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# 3D-QSAR, molecular docking, and dynamics simulation of quinazoline-phosphoramidate mustard conjugates as EGFR inhibitor

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## **ABSTRACT**

To develop novel and more potent quinazoline-phosphoramidate mustard conjugates as epidermal growth factor receptor (EGFR) inhibitor, three-dimensional quantitative structure-activity relationship [comparative molecular field analysis (CoMFA) and comparative molecular similarity indices analysis (CoMSIA)] combined with molecular docking were performed. A series of 13 compounds in the training set gave  $q^2$  values of 0.577 and 0.537, as well as  $r^2$ values of 0.926 and 0.921 for CoMFA and CoMSIA models, respectively. The contour maps that were produced by the CoMFA and CoMSIA models revealed that steric, electrostatic, and hydrophobic fields were crucial in the inhibitory activity of quinazoline-phosphoramidate derivatives. Based on the CoMFA and CoMSIA models, several novel EGFR inhibitors were designed, which established crucial interactions at the ligand binding domain of EGFR. Nearly, 100 ns MD simulation indicated the stability of the designed compounds at 100 ns, while molecular mechanics-Poisson Boltzmann surface area calculation showed that the designed compound had a higher affinity than that of the parent compound.

## INTRODUCTION

The non-small cell lung cancer (NSCLC) represents the most prevalent lung cancer worldwide with less than 20% of 5-year survival rate after diagnosis (Chen et al., 2018; Gaber et al., 2018; Hao et al., 2018). The epidermal growth factor receptor (EGFR) is a member of the protein kinase family, which is a clinically validated target for NSCLC treatment. EGFR is involved in various cellular signaling cascades, which are crucial in cell growth, proliferation, survival, and migration. Due to its crucial role, there have been continuous efforts to find a small molecule that is able to inhibit EGFR, particularly for the NSCLC treatment. Erlotinib and Gefitinib were considered as the first-generation of EGFR inhibitors, which were used for the treatment of NSCLC (Bonomi, 2003; Vansteenkiste, 2004).

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However, it was known that the T790M point mutation in the EGFR, which was the substitution of Thr790 with Met residue, had induced acquired resistance after a median of 10-14 months to most NSCLC patients of first-generation EGFR inhibitors (Kobayashi et al., 2005). Furthermore, Afatinib and canertinib (Ou, 2012), two of several second-generation EGFR inhibitors, were approved by FDA for the treatment of metastatic NSCLC patients. However, the nonselective inhibition against wild type of EGFR has limitations in their clinical use (Gaber et al., 2018). The third-generation EGFR inhibitors such as rociletinib and avitinib (Walter et al., 2013) were developed; however, it was reported that the hyperglycemia was observed in NSCLC patients who used Rociletinib (Chen et al., 2018; Yver, 2016).

Lin et al. (2017) designed and synthesized a series of phosphoramide mustard functionality, which was incorporated into the quinazoline scaffold, and their potential as EGFR inhibitors for the treatment of lung cancer was investigated. It was found that the designed compound could inhibit EGFR with IC<sub>50</sub> at the nanomolar range and showed no acute toxicity to mice at a single dose up to 900 mg/kg. It was concluded that the

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designed compound posed a potential as EGFR inhibitor. Based on these results, the present study was aimed to build a model of three-dimensional quantitative structure-activity relationship (3D-QSAR) including comparative molecular field analysis (CoMFA) (Cramer *et al.*, 1988) and comparative molecular similarity indices analysis (CoMSIA) (Klebe *et al.*, 1994) of quinazoline–phosphoramidate mustard conjugates. Using the built model, a novel compound was proposed, and molecular docking and molecular dynamics simulation were then used to check the conformational stability of the newly proposed compound in the binding site of EGFR.

#### MATERIALS AND METHODS

17 compounds of quinazoline–phosphoramidate mustard conjugates were selected based on the literature study (Lin *et al.*, 2017). Based on a random selection, the compounds were grouped into a training set (13 compounds) and a test set (four compounds) (Table 1) by considering structural diversity and distribution of biological data. The inhibitory activity [IC $_{50}$  (nM)] values were converted to the logarithmic scale (pIC $_{50}$ ), where pIC $_{50}$  = -Log IC $_{50}$  (Table 2). The training set was used to develop 3D-QSAR, including CoMFA and CoMSIA, while the test set was used to evaluate the model's validity. All structure sketching, optimization, and modeling were conducted with SYBYL-2.1 program package (Tripos, Inc.).

#### Minimization and alignment

Each compound was sketched and energy-minimized using Tripos molecular mechanic force field and Powell method (Clark et al., 1989), while charges were assigned using Gasteiger–Huckel method (Purcell and Singer, 1967) in the SYBYL-X 2.1. The minimization was conducted using energy convergence threshold and maximum iterations of 0.5 kcal/mol and 1,000 cycles, respectively. Superpositioning of ligands was conducted based on the core structure, N-phenylquinolin-4-amine, by employing compound 12 as a template, since it was the most active compound. Figure 1 depicts the superimposed structures of aligned molecules.

#### CoMFA and CoMSIA

Both CoMFA and CoMSIA were developed employing SYBYL-X 2.1 (Tripos, Inc., USA). CoMFA steric field was developed based on van der Waals interaction using Lennard-Jones potential, while CoMFA electrostatic field was based on Coulombic potential. Both CoMFA fields were generated at each lattice point of a grid box of 2.0 Å. Cut-off values of 30 kcal/mol were set for both steric and electrostatic fields (Ståhle and Wold, 1988).

In CoMSIA, a distance-dependence Gaussian-type of the physicochemical property has been adopted to avoid any singularities at the atomic position (Klebe *et al.*, 1994). Similar

Table 1. Chemical structure of quinazoline-phosphoramidate mustard conjugates and their inhibitory activities.

$$Cl$$
 $O$ 
 $HN$ 
 $R_2$ 
 $N-P-O$ 
 $NH_2$ 
 $R_1$ 
 $N$ 

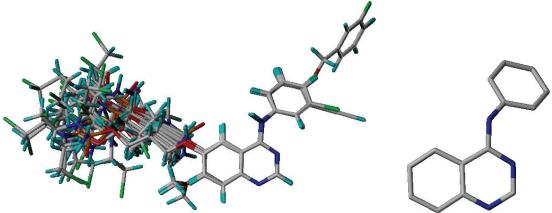
Comp -		Substituent				
	R <sub>1</sub>	R <sub>2</sub>	n	X	pIC <sub>50</sub>	
1*/10a	Н	3-Br	0	Single bond	8.0362	
2/10b	Н	3-Br	2	O	8.3665	
3/10c	Н	3-Br	3	O	8.5229	
4/10d	Н	3-Br	4	O	8.1308	
5*/10e	Н	3-Br	5	O	8.6383	
6*/10f	Н	3-C≡C	4	O	7.9586	
7/10g	Н	3-Cl-4-F	4	O	8.2218	
8/10h	Н	3-Cl-4-[(3-F-benzyl)oxo]phenyl	4	O	7.5686	
9/10i	Н	3-C≡C	3	O	8.1427	
10/10j	Н	3-Cl-4-F	3	O	8.2441	
11*/10k	Н	3-Cl-4-[(3-F-benzyl)oxo]phenyl	3	O	7.1249	
12/101	OMe	3-Br	3	O	9.5229	
13/10m	OMe	3-Cl-4-F	3	O	9.301	
14/10n	OMe	3-Cl-4-[(3-F-benzyl)oxo]phenyl	3	O	7.6576	
15/10o	Н	3-Br	3	CONH	8.4815	
16/10p	Н	3-Cl-4-F	3	CONH	8.0605	
17/10q	Н	3-Cl-4-[(3-F-benzyl)oxo]phenyl	3	CONH	7.585	

<sup>\*</sup>Test set.

G	A storal mIC (mIM)	Predicted pIC <sub>50</sub> (nM)				
Comp	Actual pIC <sub>50</sub> (nM)	CoMFA	Residual	CoMSIA	Residual	
1*	8.0362	8.3865	-0.3503	8.3710	-0.3348	
2	8.3665	8.4324	-0.0659	8.5155	-0.1490	
3	8.5229	8.506	0.01690	8.4959	0.0270	
4	8.1308	8.1592	-0.0284	8.1159	0.0149	
5*	8.6383	8.6469	-0.0086	8.4700	0.1683	
6*	7.9586	8.2951	-0.3365	8.0713	-0.1127	
7	8.2218	8.1785	0.0433	8.1626	0.0592	
8	7.5686	7.5567	0.0119	7.5942	-0.0256	
9	8.1427	7.9883	0.1544	7.9287	0.2140	
10	8.2441	8.3451	-0.1010	8.4419	-0.1978	
11*	7.1249	7.5780	-0.4531	7.5708	-0.4459	
12	9.5229	9.5269	-0.0040	9.4847	0.0382	
13	9.3010	9.2933	0.0077	9.2592	0.0418	
14	7.6576	7.6575	0.0001	7.6443	0.0133	
15	8.4815	8.3488	0.1327	8.3475	0.1340	
16	8.0605	8.1586	-0.0981	8.1859	-0.1254	
17	7.5850	7.6546	-0.0696	7.6296	-0.0446	

**Table 2.** The observed pIC<sub>50</sub>s and predicted pIC<sub>50</sub>s of the training and test set molecules.

<sup>\*</sup>Test set molecules



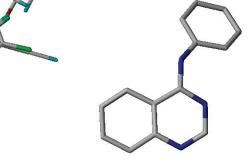


Figure 1. 3D-QSAR structure superposition and alignment of the training set (left) and common substructure used for alignment (right).

standard parameters and no arbitrary cutoff limits constructed for CoMFA field calculation were used for the calculation of CoMSIA field including steric (S), electrostatic (E), hydrophobic effects (H), and hydrogen bond donor (HBD) (D) and hydrogen bond acceptor (HBA) (A).

The partial least squares method with leave-one-out cross-validation was employed to correlate the CoMFA electrostatic and steric fields and CoMSIA electrostatic, steric, HBD, HBA, and hydrophobic properties, each with EGFR inhibitory activity (Bush Bruce and Nachbar, 1993). The value of filtering (σ) column was set to lower than 2.0 kcal/mol to improve the signal-to-noise ratio. The optimal number of principal component (ONC) was obtained by applying LOO cross-validation, which was then utilized to derive the final CoMFA and CoMSIA models. The following equation was utilized to calculate the cross-validation coefficient of  $q^2$  value:

$$q^2 = 1 - \frac{\sum (y - \hat{y})^2}{\sum (y - \bar{y})^2}$$

Where v and  $\hat{v}$  are observed and predicted activities of compound i, respectively, and  $\bar{y}$  is the average observed activity of the compound in the training set. The best QSAR model was justified on the basis of high  $q^2$ , conventional correlation coefficient  $R^2$  values ( $q^2 > 0.50$  and  $R^2 > 0.60$ ), low standard error estimation (SEE), and an optimal number of component values.

#### Molecular docking

The Surflex-Dock module in SYBYL was utilized to perform molecular docking to clarify the binding mode of the compounds. The X-ray crystal structure of EGFR that was cocrystallized with Erlotinib (PDB entry code: 1M17, resolution: 2.6 Å) was taken from the RCSB Protein Data Bank. The protein structure was prepared by removing water molecules and cognate ligand and adding polar hydrogen atoms. Protomol was generated based on ligand mode, which represented a three-dimensional space in which ligand make potential interaction with every binding site.

Automatic docking was applied for molecular docking. Other parameters were left at default. Total-score of Surflex-Dock, which was expressed in the negative logarithm of the dissociation constant,  $-\log 10~(K_{\rm d})$ , was used to represent binding affinities. The docked conformation of the ligand was generated after docking, where those with the highest scores were selected as the docking results. Each compound was energetically minimized using the Tripos force field and the Powell algorithm with a convergence criterion of 0.05 kcal/mol A° and Gasteiger–Huckel charges.

## Molecular dynamics simulation and MM-PBSA calculation

Molecular dynamics (MD) simulation of 100 ns was performed for Compound 12 (Comp12), RA1, RA2, RA3, and RA4, each complexed with EGFR. The details of MD protocol follow our previous work (Arba *et al.*, 2018a). Molecular mechanics-Poisson Boltzmann Surface Area (MM-PBSA) (Kollman *et al.*, 2000) calculation was performed to predict the binding free energy of Comp12, RA1, RA2, RA3 and RA4 to EGFR. The MM-PBSA calculation was performed using 1,500 snapshots taken from 85 to 100 ns. Details of MM-PBSA protocol were explained in Kollman *et al.* (2000), Arba *et al.* (2018b) and Arba *et al.* (2018c).

# RESULTS AND DISCUSSION

#### CoMFA model

The quinazoline–phosphoramidate mustard conjugates were utilized to conduct the CoMFA study. The CoMFA model calculated from the training set exhibited good cross-validated correlation coefficient with  $q^2 = 0.577$ ,  $r^2 = 0.982$ , F = 166.591, and SEE = 0.09, with three ONC. The external validation of the CoMFA model using the test set showed good predicted correlation

coefficient  $r^2_{\text{pred}} = 0.926$ , which indicated the predictive ability of the model. Table 2 shows the actual and predicted  $\text{pIC}_{50}$  as well as the residuals of the training and test set compounds, while Table 3 shows the statistical parameters associated with the CoMFA model. The steric and electrostatic field contributions were found to be 37.4:62.6, which indicated the significant contributions of both fields on ligand-receptor interaction.

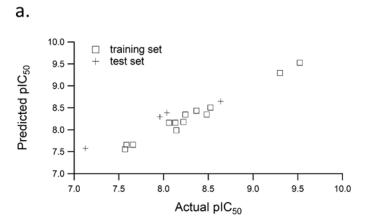
#### CoMSIA model

Compared to CoMFA, CoMSIA defines five interaction fields, i.e., steric, electrostatic, hydrophobic, HBD, and HBA. The final CoMSIA model gave cross-validated correlation coefficient  $q^2 = 0.537$ ,  $r^2 = 0.964$ , and SEE=0.13 for three numbers of component. The external validation of the test set resulted in  $r^2_{\text{pred}} = 0.921$ , which indicated the predictive ability of the CoMSIA model. The field contribution values for steric, electrostatic, hydrophobic, HBD, and HBA were 8.6%, 31.7%, 20.7%, 21.2%, and 17.8%, respectively. Figure 2 exhibits the relationship between the data of predicted and observed activity for CoMFA and CoMSIA models.

Table 3. PLS statistics of CoMFA and CoMSIA models.

Statistical parameter	CoMFA	CoMSIA
$q^2$	0.577	0.537
$t^2$	0.982	0.964
$\mathcal{F}_{\mathrm{pred}}$	0.926	0.921
F	166.591	62.404
SEE	0.0904529	0.129642
N	3	3
Fraction	CoMFA	CoMSIA
Steric	0.374	0.086
Electrostatic	0.626	0.317
Hydrophobic	-	0.207
Hbond donor	-	0.212
Hbond acceptor	-	0.178

 $Q^2$  = cross-validated correlation coefficient, N = optimum number of components,  $R^2$  = non-cross-validated correlation coefficient,  $r^2_{\text{ext}}$  = external validation correlation coefficient, SEE = standard error of the estimate, F = F-test value.



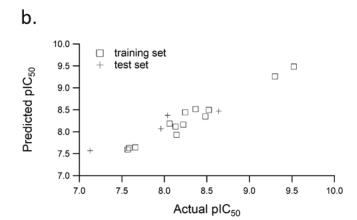


Figure 2. Scatter plot diagram between actual pIC<sub>50</sub> versus predicted pIC<sub>50</sub> for (a) CoMFA and (b) CoMSIA.

## Graphical interpretation of CoMFA and CoMSIA

CoMFA and CoMSIA contour maps were generated to rationalize the regions in 3D space around the molecules for increasing the inhibitory activity. The CoMFA steric and electrostatic contour maps are shown in Figure 3, while the corresponding CoMSIA contour maps are shown in Figure 4. The most active compound 12 was used as the reference structure.

#### CoMFA contour maps

CoMFA steric interactions are represented by green and yellow colored contours, while CoMFA electrostatic interactions are shown with red and blue colored contours. The bulky substituents are favorable in the green regions of steric contours for enhancing the inhibitory activity, while those in yellow regions may lead to a decrease in inhibitory activity. Meanwhile, in the map of the electrostatic field, the blue contours indicate that electropositive charges are favored for inhibitory activity, while the red contour designates an increase in inhibitory activity of the electronegative charges. Compound 12 (Comp12) was utilized to explain the contour map. Figure 3 exhibits CoMFA steric and electrostatic contour maps. The steric field contour shows that the small-sized green contour could be observed near  $R_1$  position, which indicates that the addition of small bulky groups near those green regions would increase the inhibitory activity. On the other hand, the yellow contour around the  $R_2$  position indicates that hydrophobic or bulky group substitution near the yellow regions is favored for increasing the activity. The CoMFA steric interaction agrees well with the experimental results. For instance, compound 11 {R<sub>1</sub> = H, R<sub>2</sub> = 3-Cl-4-[(3-F-benzyl)oxo]phenyl, n = 3, X = O}, compound 8 {R<sub>1</sub> = H, R<sub>2</sub> = 3-Cl-4-[(3-F-benzyl)oxo]phenyl, n = 4, X = O}, compound 17 {R<sub>1</sub> = H, R<sub>2</sub> = 3-Cl-4-[(3-F-benzyl)oxo]phenyl, n = 3, X = CONH}, and compound 14 {R<sub>1</sub> = OMe, R<sub>2</sub> = 3-Cl-4-[(3-F-benzyl)oxo]phenyl, n = 3, X = O} had the lowest activities as indicated by pIC<sub>50</sub> values of 7.1249, 7.5686, 7.585, and 7.6576, respectively. On the other hand, only red contour was observed in the electrostatic contour map near R<sub>1</sub> substituent, which indicates that electronegative substituent at the position would increase the inhibitory activity. This phenomenon explains the reason why compound 14 (R<sub>1</sub> = OMe) displayed low inhibitory activity (IC<sub>50</sub> = 22 nM).

## CoMSIA contour map

The COMSIA steric field contour maps are depicted in Figure 4a. The green contour around  $R_1$  means that sterically, bulky groups are favorable for increasing the inhibitory activity. In contrast, a yellow contour near the phenyl group means that bulky group substituent in that position may decrease the inhibitory activity.

In the CoMSIA electrostatic field (Fig. 4b), the blue color designates the positively charged groups that are favored for inhibitory activity, while the red contour denotes the negatively charged groups, which are favored for improving the inhibitory activity. CoMSIA hydrophobic contour map is shown in Figure 4c. The hydrophobic yellow contours can be observed around the C3 position of R<sub>2</sub>, which indicates that replacing this position

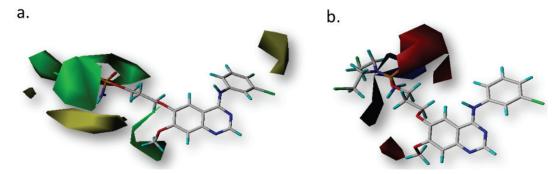


Figure 3. CoMFA steric (a) and electrostatic (b) contour maps with 2 Å grid spacing. Compound 12 was displayed in the background. In steric and electrostatic fields, the contribution of green and blue contours, respectively, is 80%, while those of yellow and red, respectively, is 20%.

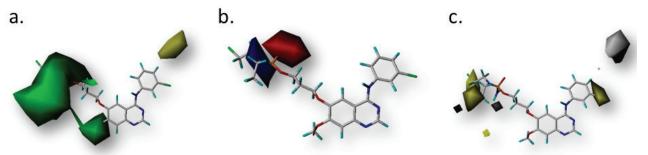


Figure 4. CoMSIA std\*coeff contour maps with compound 12 as a background. (a) Steric fields: contribution of green contours is 80%, while yellow contours are 20% contribution. (b) Electrostatic fields: blue contours are 80% contribution, while red contours are 20% contribution. (c) Hydrophobic fields: yellow contours are 80% contribution, while white contours are 20% contribution.

with hydrophobic groups may increase the activity. In contrast, the white contour around the C4 position of R<sub>2</sub> indicates that hydrophobic groups are not favored for increasing the activity.

In the CoMSIA HBA (Fig. 5a), the magenta contour represents that the hydrogen bond acceptor groups are favorable for increasing the inhibitory activity, while the red contour implies that the HBA groups would decrease the inhibitory activity. In the CoMSIA HBD (Fig. 5b), HBD groups are favored in the cyan contour for increasing the inhibitory activity, while the purple contour denotes that HBD groups are not preferred. Overall, electrostatic field contributes the most to the CoMSIA model as shown in Table 3, which indicates that the electrostatic field is the most influencing factor to the inhibitory activity of the quinazoline–phosphoramidate mustard conjugates. This result agrees well with the CoMFA model, which shows that the electrostatic field is more important to the inhibitory activity than the steric field (Table 3).

## Design for new compound

Based on the proposed 3D CoMFA and CoMSIA models, new compounds are designed using compound 12 (Comp12) as a template. All new compounds were minimized and aligned to the database, and then docked into the active site of EGFR. The

binding affinities (total score) of the newly designed compounds were higher than that of Comp12. Table 4 depicts the structures of the newly designed compound using the predicted  $pIC_{50}$  CoMFA and CoMSIA as well as total scores, while Figure 6 shows predicted  $pIC_{50}$  values between CoMFA and CoMSIA models.

#### Molecular docking

To explore the interaction of Comp12, RA1, RA2, RA3, and RA4 with EGFR, molecular docking using surflex-dock was applied. The results showed that all Comp12, RA1, RA2, RA3, and RA4 formed hbond interactions with Met769. In the crystallographic pose of erlotinib, Met769 was observed in direct hbond. Additional hbond with Thr766 was established with Comp12, RA1, and RA3, which was detected in the crystal structure of erlotinib through water-mediated hbond. Besides, RA1 and RA2 formed hbonds with Lys721, while Comp12 and RA1 share the same hbond with Thr830. Figure 7 depicts the docked positions of Comp12, RA1, RA2, RA3, and RA4 in the binding site of EGFR.

#### MD simulation and MM-PBSA calculation

MD simulation of 100 ns was performed on each complex of Comp12, RA1, RA2, RA3, and RA4, with EGFR.

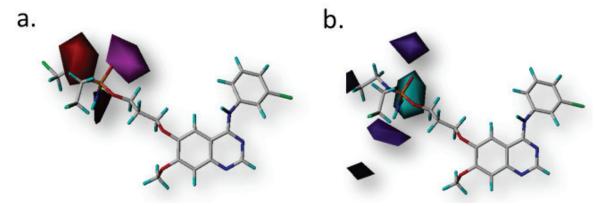


Figure 5. CoMSIA std\*coeff. contour maps with compound 12 as a background. (a) H-bond acceptor fields: the magenta contour is 80% contribution, while the red contour is 20% contribution. (b) H-bond donor fields: the cyan contour is 80% contribution, while purple is 20% contribution.

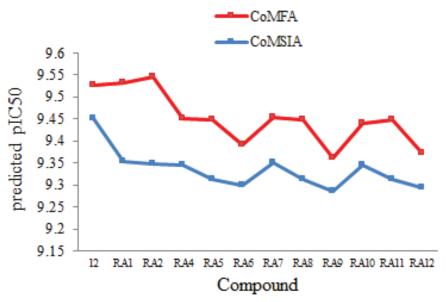
**Table 4.** Chemical structures of newly designed compounds and their predicted pIC<sub>50</sub> CoMFA and CoMSIA models as well as total scores.

$$\begin{array}{c|c} CI & O & HN \\ N-P-O & O & N\\ NH_2 & R_2 & N \end{array}$$

Compound	Substitu	ent	Predic	Total Score		
Compound -	$\mathbf{R}_{_{1}}$	$\mathbf{R}_{2}$	CoMFA	CoMSIA	— Total Score	
Comp12	Br	OCH <sub>3</sub>	9.5269	9.4499	5.7583	
RA1	СООН	OCH <sub>3</sub>	9.5322	9.3532	11.4093	
RA2	COCH <sub>3</sub>	OCH <sub>3</sub>	9.5451	9.3476	8.4734	
RA3	COOCH <sub>3</sub>	$OC_2H_5$	9.3906	9.2980	7.8309	
RA4	СООН	$OC_3H_7$	9.4528	9.3495	8.1122	

Figure 8a shows the RMSD of the heavy atoms of EGFR during 100 ns. It shows stable conformations of both complexes after 10 ns. To assess the flexibility of amino acid residues during 100 ns, RMSF plot was measured (Fig. 8b). The results showed that the two complexes had a similar pattern of residual movements. High flexibility was observed in the amino and carboxy terminals of the protein, while the majority of the residues showed rigid flexibility, which indicated that the ligand binding did not induce a large change in protein conformation.

To assess the affinity of each compound to EGFR, MM-PBSA calculation was performed (Table 5). It can be inferred from Table 5 that electrostatic, van der Waals, and non-polar energy of desolvation were favorable for ligand binding in both complexes. Whereas, the polar energy of desolvation was not favored, resulting in unfavorable net electrostatic energies. Interestingly, the predicted total binding free energy of RA2 ( $\Delta G = -38.32 \text{ kcal/mol}$ ) was lower than that of Comp12 ( $\Delta G = -30.48 \text{ kcal/mol}$ ), which indicated the good potential of the designed compound.



**Figure 6.** The pIC<sub>50</sub> values between CoMFA and CoMSIA models.

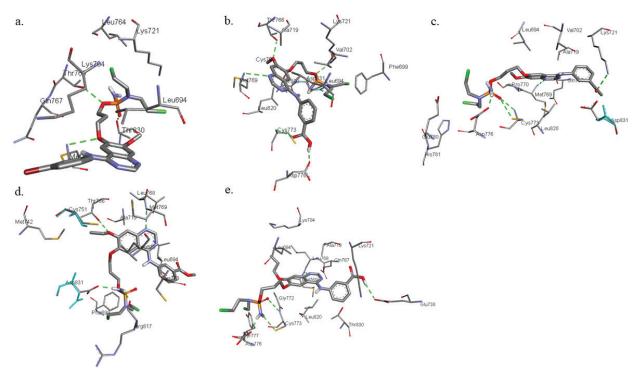


Figure 7. The docked poses of (a) Comp12, (b) RA1, (c) RA2, (d) RA3, and (e) RA4.

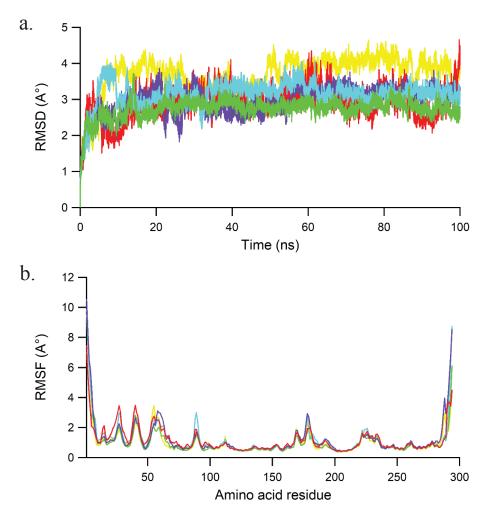


Figure 8. (a) RMSD value of each ligand-EGFR complex during 100 ns dynamics runs calculated from heavy atoms of protein for Comp12 (red), RA1 (yellow), RA2 (green), RA3 (blue), and RA4 (purple). (b) RMSF plot of each amino acid residue during 100 ns dynamics simulation for Comp12 (red), RA1 (yellow), RA2 (green), RA3 (blue), and RA4 (purple).

 $\Delta E_{\text{ele}}$  $\Delta E_{
m VDW}$  $\Delta E_{ ext{PBCAL}}$  $\Delta E_{\underline{ ext{PBSUR}}}$  $\Delta E_{\underline{ ext{PBTOT}}}$ Comp Comp12  $-23.19 \pm 9.28$  $-49.74 \pm 5.07$  $47.43 \pm 10.23$  $-4.98 \pm 0.29$  $-30.48 \pm 4.81$ RA1  $-59.04 \pm 8.99$  $-46.10 \pm 3.23$  $85.57 \pm 7.99$  $-5.09 \pm 0.16$  $-24.66 \pm 4.75$  $-53.28 \pm 10.93$  $-62.04 \pm 3.69$  $82.89 \pm 9.21$  $-5.89 \pm 0.14$  $-38.32 \pm 6.13$ RA2 RA3  $-26.13 \pm 10.21$  $-52.13 \pm 3.37$  $54.65 \pm 7.73$  $-5.33 \pm 0.24$  $-28.93 \pm 5.03$ RA4  $-35.09 \pm 9.28$  $-55.06 \pm 4.20$  $70.18 \pm 11.56$  $-5.70 \pm 0.27$  $-25.67 \pm 6.37$ 

Table 5. The binding free energy terms (kcal/mol) of each ligand bound to EGFR.

## CONCLUSION

In the current study, CoMFA and CoMSIA analysis and molecular docking were performed to explore the structure-activity relationship of novel quinazoline-phosphoramidate mustard conjugates as an EGFR inhibitor. Both CoMFA and CoMSIA models were valid with acceptable statistical criteria. Using CoMFA and CoMSIA contour maps, new compounds were designed and docked to the binding site of EGFR. The docked position of the newly designed compounds showed key interactions with the active site residues of EGFR, which was

stable during 100 ns of MD simulation. The MM-PBSA binding energy analysis shows that one of the new compounds had higher affinity than that of the parent compound, thus providing a good candidate for further drug discovery research.

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