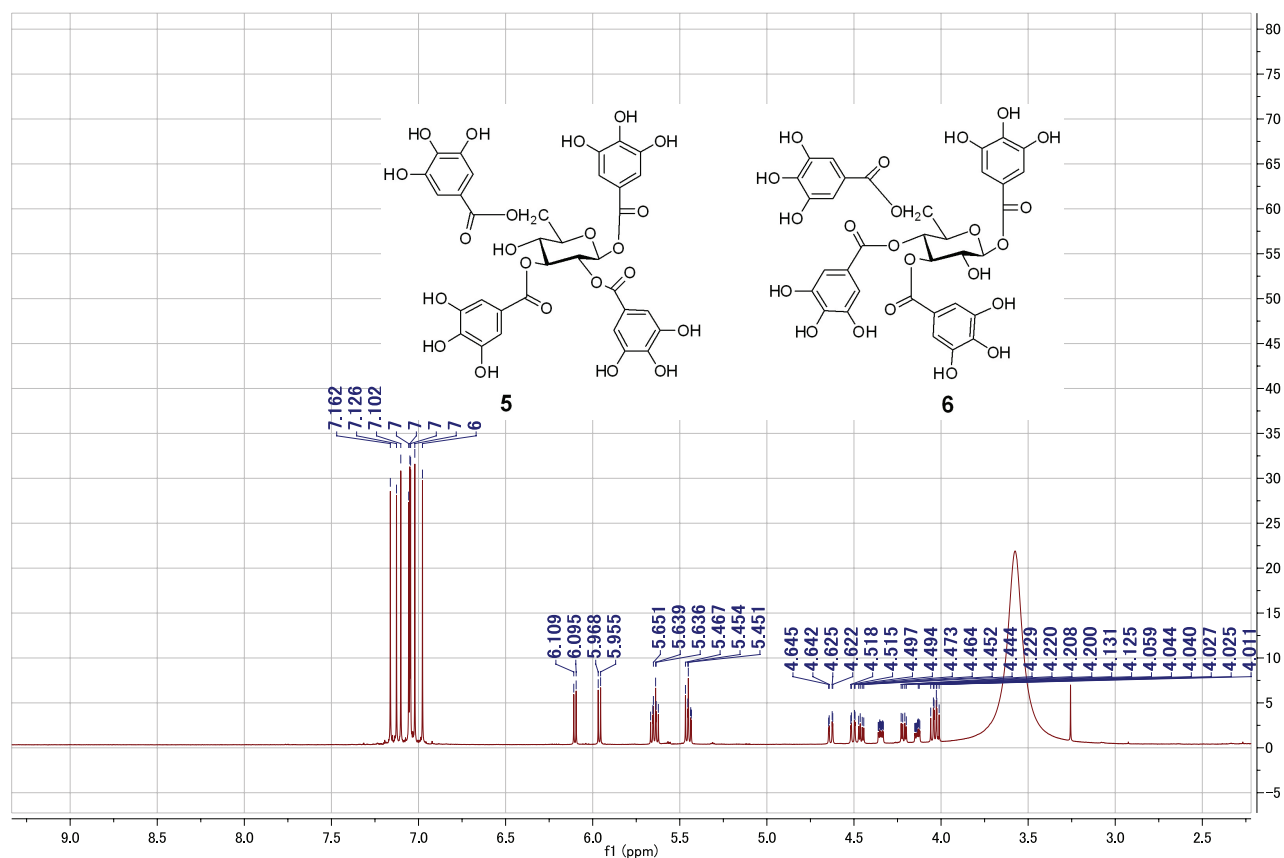
S7. ^1H - ^1H COSY spectrum of **4** [600 MHz, (acetone- d_6 - D_2O , 9:1), 27 °C].



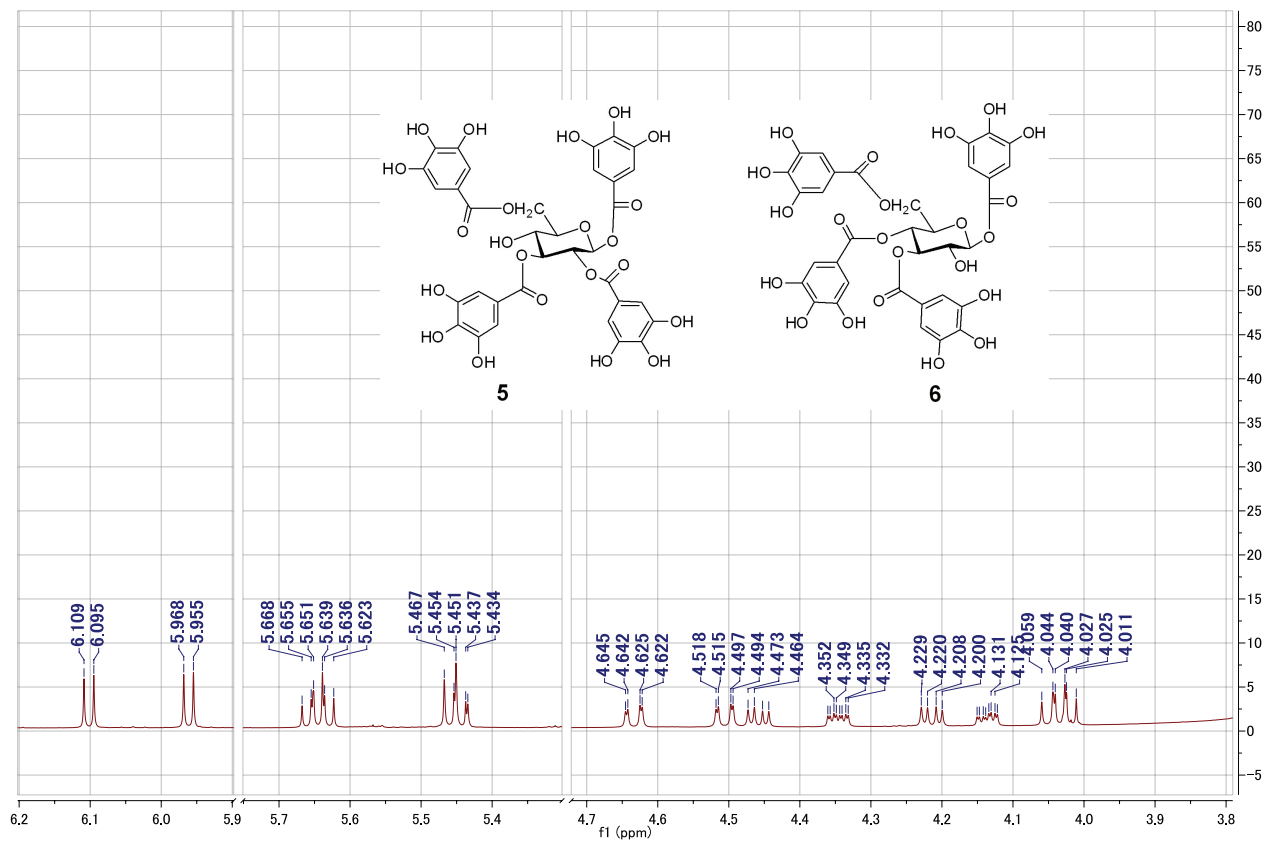
S8. ^1H NMR spectrum of **4** [600 MHz, (acetone- d_6 - D_2O , 9:1), 27 °C].



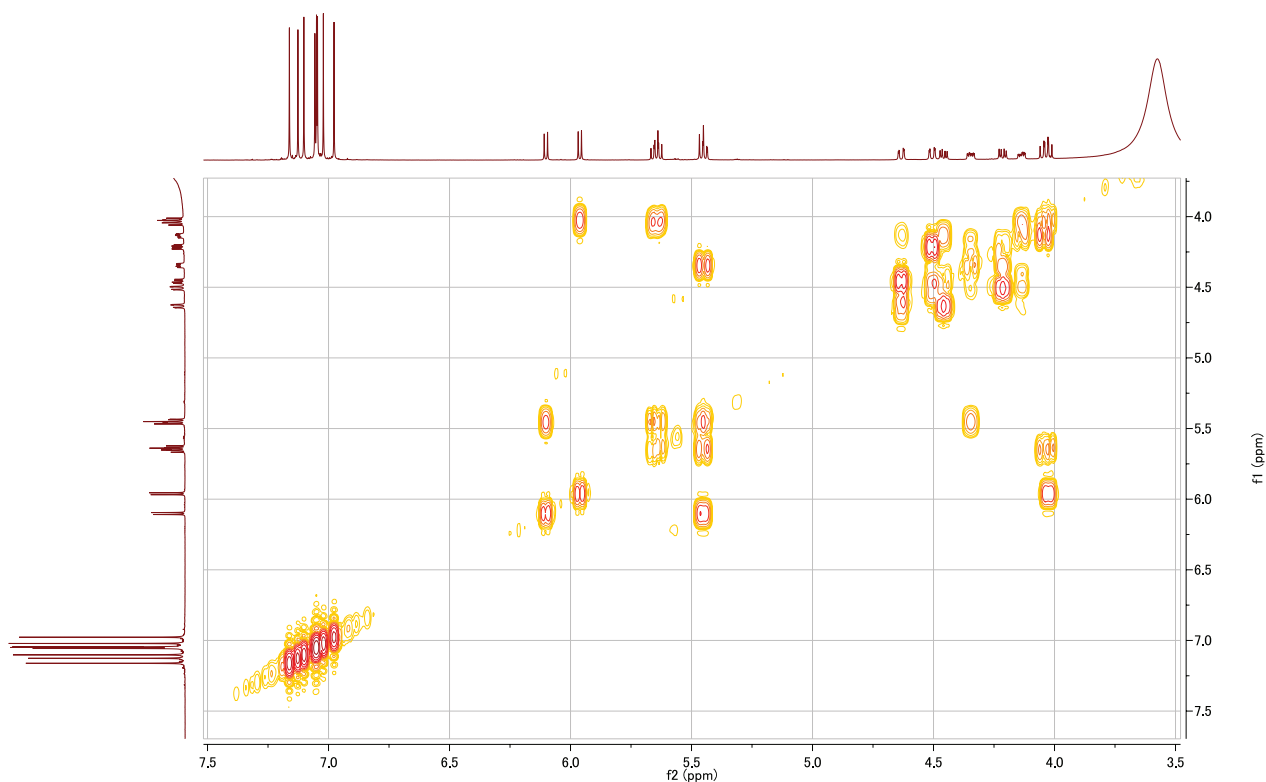
S9. ^1H - ^1H COSY spectrum of **4** [600 MHz, (acetone- d_6 - D_2O , 9:1), 27 °C].



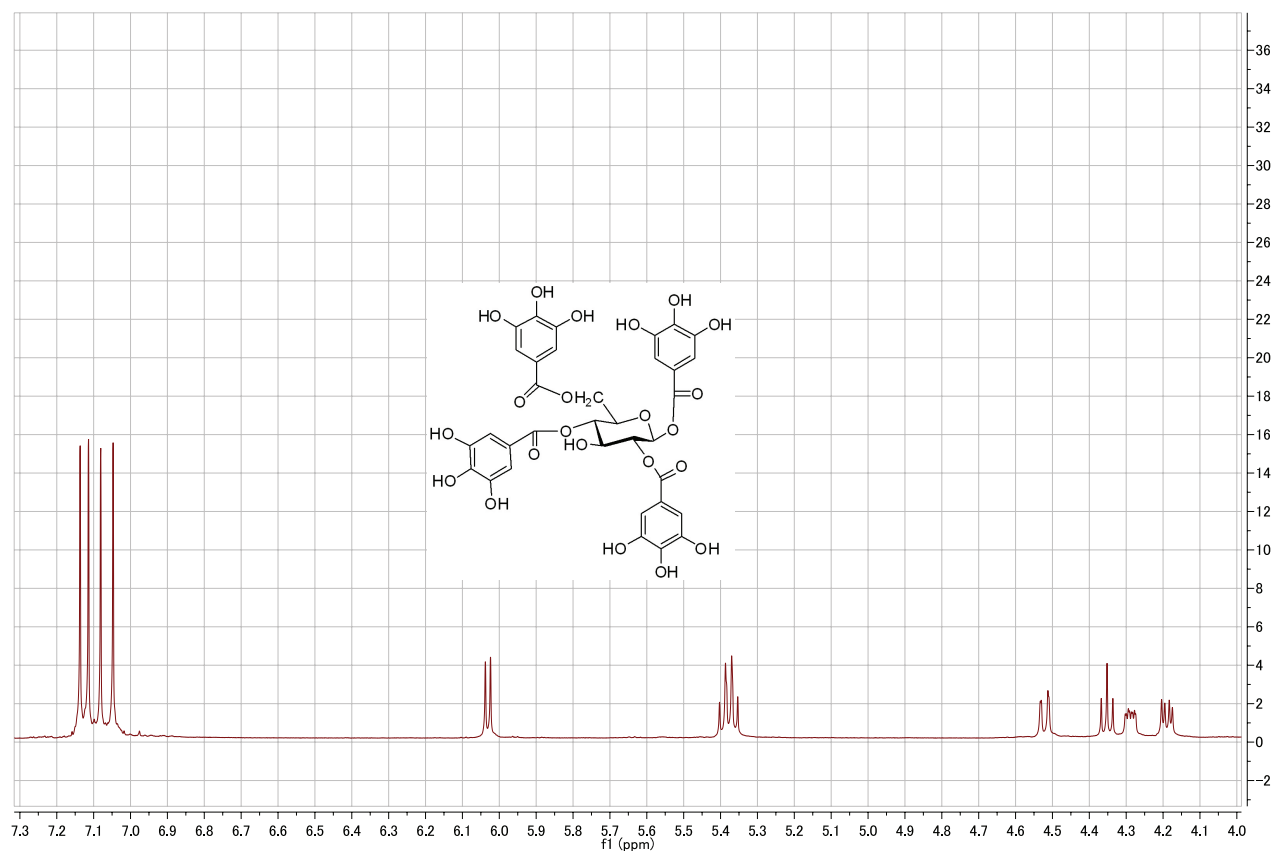
S10. ^1H NMR spectrum of mixture of **5** and **6** [600 MHz, (acetone- d_6 - D_2O , 9:1), 27 °C].



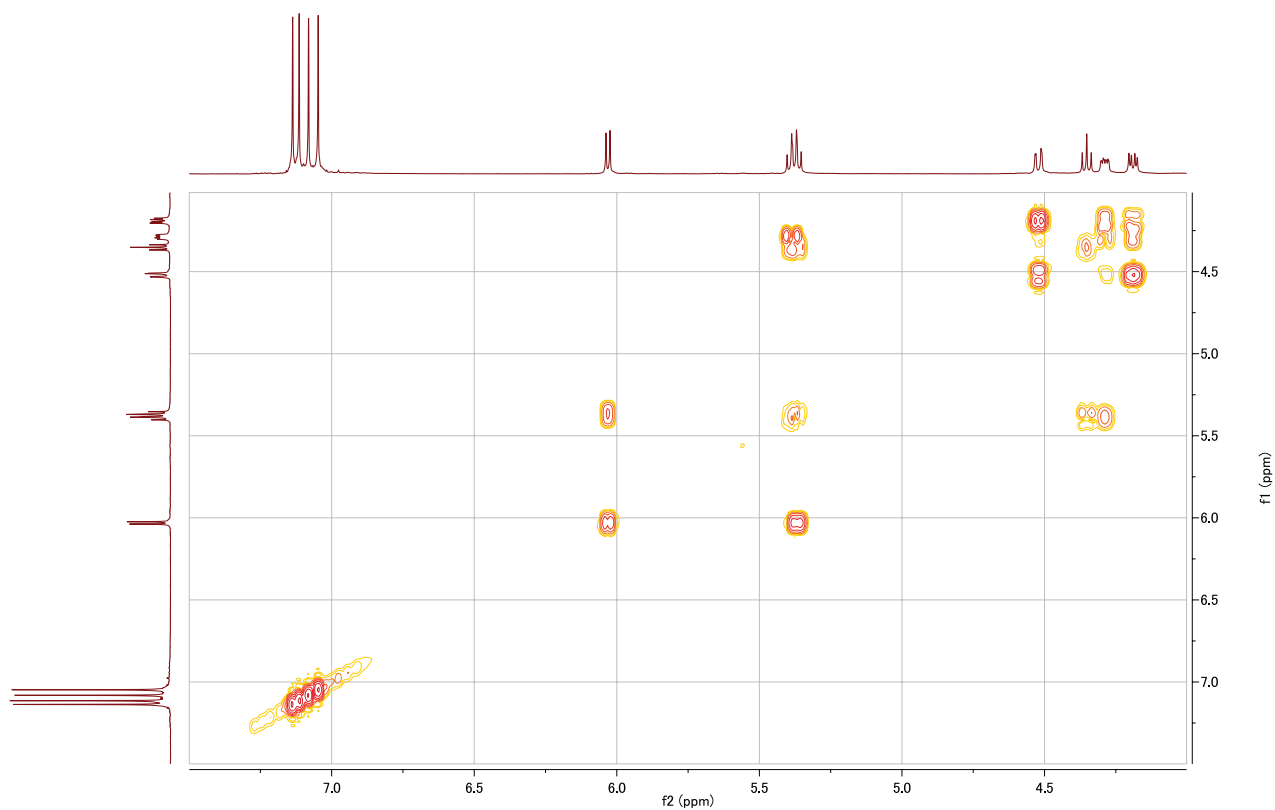
S11. Expanded sugar protons region of ^1H NMR spectrum of mixture of **5** and **6** [600 MHz, (acetone- d_6 - D_2O , 9:1), 27 $^\circ\text{C}$].



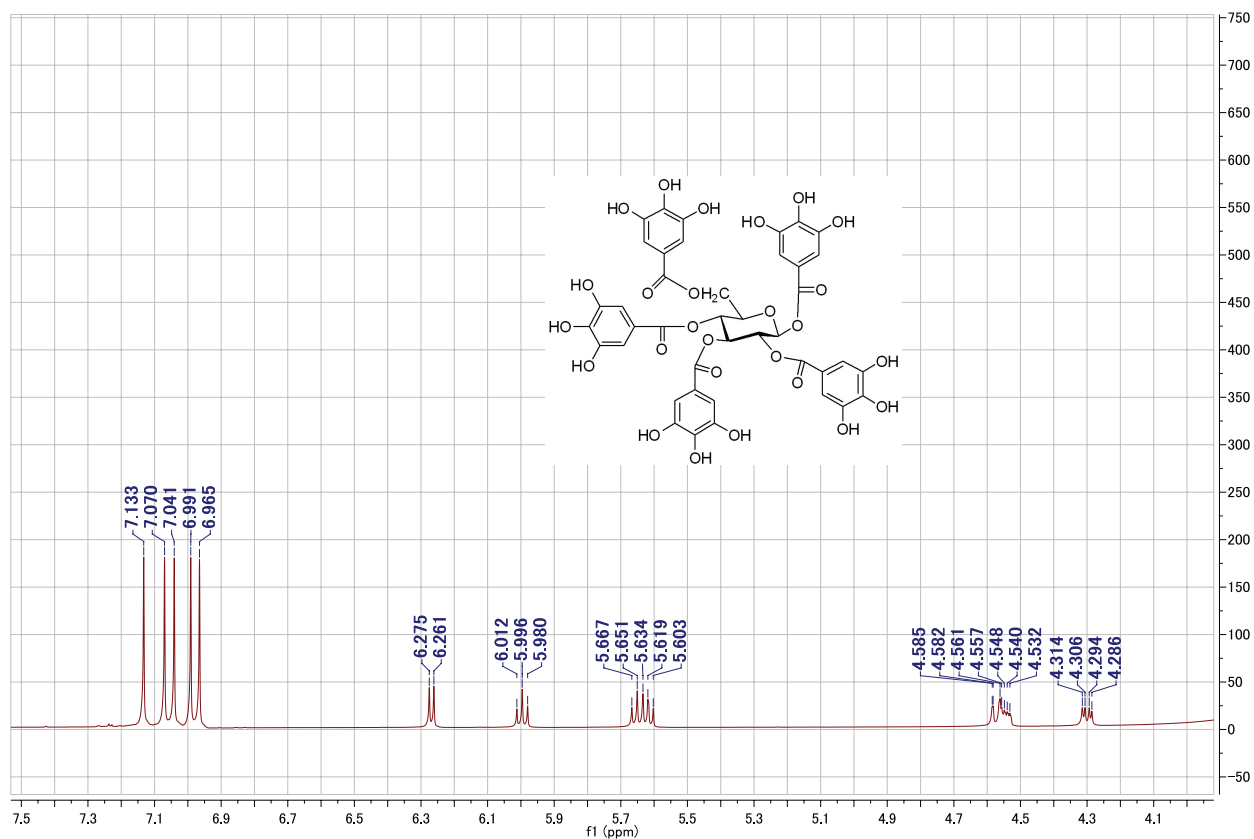
S12. ^1H - ^1H COSY spectrum of mixture of **5** and **6** [600 MHz, (acetone- d_6 - D_2O , 9:1), 27 $^\circ\text{C}$].



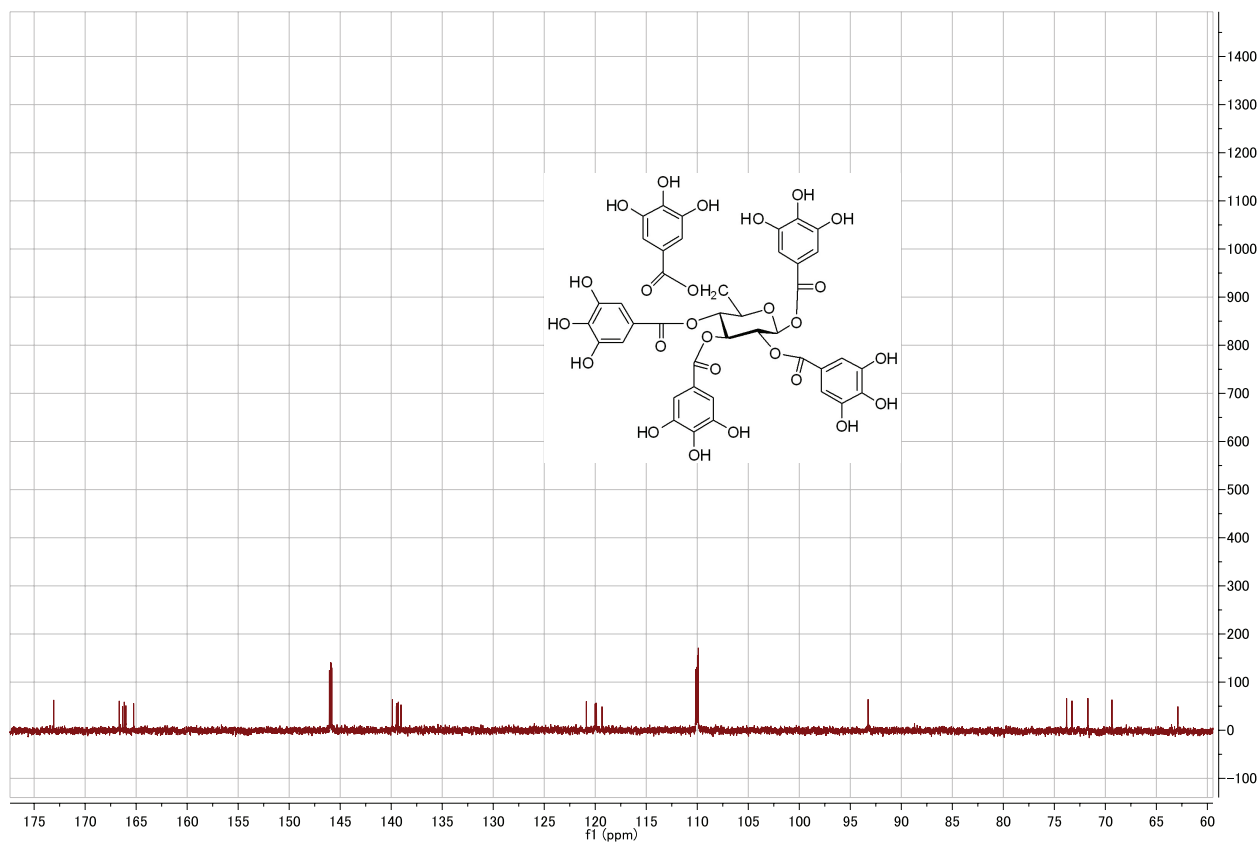
S13. ¹H NMR spectrum of 7 [600 MHz, (acetone-*d*₆ - D₂O, 9:1), 27 °C].



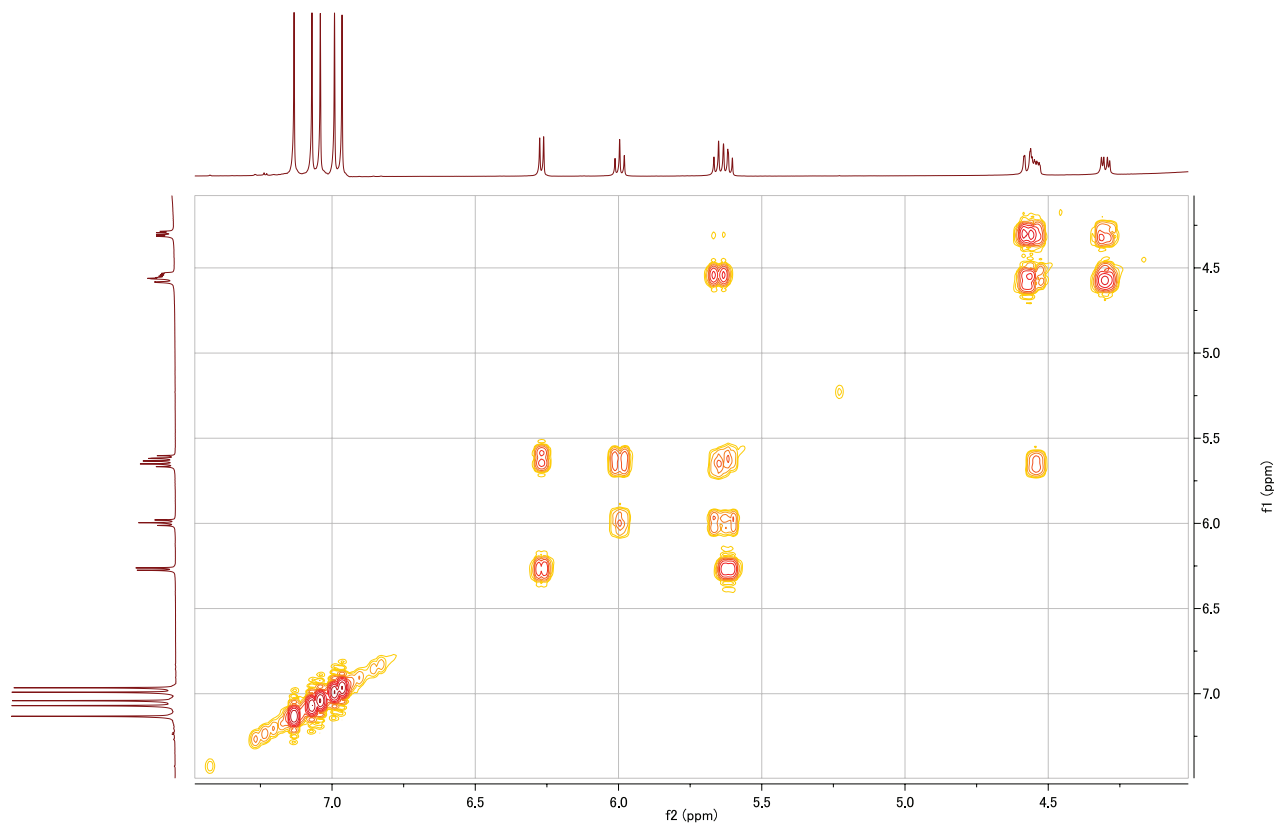
S14. ¹H - ¹H COSY spectrum of 7 [600 MHz, (acetone-*d*₆ - D₂O, 9:1), 27 °C].



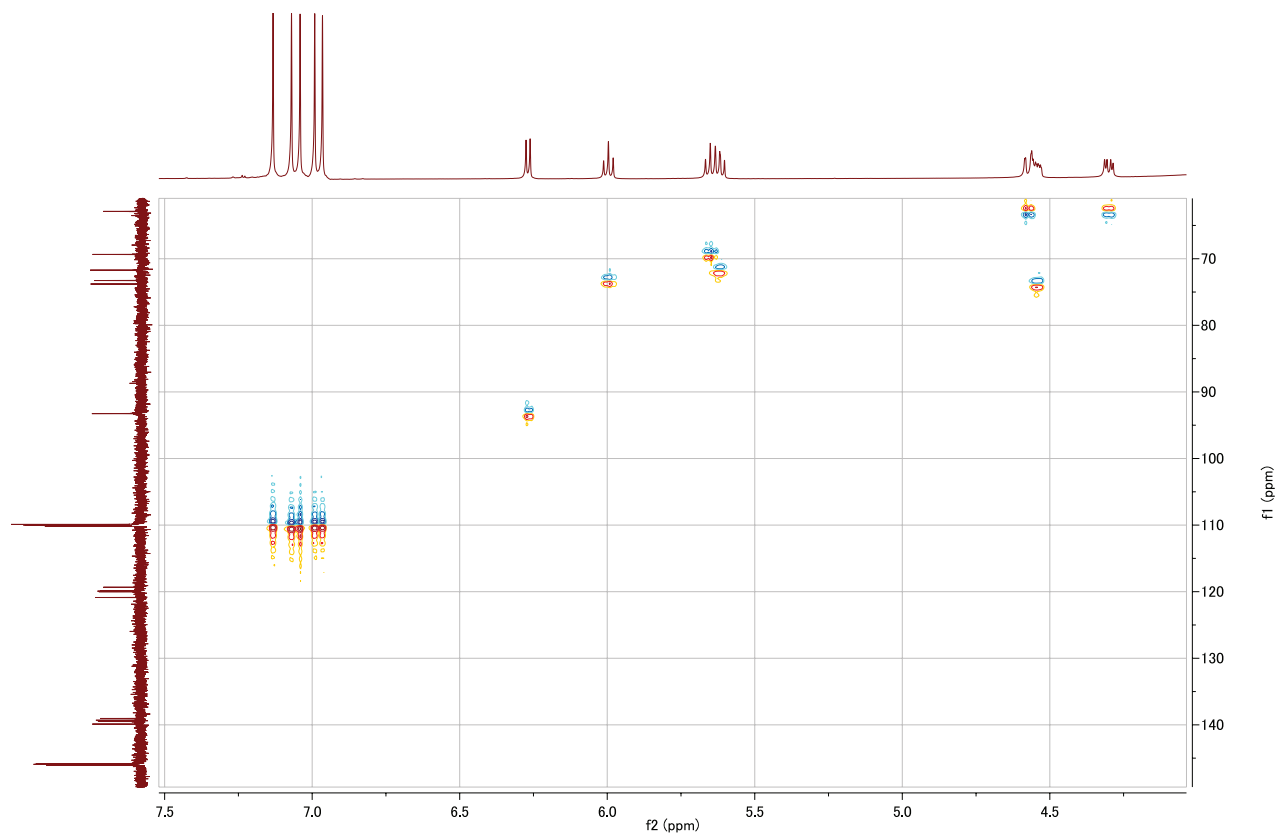
S15. ¹H NMR spectrum of **8** [600 MHz, (acetone-*d*₆-D₂O, 9:1), 27 °C].



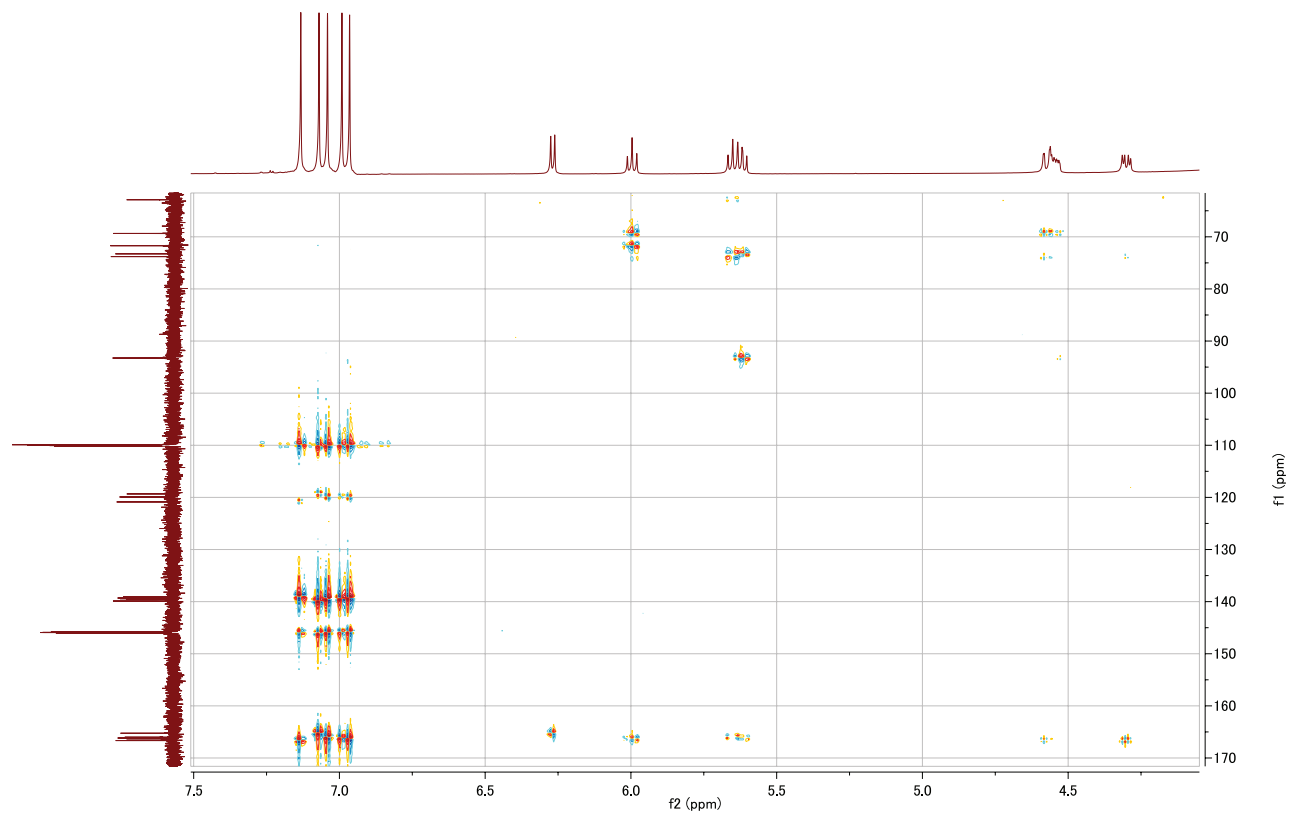
S16. ¹³C NMR spectrum of **8** [151 MHz, (acetone-*d*₆-D₂O, 9:1), 27 °C].



S17. ¹H-¹H COSY spectrum of **8** [600 MHz, (acetone-*d*₆-D₂O, 9:1), 27 °C].



S18. HSQC spectrum of **8** [600 MHz, (acetone-*d*₆-D₂O, 9:1), 27 °C].



S19. HMBC spectrum of **8** [600 MHz, (acetone- d_6 -D₂O, 9:1), 27 °C].