

# *In vitro* evaluation of photoprotective potential of the different solvent extracts of *Graptophyllum pictum* leaves

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## ABSTRACT

Natural substances extracted from plants have been gaining attraction as protective agents due to their safety and responsible for multiple biological effects on skins. The present work evaluates the photoprotective potential of different solvent extracts of *Graptophyllum pictum* (*G. pictum*) leaves, and the comparison of its Sun Protection Factor (SPF) value against photo-stability and thermal-stability under a 21-day of storage process. Preliminary phytochemical screening was also performed. The dried powdered leaves were extracted by cold maceration method, using five different solvent, i.e. methanol, ethanol, chloroform, ethyl acetate and hexane. The SPF of all extracts were analyzed by ultraviolet (UV) spectrophotometry. The SPF results of the five types of extracts were found to have significant differences ( $P < 0.001$ ). Methanol extract displayed the highest SPF value ( $15.303 \pm 0.045$ ), while the lowest SPF value was showed by ethyl acetate extract. SPF reductions of all extracts under sun exposure condition were higher compared to the extracts placed in dark conditions at room temperature. As compared, the ethanol extract displayed better chemical and thermo-stability with higher *in vitro* SPF at 200  $\mu\text{g/ml}$ . These findings suggested that ethanol was the solvent of choice for yielding high levels of photoprotective ingredients from *G. pictum* which can be used to formulate effective sunscreen preparations.

## INTRODUCTION

Efficacy of sunscreen products is historically assessed through the determination of the so-called SPF, which is defined as the ultraviolet (UV) energy required to produce a minimal erythema dose (MED) on a protected skin versus the UV energy required to produce an MED on an unprotected skin. It is a measurement accepted worldwide to indicate how many times longer a person wearing sun protection in the sun can withstand without getting burned as opposed to not wearing any sun protection at all (Mbanga *et al.*, 2015; Wagemaker *et al.*, 2011). The UV light is classified into three major regions: UV-A (320–400 nm), UV-B (290–320 nm) and UV-C (200–290 nm). UV-C is most biologically damaging, but it gets effectively filtered by the ozone layer before reaching earth. Both UV-A and UV-B are

not completely filtered out by the ozone layer and are responsible for the damage due to sunburn and pyrimidine dimers (Napagoda *et al.*, 2016; Singh and Sharma, 2016). UV-B is approximately 1000 times more effective than UV-A in inducing erythema, most sunscreens contain compounds which absorb radiations in the UV-B region (Shenoy *et al.*, 2010; Van *et al.*, 2017). The application of the sunscreen products that contain UV absorbing, reflecting or scattering active molecules is the most popular practice in the present day to reduce the amount of UV radiation penetrating the skin (Napagoda *et al.*, 2016). Currently, the available sunscreen products with varying SPF often contain synthetic chemical compounds such as titanium oxide, benzophenone, bisdisolizole disodium (Fonseca, 2013) to cover a wide range spectrum of UV radiation protection. Unfortunately, these synthetic compounds are found to be potentially hazardous to health as they can cause skin diseases. Therefore, there is need to explore various natural and non-toxic sun protection properties from plant resources particularly to complement the current use of synthetic compounds.

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*Graptophyllum pictum* (*G. pictum*), or locally known as “Puding” or purple leaf is a well-known traditional medicinal plant in Malaysia. The leaves of *G. pictum* are commonly consumed as culinary dishes by the Malay community (Singh *et al.*, 2015a). Traditionally, *G. pictum* was consumed to enhance fertility, treat ulcers and constipation (Ministry of Health Republic of Indonesia, 2010). *In vitro* studies found that *G. pictum* is rich with phytochemical constituents and it possesses anti-oxidation (Singh *et al.*, 2015b; Winata, 2011), antidiabetic (Ogbonnia *et al.*, 2011), anti-implantation (Andrianto, 2015) and anti-plaque (Wahyuningtyas, 2005) properties. Though various pharmacognostical and phytochemical studies had been done on *G. pictum*, the leaf extract of this plant had notably not been previously assessed on the sun protective ability. Scientists believe that the chemical components on some plants, like anthocyanins, proanthocyanidin and carotenoids may act as a natural “sunscreen,” protecting the cells from too much light (Koracand Khambhoja, 2011; Stahl and Sies, 2007). In such, the leaves of *G. pictum* with strong brown and purple color are suspected to have high sun protective properties (Saewan and Jimtaisong, 2013).

According to the literature review, around 50 plant species from various plant parts that had been tested for *in vitro* SPF and the sun protection efficacy was found to be in the range of 0.1 – 39 at the concentration of 200 – 1000 µg/ml (Gajardo *et al.*, 2016; Gonçalves *et al.*, 2015). It was noted that various plant parts were extracted only with either methanol or ethanol while other organic solvents which could affect extractable the sun protective ingredients were not well explored. In addition, to the best of our knowledge, only two studies had successfully been investigated and reported to date on the photo-stability and thermal-stability of plant extracts (Fonseca, 2013; Napagoda *et al.*, 2016). The present work was aimed on evaluation of photoprotective potential of various extracts of *G. pictum* leaves, and to compare the SPF values from each extract against photo-stability and thermal-stability under a 21-day of storage process.

## MATERIALS AND METHODS

### Plant Material

Fresh leaves of *G. pictum* were collected from *Kuala Pilah* district which is located in the southern Malaysian state of *Negeri Sembilan*. The plant material was identified by Dr Fatimah Mohamed from Universiti Pendidikan Sultan Idris, *Tanjung Malim*, Perak. The plant material was thoroughly cleaned and dried under shade for four days. Dried plants were powdered using an electrical grinder (Panasonic Mixer Grinder MX-AC400W). The powder was then stored in an airtight sample bottle for further analysis.

### Preparation of Extracts

The dried powdered of the *G. pictum* leaves (10 grams per solvent) were extracted by cold maceration in methanol, ethanol, hexane, ethyl acetate, and chloroform respectively for three days. The extracts were concentrated under reduced pressure

using a rotary evaporator until dryness to yield five different solvent extracts.

### Preliminary phytochemicals screening

The phytochemical screening of each extract (1g/mL in mother solvent) was performed according to the standard phytochemical screening method (Prashant *et al.*, 2011) and the observations were recorded.

### Determination of the *in vitro* Sun protection factor (SPF)

The *in vitro* SPF was determined by following the modified spectrophotometric method (Fonseca, 2013). Each extract was diluted in the mother solvent respectively at a final concentration of 200 µl/ml. The samples were placed in (i) dark conditions at room temperature (27 °C) and (ii) sunlight exposure at temperature higher than 27°C. The measurement of SPF was evaluated at 7<sup>th</sup>, 14<sup>th</sup> and 21<sup>st</sup> day interval period, by using UV spectrophotometry (Perkin-Elmer) from 290 to 320 nm, with intervals of 5 nm. The readings were taken in triplicate and the determinations were made at each point. The SPF was calculated according to the equation (1).

$$SPF = CF \times \sum_{290}^{320} EE(\lambda) \times I(\lambda) \times Abs(\lambda) \quad (1)$$

Where the obtained absorbance values between 290 and 320 nm were multiplied with the respective  $EE(\lambda)$  values. Their summation was taken and multiplied with the correction factor (10) to obtain the SPF values. The data were expressed as  $\pm$  standard deviation. All data were analyzed statistically by the One-Way ANOVA and T-Test analyses to analyse SPF against various days of exposure. The analysis was performed using the statistical package for Social Sciences version 17.0 (a window software for data analysis) and the level of significance was set at  $p=0.05$ .

## RESULT AND DISCUSSION

### Preliminary Phytochemical screening

Preliminary phytochemical screening of the crude extracts of leaves of *G. pictum* revealed the presence of different kind of chemical groups as summarized in Table 1. Leaves of *G. pictum* had resulted in the identification of carbohydrate, flavonoids and alkaloids, for both methanol and ethanol extracts. Steroids and glycosides were found to be present in chloroform, ethyl acetate and hexane extracts. This result supplies the additional information as current literature only reports of phytochemical constituents from ethanol and petroleum ether extract (Singh *et al.*, 2015a). However, negative results of phytochemical constituents were reported for hexane and ethyl acetate extract (Jiangseubchatveera *et al.*, 2017). All types of solvent extractions gave negative of volatile oil indication. This being attributed to the very low composition of volatile oil in leaves, which is only extractable through distillation process (Jiangseubchatveera *et al.*, 2015).

**Table 1:** Phytochemical screening of various extracts of *G. pictum* leaves.

	Methanol	Ethanol	Chloroform	Ethyl Acetate	Hexane
<b>Alkaloid</b>					
Mayer test	+	+	-	-	-
Wagner test	+	+	-	-	-
<b>Carbohydrate</b>					
Molisch test	+	+	-	-	-
<b>Steroids</b>					
Liebermann-Burchard test	-	-	+	+	+
Salkowski	-	-	+	+	+
<b>Cardiac glycosides</b>					
Keller Killani's test	-	-	+	+	+
Baljet test	-	-	+	+	+
<b>Flavonoids</b>					
Ferric chloride test	+	+	-	-	-
Lead acetate test	+	+	-	-	-
<b>Volatile oil</b>					
Filter paper	-	-	-	-	-
NaOH	-	-	-	-	-

Remarks: + presence, - absence.

The results of the evaluation of *in vitro* SPF of various extract of *G. pictum* leaves are shown in Table 2. All reported SPF results were found significantly different ( $P < 0.001$ ). Among the five extracts tested, methanol extract had displayed the highest SPF value as  $15.303 \pm 0.045$ . SPF values of ethanol and chloroform extract were  $13.423 \pm 0.004$  and  $13.181 \pm 0.008$ , respectively. The ethyl acetate extract showed the lowest SPF value as  $11.656 \pm 0.001$ . SPF values of methanol and ethanol extract are comparatively higher than reported in previous studies (Costa *et al.*, 2015; Gajardo, 2016; Gonçalves *et al.*, 2015; Napagoda *et al.*, 2016). Whereas, SPF value obtained from hexane, chloroform and ethyl acetate had never been reported before.

**Table 2:** SPF of various extracts of *G. pictum* leaves.

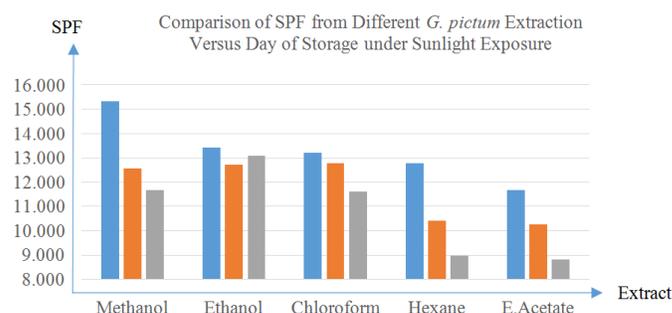
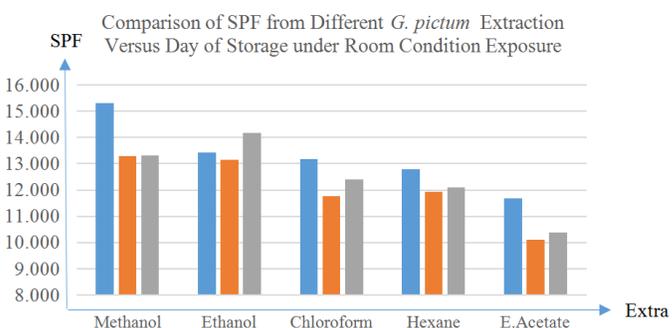
Extracts	SPF	P-value
Methanol	$15.300 \pm 0.005$	0.000
Ethanol	$13.423 \pm 0.004$	
Chloroform	$13.181 \pm 0.008$	
Hexane	$12.777 \pm 0.001$	
Ethyl acetate	$11.657 \pm 0.001$	

According to the guidelines of international regulatory agencies, only SPF value equal or greater than 6 is suitable for use in cosmetic products (Costa *et al.*, 2015). Hence, the results suggested that the *G. pictum* extract can be considered as a promising active ingredient because of its high SPF value in low concentration ( $200 \mu\text{g/ml}$ ). If higher SPF is required for formulation, it can be achieved by reducing the dilution factor when preparing the *G. pictum* extract because SPF is found to be concentration-dependent (Costa *et al.*, 2015). One of the most important factors affecting the extraction efficiency of bioactive compounds from plant materials and their consequent health benefits is the extraction solvent (Van *et al.*, 2017). The better solubility of photo protective ingredient in ethanol and methanol

can be attributed to its chemical structure of flavonoids and alkaloids. The characteristic conjugated system in these phytochemicals enables it to absorb high intensity of UV rays (Saewan and Jimtaisong, 2013), hence producing better SPF. The lower *in vitro* SPF readings in chloroform, hexane and ethyl acetate extracts is possibly due to the presence of steroids, cardiac glycoside and other non-tested secondary metabolites such as chlorophyll and carotenoid. The higher SPF value of chloroform extract may be attributed due to the higher concentration of photo-protective ingredient compared to hexane and ethyl acetate extracts. Although no specific literature is available on the correlation of these compounds to SPF, the common chemical feature of the steroid, cardiac glycoside structure with less number of carbon conjugated system is less capable of absorbing UV light.

#### Determination of the *in vitro* Sun protection factor (SPF)

The comparison of SPF values from different extracts of *G. pictum* leaves versus the storage period under room temperature and sunlight exposure were presented in Figure 1 and 2. In terms of thermal and photo stability, there is an overall consistent reduction in the sun protection capability of each extract from days 7, 14 to 21.

**Fig. 1:** Comparison of SPF from different *G. pictum* extracts versus day of Storage under sunlight exposure**Fig. 2:** Comparison of SPF from different *G. pictum* extracts versus Day of Storage under room condition exposure

The total percentage of reduction of SPF values from different extracts of *G. pictum* leaves under room condition and sunlight exposure were summarized in Table 3.

**Table 3:** Comparison of total % reduction of SPF.

Exposure	Day	Methanol	Ethanol	Chloroform	Hexane	EthylAcetate
sunlight	7 <sup>th</sup>	15.300±0.005	13.423±0.004	13.181±0.008	12.777±0.001	11.657±0.001
	14 <sup>th</sup>	12.551±0.006	12.697±0.010	12.779±0.003	10.384±0.008	10.250±0.005
	21 <sup>st</sup>	11.653±0.002	13.080±0.035	11.606±0.004	8.968±0.004	8.797±0.005
	Reduction (%)	24%	3%	14%	42%	25%
Room	7 <sup>th</sup>	15.303±0.006	13.423±0.000	13.180±0.010	12.780±0.005	11.660±0.000
	14 <sup>th</sup>	13.273±0.058	13.137±0.006	11.770±0.006	11.933±0.006	10.090±0.000
	21 <sup>st</sup>	13.297±0.006	14.160±0.006	12.397±0.006	12.087±0.000	10.360±0.000
	Difference (%)	13%	0%	6%	5%	11%

The SPF reductions of all extracts under sun exposure condition were higher compared to the extracts placed in dark conditions at room temperature.

Heat and light acceleration induced conjugated a structural change, the degradation and oxidation stress of the extracted chemical constituents. However, without heat and UV exposure, the reduction of SPF was mainly caused by oxidation of a prolonged storage process.

Although all the extracts were found to have reduction in sun protective ability, ethanol extract was found to be the most stable sample in the 21-day storage duration. Table 3 showed that SPF of ethanol extract only reduced 3% and 0% under sunlight exposure and room condition respectively. Hence, at 21<sup>st</sup> day, the SPF of methanol extract was the highest in comparison to other extracts.

The stability of ethanol extract can be due to high polarity of ethanol solvent which capable of extracting broad polar and semi-polar constituents, and hence, the ability to produce a better sun protection coverage. Furthermore, the synergistic effect of various compounds is found capable of providing broader protection.

The good photoprotective potential was observed on ethanol extract of *G. pictum* leaves, however, they cannot be used as stand-alone photoprotective agent. *In vitro* SPF tests are usually served as a preliminary screening to choose the best extract. Further studies like antioxidant activity studies and *in-vivo* SPF testing are suggested. The photoprotective compounds in the leaves can be isolated and be used in formulations of natural sunscreen, as well as replacing the synthetic sunscreens.

## CONCLUSION

This study indicated that the *G. pictum* leaves sun protective ability is well related to the choice of solvent extraction used. These findings suggested that ethanol was the solvent of choice for yielding high levels of photo-protective ingredients and better stability in 21-day storage condition, which can be used to formulate effective sunscreen preparations.

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## CONFLICT OF INTEREST

The authors declare no conflict of interest.

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