

Antioxidant and Antimicrobial Activities of the Leaf Extract of *Salvia palaestina*

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ARTICLE INFO

Article history:

Received on: 25/10/2015

Revised on: 16/11/2015

Accepted on: 20/12/2015

Available online: 26/01/2016

Key words:

Antioxidant, DPPH, Antimicrobial *Salvia palaestina*, Essential oil, GC-MS.

ABSTRACT

Salvia palaestina (Lamiaceae) is listed as native plant in *Flora palaestina*. In the Palestinian kitchen, its leaves are virtually used on a daily basis with the breakfast tea to add its distinct pleasant flavor and aromatic aroma. In order to investigate the antioxidant and antimicrobial potential of cultivated *S. palaestina*, the leaves of the herb were collected, air-dried, steam distilled and the culminated essential oils were subjected to GC-MS analysis. About two dozens of volatile and semivolatile phytochemicals were separated and identified. The principal components were eucalyptol (47.09%) and camphor (8.73%). The antioxidant activity of the oil was estimated by 2, 2'-diphenyl-1-picrylhydrazyl (DPPH) method. The IC₅₀ was 2.333 mg/ml after 30 min, while after 90 min it was 1.585 mg/ml, meaning that the antioxidant activity of the oil increased with time. Linear relation between concentration and activity was observed. The antimicrobial activity was examined by the disc diffusion method. The antimicrobial activity of 5 µl of the essential oil was found to be greater than the activity of gentamicin in the case of *Staphylococcus aureus* while it was nearly the same as gentamicin against *E. coli*. Furthermore, this concentration was two times more active than nystatin against *Candida albicans*.

INTRODUCTION

Infectious diseases represent a significant cause of morbidity and mortality worldwide and the developing countries in particular (Enwonwu and Salako, 2000). Consequently, pharmaceutical companies were motivated to develop new antimicrobial drugs because of the emergence of microorganisms resistant to available antimicrobials. It is obvious that bacterial species have the genetic ability to develop and transmit resistance against currently existing antimicrobials. This can be witnessed from the frequent reports on the isolation of bacteria that became multi-resistant to available antimicrobials at the recommended therapeutic dose. Thus, recently, natural products with antimicrobial activity have gained more attention due to safety concerns and to increasing resistance to available antibiotics (Palombo and Semple, 2002). Developments in biomedical science emphasize the involvement of free radicals in the pathophysiology of many diseases, particularly, degenerative diseases.

They can occur as a consequence of cellular damage by free radicals, which are produced either from normal cell metabolisms or from external sources including pollution, cigarette smoking, radiation and medication, etc. (Devasagayam *et al.*, 2004). Free radicals are generally reactive oxygen species (ROS) and reactive nitrogen species (RNS). Both have toxic and beneficial effects and eventually balance is vital to achieve good health. Low or moderate levels of both, ROS and RNS, have positive impact on cellular responses and immune function. At high levels, oxidative stress is generated, which can damage cell structures and have great effect on developing chronic and degenerative conditions such as cancer, arthritis, aging, autoimmune disorders, cardiovascular and neurodegenerative diseases (Emerit and Michelson, 1982, Gutteridge and Halliwell, 1992, Valko *et al.*, 2006, Halliwell, 2001, Liao and Yin, 2000).

Thus, antioxidants may play an important role in disease prevention and they are defined as compounds that inhibit or delay the oxidation of other molecules by inhibiting the initiation or propagation of oxidizing chain reactions. Their antioxidant activity is based on their ability to donate hydrogen atoms to free radicals and they are usually phenolic compounds with potent scavenging activity (Aruoma, 1994).

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Antioxidants are not only important in the medical field but are equally important in the food industry, which is evident in the increasing number of researches involving plants and food antioxidants (Bohm *et al.*, 1998, Sanches-Silva *et al.*, 2014). The beneficial medicinal effects of plants resulted mostly from secondary metabolites such as alkaloids, tannins, and polyphenolic compounds, which are synthesized and localized in certain parts in the plant (Stary, 1996). It was believed that these secondary metabolites might exert their action by resembling endogenous metabolites, hormones, ligands, signal transduction molecules or neurotransmitters (Briskin, 2000). As with other members of Lamiaceae family, *Salvia* species were reported to possess antimicrobial activity especially due to the presence of linalool and eucalyptol active ingredients (Sonboli *et al.*, 2006). The commonly synthetic chemicals used as antioxidants such as butylated hydroxyanisole (BHA) and butylated hydroxytoluene (BHT) have possible toxicity, which limit their use (Papap, 1999). Thus, natural antioxidants are highly demanded due to safety concerns (Lobo *et al.*, 2010). For example, the essential oil of *Salvia* species is used in some countries as antioxidant due to its scavenging activity of free radicals to prevent and repair related degenerative diseases (Miguel, 2011). The antimicrobial activity of *Salvia's* essential oils was pronounced against Gram-positive more than Gram-negative bacteria (Sonboli *et al.*, 2006, Miguel, *et al.*, 2011, Fraternali *et al.*, 2012, Kamatou *et al.*, 2008). *Salvia* (Lamiaceae) is traditionally acknowledged in the Mediterranean region for its pharmacological benefits. *S. palaestina* (*Meramia Falastinia* in Arabic) is listed as native plant in *Flora Palaestina*. In the Palestinian kitchen, *S. palaestina* is virtually used in daily basis with the breakfast tea to add its distinct pleasant flavor and aromatic aroma. To the best of our knowledge, there has never been reported information about the antioxidant or antibacterial activities of the essential oil from cultivated *S. palaestina* (Lamiaceae) leaves growing in Palestine. The current study therefore is conducted to separate and identify the chemical composition and to assess the antioxidant and antibacterial potential of the essential oil of *S. palaestina* and to use the gas chromatography combined with mass spectrometry (GC-MS) in the electron impact mode to reveal the components identities.

MATERIALS AND METHODS

Reagents

GC grade n-hexane solvent and anhydrous sodium sulfate salt were purchased from Sigma-Aldrich Inc. (USA). Kovats retention index (KI) reagent that consist of alkane standard mixture were between C₁₀-C₄₀ (even numbered) were purchased from Fluka, Switzerland. All the reference standards used in the research were kindly supplied by the Central Public Health Laboratory, Ministry of Health, Ramallah, Palestine.

Plant materials

Cultivated *S. palaestina* leaves were collected from Ya'bad, Jenin in the northern part of Palestine in May 2013. The

plant species is properly authenticated by Prof. Khalid Sawalha, the director of the biodiversity research laboratory, Al-Quds University. Leaves were air dried in the absence of light at room temperature for a about one week until constant weight is achieved.

Instrumentation

Essential oils were analyzed using Perkin Elmer, Clarus Gas Chromatography connected to Clarus 600 C mass spectrometer (USA). The GC-MS was operated in the electron impact ionization mode (EI) at 70 eV. Perkin Elmer autosampler was used with 2ml vials. The GC is equipped with a fused silica capillary column; DB-5 MS consisted of (5% diphenyl polysiloxane, 95% dimethyl polysiloxane) 28 m x 0.25 mm, coating film thickness is 0.25 µm (Restck, USA).

GC-MS chromatographic conditions

Perkin Elmer GC-MS at electron impact mode (EI) was used. The flow rate of the carrier gas was 1 ml He/min. Injector temperature was set at 235°C, the source temperature was at 250°C and the interface temperature was at 260°C. Split ratio of 1:20 was adopted during the entire analysis. The column gradient temperature was held at 50°C for 2 minutes, then raised from 50°C to 180°C at a ramp of 5°C/min and from 180° to 280°C at a ramp rate of 15°C/min and held there for extra 5min. Solvent cut time of 4.5 minutes was used to eliminate the solvent gigantic peak. The mass range was from m/z 50 to 480 Da, and of scan interval of 0.2 seconds. The obtained mass spectra were interpreted by comparing with the Mass Spectral Library of the National Institute of Standards and Technology (NIST).

Steam distillation of Essential oils

The essential oils of the *S. palaestina* leaves were isolated by steam distillation using Clevenger type apparatus for three hours: The water distillate was extracted twice with 100 ml hexane. The hexane fractions were combined and dried over anhydrous sodium sulfate. Then, 300 µL of hexane extract was diluted to 1 ml with hexane and 1 µL of the resulted diluted sample was injected to GC-MS using optimized method. The oil was obtained by evaporation of the hexane by rotary evaporator.

Antioxidant activity

Electrons donation ability of *S. palaestina* essential oils was measured from the bleaching of purple colored methanol solution of 2, 2'-diphenyl-1-picrylhydrazyl stable radical (DPPH) using spectrophotometric assay. After incubation period, the absorbance was measured at 517nm. Different concentrations of *S. palaestina* essential oil in methanol were prepared, 50 microliters of each concentration were added to 2 ml of DPPH solution (DPPH concentration was 6×10⁻⁵ M), final concentrations ranging from 0.122 to 1.35 mg/ml all samples were warped with aluminum foil and kept in dark place (drawer). Absorbance at wavelength 517 nm was measured at three different time points,

namely after 30 min, 1 hour and 1.5 hour. Percentage of the antioxidant scavenging activity (A_i %) was calculated from the following equation and then plotted against concentration to calculate the $A_{i\%}$ (Brand-Williams *et al.*, 1995).

$$(A_i\%), \text{DPPH radical scavenging activity } \% = \left[1 - \left(\frac{A_s}{A_c} \right) \right] \times 100\%$$

where, A_i is the antioxidant index, A_s is the sample absorbance, A_c is the control absorbance. The control used was DPPH while the blank was methanol. IC_{50} (inhibitory concentration), which is defined as the concentration of sample required to inhibit the formation of DPPH radicals by 50% was calculated. Tert-butyl-4-hydroxy toluene (BHT) was used as positive control, series of concentrations were prepared, and 5 μ L of each concentration was taken and to 2 ml of DPPH was added as in *S. palaestina* essential oils, final concentrations for BHT ranged from 0.015 to 0.125 mg/ml.

Antimicrobial activity

Reagents

Nutrient agar (Difco), sabouraud dextrose agar (Difco) for *Candida albicans*, 0.9 % sodium chloride AR solution, ciprofloxacin standard, gentamicin standard, nystatin standard, barium chloride AR and sulfuric acid AR were kindly supplied by the Central Public Health Laboratory, Ministry of Health, Ramallah, Palestine.

Microbial strains

The essential oil antimicrobial activity was evaluated against a panel of American Type Culture collection micro organisms, including the following strains; *Staphylococcus*

aureus (ATCC#25923), *Staphylococcus epidermidis* (ATCC#12228), *Candida albicans* (ATCC#10231), *E.coli* (ATCC#8739) and *Salmonella Typhimurium* (ATCC#14028). All these strains were from Becton Dickinson, France.

Procedure

The test was carried out using disk diffusion method. 5 μ l of (100%) *S. palaestina* essential oil/positive control was added to each disk. Four disks were spread on the surface of the media in each plate. Plates were incubated at 37 °C for 24 hours in case of bacteria while for *Candida albicans* it was incubated at 25 °C for 72 hours. Then, zones of inhibition were measured and the results were documented. Solutions of gentamicin, ciprofloxacin and nystatin (10 μ g/ml) were prepared and were used as positive control. The final concentration of both bacteria and fungi on each plate was about 1.5 x 10⁶ CFU/ ml.

RESULT AND DISCUSSION

Twenty components were identified by GC-MS analysis. Figure 1 represents the total ion chromatogram (TIC) from GC-MS, while figure 2 represents the detected components of *S. palaestina* and their abundance. The abundant components were the oxygenated monoterpene eucalyptol (47.09%) and camphor (8.73%). Other components that were present but to a lesser extent include β -caryophyllene (2.88%), α -terpineol (2.16%), β -thujene (1.77%) and epiglobulol (1.25%). All the compounds were identified by matching their MS spectra with NIST MS spectral library database and by calculating their corresponding Kovats indices (KIs) and comparing them with those reported in the literature using the same DB-5 capillary column.

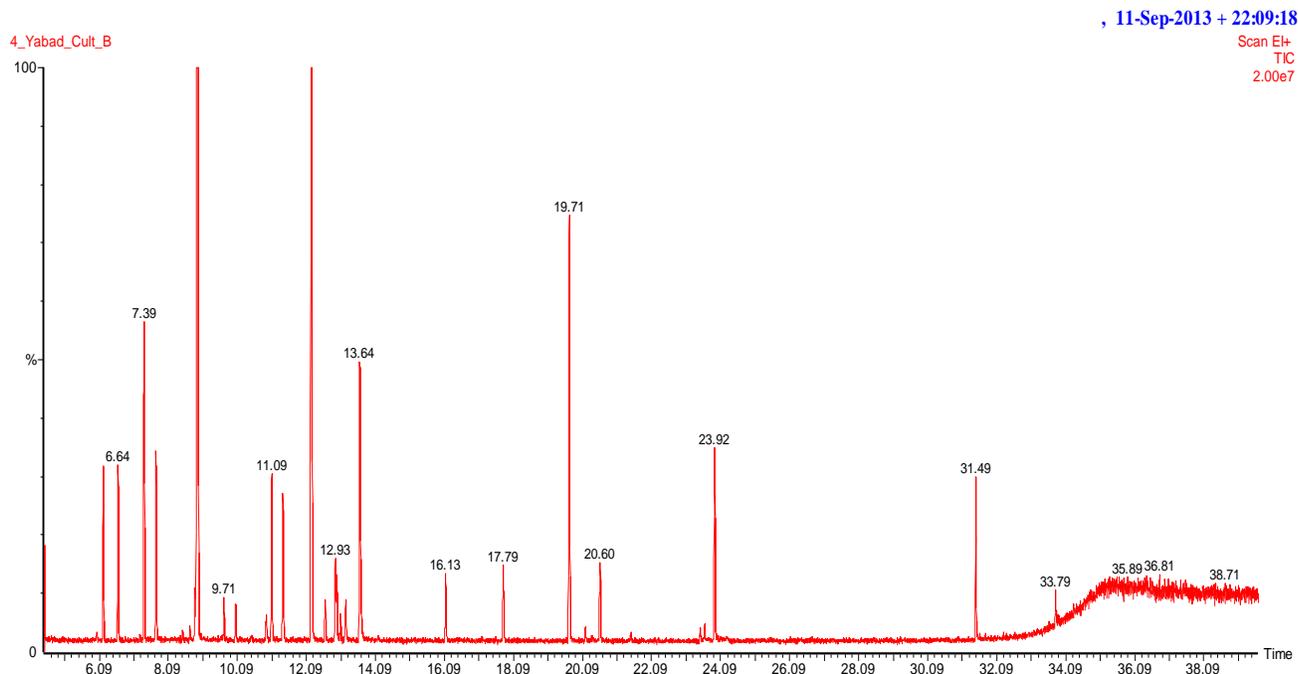


Fig. 1: TIC GC-MS of cultivated *S. palaestina* essential oil collected from Ya'bad, Jenin.

S. palaestina components

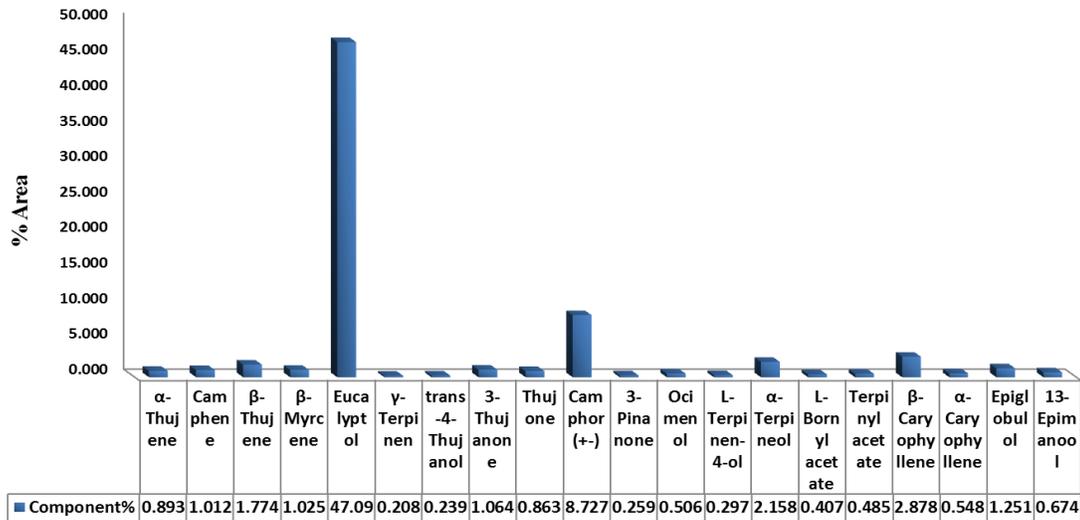
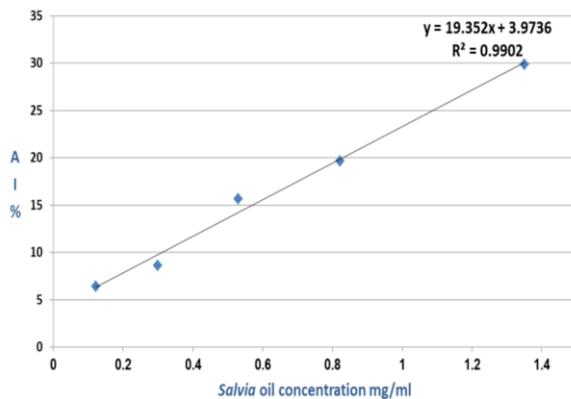


Fig. 2: Histogram of the percentage of essential oil components of *S. palaestina* from Ya'bad, Jenin as determined by GC-MS.

Antioxidant activity of *S. palaestina* oil after 30 min.



Antioxidant activity of BHT after 30 min.

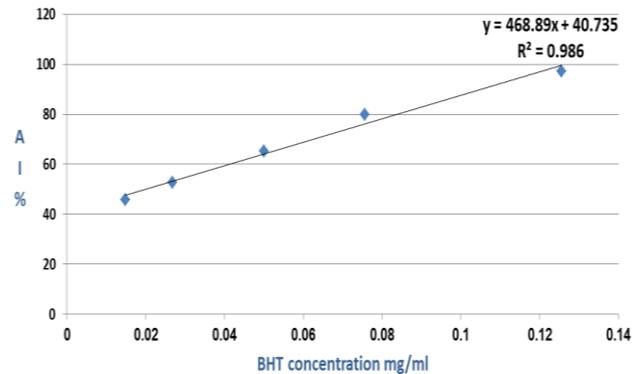


Fig. 3: (A) Antioxidant activity of Palestinian *S. palaestina* oil, (B) antioxidant activity of the BHT after 30 mins.

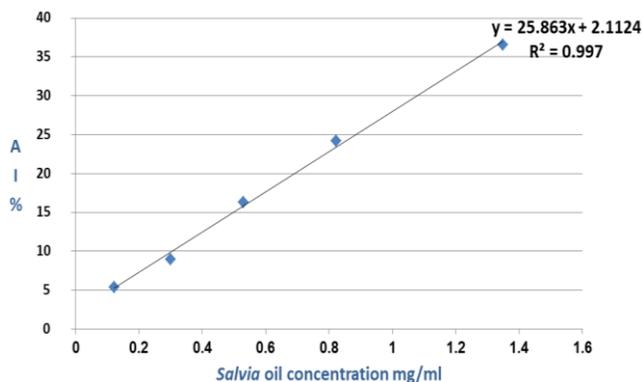
Antioxidant activity of *S. palaestina*

All A_i% values were plotted against their corresponding concentrations for both *S. palaestina* oil and the positive control in three time intervals of 30, 60 and 90 minutes respectively (Figures 3-5). Linear relation between concentration and antioxidant activity was observed. The inhibition concentration (IC₅₀) was calculated and it was 2.333 mg/ml for the oil and 0.02 mg/ml for BHT, which means that the antioxidant activity of the oil is less than that of the positive control. IC₅₀ after 60 min. was calculated and it was 1.852 mg/ml for *S. palaestina* oil, while it was 0.004 mg/ml for BHT, which means that the activity of the oil is still less than that of the positive control. IC₅₀ for *S. palaestina* oil was 1.585 mg/ml, while it was excluded for BHT because of unharmonized values. IC₅₀ values for both sample and positive control were represented in figure 6. It is obvious from the histogram that *S. palaestina* oil has less antioxidant activity than that of the positive control; hence, the activity of 2.333 mg/ml of

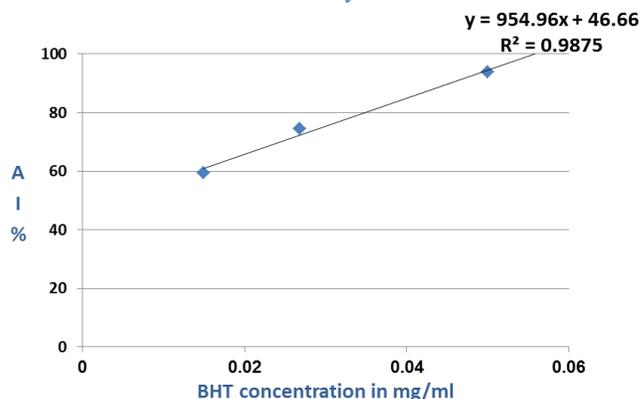
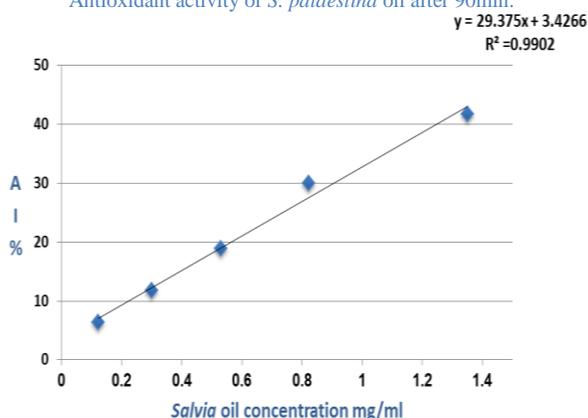
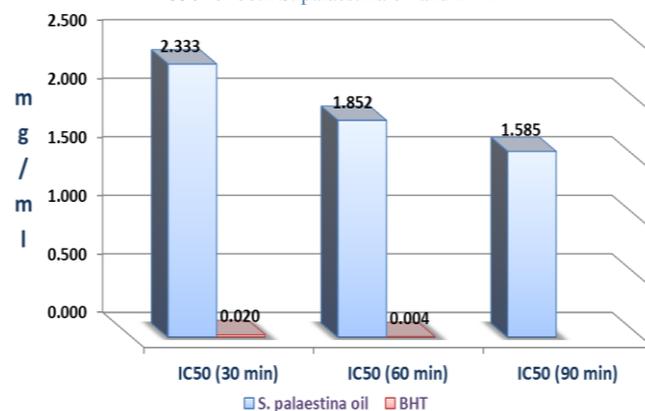
the oil is equal to 20 µg/ml of BHT after 30 min. Furthermore, after 60 min. 1.852 mg/ml will give the same antioxidant activity as that of 4 µg/ml of BHT. Within the tested range of concentrations, free radical scavenging capacity of the tested *S. palaestina* oils increased in a concentration dependent manner.

DPPH scavenging activities increased significantly with increasing the concentration from 0.122 to 1.35 mg/ml of *S. palaestina* essential oil.

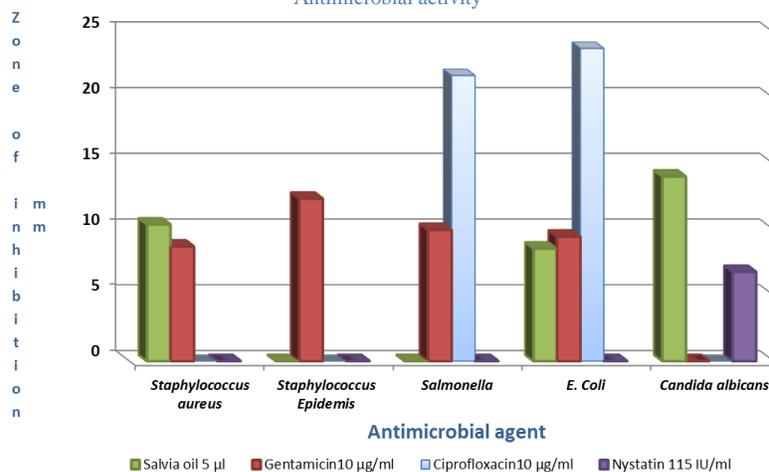
In addition, the activity of the oil increased with time, thus to achieve a good antioxidant activity, sufficient time is required. The activity was mainly attributed to the major effective components i.e., oxygenated terpenes such as eucalyptol and camphor. Preliminary GC-MS screening indicated the presence of these compounds in *S. palaestina*'s from different other locations in the Mediterranean region, which are considered as free radical scavengers (Al-Qudah *et al.*, 2014, Gürsoy *et al.*, 2012, Tenore *et al.*, 2011).

Antioxidant activity of *S. palaestina* oil after 60min.

Antioxidant activity of BHT after 60min.

Fig. 4: (A) Antioxidant activity of the *S. palaestina* oil, (B) antioxidant activity of the BHT after 60 mins.Antioxidant activity of *S. palaestina* oil after 90min.Fig. 5: Antioxidant activity of *S. palaestina* oil after 90 minIC50 for both *S. palaestina* oil and BHTFig. 6: IC₅₀ for both *S. palaestina* oil and the positive control

Antimicrobial activity

Fig. 7: Antimicrobial activity of cultivated *S. palaestina* oil.

Although native *S. palaestina* (at vegetative stage) does not have the antioxidant activity as that of BHT but currently, there is considerable interest in new natural antioxidants to replace the synthetic especially in food and cosmetic products. Several studies were conducted on the available synthetic antioxidants BHT and BHA to evaluate the safety of both substances. It turned

out that long exposure might cause thyroid, liver, kidney dysfunctions and might also affect the lung function and blood coagulation. Moreover, it was suggested that high doses of BHT might mimic estrogen, the main female sex hormone, causing reproductive system dysfunctions (Clapp *et al.*, 1979, Kahl and Kappus, 1993).

Antimicrobial activity

The antimicrobial activity of 5 μ l of *S. palaestina* essential oil was examined on gram positive bacteria (*Staphylococcus aureus*, *Staphylococcus epidermis*), gram negative bacteria (*Salmonella typhimurium*, *E. coli*) and fungus (*Candida albicans*) in the presence of positive control (gentamicin, ciprofloxacin and nystatin) by using disc diffusion method. The zones of inhibition were measured and the average results of zones of inhibition were summarized in Table 1. Comparison between antimicrobial activities is illustrated in Figure 7. It is clear that 5 μ l of *S. palaestina* essential oil exhibits notable antimicrobial activity against some gram positive bacteria (*Staphylococcus aureus*) and some gram negative bacteria (*E. coli*). This observed activity is superior to gentamicin in the case of *Staphylococcus aureus* and almost nearly the same as that of gentamicin in the case of *E. coli*. Moreover, this small volume of *S. palaestina* oil was two times more effective than nystatin in the case of *Candida albicans*. Conversely, this volume of tested oil does not exhibit any activity neither on *Staphylococcus epidermis* nor on *Salmonella typhimurium*. Regardless of effectiveness, the usage of gentamicin is restricted by its toxicity, which includes ototoxicity and nephrotoxicity. This reported toxicity remains a major problem in clinical use (Andreu *et al.*, 1985, Dulong *et al.*, 1988).

Table 1: The antimicrobial activity of *S. palaestina* essential oil (average inhibition zone in (mm)).

	<i>Staphylococcus aureus</i>	<i>Staphylococcus epidermis</i>	<i>Salmonella typhimurium</i>	<i>E. coli</i>	<i>Candida albicans</i>
<i>S. palaestina</i> oil 5 μ l	10.39	ND	ND	8.57	14.04
Gentamicin 10 μ g/ml	8.72	12.36	9.98	9.47	NT**
Ciprofloxacin 10 μ g/ml	NT*	NT	21.73	23.79	NT
Nystatin 115 IU/ml	NT	NT	NT	NT	6.79
Blank	ND	ND	ND	ND	ND

*ND: Not detected, **NT: Not tested

It is well known that nystatin resistance was reported after gradual exposure, seven isolates of *Candida* species became resistant (Athar and Winner, 1971). In addition, it has no appreciable activity against bacteria, protozoa, or viruses. It is contraindicated in patients with history of hypersensitivity which might include the following side effects; tachycardia, bronchospasm, facial swelling, sever skin reactions which all required immediate emergency admission. Minor adverse reactions might include diarrhea (in some cases bloody diarrhea), vomiting, nausea, rashes, urticaria, Stevens-Johnson syndrome, and nonspecific myalgia (available at <http://www.drugs.com>). Although ciprofloxacin's antimicrobial activity is greater than that of 5 μ l of the oil against tested organisms, but it's contraindicated in ages less than 18 years and its adverse reactions might be very severe such as tendon rupture which was reported recently in a number of cases. Achilles tendon rupture has been reported to the FDA in 25 cases. Some ruptures have also occurred in hands and

shoulder (available at <http://www.drugs.com>). In general, the essential oils are hydrophobic in nature, and it was proposed that the cell membrane is the primary target of their antimicrobial action. *S. palaestina* essential oil seems to accumulate in the cell membrane causing leakage of enzymes, ions and metabolites (Inoue *et al.*, 2004). In the case of gram negative bacteria there might be resistance to the essential oil due to the additional outer membrane of their cell wall, which acts as a barrier to many substances including antibiotics (Palombo and Semple, 2001). Furthermore, the antimicrobial activity of *S. palaestina* essential oil on *Candida* is mainly due to the antimicrobial constituents especially eucalyptol, which are capable of changing the structure and moisture of mucous membranes of fungal cells, interfering with the respiratory processes, and thus eliminating the pathogen (Chen *et al.*, 2013). It was suggested that the antimicrobial activity of *S. palaestina* oil from other locations in the Mediterranean was probably due to its constituents (thujone, eucalyptol, camphor, camphene and caryophylline). However, when comparing the antimicrobial activity of each constituent with the antimicrobial activity of the whole oil, the oil activity was superior. This indicates that their antimicrobial effect probably involves some type of synergism between many constituents (Fraternal *et al.*, 2012, Mitić-Čulafić *et al.*, 2005). Thus, the overall therapeutic effect of *S. palaestina* oil could be attributed to the synergistic interactions of individual components and to the anti-inflammatory approved effect of this oil, which can lead to easier passage of the essential oils through mucous membrane.

CONCLUSION

The antioxidant activity results revealed that the activity of *S. palaestina* oil is less than that of the positive control (BHT). Nonetheless, it increased with time thus special sustained release formulations might be helpful. Moreover, *S. palaestina* oil revealed antimicrobial activity against both bacteria and fungi. It was superior to some available antibiotics in certain cases. Small volume of 5 μ l of essential oil exhibits notable antimicrobial activity against some gram positive bacteria (*Staphylococcus aureus*) and some gram negative bacteria (*E. coli*). The observed activity of essential oil (100%) was superior to gentamicin (10 μ l/ml) in case of *Staphylococcus aureus* and almost the same in case of *E. coli*. Furthermore, this small volume of oil (100%) was two times more effective than nystatin (115 IU/ml) in case of *Candida albicans*. This promising result is important due to the increasing resistance against available antimicrobial agents in addition to its known toxicity.

ACKNOWLEDGEMENT

We would like to thank the Central Public Health Laboratory CPHL staff, Ministry of Health in Ramallah for providing the GC-MS instrument for the analysis. Special thanks to Dr. Asad Ramlawi, Deputy Minister, Ministry of Health for his continuous support. Thanks are extended to Mr. Ibrahim Salem for facilitating this research at the Ministry of Health in Ramallah.

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How to cite this article:

Sabbobeh R, Hejaz H, Jahajha A, Al-Akhras S, Al-Jaas H. and Abu-Lafi S. Antioxidant and Antimicrobial Activities of the Leaf Extract of *Salvia palaestina*. *J App Pharm Sci*, 2016; 6 (01): 076-082.