Solid-phase extraction for RP-HPLC/UV determination of ziprasidone at presence its main metabolite in urine

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ABSTRACT

The rapid and accurate high-performance liquid chromatography method with ultraviolet detection (210 nm) for the determination of Ziprasidone in rat urine was developed. Simple protocol of solid-phase extraction of ziprasidone from urine sample at presence its main metabolites was worked out. Vardenafil was used as an internal standard. Reversed phase column LUNA® C18(2) 100A 250 mm × 4.6 mm × 5 μm was used with an isocratic mobile phase consisting of acetonitrile : 0.5 % triethylamine (30: 70) and 0.1 M phosphoric acid (pH 2.5). Developed conditions of HPLC analysis provide high efficiency of a system and good separation of analytes. The method was validated and applied to rat urine samples after modelling intoxication. The intra- and inter-day precision was ≤15% with recovery about 95 %. A linear range of 1 μg/mL to 200 μg/mL was established. LOD and LOQ were 0.2 and 0.5 μg/mL, respectively. Obtained results indicate that the described procedure of sample pretreatment with automatization possibility allows obtaining good results for ziprasidone. This method is sensitive, precise and repeatable enough to be used in toxicological casework.

INTRODUCTION

Ziprasidone, (5-[2-[4-(1,2-benzisothiazol-3-yl)-1-piperazinyl][ethyl]-6-chloro-1,3-dihydro-2H-indol-2-one) is widely used for the treatment of both positive and negative symptoms of schizophrenia (Shin et al., 2011; Strom et al., 2011; Pfizer Inc., 2014). Ziprasidone is a representative of atypical antipsychotic drugs with a unique receptor profile and ability to inhibit the reuptake of serotonin and norepinephrine. The interaction with the 5-HT2A and 5-HT1A, 5-HT2C, 5-HT1D, 5-HT7 receptors and type 2(D2) dopamine receptors ziprasidonehas antipsychotic, antidepressant and anxiolytic effects (Asif, 2016). The recommended dose for the treatment of acute conditions is 40-80 mg twice daily (Pfizer, 2014). In the case of overdose or combining it with drugs, which cause prolongation of the QT interval, the cardiotoxic effect of the drug is observed (Wenzel-Seifert et al., 2011; Timour et al., 2012; Beach et al., 2013; Leonard et al., 2013; Hasnain and Vieweg, 2014). Cardiovascular toxicity is the most frequent cause of specified or sudden cardiac death in patients with schizophrenia (Marano et al., 2011; Söderberg et al., 2016).

Ziprasidoneis intensively metabolized by CYP3A4 and aldehyde oxidase – about 5% of the dose of the drug is excreted in the urine (<1%) and faeces (<4%) unchanged (Levine and Ruha., 2012). The major circulating metabolites of ziprasidonare S-methyl-dihydroziprasidone, ziprasidone sulfoxide, benzisothiazole piperazine (BITP) sulfoxide and BITP sulfone. (Xie, 2014). S-methyl-dihydroziprasidone and ziprasidone sulfoxide in tests in vitro reveal properties that may indicate the action that extends the QT interval, so determining their content in biomaterial has toxicological significance as well (Pfizer Inc., 2000).
Several authors describe techniques for determining the presence of ziprasidone in biological material (plasma, serum, brain) using LC/MS, HPLC/MS, HPLC/UV, HPLC with fluorescence detector, UPLC/UV and the potentiometric method (Garcia et al., 2011; Mercolini et al., 2011; Marghade et al., 2012; Sampedro et al., 2012; Sreedasyam et al., 2012; Rao et al., 2013; Mercolini et al., 2014; Sistik, 2016; Wang and Li, 2017). Preparation and analysis of samples in the described techniques were carried out in several steps and were generally time-consuming. Elimination time of ziprasidone is short and concentrations of analytes and their metabolites in urine can be higher than in blood after few hours of ingestion. Therefore, urine and hair samples are better biosamples because drugs can be detected even days or up to weeks after the intake and, undoubtedly, the urine screening may give better results (Scott, 2009). Unfortunately, there are no described methods of isolation and identification of ziprasidone from urine samples that would be suitable for the purpose of post-mortem toxicological investigation. Summing up these factors, the relevance of this study is due to the need to develop a simple, rapid, accurate, and sensitive technique for sample preparation to detect ziprasidone in urine using the HPLC/UV. The objective of our study is elaborating conditions for identification and quantification of ziprasidone by HPLC with UV detection in samples obtained during purification of extracts from urine by solid phase extraction.

**MATERIALS AND METHODS**

**Chemicals and Reagents**

A standard sample of ziprasidone (≥98.0% Sigma-Aldrich, USA) was used to produce a series of stock solutions, as internal standard – Vardenafil (≥98.0% Sigma-Aldrich, USA). Methanol, acetonitrile and phosphoric acid, all “HPLC” grade (≥99.9%, Sigma-Aldrich, USA) were used to prepare mobile phase, and triethylamine (for HPLC, ≥99.5%, Fluka). All other chemicals were of analytical grade. Double distilled water obtained in Milli-Q purification system (Millipore; Vienna, Austria) was used to prepare triethylamine and calcium chloride solutions. Test animals were injected with the content of Zeldox capsules (Pfizer, New York, USA) containing 40 mg of ziprasidone and triethylamine (for HPLC, ≥99.5%, Fluka). All other chemicals were of analytical grade.

**Instrumentation and Chromatographic conditions**

Preparation of urine samples for the analysis was performed using Strata-X 30 mg cartridges (Phenomenex). Identification and quantification of ziprasidone were performed with liquid chromatography Thermo Dionex ultimate 3000 UHPLC. The system consists of an UltiMate 3000 RS pump, an UltiMate 3000 RS autosampler and an UltiMate 3000 RS column compartment (Dionex, Olten/Switzerland).

Chromatographic separation of analytes was performed with a reverse phase column LUNA@ C18(2) 100 Å 250 mm×4.6 mm × 5 μm, at 25°C. The composition of the mobile phase: acetonitrile-0.5% triethylamine (30:70), pH 3.0. The acidic pH was adjusted 0.1 M phosphoric acid. The isocratic flow rate of the mobile phase was 1 ml/min. The volume of injected sample was 10 μl. Spectrum of ziprasidone was recorded in range 190 – 350 nm. Detection was performed with LED UV detector (UV-VIS, DAD, FLD) at 210 nm. The results were processed with Chromeleon® Chromatography Data System software (Version 7.2.0.3765, Thermo Fisher Scientific). Organic solvents were evaporated with A TurboVap evaporator (Zymark; Hopkinton, MA). Urine samples were centrifuged in Sigma 3-16 K centrifuge with cooling. Filtration of the mobile phase and samples before the injection into chromatograph was performed using PTFE membrane syringe filters with diameter of 13 mm and a pore size of 0.2 microns by STL-labortechnics (Czech Republic).

**Preparations of standard and sample solutions**

Primary standard solutions of ziprasidone and vardenafil (internal standard) were prepared separately by dissolving an accurately weighed amount of standard substances (25 mg) with methanol in 25 ml volumetric flask to get the final concentration of 1 mg/ml. All standard solutions were stored at 4°C in the dark place for 30 days. Stock solution of ziprasidone were obtained by diluting 5 ml of the respective standard solutions with 20 ml of the solvent (methanol) to obtain the concentrations 200 μg/ml. Stock solutions of vardenafil (50 μg/ml) were prepared by diluting 1.25ml of appropriate standard solution with methanol in 25 ml volumetric flask. Three parallel series of model samples of urine with ziprasidone concentration of 1.0 to 200.0 μg/ml (1.0; 5.0; 10.0; 20.0; 30.0; 100.0, 150.0 and 200.0 μg/ml) were prepared for the analysis. 2 ml of pure rats’ urine were spiked with 100 μl of internal standard stock solution and with 10.0, 50.0, 100.0, 200.0, 300.0 μl of stock solutions of ziprasidone; 200.0, 300.0, 400.0 μl of standard solutions. Samples were kept for 3 minutes in the orbital mixer of ”Vortex” type and then incubated at 37°C for 60 minutes. The resulting solutions were used to construct the calibration curve by the method of internal standard in the concentration range of 1-30 µg/ml and 30-200 µg/ml. Similarly, quality control samples were prepared for method validation (QC) – pure rats’ urine was spiked with ziprasidone to obtain concentrations of 5, 100 and 200 μg/ml. Model samples and quality control samples were subject further sample preparation by the approach described below.

**Sample collections**

To conduct the experimental study a group of twelve white rats (230-240 g) aged 2-3 months was used. Half of them was the study group, the other half was the control one. Adult rats were obtained from the Laboratory Animal Center (Vivarium) at Danylo Halytsky Lviv National Medical University. Experiments adhered to ethical standards and approved by the ethical committee of mentioned university (Approval File No. 1/2015). The animals were kept under a 12 h light-dark cycle with free access to food. Animals in the study group received...
ziprasidone (174 mg/kg) in three divided doses during the day. The drug was administered intragastrically in the form of an aqueous suspension (667 µg/ml in 6 ml) with the addition of Tween-80 as stabilizing agent. The choice of dose to model poisoning was resulting from the literature data (Lee et al., 2013; Hovda et al., 2016). Control animals received the same amount of solvent (water). The collection of daily urine for the analysis was performed from the first injection. The biological material was stored at-20°C.

**Sample preparation**

2 ml of rats’ urinewere spiked with 100 µl of internal standard solution (50 µg/ml) and 0.2 ml of 20% solution of calcium chloride was added to precipitate of uric acid salts, and centrifuged for 15 min (5000 × g). The supernatant was quantitatively selected and passed through a solid phase extraction cartridges Strata-X (30 mg; Phenomenex, UK). The cartridges were activated with 1 ml of methanol and 1 ml of distilled water before using. After the sample displacement, the columns were washed sequentially on 1 ml of phosphate buffer (pH 7.8) and distilled water with the flow rate of all liquids through the cartridge 1 ml/min. Ziprasidone was eluted with 2 ml of methanol after drying the sorbent in a stream of nitrogen for 5 minutes. The volume of the solution was adjusted to 5 ml of methanol. 1 ml of this solution was evaporated to 500 µl of methanol and 10 µl of the resulting solution was injected into the chromatograph. In case quantification of ziprasidone at therapeutic doses, we recommend to elute analyte from SPE cartridges with 2 ml methanol, evaporate it to 500 µl of methanol and injected 10 µl of the resulting solution into the chromatographic system. This allows to concentrate the sample in five times and to increase the limit of ziprasidone determination in biological material at the level of 125 ng in 1 ml of urine. Samples were mixed using a shaker of “Vortex” type for 2 minutes and filtered through membrane filters PTFE 0.2 µm. Control urine samples were studied in parallel by the same scheme. Extraction recovery was evaluated by performing SPE in triplicate on urine samples containing ziprasidone at three concentration levels of the analyte and IS.

Method validation was performed by such characteristics as specificity, accuracy and precision, linearity, repeatability, and matrix effects according to International Conference on Harmonization (ICH) Guidelines (ICH, 2005). An important part of method validation is to determine the indicators of the suitability of the system. To this end, the following parameters were calculated: a number of theoretical plates, peak asymmetry, and repeatability of retention times and peak areas. To prove the specificity of the technique chromatograms of a stock, tested, and placebo solution was compared. Selectivity was studied by analyzing urine samples from several different laboratory rats. The lower limit of quantitation (LLOQ) was defined as the minimal concentration of ziprasidone that can be precisely measured (coefficient of variation (CV) of less than 20%) and is determined by 5 samples of model tests independent of this curve. Limit of detection (LOD) was determined by 5 samples of model tests with signal to noise ratio of at least 3:1.

The efficacy of extraction and matrix effects were studied by model samples with three levels of concentrations (low, medium, high). Accuracy and precision of the method were determined by analyzing a series of three solutions with concentrations of 5, 100, 200 µg/ml (n=6). Samples were analyzed on the day of preparation and 24 hours after preparation (samples were stored at 4°C). Short-term and long-term stability of ziprasidone was studied by model urine samples containing 1 µg/ml and 200 µg/ml of a drug. Ziprasidone stability during three freeze-thaw cycles was determined as well. 10% degradation criterion was used to confirm the stability.

**RESULTS**

Ziprasidone identification was performed by retention time, which was 9.6±0.1 min and by UV spectrum under described analysis conditions. The retention time of internal standard (vardenafil) was 7.6±0.1 min. The proposed method of extraction and detection of ziprasidone was used to analyze urine samples of laboratory rats that were given the drug under study. Concurrently the results were compared that were obtained after the analysis of model samples of urine containing ziprasidone (Figure 1) and samples of biological fluid of test animals (Figure 2) with blank samples. Most of urine components were eluted from the chromatographic column long before the time of ziprasidone release because no interaction between them was observed. The suggested technique of the extraction of the drug enabled to achieve high efficiency in isolation and purity of samples obtained, as evidenced by minor variations in the zero line in the chromatograms. It was experimentally determined that the ziprasidone recovery from an aqueous solution after solid phase extraction was 97.5-98.2%. The efficiency of extraction of ziprasidone from urine was determined by analyzing model urine samples (n=9) with three levels of drug concentrations (low, medium, high). The overall recovery of ziprasidone in the assay was above 95%. Validation parameters of ziprasidone isolation from model urine samples by SPE are presented in Table 1. There is a clear separation of ziprasidone peaks with its metabolites in the chromatograms obtained from urine samples of laboratory rats (Figure 3). Conclusions about the ziprasidone metabolites presence in extracts from the urine of test animals were made comparing the chromatograms of pure urine samples, urine with the standard solution of ziprasidone addition, and urine of animals treated with the drug. We have identified four metabolites observed in all urine samples of the animals treated with ziprasidone and were absent in chromatograms of control urine samples. Their retention time was 9.08, 10.2, 10.75 and 13.43 min. The proposed method enabled to identify mean 46.2 µg of ziprasidone in 1 ml of daily urine of animals. The total content of metabolites constituted about 28% of drug concentration level. The minimum concentration that can be determined in biomaterial is 0.125µg/ml.
Fig. 1: Chromatograms of model urine samples, containing 30 µg/ml ziprasidone (2) and 10 µg/ml internal standard (1).

Fig. 2: Chromatogram of drug-free urine samples.

Fig. 3: A representative chromatogram of urine samples of rats treated with ziprasidone. 1 – IS; 3 – Ziprasidone. (Peak area of ziprasidone corresponds to 37µg/ml, and peak area of vardenafil - 10µg/ml.)

Table 1: The extent and efficacy of extraction and matrix effects in the preparation of samples.

<table>
<thead>
<tr>
<th>Concentration added (µg/ml)</th>
<th>Extraction recovery (%) ± SD</th>
<th>Process efficiency (%) ± SD</th>
<th>% CV (RSD)</th>
<th>Matrix effect (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>5</td>
<td>95.58 ± 2.07</td>
<td>95.25 ± 2.07</td>
<td>2.17</td>
<td>100.80</td>
</tr>
<tr>
<td>100</td>
<td>95.15 ± 1.85</td>
<td>95.15 ± 1.85</td>
<td>1.94</td>
<td>100.11</td>
</tr>
<tr>
<td>200</td>
<td>95.08 ± 1.44</td>
<td>95.05 ± 1.44</td>
<td>1.51</td>
<td>100.17</td>
</tr>
</tbody>
</table>
System suitability
To study system suitability 10 µl of the stock solution of ziprasidone was injected into the chromatograph (n=6) and determined the following parameters: the number of theoretical plates, the number of theoretical plates per meter, coefficient of asymmetry and height equivalent to theoretical plate. Measurement results are presented in Table 2.

The figures in Table 2 confirm high efficacy and selectivity of the proposed system compared to the described before.

The specificity of the proposed method is confirmed by the absence of peaks in the retention time of ziprasidone and its metabolites in chromatograms of control samples of urine.

Table 2: System suitability parameters for ziprasidone.

<table>
<thead>
<tr>
<th>Parameters</th>
<th>Values</th>
</tr>
</thead>
<tbody>
<tr>
<td>Theoretical Plates (n)</td>
<td>17112</td>
</tr>
<tr>
<td>Theoretical plates per meter (N)</td>
<td>68448</td>
</tr>
<tr>
<td>Height equivalent to theoretical plates [HETP] (mm)</td>
<td>0.01</td>
</tr>
<tr>
<td>Tailing factor</td>
<td>1.06</td>
</tr>
</tbody>
</table>

Linear range.
The area ratio of ziprasidone peaks to the internal standard in urine samples was linear in relation to the concentration of ziprasidone within 1-30 µg/ml and 30-200 µg/ml. In the concentration range of 1-30 µg/ml the calibration graph is described as \( Y = 0.0827 \times X + 0.00313 \) (\( R^2=0.9996 \)), and in the range of 30 to 200 µg/ml, this dependence is \( Y = 0.05196 \times X + 0.75165 \) (\( R^2=0.9988 \)), where \( Y \) is the area ratio of ziprasidone peaks to the internal standard and \( X \) is the ziprasidone concentration, µg/ml. Results obtained using the calibration graph for model samples of urine are presented in Table 3.

Table 3: Accuracy and precision of the method for determining the concentration of ziprasidone in model urine samples.

<table>
<thead>
<tr>
<th>Ziprasidone added in sample, µg/ml</th>
<th>Intra-day (n=6)</th>
<th>Inter-day (n=6)</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>found µg/ml ± SD (mean value)</td>
<td>Accuracy (%)</td>
</tr>
<tr>
<td>1</td>
<td>0.99 ± 0.06</td>
<td>99.30</td>
</tr>
<tr>
<td>5</td>
<td>4.96 ± 0.28</td>
<td>99.20</td>
</tr>
<tr>
<td>10</td>
<td>9.91 ± 0.51</td>
<td>99.10</td>
</tr>
<tr>
<td>20</td>
<td>19.81 ± 1.01</td>
<td>99.05</td>
</tr>
<tr>
<td>30</td>
<td>29.67 ± 1.40</td>
<td>98.90</td>
</tr>
<tr>
<td>100</td>
<td>98.60 ± 3.55</td>
<td>98.60</td>
</tr>
<tr>
<td>150</td>
<td>147.30 ± 5.08</td>
<td>98.20</td>
</tr>
<tr>
<td>200</td>
<td>195.50 ± 6.49</td>
<td>97.75</td>
</tr>
</tbody>
</table>

LOD and LOQ were determined by the ratio of signal intensity of analyte to noise signal. Limit of quantitation was set as the lowest concentration that provides the signal/noise ratio above 3:1 for the detection and 5:1 for quantitation. Their values were 0.2 µg/ml and 0.5 µg/ml in methanol solution, respectively.

Accuracy and precision
Results to determine the accuracy and precision of the HPLC method were within acceptable limits. The results presented in Table 2 show that the method is correct and reproducible for repeated tests of ziprasidone in urine within one day and on different days.

Stability
Stock solutions of ziprasidone maintained stability for at least 1 month when stored at 4°C. Model urine samples with ziprasidone maintained stability at 60±5°C. Three freeze-thaw cycles of model urine samples and storage at room temperature for 18 hours demonstrated the stability of ziprasidone in urine. When testing ziprasidone stability the loss of the drug does not exceed acceptable limits when stored in autosampler for 72 hours. The research results are summarized in Table 4.

Table 4: Stability of ziprasidone in urine.

<table>
<thead>
<tr>
<th>Storage conditions of samples</th>
<th>Concentration of ziprasidone in the sample (µg/ml)</th>
<th>Found concentration (µg/ml)</th>
<th>% changes</th>
</tr>
</thead>
<tbody>
<tr>
<td>Three freeze-thaw cycles</td>
<td>1.00</td>
<td>0.982</td>
<td>-1.80</td>
</tr>
<tr>
<td>cycles (-20 °C / room temperature)</td>
<td>200.00</td>
<td>197.200</td>
<td>-1.40</td>
</tr>
<tr>
<td>Storage at room temperature for 18 hours</td>
<td>1.00</td>
<td>0.955</td>
<td>-4.50</td>
</tr>
<tr>
<td>Stability when stored in autosampler for 72 hours</td>
<td>1.00</td>
<td>0.957</td>
<td>-4.30</td>
</tr>
<tr>
<td>Prolonged storage of urine samples with ziprasidone for 60 days</td>
<td>1.00</td>
<td>0.922</td>
<td>-7.80</td>
</tr>
<tr>
<td></td>
<td>200.00</td>
<td>187.600</td>
<td>-6.20</td>
</tr>
</tbody>
</table>

DISCUSSION
The primary purpose of this study was to develop a simple, rapid, accurate, and sensitive technique for sample preparation to quantify ziprasidone in urine using the HPLC/UV. It was found that content of ziprasidone in the daily urine of test animals was about 0.67 %; the total percentage of identified metabolites in relation to ziprasidone concentration was approximately 28%.

The results from this study showed that SPE allowed to eliminate the influence of endogenous contaminants from biological fluids and to concentrate the sample, which in turn provided the ziprasidone determination limit in the biological material at the level of 0.125 µg in 1 ml of urine. The suggested analysis technique enables to determine both therapeutic and toxic doses of ziprasidone providing an accurate determination of concentrations of the drug in the urine sample to confirm the poisoning or fatality. The described methods of isolation and identification of ziprasidone from urine samples would be suitable for the purpose of post-mortem toxicological investigation.

CONCLUSIONS
The simple and quick method to determine ziprasidone in the urine of white rats by HPLC/UV in reverse phase column
LUNA® C18(2) was described. The suggested conditions of chromatographic analysis provide rapid extraction of ziprasidone with the matrix components and metabolites, as well as high selectivity, reproducibility and asymmetry of peaks. The results of method validation showed good precision, accuracy and reproducibility. The method maintains linearity in a much larger concentration range than is usually described in the studies making it suitable for chemical and toxicological analysis.

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